### U.S. Tire Manufacturers Association

Please find attached comments from the U.S. Tire Manufacturers Association in response to Washington Ecology's Safer Products for Washington Cycle 2 Draft Priority Products Report. Please don't hesitate to reach out with any questions.



December 31, 2024

Submitted electronically via <u>SaferProductsWA@ecy.wa.gov</u> Kim Morley, Safer Products for Washington Project Manager Washington Department of Ecology, HWTR Program P.O. Box 47600 Olympia, WA 98504-7600

# Re: Draft Identification of Priority Products Report to the Legislature, Safer Products for Washington Cycle 2, Implementation Phase 2

#### Dear Ms. Morley:

The U.S. Tire Manufacturers Association (USTMA) appreciates the opportunity to provide comments on Washington Department of Ecology's (Ecology) Draft Identification of Priority Products Report (hereinafter referred to as the "Draft Report") to the Legislature under the Safer Products for Washington program. USTMA is the national trade association of tire manufacturers that produce tires in the United States. Domestic tire manufacturing is responsible for more than 291,000 jobs and has an annual economic footprint of \$170.6 billion in the United States. The tires from our member companies make mobility possible and keep the U.S. economy moving. USTMA advances a sustainable tire manufacturing industry through a commitment to science-based public policy advocacy.

USTMA members are dedicated to sustainable practices in every aspect of their businesses and embrace a shared responsibility of helping to achieve a more sustainable society. As part of this, we are committed to understanding any potential impacts of our tires on the environment. USTMA is pleased to provide input on this Draft Report, which includes N-(1,3-dimethylbutyl)-N'-phenyl-p-phenylenediamine (6PPD) in artificial turf. USTMA has previously provided comments in response to the inclusion of 6PPD in motor vehicle tires in the agency's Draft Identification of Priority Chemicals Report to the Legislature, phase one of Ecology's safer products cycle. We have included these comments as Attachment A.

#### 6PPD serves a critical role in the safety and reliability of motor vehicle tires

The use of 6PPD in tires serves an essential safety function, protecting the components of the tire from attack by ozone and oxygen, and has been used for decades. Without 6PPD, a tire's integrity would be severely and quickly compromised, jeopardizing driver and passenger safety. 6PPD is currently used in all USTMA member passenger, light truck, truck and bus radial, and motorcycle tires. USTMA is not aware of any new motor vehicle tires available today that do not contain 6PPD.

Since 6PPD-quinone (6PPDQ), a transformation product of 6PPD, was first identified in Tian et al., 2021<sup>1</sup>, USTMA has engaged with Washington State, federal, and Tribal agencies, researchers, and other stakeholders to identify and support existing and future research related to 6PPDQ and to ensure research utilizes the most robust methodologies. USTMA continues to support the use of the best available, peer-reviewed science to inform regulatory actions. The Association's members are committed to working with partners, including Washington Ecology, to fill knowledge gaps in existing research.

# Washington Ecology should work closely with California's Department of Toxic Substances' Safer Consumer Products Program.

After the Tian et al. study was published, USTMA sought a review of 6PPD in tires under California's Department of Toxic Substances Control's (DTSC) Safer Consumer Products (SCP) Program. A review of 6PPD in tires under the SCP program provides a scientific, regulatory framework to analyze whether alternatives exist that will enable tire manufacturers to ensure both tire and environmental safety. DTSC added 6PPD in tires to the Priority Products Workplan in early 2021 and since that time USTMA has convened a consortium of 36 tire companies to work diligently towards identifying alternative chemicals that provide the same level of safety to protect motorists.

In August 2024, DTSC approved the USTMA 6PPD Consortium Preliminary (Stage 1) Alternatives Analysis Report (Attachment B), which identified seven possible candidates to replace 6PPD that warrant further analysis. The second phase in the alternatives analysis process is for the Consortium to embark on an in-depth analysis that refines the relevant factors and product function descriptions of the first stage and expands the analysis to consider additional impacts, including life cycle and economic effects. USTMA will provide DTSC with an interim update in August 2025 and submit a final report in August 2026.

The safety, performance and sustainability of our products remains our uncompromising priority and there is important work to be done to ensure any potential alternatives meet that high standard. The chemicals present in tires today all perform specific and integrated functions, and tire composition cannot responsibly be modified without great care, including extensive and rigorous testing. Any alternative identified must continue to ensure compliance with Federal Motor Vehicle Safety Standards and other consumer, vehicle, and tire manufacturer requirements.

Based on the experience of Consortium members, the tire research and development, design, and performance testing process for a tire using existing, commercially produced materials known to perform as necessary in tires, can take a minimum of 4 to 6.5 years. In the tire design process, each step may be repeated multiple times until an acceptable design is

<sup>&</sup>lt;sup>1</sup> Tian, Z; Zhao, H; Peter, KT; et al. 2021. A ubiquitous tire rubber-derived chemical induces acute mortality in coho salmon. Science 371(6525):185-189. doi: 10.1126/science.abd6951.

achieved, which can significantly extend the design process. Any challenges encountered while conducting a step in the tire design process may require development to go back to an earlier stage.

In the case of replacing 6PPD, once a suitable new candidate antidegradant is identified, an additional 4 years (minimum) of limited-scale field testing would be required to ensure performance as a tire ages. After satisfactory results are obtained from field testing, additional time would be needed for deployment of the new antidegradant in tires for the market. USTMA members are committed to moving the process forward as promptly as possible within the bounds of responsible tire design and development.

The Safer Products for Washington enacting legislation encourages Ecology to consider actions taken by other states. When determining regulatory actions, Ecology is encouraged to consider whether a "restriction would be consistent with regulatory actions taken by another state or nation on a priority chemical or members of a class of priority chemicals in a product." <u>RCW 70A.350.040(4)(b)</u>. Although the product in question under DTSC's SCP Program is 6PPD in motor vehicle tires, the process is similar and directly linked to Ecology's work. At the end of both processes, each agency will decide what regulatory response, if any, to undertake. Additionally, the work under the SCP Program is already further along than Ecology's Safer Products Program.

Given the similarities between the two programs and the ongoing alternatives analysis in California, USTMA strongly recommends that Ecology closely coordinate with DTSC to ensure a thorough and consistent assessment, and leverage what is learned throughout the alternatives assessment process. USTMA is committed to working with regulatory bodies, including Ecology, as agencies move forward through these processes.

# Aspects of Ecology's Draft Technical Supporting Documentation for the Draft Report require additional considerations and clarifications

After careful review of the technical supporting document, USTMA identified multiple areas that would benefit from additional considerations by Ecology to improve the Draft Report's accuracy and clarity.

# I. The evidence base to support a conclusion of potential exposure to PFAS from crumb rubber used as infill in artificial turf is scarce and does not support the Priority Product designation.

The Draft Technical Supporting Documentation for Priority Products: Safer Products for Washington Cycle 2 Implementation Phase 2 (hereinafter referred to as the "Technical Document"), Chapter 2 states that PFAS have been reported in different components of artificial turf, including crumb rubber. However, Ecology references only two peer-reviewed scientific publications to support this assertion, one of which examined a PFAS compound (8:2 FTOH) which is not used in tire manufacturing (Zuccaro et al. 2023)<sup>2</sup> and another in which no specific PFAS were reported (Lauria et al.  $(2022)^3$ .

Zucarro et al. (2023) characterized their research as a pilot study to establish an extraction analysis method for the detection of fluorotelomer alcohols in artificial turf components. The authors measured fluorotelomer alcohols in extracts from a single sample of new artificial turf fibers and a single sample of new crumb rubber pellets in fully sealed packaging. They reported fluorotelomer alcohol (8:2 FTOH) in crumb rubber at 110 ng/g rubber. The only explanation Zuccaro et al. offered for finding 8:2 FTOH in the crumb rubber was that there are "many chemicals" in end-of-life tires, although the authors did not explain why it was also measured in the turf fibers. USTMA members report that no fluorotelomer alcohols, including 8:2 FTOH, are used in manufacturing tires; thus the authors' explanation is not reliable. Considering that Zuccaro et al. (2023) only analyzed a single sample and that fluorotelomer alcohols are common in many consumer products (Titaley 2024)<sup>4</sup>, it is possible that there could have been 8:2 FTOH contamination in the crumb rubber sample from the packaging or during the sample handling. Zuccaro et al. stated that their study provides "preliminary evidence" of the presence of 8:2 FTOH and therefore is an insufficient basis for concluding that crumb rubber used as infill in artificial turf fields is a source of PFAS exposure.

Lauria et al. (2022) reported results of analysis for PFAS in artificial turf backing, infill, and blades. This study measured specimens from in-place artificial turf throughout Sweden but included only 2 samples of SBR-crumb rubber. The researchers analyzed the 2 samples for total fluorine (TF), total oxidizable precursor assay (TOPA), extractable organic fluorine (EOF) and 23 target PFAS compounds. The TF measurements showed the presence of fluorine in both SBRcrumb rubber samples (43115 ng/g and 36183 ng/g), although EOF concentrations were only detected in 1 of the 2 samples (45.2 ng/g), and total PFAS was also only detectable in 1 of the 2 crumb rubber samples (0.173 ng/g). In addition, the TOPA results indicated negligible formation of perfluoroalkyl carboxylic acids (PFCAs) following oxidation. The researchers stated that "target PFAS were detected intermittently and at low concentrations in infill." The authors concluded that the collective results suggest that fluorine in the synthetic turf materials "consists mostly of non-extractable, non-PFAA precursors, such as fluoropolymers." The results of these two studies and those of the non-peer reviewed reports that did not include crumb

<sup>&</sup>lt;sup>2</sup> Zuccaro et al. 2023. Assessing extraction-analysis methodology to detect fluorotelomer alcohols (FTOH), a class of perfluoroalkyl and polyfluoroalkyl substances (PFAS), in artificial turf fibers and crumb rubber infill. Case Studies in Chemical and Environmental Engineering 7:100280.

<sup>&</sup>lt;sup>3</sup> Lauria et al. 2022. Widespread occurrence of non-extractable fluorine in artificial turfs from Stockholm, Sweden. Environmental Science & Technology Letters 9:666-672.

<sup>&</sup>lt;sup>4</sup> Titaley. 2024. Chemical transformation, exposure assessment, and policy implications of fluorotelomer alcohol partitioning from consumer products to the indoor and outdoor environment—from production to end-of-life. Environmental Science: Advances 3:1364-1384.

rubber infill as test materials TRC (2022)<sup>5</sup> and Ecology Center (2019)<sup>6</sup>, do not provide compelling evidence that non-polymeric PFAS compounds are likely to be available from crumb rubber derived from end-of-life tires.

Given this dearth of data, USTMA recommends that Ecology reconsider including PFAS in artificial turf under the agency's priority products.

# II. The Technical Document does not accurately reflect the potential for environmental or human health exposure to 6PPD from artificial turf.

Chapter 2 of the Technical Document cited several studies which reported the presence of 6PPD in crumb rubber from artificial turf fields. Namely, Duque-Villaverde et al. (2024)<sup>7</sup>, Kawakami et al. (2022)<sup>8</sup>, Schneider et al. (2020a)<sup>9</sup>, and Zhao et al. (2023<sup>10</sup>, 2024<sup>11</sup>), all of which used aggressive organic solvents to extract compounds from crumb rubber and analyzed the extracts with gas or liquid chromatography with mass spectrometry. The presence, however, of 6PPD in crumb rubber as measured in extracts does not directly translate to the potential for exposure or risk. Rather, studies which investigate the potential for leaching in water or biological fluids would be more relevant as indicators of the potential for exposure and risk. Leachate to water would indicate the potential for migration in the environment, and leachate to biological fluids would represent the relative potential for uptake to humans.

For example, Schneider et al. (2020b)<sup>12</sup> reported that only approximately 7% of 6PPD in crumb rubber could migrate to artificial sweat, and no measurable 6PPD migrated to artificial saliva or gastric juice. Schneider et al. (2020c)<sup>13</sup> further concluded that the potential human exposures to 6PPD from end-of-life tire crumb rubber in artificial turf did not pose an

<sup>&</sup>lt;sup>5</sup> TRC. 2022. Technical Memorandum: Evaluation of PFAS in Synthetic Turf. Available at:

https://www.cityofportsmouth.com/sites/default/files/2022-

<sup>06/</sup>Technical%20Memorandum\_Portsmouth\_Final.pdf

<sup>&</sup>lt;sup>6</sup> Ecology Center. 2019. Toxic "Forever Chemicals" Infest Artificial Turf. <u>https://www.ecocenter.org/toxic-forever-chemicals-infest-artificial-turf</u>

<sup>&</sup>lt;sup>7</sup> Duque-Villaverde et al. 2024. Recycled tire rubber materials in the spotlight. Determination of hazardous and lethal substances. Science of The Total Environment 929:172674.

<sup>&</sup>lt;sup>8</sup> Kawakami et al. 2022. Characterization of synthetic turf rubber granule infill in Japan: Rubber additives and related compounds. Science of The Total Environment 840:156716.

<sup>&</sup>lt;sup>9</sup> Schneider et al. 2020a. ERASSTRI - European Risk Assessment Study on Synthetic Turf Rubber Infill – Part 1: Analysis of infill samples. Science of The Total Environment 718:137174.

<sup>&</sup>lt;sup>10</sup> Zhao et al. 2023. Screening p-Phenylenediamine Antioxidants, Their Transformation Products, and Industrial Chemical Additives in Crumb Rubber and Elastomeric Consumer Products. Environmental Science & Technology 57(7):2779–2791.

<sup>&</sup>lt;sup>11</sup> Zhao et al. 2024. Occurrence and Oxidation Kinetics of Antioxidant p -Phenylenediamines and Their Quinones in Recycled Rubber Particles from Artificial Turf. Environmental Science & Technology Letters 11(4):335–341.

<sup>&</sup>lt;sup>12</sup> Schneider et al. 2020b. ERASSTRI - European Risk Assessment Study on Synthetic Turf Rubber Infill – Part 2: Migration and monitoring studies. Science of The Total Environment 718:137173.

<sup>&</sup>lt;sup>13</sup> Schneider et al. 2020c. ERASSTRI - European Risk Assessment Study on Synthetic Turf Rubber Infill – Part 3: Exposure and risk characterization. Science of The Total Environment 718:1377721.

unacceptable health risk to humans. In addition, McMinn et al. (2024)<sup>14</sup> was not able to detect 6PPD in water from their crumb rubber leachate experiments, although they did report 6PPD in leachate to gastric and gastrointestinal fluids. Not only do extraction studies fail to convey the potential for 6PPD migration to the environment or uptake to humans, but such leachate studies have not produced consistent results. Therefore, additional research is required to better understand the potential for 6PPD exposure and risk in the environment and to humans.

USTMA recommends that Ecology add the above research regarding the leaching potential and bioaccessibility to provide a more complete picture of the science.

# III. The conceptual exposure model in Figure 2 is not accurate and should be separated into 2 CEMs – one for PFAS and one for 6PPD.

Figure 2 in the Technical Document illustrates potential exposure pathways for 6PPD and PFAS from artificial turf fields. Although this figure provides qualitative analysis of pathways, 6PPD and PFAS have different physical-chemical properties, behave differently in the environment, and interact differently with humans and other species. For instance, the figure links exposure to fish as an exposure pathway to sensitive populations via seafood consumption. While this may be plausible for PFAS, it is not as clear for 6PPD. The bioaccumulation potential for 6PPD is not well understood, though some studies have estimated it to be low to moderate (ITRC 2024)<sup>15</sup>. Furthermore, although Wei et al. (2024)<sup>16</sup> suggested that 6PPD could be bioaccumulative, the authors found 6PPD in only 3.2% of their biota samples. It is possible that the low detection rate in this study can be attributed to the rapid hydrolysis half-life of 6PPD also puts into question the figure's suggestion that the outdoor environment (i.e., water, sediment, and soil) is a relevant exposure pathway to sensitive populations.

USTMA recommends that the conceptual exposure model illustrated in Figure 2 be revised to differentiate relevant exposure pathways for PFAS and 6PPD.

<sup>&</sup>lt;sup>14</sup> McMinn et al. 2024. Emerging investigator series: in-depth chemical profiling of tire and artificial turf crumb rubber: aging, transformation products, and transport pathways. Environmental Science Processes & Impacts 26(10):1703-1715.

<sup>&</sup>lt;sup>15</sup> https://6ppd.itrcweb.org/3-chemical-properties/#3\_5

<sup>&</sup>lt;sup>16</sup> Wei et al. 2024. First evidence of the bioaccumulation and trophic transfer of tire additives and their

transformation products in an estuarine food web. Environmental Science & Technology 58(14):6370-6380.

<sup>&</sup>lt;sup>17</sup> https://echa.europa.eu/registration-dossier/-/registered-dossier/13653/6/2/3

<sup>&</sup>lt;sup>18</sup> Di et al. 2022. Chiral perspective evaluations Enantioselective hydrolysis of 6PPD and 6PPD-quinone in water and enantioselective toxicity to *Gobiocypris rarus* and *Oncorhynchus mykiss*. Environment International 166:107374.

# IV. Terminology regarding tire wear particles and crumb rubber are not accurate and should be corrected.

Chapter 2 repeatedly claims that artificial turf releases "tire wear particles". However, tire wear particles are different from crumb rubber, and the two materials should not be confused with each other. Tire wear and road wear particles (TRWP) are generated from friction between tires and the road surface. Crumb rubber, however, is intentionally produced from end-of-life tires with size and chemical profiles that differ from TRWP.

USTMA recommends that references to tire wear particles as being representative of crumb rubber properties should be removed from the document.

#### Conclusion

USTMA supports the use of sound science and peer-reviewed data to inform regulatory actions. In summary, USTMA strongly recommends that Ecology work closely with DTSC as both agencies move forward in their respective programs addressing priority products. Additionally, the reconsiderations, additions, and clarifications outlined above will help to ensure that Ecology's report is based on sound science and peer-reviewed data. USTMA's members are committed to collaborating with Washington Ecology and others to better understand 6PPD and 6PPDQ and continue to fill knowledge gaps in existing research.

We thank Ecology again for the opportunity to provide comments on this Draft Report and are happy to answer any questions that the agency may have. We look forward to working with Ecology as this report moves forward.

Respectfully submitted,

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Stephanie Schlea Vice President, Environment, Health, Safety, and Sustainability

**Attachment A:** 

USTMA Comments on Washington Ecology's Draft Identification of Priority Chemicals Report to the Legislature, Safer Products for Washington Cycle 2, Implementation Phase 1

July 14, 2023



July 14, 2023

Washington Department of Ecology Hazardous Waste and Toxics Reduction Program PO Box 47600 Olympia, WA 98504-7600

Re: Draft Identification of Priority Chemicals Report to the Legislature, Safer Products for Washington Cycle 2, Implementation Phase 1 (see - <u>https://apps.ecology.wa.gov/publications/documents/2304038.pdf)</u>

#### I. Overview

The U.S. Tire Manufacturers Association (USTMA) and our member companies appreciate the opportunity to provide comments on the Department of Ecology's Draft Identification of Priority Chemicals Report to the Legislature, Safer Products for Washington Cycle 2, Implementation Phase 1 ("Draft Report").<sup>1</sup> USTMA is the national trade association for tire manufacturers that produce tires in the U.S. and are responsible for more than 291,000 jobs, and have an annual economic footprint of \$170.6 billion in the United States. USTMA advances a safe and sustainable tire manufacturing industry through a commitment to science-based public policy advocacy. The tires from our member companies make mobility possible and keep the U.S. economy moving.

Safer Products for Washington is implemented through a four-phase cycle that repeats every five years. Ecology's Draft Report identifies priority chemicals and chemical classes for the first phase of the second cycle of Safer Products for Washington implementation. Ecology's Draft Report identifies 6PPD as a priority chemical with respect to sensitive species and populations.

Separately, the Washington State Legislature has tasked Ecology with preparing an Alternatives Assessment ("AA") on 6PPD in motor vehicle tires to identify compounds with the potential to replace 6PPD in these products. USTMA is submitting separate comments to Ecology on its Draft 6PPD AA hazard criteria document, which are incorporated by reference in USTMA's comments on the Draft Report.<sup>2</sup>

USTMA would like to emphasize the following comments on Ecology's Draft Report:

II. Protection materials, such as 6PPD, are essential for tire performance and safety and any potential alternative must continue to ensure compliance with Federal Motor Vehicle Safety Standards ("FMVSS") and additional industry and consumer performance requirements.

<sup>&</sup>lt;sup>1</sup> USTMA members include: Bridgestone Americas, Inc., Continental Tire the Americas, LLC; Giti Tire (USA) Ltd.; The Goodyear Tire & Rubber Company; Hankook Tire America Corp.; Kumho Tire Co., Inc.; Michelin North America, Inc.; Nokian Tyres; Pirelli Tire North America; Sumitomo Rubber Industries, Ltd.; Toyo Tire Holdings of Americas Inc. and Yokohama Tire Corporation.

<sup>&</sup>lt;sup>2</sup> 6PPD Alternatives Assessment Hazard Criteria document available here:

https://apps.ecology.wa.gov/publications/documents/2304036.pdf

Tire manufacturers are required by law to certify to the National Highway Traffic Safety Administration ("NHTSA") that every tire they manufacture meets safety, durability, and other performance requirements prior to their sale to the consumer. The Safety Act, 49 U.S.C. §§ 30103-30105 et seq., explicitly preempts any state law or regulation that conflicts with a NHTSA regulation relating to "safety." The rationale, simply put, is that vehicles are a significant means of transportation for citizens and freight. They travel from one state to another and between countries. The absence of a uniform set of safety rules would allow one state to impose arbitrary requirements that could significantly impact interstate commerce. Thus, Ecology's ability to impart a material change required under the Safer Products for Washington law would be significantly limited by the safety and performance requirements that tire manufacturers must meet.

#### III. 6PPD provides critical functions in manufacturing safe and durable tires.

Potential alternatives to 6PPD must provide the same critical functions that 6PPD provides in a tire including:

- Optimal migration rate and diffusion in rubber compounds
  - Adequate solubility and diffusivity in rubber compounds is also referred to as migration and mobility.
  - Continuously present at the surface of the tire to ensure protection of the rubber formulations from degradation due to ozone.
  - Available in rubber formulation over a tire's entire life cycle to ensure protection of the rubber
- Protection against ozone
  - Readily reactive with ozone to prevent crack formation on the surface of the rubber, but not too reactive in order to prevent premature depletion
- Protection against oxygen
  - Reactive with oxygen to prevent hardening of the rubber, loss of strength, and improve tire wear
- Protection against fatigue
  - Reactive with the free radicals generated by the breaks in polymer during flexing. These free radicals can break the polymer chains and crosslinks in the rubber compound that would lead to a loss of strength
- Manufacturing Impact
  - No adverse effects on the processability of rubber compounds
  - Resistance to temperatures encountered during the tire manufacturing process
- No adverse effects on tire safety and performance

# IV. USTMA supports Ecology's analysis of potential 6PPD alternatives, provided it is conducted consistently with the statutory requirements and Ecology's own precedent

In 2019, the Washington State Legislature directed Ecology to implement a regulatory program to reduce toxic chemicals in consumer products (Chapter <u>70A.350</u> RCW). The implementation program is called Safer Products for Washington. The statute provides that Ecology "may not identify the following as priority consumer products … motorized vehicles, including on and off-highway vehicles, such as all-terrain vehicles, motorcycles, side-by-side vehicles, farm equipment, and personal assistive

mobility devices." RCW70A.350.030(5)(a)(vi). The statute further defines "consumer product" to mean "any item, including any component parts and packaging, sold for residential or commercial use." RCW 70A.350.010(1). Appendix G of Ecology's Final Regulatory Determinations ("Final Report") report states that Ecology will not identify "motorized vehicles, including on and off-highway vehicles, such as all-terrain vehicles, motorcycles, side-by-side vehicles, farm equipment, and personal assistive mobility devices" as priority consumer products. Final Report at 364. Under the statute, component parts associated with motorized vehicles (e.g., tires) are exempt. USTMA is concerned that any regulatory action arising out of an alternatives assessment regarding tires as priority products would be outside the scope of Ecology's authority.

Ecology may only restrict the use of a priority chemical in a priority consumer product, if safer alternatives are feasible and available. RCW70A.350.040(3)(a). Ecology has stated that to be considered feasible, an alternative must meet at least one of the following criteria:

- Already used for the application of interest or a similar application.
- Marketed for the application of interest or a similar application.
- Identified as feasible by an authoritative body.
- To be available, an alternative must be either:
  - Currently used for the application of interest; or
  - Offered for sale at a price that is close to the current.

Any alternative to 6PPD identified by Ecology must be feasible and available in the context of safety and performance requirements that tires must meet. An alternative that does not enable a tire to meet safety and performance requirements would be considered per se not feasible and available. Ecology's recently promulgated rule for Cycle 1 recognized limits on Ecology's authority. For example, Ecology's determination reports recognize the limits on its regulatory authority. In the Final Report, Ecology states that "[i]f at any point federal action preempts our ability to implement the restrictions...we will require reporting of priority chemicals in those priority products." Final Report at 25. Ecology, for example, recognized the limits imposed on its regulatory authority as a result of an exemption that the Environmental Protection Agency ("EPA") established for inadvertently generated PCBs and decided not to implement a restriction on inadvertent PCBs in paints and inks. Final Report at 29. To support its determination, Ecology noted that the "only other option for a restriction would be implementing rules identical to EPA." Final Report at 29.

Ecology's report identifying priority chemicals for Cycle 2 ("Priority Chemicals Report") noted that it narrowed its list of chemicals to the seven classes identified "by deprioritizing those with existing effective regulatory structures and prioritizing those with potential for: equitably reducing exposure; preventing regrettable substitutions; reducing environmental persistence; reducing carcinogens, mutagens, reproductive and developmental toxicants, and endocrine disruptors; and reducing production and release volumes." Priority Chemicals Report at 23; *see also* <u>Safer Products for</u> <u>Washington Draft Priority Chemicals for Cycle 2 webinar</u>, at Slide 14 (Ecology deprioritizes chemicals with existing, effective regulatory structures). Moving forward, Ecology needs to recognize and consider the safety and performance requirements that tires must meet and how that impacts Ecology's ability to regulate 6PPD under both federal preemption analysis and Ecology's feasibility analysis.

### V. USTMA welcomes the opportunity to work with Ecology to provide information on the performance of possible alternatives to ensure driver safety

In December 2020, a research paper by Tian et al. 2020 was published that suggests a link between 6PPD-quinone and coho salmon mortality.<sup>3</sup> 6PPD-quinone is not used in tire manufacturing. It is a transformation product of 6PPD that may form when 6PPD reacts with oxygen and/or ozone. 6PPD is an antioxidant and antiozonant that helps prevent the degradation and cracking of rubber compounds caused by exposure to oxygen, ozone, temperature fluctuation, and flexing induced fatigue. These benefits of 6PPD are critical to effective tire endurance and thus ultimately to motor vehicle safety. That said, as a science-driven industry committed to safety and environmental stewardship, we take the findings of this study seriously. In December 2020, USTMA requested that the California Department of Toxic Substances Control ("DTSC") include 6PPD in tires on the 2021-2023 Priority Products Work Plan for the Safer Consumers Products Regulation (SCPR). USTMA sought a review of 6PPD in tires in California rather than Washington, because at the time the Tian et al. study was released, DTSC was in the process of revising its Priority Products Work Plan and the SCPR provided the fastest path forward to complete an alternatives analysis on 6PPD. A review of 6PPD in tires under the SCPR provides a scientific, regulatory framework to analyze whether alternatives exist that will enable tire manufacturers to ensure both tire and environmental safety. DTSC added 6PPD in tires to the Priority Products Workplan in early 2021 and since that time, USTMA has worked to support a review of 6PPD in tires under the SCPR.

The Safer Products for Washington enacting legislation encourages Ecology to consider actions taken by other states. For example, when selecting priority consumer products, Ecology is required to consider certain factors, including "[i]f another state or nation has identified or taken regulatory action to restrict or otherwise regulate the priority chemical in the consumer product." RCW 70A.350.030(2)(e). When determining regulatory actions, Ecology is encouraged to consider whether a "restriction would be consistent with regulatory actions taken by another state or nation on a priority chemical or members of a class of priority chemicals in a product." RCW 70A.350.040(4)(b).

During Cycle 1 of the Safer Products for Washington Program, Ecology acknowledged relevant activities in other states. For example, Ecology's Regulatory Determinations Report to the Legislature stated that to identify safer alternatives, Ecology used – among other things – "[e]xisting alternatives assessments."<sup>4</sup> Ecology also cited to and used DTSC studies developed under the California SCPR<sup>5</sup> as well as relevant material restrictions in other states.<sup>6</sup>

Given the similarities between an ongoing alternatives analysis in California and the process on which Ecology is embarking, Ecology should closely coordinate with DTSC to ensure a thorough and consistent assessment and analysis of the potential alternatives. After DTSC finalizes its designation of automotive tires containing 6PPD as priority products in California, this will trigger an obligation for the tire industry to develop and submit to DTSC an alternatives analysis. This alternatives analysis is already well underway – USTMA is not waiting for DTSC to finalize the designation before starting it. Once the alternatives analysis is complete, DTSC will evaluate it and decide whether it meets applicable legal requirements. DTSC will also decide whether to undertake a regulatory response regarding automotive

<sup>&</sup>lt;sup>3</sup> Tian Z et al. (2021). A ubiquitous tire rubber–derived chemical induces acute mortality in coho salmon. Science. 371(6525):185–189. doi: 10.1126/science.abd6951.

<sup>&</sup>lt;sup>4</sup> Regulatory Determinations Report to the Legislature: Safer Products for Washington Cycle 1 Implementation Phase 3, at 64.

<sup>&</sup>lt;sup>5</sup> Id. at 286; 358.

<sup>&</sup>lt;sup>6</sup> *Id*. at 68.

tires containing 6PPD. USTMA is committed to providing Ecology with the same information it provides DTSC as part of the AA process on an ongoing and timely basis.

The California and Washington processes will be similar and directly linked. Both alternative analyses will involve evaluating whether potential alternatives are feasible for use in tires and would represent safer alternatives. At the end, the agencies will decide what regulatory response to undertake, if any. The processes are compatible and consistent. In California, the AA is already underway and, once completed, will be reviewed and approved by DTSC. Furthermore, because the tire industry will be directly involved in the alternatives analysis in California and currently possesses the best information about potential 6PPD alternatives, tire safety, and tire regulatory requirements, it would be helpful and efficient for Ecology to leverage the California alternatives analysis. USTMA is encouraged by Ecology's recent June 21, 2023 webinar discussing the draft priority chemicals for Cycle 2, in which Ecology acknowledged that "significant efforts by Washington, California, and other states, and the tire industry are being undertaken to identify safer alternatives to 6PPD used as an anti-degradant in vehicle tires." Safer Products for Washington Draft Priority Chemicals for Cycle 2 webinar, at Slide 48. Moving forward, USTMA requests that Ecology closely coordinate with DTSC as part of its 6PPD analysis and strongly consider adopting positions consistent with DTSC's.

### VI. The assessment and testing processes necessary to evaluate potential alternatives to 6PPD in tires, and to ensure tire safety and performance, are complex and rigorous

To identify a possible alternative that ensures both motorist and environmental safety, extensive testing is needed. A variety of laboratory screening tests must be performed for each tire component to assess functionality of the candidate alternative in the rubber compound for each tire component. These laboratory screening tests must be performed and completed with satisfactory results before a tire is built. Candidate alternatives that do not pass these initial laboratory screening tests would not be moved forward in the testing evaluation process. Candidate alternatives that do pass this initial screening testing would be moved on to the evaluation process which would include building a tire and performing a multitude of performance and safety tests before the tire could be assessed on a vehicle. Tires containing the candidate alternative would need to meet these performance and safety tests before being assessed for further testing on vehicles. The testing to evaluate candidate chemicals in tires is extensive and required to ensure compliance with applicable Federal Motor Vehicle Safety Standards (FMVSS) and other tire performance and safety requirements.

# VII. The persistence and bioaccumulation potentials of 6PPD have been mischaracterized throughout Ecology's Draft Report and should be corrected.

In various sections throughout Ecology's Draft Report, 6PPD is stated to be potentially persistent and bioaccumulative, however 6PPD is not characterized as either according to regulatory agencies or other authoritative bodies including OECD, ECHA, Environment Canada and Chemicals Inspection & Testing Institute of Japan (CERI). In the Hazard of 6PPD and the Environmental Fate sections of Ecology's Draft Report, should be revised to accurately reflect the designations for 6PPD.

In the Hazard of 6PPD section of Ecology's Draft Report, the designation of high bioaccumulation is based on GreenScreen criterion, not regulatory criterion. For example, because the bioconcentration factor for 6PPD is less than 2000, it is not given the bioaccumulation classification in EU under REACH or by Env. Canada. Similarly, OECD 2004 concluded that 6PPD is not bioaccumulative. Testing data provided in the REACH dossier states that 6PPD hydrolyses rapidly within a half- life of about 8 hours,

hence it is not persistent. As such experimental data for the hydrolysis products 4 hydroxydiphenylamine, N-phenyl-p-benzoquinone monoimine and 1,3 -dimethylbutylamine were taken into account. A QSAR model for 6PPD yielded a BCF of 569. Even considering uncertainties using a QSAR, this value indicates that the original substance does not meet the bioaccumulation criterion of 2000. Additionally, the bioconcentration factor of the two main hydrolysis products of 6PPD, 4-anilinophenol (4-hydroxydiphenylamine) and its oxidized form N-Phenylphenyl-p-benzoquinone monoimine, were investigated by y the Chemicals Inspection & Testing Institute of Japan (CERI, 1995, National Institute of Technology and Evaluation 2002) according to OECD Guideline 305 C, using Cyprinus carpio as test organism. The BCFs ranged from 3.3 - 49 and < 1.2 - 23, and 1.7-17, respectively and therefore do not meet the criterion for classification as bioaccumulative. The statement that there is not much known about the hazards of 6PPD-quinone seems overly broad. Over the past 2.5 years since the chemical was discovered, quite a bit has been learned about the ecotoxicity. This sentence should be rephrased to indicate that information regarding the hazards of 6PPD-q is emerging, with most of the currently available hazard data focused on aquatic species.

In the Environmental Fate section of Ecology's Draft Report, the Castan et al. 2022 study is referenced as evidence of potential 6PPD bioaccumulation, however, on the contrary, the study does not demonstrate that 6PPD bioaccumulated. In fact, the researchers reported that the uptake into the lettuce leaves was followed by a rapid concentration decrease. Over the course of 14 days of exposure,6PPD peaked at 7 days and then decreased to concentrations close to the limit of quantification (LOQ). In a similar manner, the Ji et al. 2022 study was also referenced as evidence of bioaccumulation of 6PPD, however the study does not show bioaccumulation. Rather, the researchers detected the presence of 6PPD in 2 out of 10 fish that were tested and 6PPD-quinone was detected but at concentrations less than the limit of quantification in 1 of 10 fish. This study involved a small sampling and analysis of fish from a food market in Beijing China and was not a guideline study to determine bioaccumulation (OECD Method 305).

#### VIII. Correction is needed to Referenced Hazard Assessment

Please correct the information in Table 29 to reflect that 6PPD-quinone has a CAS Number. It is 2754428-18-5.

#### IX. USTMA recommends that Ecology revise the Human Exposure section of Ecology's Draft Report to ensure it accurately reflects the findings in the published literature.

USTMA asks that Ecology clarify the findings for the citation to Armada et al. 2023. The authors detected 6PPD and 6PPD-quinone in the synthetic gastrointestinal fluids extract of the crumb rubber but did not quantify the amount of either substance. USTMA suggests adding a sentence related to the Schneider et al.(2020a)<sup>7</sup> study wherein the authors measured the bioaccessibility of 6PPD in synthetic gastrointestinal fluids extract from crumb rubber and reported the substance had very low bioaccessibility; approximately 0.58% was bioaccessible. Further, the authors concluded in a risk assessment of children potentially exposed to the crumb rubber from use in artificial turf fields that the

<sup>&</sup>lt;sup>7</sup> Schneider, K; de Hoogd, M. ; Haxaire, P.; Phillips, A.; Bierwisch, A.; Kaiser, E. (2020) (a). ERASSTRI – European Risk Assessment Study on Synthetic Turf Rubber Infill – Part 2: Migration and monitoring studies. Science of the Total Environment: 718 (2020) 137173.

risk ratio was less than 0.2 and therefore demonstrated a low potential for health risk (Schneider et al. 2020b)<sup>8</sup>.

## X. USTMA recommends that Ecology revise the Environmental Monitoring Data section of Ecology's Draft Report to ensure it accurately reflects the findings in the published literature.

The first sentence in the Environmental Monitoring Data section which states that 6PPD and its transformation products "are likely present in almost all media worldwide" is grossly overstated. In fact, there are very few studies and samples for 6PPD and 6PPD-quinone that have been reported in the published literature and the detection frequency has been very low in some media. For example, Rauert et al. 2022 reported non-detects at several of the sampling sites in Australia and Johannessen reported that all air samples from around the world were either non-detect or less than the limit of quantification for 6PPD. As such USTMA recommends that the sentence be revised to reflect the current state of knowledge.

The second paragraph in this section is misleading with respect to the comparison of measured concentrations of 6PPD-quinone to the LC50 for coho salmon. The sentence "Again, these reported concentration often are above the LC50 values of sensitive species," is inaccurate because of the studies cited, only Johannessen et al. 2022 measured concentrations greater than the LC50s for coho salmon and other sensitive species. H.Y. Zhang et al. 2023 and Rauert et al., 2022 did not measure concentrations above the coho salmon LC50. R. Zhang et al. 2023 measured 6PPD-quinone in WWTP effluents in 1 out of 118 samples and the measurement did not exceed the LC50 for coho.

# XI. USTMA recommends that Ecology revise the Potential to Contribute to Adverse Effects section of Ecology's Draft Report to ensure it accurately reflects the findings in the published literature.

In sensitive species discussion, USTMA recommends that citations be provided for the studies associated with this sentence:

"Although 6PPD-q shows extremely high acute toxicity to some species, including coho salmon, brook trout, and rainbow trout, it shows much lower acute toxicity to other species, such as zebrafish. 6PPD-q was not acutely toxic, even at high concentrations, to species closely related to coho salmon, such as Atlantic salmon, chum salmon, and sockeye salmon."

USTMA disagrees with the statement that chronic effects of 6PPD-quinone to aquatic species have been observed and that those effects occur at much lower concentrations than the LC50 value based on the studies cited. In fact, Ji et al 2022 was not a chronic study - it was for 12 hrs and the lowest concentration tested was 50 ug/L, which is well above the LC50 values for 6PPD-quinone and coho salmon. Also Varshney et al. 2022 was an acute study, not a chronic study and the lowest tested concentration of 1 ug/L was above the coho LC50, and furthermore effects were not seen until 10 ug/L.

#### XII. Conclusion

USTMA thanks Ecology for the opportunity to provide comments on the Draft Identification of Priority Chemicals Report to the Legislature, Safer Products for Washington Cycle 2, Implementation

<sup>&</sup>lt;sup>8</sup> Schneider, K.; Bierwisch, A.; Kaiser, E. (2020)(b). ERASSTRI – European Risk Assessment study on synthetic turf rubber infill – Part 3: Exposure and risk characterisation. Science of the Total Environment. 718 (2020) 137721

Phase 1. We welcome the opportunity for continued dialogue with Ecology on this topic and the opportunity to discuss our comments in greater detail. If you have any questions, please contact Jamie McNutt (jmcnutt@ustires.org; 202-682-4845).

Attachment B:

Preliminary (Stage 1) Alternatives Analysis Report Motor Vehicle Tires Containing N-(1,3-dimethylbutyl)-N'-phenyl-p-phenylenediamine

(6PPD)

July 17, 2024

### Preliminary (Stage 1) Alternatives Analysis Report Motor Vehicle Tires Containing N-(1,3dimethylbutyl)-N'-phenyl-p-phenylenediamine (6PPD)

Prepared for U.S. Tire Manufacturers Association (USTMA) 1400 K St NW #900 Washington, DC 20005

July 17, 2024



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### **Abbreviations**

6PPD 6PPDQ 6QDI 7PPD 77PD AA ASTM BCF CAFE CaIDTSC CAFE CAFE CAIDTSC CARB CAS CASRN CCPD CCR CPPD CPSC CSI DAPD DNA DNPDA DNPDA DOPD DTPD DTPD DTSC ECHA	<ul> <li>N-(1,3-Dimethylbutyl)-N'-phenyl-p- phenylenediamine</li> <li>N-(1,3-Dimethylbutyl)-N'-phenyl quinone diimine</li> <li>N-(1,4-Dimethylpentyl)-N'-phenyl-p-phenylenediamine</li> <li>N,N'-Bis(1,4-dimethylpentyl)-p-phenylenediamine</li> <li>Alternatives Analysis</li> <li>American Society for Testing and Materials</li> <li>Bioconcentration Factor</li> <li>Corporate Average Fuel Economy</li> <li>California Department of Toxic Substances Control</li> <li>California Air Resources Board</li> <li>Chemical Abstracts Service Number</li> <li>Chemical Abstracts Service Registry Number</li> <li>N,N'-Dicyclohexyl-p-phenylenediamine</li> <li>California Code of Regulations</li> <li>N-Cyclohexyl-N'-phenyl-p-phenylenediamine</li> <li>U.S. Consumer Product Safety Commission</li> <li>Chemical Scoring Index</li> <li>Diaryl-p-phenylene diamine</li> <li>Deoxyribonucleic Acid</li> <li>N,N'-Di-2-naphthyl-p-phenylenediamine</li> <li>4,4'-Dioctyldiphenylamine</li> <li>N,N'-Ditolyl-p-phenylenediamine</li> <li>N,N'-Ditolyl-p-phenylenediamine</li> <li>Department of Toxic Substances Control</li> </ul>
EPDM	Ethylene Propylene Diene Monomer
EU FHSA	European Union Federal Hazardous Substances Act
FMVSS	Federal Motor Vehicle Safety Standards
GHS	Globally Harmonized System of Classification and Labelling of Chemicals
GPC	Global Product Classification
GRAS	Generally Recognized as Safe
GWP	Global Warming Potential
HCI	Hydrochloric Acid
Hg	Mercury
HSDB	National Library of Medicine's Hazardous Substance Data Base
IMAP	Australia Inventory Multi-Tiered Assessment and Prioritisation
IPCC	Intergovernmental Panel on Climate Change
IPPD	N-Isopropyl-N'-phenyl-p-phenylenediamine
К <sub>ос</sub>	Log Organic Carbon Partition Coefficient
K <sub>ow</sub>	Log Octanol-Water Partition Coefficient Median Lethal Concentration
LC <sub>50</sub>	

LCA	Life Cycle Assessment
LOLI	Underwriters Laboratories, Inc.'s List of Lists
MCDA	Multi-Criteria Decision Analysis
NBC	Nickel Dibutyldithiocarbamate
NESHAP	National Emission Standards for Hazardous Air Pollutants
NHTSA	National Highway Traffic Safety Administration
ODI	Office of Defects Investigation
ODP	Ozone-Depleting Potential
OE	Original Equipment
OECD	Organisation for Economic Co-operation and Development
OEM	Original Equipment Manufacturer
OPERA	Open (Quantitative) Structure-Activity/Property Relationship App
OVSC	Office of Vehicle Safety Compliance
PBT	Persistent, Bioaccumulative, and Toxic
PPDs	Paraphenylene Diamines
PPE	Personal Protective Equipment
pphm	Parts Per Hundred Million
QSAR	Quantitative Structure-Activity Relationship
RE	Responsible Entity
REACH	Registration, Evaluation, Authorisation, and Restriction of Chemicals
RKIS	Rotary Kiln Incinerator Simulator
RRC	Rolling Resistance Coefficient
SBR	Styrene Butadiene Rubber
SCP	Safer Consumer Products
SDS	Safety Data Sheet
SMART	Shape Memory Alloy Radial Technology
SMILES	Simplified Molecular Input Line Entry System
TAC	California's Toxic Air Contaminant List
TAPDT	2,4,6-tris-(N-1,4-dimethylpentyl-para-phenylenediamino)-1,3,5 triazine
TDF	Tire-Derived Fuel
TIN	Tire Identification Number
TMQ	Poly(1,2-dihydro-2,2,4-trimethyl-quinoline)
TREAD Act	Transportation Recall Enhancement, Accountability, and Documentation Act
TRWP	Tire and Road Wear Particles
TSCA	Toxic Substances Control Act
UVCBs	Unknown or Variable Compositions, Complex Reaction Products, and Biological Materials
UL	Underwriters Laboratories, Inc.
UN	United Nations
URMS	Urban Runoff Mortality Syndrome
US DOT	United States Department of Transportation
USGS	United States Geological Survey
US EPA	United States Environmental Protection Agency
US FDA	United States Food and Drug Administration
USPTO	United States Patent and Trademark Office
USTMA	U.S. Tire Manufacturers Association
UTQGS	Uniform Tire Quality Grading Standards
VOC	Volatile Organic Compound
WA DOE	Washington State Department of Ecology

### **Executive Summary**

Effective October 1, 2023, The California Department of Toxic Substance Control (DTSC) listed motor vehicle tires containing N-(1,3-dimethylbutyl)-N'-phenyl-p-phenylenediamine (6PPD) as a "priority product" under the Safer Consumer Products (SCP) Regulations.

This Stage 1 Alternatives Analysis (AA) report was prepared under the SCP Regulations on behalf of a Consortium<sup>1</sup> comprising some, but not all, manufacturers of the Priority Product for sale in California. As conceived by Gradient and the Consortium, the initial goal of an AA is to answer the following question: Do potentially safer, functionally acceptable, and technically feasible alternatives to the Priority Product exist that should be given a more in-depth consideration to determine if they qualify as acceptable alternatives?

This Stage 1 AA was based on available information and sought to determine whether there are possible alternatives to the priority product that should be considered in greater depth to evaluate if they are suitable alternatives to replace the priority product under regulatory guidelines (CalDTSC, 2017a). Important elements of this work were considering the requirements (legal, regulatory, or otherwise) for the priority product, determining the function of the chemical of concern in the priority product, determining whether simple elimination was possible and assessing relevant factors to identify those that would suggest a material difference exists that could affect the decision as to whether a possible alternative is a suitable replacement for the priority product. This last element required compiling extensive information on the potential hazards, potential performance, and chemical and physical properties of the possible alternatives.

6PPD is used in tires as an antidegradant, protecting the components of the tire from attack by ozone, oxygen, thermal degradation, and mechanical fatigue, *etc.* In late 2020, it was first reported that when it reacts with ozone, 6PPD forms a degradation product, 6PPD quinone (6PPDQ) (Tian *et al.*, 2021); this reaction with ozone is part of the way in which 6PPD protects tire rubber from degradation. Without 6PPD, tires will quickly develop cracks and fractures as the rubber polymer is degraded. The antidegradant function of 6PPD in tires is therefore essential to their safe use, and elimination of 6PPD without replacement is not an option.

One way 6PPD and 6PPDQ may enter the environment is through tire and road wear particles (TRWP), which are produced as the tire grips the road surface during driving. Some 6PPD and 6PPDQ on the tire surface may also be washed off the tire by rain or vehicle washing. US EPA has also noted uncertainty about levels of 6PPDQ exposure to the environment from tires relative to other potential sources (Freedhoff, 2023).

Recent laboratory studies stated that 6PPDQ is acutely toxic to coho salmon, and those studies suggest that 6PPDQ exposure from stormwater runoff, under certain conditions, may result in mortality of these fish in streams and rivers located near roadways (Tian *et al.*, 2021). In its Product-Chemical Profile for 6PPD in motor vehicle tires (CalDTSC, 2022), DTSC notes that "Behavioral symptoms of URMS [Urban Runoff Mortality Syndrome] occur within a few hours of exposure to urban runoff and include 'erratic surface swimming, gaping, fin splaying, and loss of orientation and equilibrium' (Scholz *et al.*, 2011)."

<sup>&</sup>lt;sup>1</sup> The Consortium refers to the group of Responsible Entities that prepared this Stage 1 AA in accordance with the SCP Regulations.

Some other salmonid species may be similarly affected under similar circumstances, although with lower toxicological potency than coho, while other species apparently exhibit negligible toxicological susceptibility (see for example Brinkmann *et al.*, 2022; Hiki and Yamamoto, 2022; Greer *et al.*, 2023a). Studies have shown that the closely related Chinook salmon (see for example Lo *et al.*, 2023) and sockeye salmon (see for example, Greer *et al.*, 2023a) are much less susceptible.

The biological mechanisms by which the toxicity occurs, and why it affects some species and not others, is not yet known but is the subject of active research. To date, scientific studies report that coho salmon remain the most sensitive of the species examined (*e.g.*, Brinkmann *et al.*, 2022; Lo *et al.*, 2023; Foldvik *et al.*, 2024). In its Product-Chemical Profile for Motor Vehicle Tires Containing 6PPD, DTSC provides a number of possible mechanisms of action for 6PPDQ causing URMS but notes that these require further study (CalDTSC, 2022). With respect to non-aquatic toxicity, DTSC also notes that, "6PPD has been shown to be acutely toxic through oral and dermal routes of exposure in animal [*e.g.*, rodent] studies, although at relatively high dosages" but that, "[n]o studies are currently available to evaluate the human health effects of 6PPD-quinone exposure."

It is important to note that while vehicle tires are large consumers of 6PPD, 6PPD is also used in other rubber products. Additionally, tire manufacturers began using 6PPD in tire manufacturing in the mid-1960s to early 1970s. However, significant declines in the coho salmon populations in California were observed as early as the 1940s, pre-dating the use of 6PPD in tires by several decades (California Dept. of Fish and Game, 2002).

In this Stage 1 AA we considered several different types of alternatives to 6PPD as an antidegradant in tires: (1) other phenylene diamines (PPDs) that are the most logical and possibly easiest to implement alternatives to 6PPD, and (2) non-PPD alternatives that likely pose greater challenges in terms of incorporation into tire chemistry.

For all of the alternatives under consideration, information is incomplete regarding their potential hazards to coho and other fish species, although for many alternatives, data are available on other types of hazards to determine if those alternatives are unsuitable. Similarly, for some of the alternatives, initial bench scale data on performance as an antiozonant is available but definitive data on the ability to use the possible alternative in manufacturing a tire and data on the performance of that tire in all of the required tests are lacking. At this point in time, we can state that seven materials – 7PPD, IPPD, 77PD, CCPD, specialized graphene<sup>2</sup> (*e.g.*, Prophene<sup>TM</sup>), octyl gallate, and Irganox 1520 – warrant further evaluation as potential alternatives.<sup>3</sup> Consequently, since there do appear to be possible alternatives that merit additional consideration, a two stage AA as described by the SCP regulations is appropriate. It is expected that additional data will become available within the time frame of the second stage AA that will allow for a more detailed evaluation of a suitable alternative(s).

As required by the SCP regulations, the following is a summary of information contained in each section of the stage 1 AA report.

• Section 1 identifies the persons who oversaw the preparation of this report.

<sup>&</sup>lt;sup>2</sup> The materials referred to as graphene in this report are graphene-based materials (sometimes referred to as a graphene nanoplatelet) with a surface area not greater than 180 m<sup>2</sup>/g, and a carbon content greater than 99% and an oxygen content less than 1%. The lateral particle size of these materials is between 100 nm and 5  $\mu$ m.

<sup>&</sup>lt;sup>3</sup> This preliminary alternatives analysis outlines the process to assess whether alternatives can replace chemicals or technologies of concern based on their hazards, performance, and exposure potential. The term "hazard" as used throughout this document is used in keeping with the relevant guidance documents.

- Section 2 identifies the Consortium members submitting this report and addresses how they will be submitting supply chain information as a separate confidential business information (CBI) submittal.
- Section 3 identifies the Priority Product (motor vehicle tires containing 6PPD) and the chemical of concern (6PPD). Tables listing the manufacturers and their priority products, consistent with the product names on the submitted Priority Product Notifications (PPNs), are also included as appendix B. Section 3 identifies the function of the chemical of concern in the Priority Product (*i.e.*, antidegradation). Section 3 also discusses the many performance requirements of motor vehicle tires and identifies tests that are conducted to evaluate this performance. Key performance criteria include static and dynamic antioxidant and antiozonant operation modes, including but not limited to high speed performance, rolling resistance, endurance, wear rate, and traction in dry, wet, and snow conditions. Some of these performance criteria are related to product safety and are mandated by federal regulations. Other performance criteria are related to vehicle fuel efficiency or customer expectations (*e.g.*, tire warranty, ride comfort). Section 3 concludes with a discussion of how removal of 6PPD from tires without a functional replacement is not possible and thus an alternative performing the same function is required.

Section 4 begins with a scoping discussion that describes technologies that fall outside the scope of this AA. Alternative tire technologies, such as non-pneumatic tires, are not suitable alternatives because they would also require the use of antidegradant chemicals in their rubber compounds or cannot be currently mandated or implemented by tire manufacturers. Some of these technologies are currently theoretical and have not been demonstrated to be useable for cars, trucks, or buses. A second non-viable option is a particle collector system which would reduce, somewhat, the load of particles emitted during tire use. This option is not suitable because it is beyond the ability of tire manufacturers to mandate such technologies and also because the technology's effectiveness in reducing migration of 6PPD to the environment appears limited. Although reduction in exposure potential does constitute a viable alternative under the SCP regulations, it is not clear that this technology could reduce exposure to the extent that would be considered meaningful. A third nonviable option involves waxes and coatings. When used alone, waxes and coatings can only protect tires from ozone under static conditions. While cars are in motion (*i.e.*, dynamic ozone conditions), waxes and coatings wear off almost immediately and expose the underlying rubber to ozone attack. which causes earlier failure. DTSC's Product-Chemical Profile (CalDTSC, 2022) noted that waxes and coatings are "likely infeasible means of protecting the tire tread." Section 4 next discusses how information was obtained to identify possible alternatives to 6PPD in motor vehicle tires and the criteria used to select a subset of possible alternatives for further evaluation in this Stage 1 AA. Following this approach, over 70 candidate alternatives for 6PPD were identified (Appendix F). The section describes how each of these possible alternatives was screened and scored in terms of its likely feasibility as an antiozonant and how those that appeared promising were selected for full evaluation in this Stage 1 AA. Possible alternatives with compound effectiveness against ozone scores of 3 or 4 were evaluated further in this preliminary (Stage 1) AA. However, all possible alternatives suggested by DTSC in the Priority Product profile (CalDTSC, 2022) and all possible alternatives listed on DTSC's website (CalDTSC, 2024), including but not limited to those possible alternatives suggested by the Washington State Department of Ecology (WA Ecology, 2021) were retained for this Stage 1 AA, irrespective of their compound effectiveness against ozone score. Overall, a total of 43<sup>4</sup> possible alternatives were evaluated further in this Stage 1 AA, including 19 other phenylene diamines (PPDs) and 24 non-PPD based antidegradants.

<sup>&</sup>lt;sup>4</sup> While waxes and coatings were carried from Appendix F into the Stage 1 AA because they were suggested by DTSC, they were not evaluated further for hazards, exposure, and performance because they are non-viable alternatives when tires are in motion. Thus, they did not contribute to the count of possible alternatives evaluated in this Stage 1 AA. See Section 4.4 for more discussions on waxes and coatings.

Section 4 concludes with a discussion of factors that were considered to be relevant to this Stage 1 AA. It includes a discussion of conceptual exposure models that show how individuals and environmental receptors may be exposed to 6PPD across the tire product life cycle. It also describes what is known regarding the relevance of each life cycle aspect noted in the SCP regulations to the evaluation of different alternatives. A life cycle assessment (LCA) is available for tires but is not available for 6PPD nor for any of the possible alternatives, making quantitative comparisons among the alternatives to determine whether there is a material difference impossible. More qualitative arguments, based on raw materials used in manufacturing 6PPD and the alternatives, their chemical properties and the required properties of any alternative (e.g., lifespan of the product, ability to be recycled or repurposed) suggest that for the use, waste generation, recycling/reuse and end of life portions of the product lifecycle, there do appear to be potential material differences among the priority product and possible alternatives but this would need to be further explored in Stage 2. For other life stages (*i.e.*, raw materials extraction, intermediate materials processing, product manufacturing, product packaging, and operation and maintenance) there are unlikely to be material differences between the Priority Product and the possible alternatives. For the remaining life cycle stages (*i.e.*, resource consumption and distribution) it is unclear whether there will be differences among products because relevant data for the possible alternatives are lacking.

• Section 5 begins with a review of health hazard information for the Priority Product and the possible alternatives. Overall, all of the alternatives involve reactive molecules, which was anticipated given that the requirement is for a chemical that can scavenge ozone and oxygen. We obtained data on the hazards of the possible alternatives from two primary data sources – European Chemicals Agency (ECHA) Registration, Evaluation, Authorization and Restriction of Chemicals (REACH) dossiers and GreenScreens<sup>®</sup> conducted by ToxServices for the State of Washington, Department of Ecology. Using an adaptation of a published scoring approach, we found that 24 of the 43 possible alternatives had insufficient data to assign a hazard score. Of the 19 possible alternatives with sufficient data, 6 had total scores estimating at least 30% potential hazard reduction relative to 6PPD. These 6 chemicals included 1 PPD (77PD) and 5 non-PPDs (DLTP, TAPDT, specialized graphene, octyl gallate, and Irganox 1520). Four of these possible alternatives had hazard scores that were an order of magnitude better than 6PPD: DLTP, specialized graphene, octyl gallate, and the Irganox 1520. Note that a reduction in hazard alone does not define an appropriate alternative.

In addition to the hazard scoring, we researched and tabulated available information about the potential of the possible alternatives to affect coho salmon or related salmonid species. Data on this subject are extremely limited. Concentrations lethal to 50 percent of the population (LC<sub>50</sub>) in salmonids were available for eight of the possible alternatives (6QDI, 77PD, DTPD, CCPD, DAPD, NBC, ethoxyquin, and N-phenyl-1-naphthylamine). Two studies evaluated the quinone transformation products of five possible alternatives (77PDQ, CPPDQ, DPPDQ, DTPDQ, and IPPDQ). These studies suggest there may be lower acute toxicity of 77PDQ, CPPDQ, DPPDQ, DTPDQ, DTPDQ, and IPPDQ, and IPPDQ relative to 6PPDQ, however, these results are preliminary and unconfirmed.

Section 5 also discusses the preliminary and unpublished results of testing commissioned by USTMA and conducted by the US Geological Survey (USGS). This testing used *in vitro* (isolated cell) systems to study the potential toxicity of 6PPD and a small number of alternatives (the number of alternatives was limited so as to be able to have data to consider in the Stage 1 AA). The results of that testing showed differential toxicity relative to 6PPD, providing a preliminary indication that not all PPDs pose the same degree of hazard to coho as 6PPD.

We also explored potential environmental transformation products of the possible alternatives and examined their chemical and toxicological properties. Using ECHA dossiers as the source of

transformation product information, we found that a number of PPDs likely share the same potential breakdown products as 6PPD (*e.g.*, aniline and p-benzoquinone). The extent to which transformation actually occurs from antidegradant in TRWP is unknown. For many of the possible alternatives, transformation product information was not available in the ECHA dossier. This lack of information represents a significant uncertainty in the AA and will need to be addressed in Stage 2.

Section 5 next discusses product performance. Performance information was grouped into three different source categories: historical data from patents and other information published before 2020 (the year the Tian *et al.* [2021] publication first appeared on-line); data from recent bench scale testing of a few alternatives by Flexsys (the same alternatives tested by USGS); and recent data from patents or other sources published in 2020 up to January 2024. In tables relevant to each category, the findings for each alternative regarding potential performance are summarized and the citation to the relevant study is provided. We characterized the performance data into the following four categories:

- Some promising ozone data. These possible alternatives had positive data regarding ozone performance in screening level tests (*i.e.*, bench scale testing, not in finished tires).
- **Limited ozone data.** These possible alternatives had some positive data for ozone performance in screening level tests but there were some concerns about the study, often a lack of an appropriate control or a lack of information about controls.
- **Insufficient data, no ozone data.** The chemical lacks any data related to performance as an antiozonant. Note that a chemical having positive data as an antioxidant does not indicate the chemical can perform as an antiozonant (Akrochem, 2010).
- **Poor ozone performance.** For these possible alternatives, the available data indicate the chemical does not perform as an antiozonant or performs poorly compared to 6PPD.

Those possible alternatives that have either some promising ozone data or limited ozone data were considered to have acceptable preliminary performance and suitable for Stage 2 if they also met other Stage 1 AA criteria. Note that an indication of promising or limited ozone performance in screening type tests alone does not define an appropriate alternative.

Those possible alternatives that have either some promising ozone data or limited ozone data are eligible to be included in the Stage 2 AA if they also have hazard information.

Section 5 concludes with a review of relative exposure information for the Priority Product and the possible alternatives. We gathered chemical specific physiochemical data for all of the alternatives, as suggested in CalDTSC's "Alternatives Analysis Guide" (CalDTSC, 2017a). Some of the possible alternatives have substantially less water solubility than 6PPD (*e.g.*, DOPD, DLTP, RU997, and TAPDT) which could affect their environmental partitioning. Similarly, some have substantially higher log K<sub>ow</sub> values (an indication of partitioning into organic materials) than 6PPD (*e.g.*, DLTP, Ru997/Irgazone 997 blend, TAPDT, and DOPD) which again could result in different environmental behavior. Some also have substantially different vapor pressures (some higher, some lower) which could affect workplace exposures. While this evaluation provided some insight into the ingredient-level exposure potential of the possible alternatives, ideally, we would compare the product-level exposure data, because the ingredients are meant to react and create a structure that is distinctly different from the individual ingredients. Because the relative importance of mobility in one environmental medium *versus* another is not clear, no product-level exposure information is available at this time for any alternative.

- Section 6 presents the conclusions of the Stage 1 AA. This section describes how information on chemical hazard, performance, and exposure potential, described in detail in Section 5, are aggregated into an overall comparison table (Table 5.15) and used to determine whether a particular possible alternative should be further evaluated in the Stage 2 AA. Chemicals selected for evaluation in Stage 2 had (1) similar or reduced overall hazard relative to 6PPD based on the available information, (2) screening level performance data indicating a potential to perform in tires as an antiozonant (*i.e.*, some promising ozone data or limited data), and (3) acceptable physical/chemical properties indicative of exposure potential. The seven chemicals that met these criteria were: 7PPD, IPPD, 77PD, CCPD, a specialized graphene, octyl gallate, and Irganox 1520. Twenty-four of the 43 alternatives evaluated were eliminated from further consideration in Stage 2 because they have complete hazard data gaps. The remaining 12 possible alternatives were eliminated due to a lack of performance data or because available data indicated they would not perform well against ozone. Because the Stage 1 AA determined there were possible alternatives to the Priority Product, the report concludes that a Stage 2 AA is appropriate.
- Section 7 discusses the proposed Work Plan for the Stage 2 AA. This includes a table of expected timing for meetings with DTSC and the types of additional information that will be gathered to support the Stage 2 assessment.
- Section 8 discusses uncertainties encountered in preparing this Stage 1 AA and the potential implications these may have for the results of the Stage 2 AA. For example, any potential acute aquatic toxicity hazards reported in salmonids may not represent potential hazards or risk associated with their presence in a final vehicle tire product, as any potential hazard of these chemicals is dependent upon their potential migration from vehicle tires and TRWP, which if any, remains unclear. This section also includes suggestions for how these uncertainties can be reduced in the Stage 2 analysis.
- Section 9 lists the report references.

Appendices providing some of the supporting data and other relevant information are included at the end of the report.

### **1** Preparer Information

### **Preparer Information:**

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<sup>&</sup>lt;sup>5</sup> Submitting on behalf of a Responsible Entity group comprised of the responsible entities listed here.

### 2 Consortium Members and Supply Chain Information

#### **USTMA 6PPD Alternatives Analysis Consortium Membership**

() indicates name(s) used for PPN if different

American Kenda Rubber Industrial Co., LTD (America Kenda Rubber Ind Co.) Apollo Tires (US) Inc. (Apollo Tyres Limited) Bridgestone Americas, Inc. CEAT Ltd. China Manufacturers Alliance, LLC Continental Tire the Americas, LLC GITI Tire (USA), Ltd. (Giti Tire) Hankook Tire America Corp. Hebei Wanda Tyre Co., Ltd. Jiangsu General Science Technology Co., Ltd. JK Tyre & Industries Limited Kumho Tire U.S.A., Inc. (Kumho Tire) Linglong Americas, Inc. Maxxis International – USA (Maxxis Technology Center) (Cheng Shin USA Tech Center) Michelin North America, Inc. (+ PT. Multistrada Arah Sarana Tbk) Nexen Tire America, Inc. (Nexen Tire Corporation) Nokian Tyres Inc. (Nokian Tyres US Operations LLC) North American Commercial Tire Resources Inc. (Guizhou Tyre Co., Ltd.) Otani Radial Tire Co, Ltd and Otani Tire Co, Ltd Pirelli Tire LLC Prinx Chengshan Holdings, Ltd Prometeon Tyre Group Commercial Solutions, LLC Oingdao Sentury Tire Co., Ltd. Sailun North Americas (Sailun Group Co., Ltd) Shandong Haohua Tire Co., Ltd Shandong Jinyu Tire Co., Ltd Sumitomo Rubber Industries, Inc. The Goodyear Tire & Rubber Company Tianjin Wanda Tire Group Co., Ltd Toyo Tire Holdings of Americas Inc. Triangle Tyre Co., Ltd Yokohama Tire Corporation (+ Yokohama TWS North America, Inc.) ZC Rubber America Inc.

Information regarding supply chain is being submitted to DTSC by USTMA as confidential business information and is not included in this report.

### 3.1 Priority Product Made by Consortium Members Participating in This Alternatives Analysis Report

This Consortium comprises some but not all manufacturers of motor vehicle tires containing 6PPD. Products made by these responsible entities that fall within the scope of the priority product listing are shown in Appendix B. Requirements under 29 CFR § 1910.1200 to provide a Safety Data Sheet (SDS) do not apply to tire manufacturers for a new, finished tire, since a new tire is an article as defined in 29 CFR § 1910.1200(c). Therefore, to meet the requirement in 22 CCR § 69505.7(e)(4) to provide "any Material Safety Data Sheets and/or Safety Data Sheets related to the Priority Product," we are providing as an example a Safety Data sheet for Santoflex<sup>™</sup> 6PPD Pastilles prepared by Flexsys (Appendix C), which is publicly available on the Flexsys website. We are aware that 6PPD may be available in other forms, including liquid form, and that other 6PPD suppliers would have their own SDSs. This SDS is intended as an example to meet the regulatory requirement.

#### 3.1.1 Overview of Motor Vehicle Tire Composition and Manufacturing

Tires<sup>6</sup> are the only part of a vehicle that contacts the road, and that connection is vital in helping to keep motorists safe. Tires play an essential role in vehicle safety by transferring the driver's inputs from the vehicle to the road. Additionally, tires support the weight of the vehicle, facilitate steering for maintaining vehicle control, grip the road for acceleration and braking, and must perform in a variety of weather conditions. Tires are highly engineered products whose performance must meet applicable Federal Motor Vehicle Safety Standards, vehicle manufacturers' ride, handling, and traction criteria, rolling resistance requirements important in meeting fuel efficiency targets, and customer expectations for quality and performance.

The product the public knows as a tire is formed from various components (*e.g.*, sidewall, tread, inner liner). These components are in turn composed of different compounds, that is, mixtures of rubber polymers, also known as elastomers, and various additives. The tire components also include materials such as textile or steel cords. An important aspect of the tire is its use of vulcanized rubber. Vulcanization is a process in which heat is applied to the "green", or uncured, rubber compound causing a chemical reaction among sulfur, other chemicals, and polymers (elastomers) in the rubber compound. These reactions result in chemical bonds (cross links) between the polymer (elastomer) chains to produce cured tires.

The general structure of a passenger car tire, including some key components, is shown in Figure 3.1, and the typical structure of a radial medium commercial truck and bus tire is shown in Figure 3.2.

<sup>&</sup>lt;sup>6</sup> As used in this document, "tire" means a pneumatic radial tire used with motor vehicles (*e.g.*, passenger cars and light duty trucks, motorcycles, and heavy duty trucks and buses).

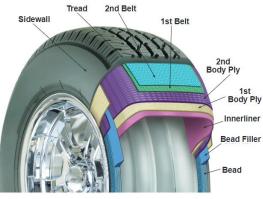


Figure 3.1 Typical Construction Features of a Figure 3 Pneumatic Radial Passenger Car Tire. Source: Radial MUSTMA. 2017a

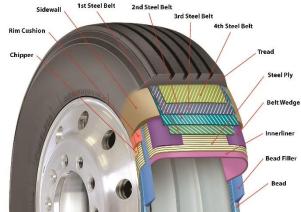


Figure 3.2 Typical Construction Features of a Pneumatic Radial Medium Commercial Truck/Bus Tire. Source: USTMA, 2017b

An explanation of some key tire components and the function they serve in a tire is given below.

- **Bead:** The tire bead is the portion (or component) of the tire that sits on the rim of the wheel. Tire beads are steel wire bundles that are coated with a specific rubber compound and secure the tire to the metal wheel.
- **Bead Filler:** A rubber compound placed above the bead that may be used between the body plies which wrap around the bead to enhance ride and handling characteristics.
- **Belts:** Typically, two belts with steel cords laid at opposing angles form a hoop under a tire's tread. Belts provide stability to the tread area of the tire, which minimizes wear and contributes to vehicle handling and traction. The steel belt is coated with a rubber compound that is called a belt coat or belt skim compound.
- **Body Plies:** Most car tires have one or two body plies, each typically comprised of textiles cords within a rubber layer. Truck and bus tires typically use steel cords for body plies. Body plies function as the base structure of the tire and provide the strength to contain the inflation pressure.
- Inner liner: A rubber compound used to retain the inflation pressure inside the tire.
- **Sidewall:** A rubber compound used to cover the body plies on the sides of the tire, which provides abrasion, scuff, and weathering resistance.
- **Tread:** Located on the road contacting portion of the tire, the tread rubber compound and tread pattern provide grip and abrasion resistance contributing to traction and treadwear.

All of these components have to be permanently bonded together in order for the tire to function properly and safely.

#### 3.1.1.1 Tire materials and tire compounding

As noted above, each of the components of a tire are composed of uniquely formulated rubber compounds and may include reinforcing materials such as steel and textiles. Compounding, the science of selecting and combining materials for a specific tire component, is complex. Categories of materials used in tire compounding include the following:

- **Natural Rubber:** Natural rubber provides specific performance characteristics to tires, such as tear and fatigue crack resistance. Some tires, especially truck and bus tires, use natural rubber in tread compounds to provide reduced rolling resistance (the resistance the tire encounters when rolling down the road, an important consideration for fuel efficiency). Natural rubber is a form of polyisoprene which is obtained by tapping rubber trees (*Hevea brasiliensis*).
- **Synthetic Polymers:** The two main synthetic rubber polymers, or elastomers, used in tire manufacturing are butadiene rubber and styrene butadiene rubber (SBR). These synthetic rubber polymers are used in combination with natural rubber. The physical and chemical properties of these rubber polymers determine the performance of each component in the tire as well as the overall tire performance. Another important synthetic rubber is halogenated polyisobutylene rubber, commonly known as halobutyl rubber, which is used in the inner liner. This material causes the inner liner to have reduced air permeability, which helps to keep the tire inflated.
- **Fillers:** Multiple grades of carbon black and coupled/uncoupled precipitated amorphous silica are used as fillers to reinforce the rubber and modify its properties resulting in improved wear performance and traction.
- Antidegradants: Antidegradants are added to rubber compounds to protect tires from overly rapid deterioration by ozone, oxygen, fatigue, and heat. Antidegradants include both antioxidants, which help to keep rubber from breaking down due to the effects of temperature and oxygen exposure, and antiozonants, which are used to impede the effects of exposure to ozone on the surface of the tire. Antidegradants in tires must serve in two load performance conditions, static and dynamic operations modes, which describe when the tire is at rest or flexing under motion, respectively.
- **Processing aids:** Bio-based oils, low aromatic petroleum oils, pine tar, and resins are the most common softening agents used in rubber compounding. Tackifying resins can be added to increase the rubber compound stickiness (tack) which helps the various tire components stick together during assembly of tire components.
- **Curing Systems:** Sulfur, chemical accelerators (often derivatives of benzothiazole), stearic acid, and zinc oxide are crucial ingredients for vulcanization, which transforms soft uncured rubber into a solid elastic article during tire curing. Curing systems not only enable vulcanization, but also shorten the vulcanization time and impact the length and number of crosslinks in the rubber matrix which in turn affects the rubber's properties.

Rubber compounds are uniquely formulated for the performance requirements needed for each tire component. For example, the rubber compound for the inner liner component of a tire is formulated to hold air inside the tire at the correct pressure when inflated; this requires specific polymers and ingredients that are unique to that purpose. The rubber compound for the tread component of a tire contacts the road, so it is formulated to meet performance expectations such as grip, wear, wet traction, snow traction, fuel efficiency, and other tire performance needs.

## 3.1.1.2 Tire manufacturing processes

The tire manufacturing process begins with the selection of polymers, fillers, oils, and other ingredients such as antidegradants, that will combine into a rubber compound to provide the exact characteristics wanted for the specific tire component. A machine called a Banbury<sup>®</sup> mixer combines the various raw materials for each compound into a homogenized batch of black material with the consistency of chewing

gum. The mixing process is computer-controlled to ensure batch-to-batch consistency. The compounded materials are then sent to machines for further processing into tire components such as sidewalls, belts, body plies, treads, or other parts of the tire.

The various tire components then come together in a machine where the tire is built from the innermost layer to the outermost layer. The uncured tire, often referred to as a "green tire" in the tire industry, is then placed inside a hot mold and inflated to press it against the mold, forming the tread pattern and the sidewall features. The tire is then vulcanized by heating it to more than 300 degrees Fahrenheit (150 degrees Celsius) for a pre-specified time which causes chemical reactions which transform the various tire components to form a finished tire.

For additional details around each specific component and their functions please see the free National Highway Traffic Safety Administration (NHTSA) resource "The Pneumatic Tire" (US DOT, 2006).

Tire manufacturing also involves compliance with various environmental and occupational safety regulations. For example, factories typically require air and water emissions permits and must comply with federal Occupational Safety and Health Administration (OSHA) regulations. For manufacturing facilities located in California, workplace warnings must be given if facilities use chemicals listed under California's Proposition 65.

## 3.1.2 Different Types of Tires

## 3.1.2.1 Passenger car and light truck tires

Passenger and light truck tires are the predominant tires in the US (see Table 3.1, below). Passenger and light truck tires can be categorized as OE (Original Equipment) which are supplied on a vehicle at its time of purchase, or replacement tires. OE tires must meet specific, often numerous and complex performance requirements specific to the vehicle manufacturer. OE tires are designed to a specific vehicle model year/make/model/trim level combination, and any changes to the materials used to manufacture OE tires, or the tire design itself, would require approval from the vehicle manufacturer. OE tires typically do not come with treadwear warranties.

Tires designed for the replacement market ("replacement tires") are designed to perform well on a wide range of vehicles – often as many as 30 different vehicle applications are appropriate for a single tire service description (tire size/speed rating/load index combination). Passenger and light truck replacement tires can be installed by a tire dealer or other tire service professional without original equipment manufacturer (OEM) approval. In the replacement market, consumers typically demand optimized treadwear and wet traction performance. In the replacement market, tire price also is a key consideration for consumers in many cases.

According to the DTSC's Product-Chemical Profile, "'motor vehicle tire' does not mean a motor vehicle on which tires have been installed" (DTSC, 2023). Tires installed on new vehicles are not part of the Priority Product definition. OE tires are considered replacement tires due to requirements in OE contracts for OE tires to be available as replacements, customer demand for OE tires in the replacement market, and to manage excess OE tire inventory. For purposes of this Stage 1 AA, OE tires are considered to be a subset of the replacement tire market and included in the analyses.

2023 U.S.	TIRE INDUS			MARY		
	Shi	pments				
	(in thous	ands of uni	its)			
	2022		2023		% Change 23/22	
	Total	Radial	Total	Radial	Total	Radial
Passenger						
Industry Original Equipment	41,616	38,817	45,657	42,723	9.7%	10.1%
Industry Replacement	213,730	213,184	219,180	218,637	2.5%	2.6%
USTMA Exports	13,962	13,932	14,462	14,412	3.6%	3.7%
Total Passenger	269,308	265,933	279,299	275,771	3.7%	3.7%
Light Truck						
Industry Original Equipment	6,260	6,250	5,856	5,846	-6.5%	-6.5%
Industry Replacement	37,241	37,082	34,253	34,162	-8.0%	-7.9%
USTMA Exports	4,349	4,349	3,842	3,842	-11.6%	-11.6%
Total Light Truck	47,849	47,680	43,952	43,851	-8.1%	-8.0%
Truck & Bus						
Industry Original Equipment	6,487	6,487	6,218	6,218	-4.1%	-4.1%
Industry Replacement	26,652	26,508	20,777	20,670	-22.0%	-22.0%
USTMA Exports	2,026	2,026	1,944	1,944	-4.0%	-4.0%
Total Truck & Bus	35,164	35,021	28,938	28,831	-17.7%	-17.7%

Notes:

Source: USTMA, 2024

Passenger and light truck replacement tires can also be divided into additional performance categories including all-season, summer, snow/winter tires, and a newer category of all-weather tires. All-season passenger and light truck tires are the most common tire type in the US and, as the name suggests, are general-purpose tires designed to perform in most climates. All-season passenger replacement tires typically come with a wear warranty from the tire manufacturer, typically in the 50,000- to 80,000-mile range. Antidegradants are a critical factor in allowing a tire to achieve these long mileage warranty periods.

Passenger and light truck tires have similar construction and utilize similar materials. However, light truck tires are designed to carry higher loads at higher inflation pressures, which requires the use of thicker rubber components, higher strength textiles and steel, and multiple body plies.

## **3.1.2.2** Truck and bus radial tires

Truck and bus radial tires differ in construction from passenger and light truck tires because the demands for truck and bus radial tire performance are more severe. Truck and bus radial tires contain steel cords as their body plies, instead of the textile plies found in passenger and light truck tires, and typically contain three or four steel belts rather than the two typically seen in passenger car and light truck tires. Depending on the application and type of service, truck and bus radial tires can last up to 150,000-300,000 miles on their original tread. Tires which are used for commercial purposes are designed to be retreaded, which is a

process to replace the tread on the tire casing. A truck and bus radial tire body (also known as a casing or carcass) may be retreaded up to three times and may last up to a total of 750,000 miles. Because the life cycle of truck and bus radial tires is much longer than that of a passenger or light truck tire, truck and bus radial tire rubber compounds typically contain higher levels of antiozonants/antioxidants (*i.e.*, 6PPD).

## 3.1.2.3 Motorcycle tires

Motorcycle tires differ in construction from both passenger and light truck tires and truck and bus radial tires due to the varying demands of the different types of motorcycles which are in use. A typical motorcycle tire for on-road application has a tread life ranging from 10,000-15,000 miles, depending on the motorcycle, driving style, and road surfaces encountered.

## 3.2 Chemical of Concern for the Priority Product

The chemical of concern for the Priority Products is N-(1,3-dimethylbutyl)-N'-phenyl-p- phenylenediamine (6PPD), Chemical Abstract Service Registry Number 793-24-8. 6PPD is within the class of PPDs and is the main antidegradant used in tires (Figure 3.3).

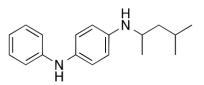


Figure 3.3 Chemical Structure of 6PPD

6PPD is used to protect tires from deterioration due to fatigue, thermo-oxidative breakdown, and from exposure to environmental degradation agents such as ozone and oxygen. These environmental degradation agents will attack the exposed tire surface and cause cracks and hardening of the rubber component throughout the tire's lifetime. In addition to the tire surface, interior portions of the tire (*i.e.*, the tire belt coat compound) can be attacked by oxygen diffusing from air inside the tire and penetrating through the grooves of the tire tread.

6PPD reacts with environmental degradants, ozone, and oxygen, faster than these degradants can react with the rubber or by quenching the reactive products of degradants and rubber, which protects the rubber products from degrading. As 6PPD reacts with the degradants, it is consumed, leaving less 6PPD in the tire. All tire compounds, except the inner liner and white rubber compounds (where used as sidewall decoration, lettering, or symbols), currently contain 6PPD as an essential antidegradant. It is important to note that 6PPD is currently used in <u>all</u> Consortium member passenger, light truck, truck and bus radial, and motorcycle tires. The Consortium is not aware of any motor vehicle tires available today that are 6PPD-free.

The adoption of the use of 6PPD in tires was a gradual process. Tire manufacturers began using 6PPD in tire manufacturing in the mid 1960's and early 1970's. In 1969, a British patent was published regarding the manufacturing of the 6PPD molecule (Davies and Neale, 1969). By 1975, 6PPD comprised 60% of the antiozonant used in tires (other, less effective PPDs were used previously)<sup>7</sup> (US EPA, 1975).

<sup>&</sup>lt;sup>7</sup> The first PPDs developed were active antiozonants but they were not as effective as 6PPD as they did not provide protection of rubber compounds for more than one and a half years. IPPD and DAPD were the first to be used in rubber compounds in the mid-

6PPD can transform into a number of reaction products when it carries out its intended function and reacts with ozone and oxygen. The reaction product of primary interest in this AA is 6PPD-quinone (6PPDQ, Figure 3.4), which was identified for the first time in December 2020 (Tian *et al.*, 2021). This same paper also suggested a link between this newly discovered substance and potential impacts to coho salmon (*Oncorhynchus kisutch*) attributed to roadway stormwater runoff. In their publication, Tian *et al.* (2021) state: "Exposures to ozone-synthesized and tire leachate–derived 6PPD-quinone (~20 ug/liter nominal concentrations) both induced rapid (<5 hours, with initial symptoms evident within 90 min) mortality...which matched the 2 to 6 hours mortality observed for positive controls." Similar findings in other laboratory studies were subsequently reported by other researchers (*e.g.*, Lo *et al.*, 2023). It should also be noted that significant declines in the coho salmon populations in California were observed since the 1940s, pre-dating the use of 6PPD in tires by several decades (California Dept. of Fish and Game, 2002).

Some other salmonid species may be similarly affected under similar circumstances, although with lower toxicological impacts than to coho, while other species apparently exhibit negligible toxicological susceptibility (see for example Brinkmann *et al.*, 2022; Hiki and Yamamoto, 2022; Greer *et al.*, 2023a). Multiple studies have shown that the closely related Chinook salmon are much less susceptible. For example, Lo *et al.* (2023) reported that "Juvenile coho were 3 orders of magnitude more sensitive to 6PPD-quinone compared with juvenile Chinook, with 24-h median lethal concentration (LC<sub>50</sub>) estimates of 41.0 and more than 67,307 ng/L, respectively."

The biological mechanisms by which the toxicity occurs, and why it affects some species and not others, is not yet known but is the subject of active research. For example, Tian et al. (2022) summarized the challenges of understanding the effects of 6PPDQ: "The mechanisms of toxicity of 6PPD-Q to coho salmon need to be characterized, while its toxicity, both acute and sublethal, to additional organisms needs to be broadly evaluated, including insights into mechanisms of species-specific sensitivities." Although scientific studies have continued since Tian et al.'s 2022 publication (e.g., Foldvik et al., 2022, 2024; Montgomery et al., 2023; Anderson-Bain et al., 2023; Ackerly et al., 2024), the mechanism of 6PPDQ's effect remains unclear. Coho salmon remain the most sensitive of the species examined. In their Product - Chemical Profile for Motor Vehicle Tires Containing 6-PPD, DTSC describes a number of possible mechanisms of action for 6-PPD causing URMS but notes that these require further study (DTSC, 2022). With respect to non-aquatic toxicity, DTSC also notes that "6PPD has been shown to be acutely toxic through oral and dermal routes of exposure in animal [e.g., rodent] studies, although at relatively high dosages" but that "[n]o studies are currently available to evaluate the human health effects of 6PPD-quinone exposure."

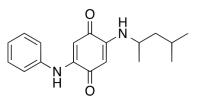


Figure 3.4 Chemical Structure of 6PPD Quinone

## 3.3 Function of the Chemical of Concern in the Priority Product

All tires contain antidegradants to prevent degradation of the rubber compounds caused by exposure to oxygen, ozone, fatigue, and elevated operating temperatures. Antioxidants and antiozonants are two classes

<sup>1960</sup>s. DAPD reacts minimally with ozone, and IPPD reacts too fast with ozone leading to premature depletion. The final PPDs to become commercialized were 6PPD, 7PPD, and 8PPD (Kuczkowski, J. A., 1990).

of antidegradant. There are in turn, two types of antioxidant and antiozonant performance that are important in tires: dynamic and static operation modes.

- Dynamic load performance: Antioxidants and antiozonants with dynamic operation modes protect the tire while it is in motion and being flexed;
- Static load performance: Antioxidants and antiozonants with static operation modes form a coating that protect the tire when it is in its resting and stationary state.

6PPD performs as an antioxidant and antiozonant in both dynamic and static operation modes. 6PPD reacts with ozone in the air to minimize the attack on the tire surface and reacts with the oxygen coming from the internal inflation pressure that degrades the belt rubber compound, thus preventing degradation of both the internal and external sides of the tire. Antidegradants are essential to ensure tire safety. Without the use of high-performing antidegradants like 6PPD, tire rubber compounds crack and degrade rapidly, creating potentially serious safety concerns (Figure 3.5).

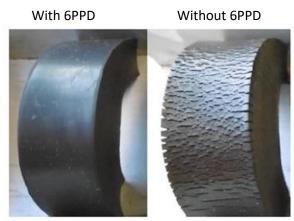


Figure 3.5 Dynamic Antiozonant Effect of 6PPD, 0.5 ppm Ozone Concentration, 40°C, 48 hours. Source: Schunk, A. 2022

The antiozonant and antioxidant properties of 6PPD are critical to creating durable tires. These properties are also important for tire longevity, leading to less demand on natural resources and energy for tire production as well as decreased tire waste. As an antidegradant reacts with degradants, its concentration in the tire is reduced; therefore, tires that are expected to last longer in the market require higher concentrations of these chemicals. 6PPD has the ability to migrate through the tire and reach the surface where it is needed to protect the tire from exposure to oxygen and ozone damage. Most importantly, the chemical migrates at the necessary rate such that the 6PPD contained in the tire can be present at the surface throughout the tire's intended lifetime. Any alternative to 6PPD would need to satisfy a similar surface availability  $\times$  time profile.

In summary, antidegradants such as 6PPD must provide the following functions:

- Protection against ozone
  - Readily reactive with ozone to prevent crack formation on the surface of the rubber, but not too reactive in order to prevent premature depletion
- Protection against oxygen

• Reactive with oxygen to prevent hardening of the rubber, loss of strength, improve tire wear, and maintain long-term durability, while not reacting so aggressively with oxygen as to cause premature depletion.

## • Protection against fatigue

• Reactive with the free radicals generated by breaks in rubber polymer during flexing. These free radicals can break the polymer chains and crosslinks in the rubber compound that would lead to a loss of strength for body plies, sidewall, and tread.

## • Optimal migration rate/ diffusion

- Adequate solubility and diffusivity in rubber compounds, also referred to as migration and mobility, which allows the chemical to move to the tire surface where it is needed to react with oxygen and ozone to ensure long term protection of the tire from oxygen and ozone damage over its life
- Available in the rubber formulation at an effective concentration over a tire's entire life cycle to ensure protection from ozone and oxygen damage

## • Be compatible with manufacturing processes

- No adverse effects on the rubber cure rate, tack, viscosity, etc.
- Resistance to temperatures encountered during the tire vulcanization process
- Be compatible with other aspects of tire safety and performance
  - The chemical cannot interfere with the function of other rubber compounding ingredients needed for safety (e.g., steel belt adhesion, cornering ability)

## 3.4 Key Performance Requirements for the Priority Product

The California Code of Regulations (CCR) §69505.5 states that an AA "shall identify the functional, performance and legal requirements of the Priority Product that must also be met by the alternatives under consideration." There are substantial requirements that tires must meet in order to be sold on the market. These requirements may be regulatory in nature (*i.e.*, safety), manufacturer-driven, or customer-driven, such as rolling resistance, wear, and other performance attributes. In addition, tires are a globally manufactured and distributed product and therefore must comply with a wide range of regulations in multiple countries.

## 3.4.1 Testing of Rubber Compounds Prior to Actual Tire Development

There are many steps that are needed to evaluate tire safety and performance. Before a tire is built and assessed for performance and safety, laboratory screening tests to evaluate the performance of various tire compounds must be completed with satisfactory results. These initial screening tests are essential to ensure that only viable compounds are used in development of tire products that then have to undergo more detailed and legally required testing.

All new rubber compounds using an alternative antidegradant must be tested and compared to a "control" or "witness" containing a standard material, in this case containing 6PPD, that has been produced at the same time and handled in the same manner. This is especially true for antidegradants, which are reactive chemistries and can be consumed during processing and aging. Each formula contains a variety of raw materials, all of which have some level of allowable range of variation. The results can also be influenced by the environmental conditions during the processing steps and during the testing; for these reasons, it is critical to have a control made at the same time.

Any alternative antidegradant would also need to be tested in multiple rubber compounds per tire component because each tire manufacturer uses different tire compounds in their products depending on the tire's intended use. The reader can find examples in the published literature (see for example, the Vanderbilt handbook [Sheridan, MF, 2010]). Accordingly, the use of different tire compounds among manufacturers influences the types of tests that they need to perform in order to demonstrate that an alternative is acceptable.

Based on the critical functions of an antidegradant, a potential list of screening tests by rubber formula type has been established by the Consortium members which includes existing modified ASTM and ISO standard laboratory methods that can be used for all the screening tests except migration. Multiple methods are mentioned in literature for assessing chemical migration (Lederer *et al.*, 1981).

Antidegradants must pass these screening tests (Table 3.2) as a first step in identifying a possible alternative.

	In		In Rubber Compound Testing		
Method	Key Parameter	Black Sidewall	Tread	Belt Coat	
ASTM D1646	Processability (viscosity and scorch)	0	0	0	
ASTM D5289-19A/ASTM D2084	Cure, Reversion	0	0	0	
ASTM D412-A, ASTM D573	Stress/Strain, Normal and Aged	0	0	0	
ASTM D624, ASTM D573	Die C Tear, Normal and Aged	0	0	0	
ASTM D2240	Durometer Hardness, Normal and Aged		0		
Lederer RCT	Migration	0	0	0	
ASTM D1149	Ozone: Static operation mode	0	0		
	Ozone: Dynamic operation mode	0			
ISO 1431-2012 (11.3)	Ozone: Intermittent Dynamic Exposure, Normal and Aged	0			
ASTM D4482-06	Fatigue to Failure	0			
ASTM D5992-96	Dynamic Properties/Viscoelastic	0	0	0	
LAT100/ASTM D5963-04/DIN 53 516	Wear		0		
ASTM D430/ASTM D813	Demattia Fatigue, Normal and Aged		0		
ASTM D2229	Wire Adhesion, Normal and Aged			0	
	Green aging			0	
	Heat aging			0	
	Oxygen aging			0	
	Steam aging			0	

Table 3.2 Potential Laboratory Screening Tests for Requirements by Rubber Compound	Table 3.2 Potential Laboratory Screening Tests for Requirem	nents by Rubber Compound	
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Notes:

ASTM = American Society for Testing and Materials; ISO = International Organization for Standardization.

## **3.4.2** Regulatory Requirements for Motor Vehicle Tires

Tires are highly regulated to ensure their safety, quality, and durability. Under the National Traffic and Motor Vehicle Safety Act of 1966 ("Vehicle Safety Act") (US Congress, 1966), automotive vehicles or motor vehicle equipment (including tires) are broadly regulated in terms of potential defects that could

impact motor vehicle safety. The Vehicle Safety Act created the National Highway Traffic Safety Administration ("NHTSA"), which promulgated the Federal Motor Vehicle Safety Standards (FMVSS) as directed by Congress. All passenger, truck and bus, trailer, and motorcycle tires sold in the United States (whether OE or replacement) must meet all applicable FMVSS (49 CFR Part 571). Additionally, passenger car tires sold in the United States must conform to the Uniform Tire Quality Grading Standards (UTQGS) (49 CFR § 575.104).

The Vehicle Safety Act has been amended several times since it was enacted in 1966. Most notably, the Transportation Recall Enhancement, Accountability, and Documentation ("TREAD") Act, enacted in 2000, added several new regulatory requirements for new motor vehicle tires (US Congress, 2000).

Tire manufacturers are required by law to self-certify to the appropriate FMVSS that every tire they manufacture meets safety, durability, and other requirements or regulations prior to sale to the consumer. NHTSA conducts periodic audits of new tires subject to FMVSS to assure compliance. The Safety Act explicitly preempts any state law or regulation that conflicts with a NHTSA regulation relating to "safety." The rationale is that vehicles travel from one state to another and between countries. The absence of a uniform set of safety rules would allow one state to impose arbitrary requirements that could significantly impact interstate commerce.

## 3.4.2.1 Passenger and light truck tires

Pursuant to the TREAD Act, NHTSA promulgated FMVSS No. 139 (49 CFR § 571.139) in 2003, which established testing requirements for new pneumatic radial tires for light vehicles. FMVSS No. 139 applies to all new pneumatic radial tires for use on motor vehicles made after 1975 with a gross vehicle weight of 10,000 pounds or less, which includes passenger and light truck tires with a tread depth of less than 18/32 of an inch. All tires are required to have dimensions within specific limits and specified markings that notify the consumer of the dimensions of the tire, the maximum load carrying capacity, the tire identification number (TIN), and that the tire is certified to meet the applicable FMVSS. FMVSS No. 139 also imposes requirements for tread wear indicators so consumers can be aware of the need for tire replacement.

FMVSS 139 requires tire manufacturers to meet the following new tire testing requirements:

- **High speed performance:** The high speed test is run on 1.70 m (67") diameter test drums<sup>8</sup>. Tires must complete the 160 km/hr (100 mi/hr) step with no visual evidence of separation in the tread, sidewall, ply, cord, inner liner, belt, or bead; chunking, open splices, cracking or broken cords and the tire pressure when measured within 15-25 minutes after the end of the test cannot be less than 95% of the initial inflation pressure.
- Endurance: The endurance test is run on 1.70 m (67") diameter test drums. All tires must complete the endurance portion, plus a 90-minute low inflation pressure step with no visual evidence of tread, sidewall, ply, cord, belt or bead separation, chunking, open splices, cracking, or broken cords, and the tire pressure, when measured within 15-25 minutes after the end of the test cannot be less than 95% of the initial inflation pressure.
- **Bead unseating resistance:** The test requires that tires retain air pressure and beads remain seated on the wheel in a test where an anvil is pressed against the tire sidewall. Wheel, tire inflation pressure, and anvil location are specified by rim diameter and tire type.

<sup>&</sup>lt;sup>8</sup> The test drum is a cylindrical structure meant to simulate the road surface. The tire is pressed against the drum and spun to simulate the effect of driving.

• **Tire strength (plunger energy):** The tire strength test requires that tires withstand a slow-moving plunger placed in the center area of the tread to a minimum level of calculated energy.

## **3.4.2.2** Uniform tire quality grading for passenger car tires

Also pursuant to Section 203 of the Vehicle Safety Act, NHTSA established the -UTQGS as a way to assist the consumer to compare various tires (49 CFR § 575.104). This regulation pertains to passenger car, SUV, and some light truck tires. Tire types excluded from UTQGS are LT-metric light truck tires, winter-type snow tires, low-volume production passenger car tires, motorcycle tires, and tires for truck and bus applications. While the UTQGS specify treadwear, traction, and temperature grades for tires within the scope of this regulation, the threshold values within these tests are informational only and are not directly linked to safety performance of motor vehicle tires.

# 3.4.2.3 Regulatory Testing Requirements for Truck and Bus Radial Tires and Motorcycle Tires

All new truck and bus radial tires (as well as some light truck tires) and motorcycle tires, are required to meet the following test requirements under FMVSS No. 119:

- **Endurance:** The test tire must complete the full endurance test with no visual evidence of tread, sidewall, ply, cord, belt or bead separation, chunking, open splices, cracking, or broken cords, and the tire pressure at the conclusion of the test cannot be less than the initial inflation pressure.
- **Tire Strength (plunger energy):** The tire strength test requires that tires withstand a slow-moving plunger placed in the center area of the tread to a minimum level of calculated energy.
- High Speed Performance (applicable only to motorcycle tires and non-speed restricted tires with rim diameter code of 14.5 or less and load range A, B, C, or D): The test tire must withstand testing at specified load over a series of increasing speeds for a set period of time. The test tire must complete the full test with no visual evidence of tread, sidewall, ply, cord, belt or bead separation, chunking, open splices, cracking, or broken cords, and the tire pressure at the conclusion of the test cannot be less than the initial inflation pressure.

## 3.4.3 Compliance with and Enforcement of NHTSA Regulations

The Vehicle Safety Act grants broad authority to NHTSA "reduce traffic accidents and deaths and injuries resulting from traffic accidents" by establishing FMVSS (US Congress, 1966). In addition, NHTSA has established regulations that address safety defects in motor vehicles and motor vehicle equipment (including tires) go beyond compliance with FMVSS.

The Vehicle Safety Act establishes a self-certification system for compliance with applicable FMVSS and UTQGS, where it is the responsibility of a manufacturer of vehicles and/or items of motor vehicle equipment, including tires, to certify that each of its regulated products is in full compliance with the performance requirements of all applicable FMVSS and consumer information regulations. This compliance burden is borne solely by the motor vehicle or equipment manufacturer.<sup>9</sup>

<sup>&</sup>lt;sup>9</sup> A manufacturer self-certifies a tire meets all applicable federal motor vehicle safety standards by molding "DOT" on the sidewall of the tire in association with the TIN, which identifies the manufacturing plant and date of manufacturer (week and year), among other information.

In addition to assuring that its tires meet all applicable FMVSS, a tire manufacturer designs and manufactures a tire to reduce the risks of a tire containing a safety defect. NHTSA regulations require manufacturers of vehicles and vehicle components to submit information to NHTSA about any FMVSS non-compliances or potential defects (49 CFR § 579).

When a tire manufacturer designs and manufactures a tire, it considers the risk of a potential tire recall. NHTSA maintains broad authority to enforce its regulations by imposing civil and criminal penalties (49 CFR § 578) and by instituting recalls of motor vehicles and motor vehicle equipment that do not meet FMVSS or contain a safety-related defect. Coupled with the NHTSA compliance surveillance program described in the previous section, the NHTSA recall authority creates a strong interest in compliance to all NHTSA requirements due to the potential damage to a company's reputation caused by a significant recall.

## 3.4.4 Additional Manufacturer Test Criteria for Highway Tires

All tire manufacturers perform a combination of voluntary outdoor vehicle tests, indoor drum tests, and technical tests in addition to all required regulatory tests. In addition to the tests outlined below, tire manufacturers may have their own proprietary test methods based on their specific tire designs and according to their market experiences and needs.

## 3.4.4.1 Indoor (drum tests) for passenger and light truck tires

Below is a list of typical indoor tests (drum tests) employed in tire development programs. Indoor laboratory tests are typically run on 1.70 m (67") diameter drums that have been an industry standard for decades. Tire manufacturers may have their own proprietary indoor drum test methods based on their specific tire designs and according to their market experiences and needs.

- **High speed performance:** Passenger tires are typically marked with a speed symbol following the maximum load rating. The tire speed symbol indicates the is the highest speed for which a tire is rated. The test method for marking a tire with a specific speed symbol is defined by UN Regulation No. 30. While most tires sold in the US contain speed symbol markings, it is an optional marking in the US from a regulatory perspective. However, most vehicles sold in the US today specify a tire fitment with a minimum tire speed symbol to meet or exceed the speed capability of the vehicle.
- Endurance: Tire endurance is a measurement of how long a tire can withstand severe conditions before displaying a condition that indicates the end of the test (damage to the tire). Endurance can be tested by varying the speed, load, inflation pressure, temperature, and/or number of cycles. The most typical tire endurance test varies the load. While FMVSS specifies an endurance test, tire manufacturers also conduct proprietary endurance tests in addition to the regulatory requirements. Some endurance tests are conducted on artificially aged tires, where the tire aging process is accelerated through the use of higher ambient temperatures and ozone and/or oxygen concentrations. These accelerated aging tests are intended to mimic the condition of tires which have been in service, including mileage and environmental exposure.
- **Rolling resistance:** The force necessary to keep a tire rolling is known as rolling resistance. To measure rolling resistance, a load is placed on the tire while it is being forced to turn by the drum and the resistance force which the tire generates to prevent it from turning is measured. In the US, this parameter first became important to vehicle manufacturers in the 1990s with implementation of more stringent Corporate Average Fuel Economy (CAFE) standards for new cars because lower tire rolling resistance equates to greater fuel economy (US DOT, 2006). For

regulatory purposes, rolling resistance is measured according to ISO 28580:2018 and is expressed in terms of rolling resistance coefficient (RRC). Rolling resistance is regulated by UN Regulation No. 117 and other governments globally. Currently, the California Energy Commission is developing a regulatory proposal to regulate rolling resistance in California pursuant to AB 844 (California State Assembly, 2003 Chapter 645).

## 3.4.4.2 Outdoor (vehicle) tests for passenger and light truck tires

In addition to indoor tests, manufacturers also test tires on actual vehicles to simulate actual (sometimes worst case) driving conditions. To some extent these tests are company specific; each tire manufacturer has proprietary formulas, manufacturing processes, and tire designs and their understanding of their products in the market and how they respond to the variety of environmental and use conditions is unique and must be assessed by each company individually. Below is a list of typical outdoor vehicle performance tests that passenger and light truck tires are subjected to.

- Wear rate: Traditionally with an outdoor test, sets of tires are driven at prescribed speeds on a known course to evaluate wear rate, usually measured in miles of travel per thousandth of an inch of tread depth loss (*i.e.*, miles per mil) or as tread loss per mileage increment (*i.e.*, mils/1000 mi).
- **Irregular wear:** During a wear test the tires are assessed for any indications of irregular wear. Uneven or abnormal wear features can significantly shorten the service life or mileage potential of tires.
- **Gravel chip/tear:** For passenger and light truck tires that are intended to be driven off road, an evaluation is conducted on vehicle on a gravel route to assess chipping and tearing of tread elements.
- **Handling dry, wet, and snow:** Handling is a result of tire/vehicle interactions in response to various driver inputs. Handling evaluations include various road conditions such as dry, wet, and snow but also everyday driving and emergency steering situations.
- **Ride comfort:** A vehicle's perceived ride comfort, whether "sporty" or "plush," can be significantly influenced by tires. The ride comfort is assessed over a variety of road conditions and can include assessment of impacts like potholes and train tracks.
- **Noise:** Tire noise can be generated from the interaction of the tire with the road. Pass-by noise is measured from the sides of the road with a vehicle traveling at a specified speed with the engine not running.
- **Endurance:** Outdoor testing for tire endurance usually involves driving a vehicle on a closed course at a specified level of loads, inflation pressures, and speeds (US DOT, 2006 224-2581).
- **Field Testing:** Tire manufacturers may also conduct field testing to obtain performance data for tires operated under real-life conditions for an extended period of time. This testing is typically performed by a contracted fleet with routine monitoring by the tire manufacturer.
- **Traction:** Dry, and wet: Specially equipped instrumented trailers with computer-controlled braking capability are towed over known skid pad surfaces. Brakes are applied gradually to cause wheel lock-up and peak and slide friction forces are recorded.
- Snow traction: Snow testing is conducted as specified by ASTM F1805. Per FMVSS No. 139 (49 CFR § 571.139), to be marked with the Alpine (snow) symbol, a tire must achieve a traction index of 112 or greater as compared to the ASTM F2493 standard reference test tire.

## **3.4.4.3** Optional technical tests for passenger and light truck tires

In addition to the typical indoor drum and outdoor performance tests, manufacturers also often test finished tires for additional technical properties using specialized equipment, depending on customer requirements and product performance specifications. Some of these tests include the following:

- Weight: OE vehicle manufacturers often specify tire weight targets as part of their requirements for meeting CAFE goals since the weight of the tire is directly related to its rolling resistance and fuel efficiency.
- Force and moment properties: A tire's cornering capability comes from the forces generated when a tire's direction of motion is different from its heading direction, causing a slip angle. This test is an indication of how the tire will perform on vehicle handling assessments.
- **Electrical resistivity:** Moving vehicles can generate static electricity which is exacerbated by low temperature and humidity.
- Uniformity: Due to material and assembly variations that occur during manufacturing and curing, small deviations in tire cross section circumferentially can result in measurable spring rate or dimensional changes, for example, an out-of-round or out of balance condition. This can have an impact on handling, noise, and ride comfort.
- Air permeation: Inner liner rubber compounds are formulated to minimize permeation of air through the tire carcass. The air permeation test measures the air loss over a specified time and conditions.
- **Dynamic ozone:** An indoor drum method where and the tire is exposed to ozone while it is running on a standard drum at a specified speed, temperature, and ozone concentration. This test can be used to correlate sidewall compound ozone cracking between a variety of market conditions and judge differences in performance of sidewall rubber compounds.
- Aged tire properties: ASTM F2838 is a method for inflating tires with a specified oxygen content, pressure, and temperature and placing them in an oven for a specific amount of time to simulate market conditions for the belt coat compound. Once a tire is aged, the belt coat compound can be cut out of the tire and tested to compare the physical properties of the compounds after aging.

## 3.4.4.4 Additional Manufacturer Test Criteria for Truck and Bus Radial Tires

Tire manufacturers may have their own proprietary test methods based on their specific tire designs according to their market experiences and needs.

- **High speed performance:** Not all truck and bus tires are marked with a speed rating. FMVSS 119 specifies that tires restricted to use at speeds of 55 mph (90 km/hr) or less must be marked to indicate this limitation. For tires which are marked, the tire speed rating is the highest speed that a tire can handle before it does not perform as designed. The test method for marking a tire with a specific speed symbol is defined by UN Regulation No. 54. While many tires sold in the US contain speed category markings, it is an optional marking in the US from a regulatory perspective.
- **Endurance:** Tire endurance is a measurement of how long a tire can withstand severe conditions before reaching its limit. Endurance can be tested by varying the speed, load, inflation pressure,

temperature, and/or number of cycles. The most typical tire endurance test varies the load. While FMVSS specifies an endurance test, tire manufacturers also conduct proprietary endurance tests in addition to the regulatory requirements.

• **Field Testing:** Tire manufacturers may also conduct field testing to obtain performance data for truck and bus tires operated under real-life conditions for an extended period of time. This testing is typically performed by a contracted fleet with routine monitoring by the tire manufacturer.

In addition to the above, truck and bus tires are tested for many of the parameters discussed in Sections 3.4.4.2 and 3.4.4.3.

## 3.4.5 SmartWay Certification for Truck and Bus Radial Tires

SmartWay<sup>®</sup> Certification is a collaborative effort between the United States Environmental Protection Agency (US EPA) and the freight industry designed to help businesses move goods in the cleanest, most energy-efficiency ways possible while reducing greenhouse gases and air pollution, and protecting public health. SmartWay<sup>®</sup> is a voluntary program outside the state of California; however, within California, the California Air Resources Board (CARB) requires fleets to operate on SmartWay<sup>®</sup> verified tires. The SmartWay<sup>®</sup> program publishes a list of new and retreaded commercial vehicle tires which have been verified to demonstrate a rolling resistance coefficient at or below certain targets set by EPA (US EPA, 2012, 2022).

## 3.4.6 Summary of Performance Testing Requirements

The paragraphs above outline the extensive set of test requirements that are required for a new tire formulation or design before it can be placed in the market. Starting with laboratory tests as shown in Table 5.14, the successful formulation then passes through several additional stages of testing before finally being tested as manufactured tires on an actual vehicle. Based on the experience of Consortium members, the tire research and development, design, and performance testing process for a tire using existing, commercially-produced materials known to perform as necessary in tires, can take a minimum of 4 to 6.5 years. In the tire design process, each step may be repeated multiple times until an acceptable design is achieved, which can significantly extend the design process. Furthermore, challenges encountered while conducting a step in the tire design process may require development to go back to an earlier stage. In the case of replacing 6PPD, once a new candidate antidegradant is identified, an additional 4 years (minimum) of limited-scale field testing would be required to ensure performance as a tire ages. After satisfactory results are obtained from field testing, additional time will be needed for deployment of the new antidegradant in tires for the market, which could take months to a few years.

## 3.4.7 Other Regulatory Requirements for the Priority Product

In addition to performance requirements, other regulatory requirements could impact the feasibility or timeline for implementation of any alternative to 6PPD in motor vehicle tires. For example, chemicals that are used in tire manufacturing need to be registered in the various jurisdiction where tires are manufactured (*e.g.*, the US, the European Union, China, South Korea). Whether or not a possible alternative is already listed on the various chemical inventories could be a significant factor in terms of the timeline for implementing an alternative. In addition, given the volumes of antidegradant that will be required for global tire production, even chemicals already present on chemical inventories may be shifted to higher production volume categories, which could trigger additional data submission requirements in terms of chemical,

environmental and toxicological properties. In particular, the need for additional toxicity testing, particularly for carcinogenicity or reproductive toxicity, could add significantly to the cost and timing of implementing any alternative (*e.g.*, a standard carcinogenicity study can take up to four years to complete). Another potential regulatory requirement is California's Proposition 65, which will become significant if any possible alternative is listed by the State of California as known to cause cancer or reproductive harm. Some of the possible alternatives also have use restrictions in the US. For example, some carbon nanotubes and graphene have Significant New Use Restrictions under the Toxic Substance Control Act (TSCA) (US EPA, 2023a), which means additional regulatory obligations would need to be considered if either were to be used in tires. The State of Washington also has announced "6PPD Alternatives Assessment Hazard Criteria," which differ from those required by the SCP regulations. The SCP regulations do not require consideration of criteria established by other states in searching for an alternative. From a practical implementation standpoint, the Washington criteria could be considered, among other factors, in the Stage 2 AA when selecting final recommended alternatives.<sup>10</sup>

## 3.5 Necessity of the Function of the Chemical of Concern in the Priority Product

Under CCR §69505.5 3(A), responsible entities, in this case, Consortium members, are required to determine whether the chemical of concern or an equivalent replacement is necessary in order to meet the product's functional and legal requirements or whether the chemical of concern can simply be eliminated from the product without replacement. As discussed in prior sections, all tires require antidegradants, including antioxidants and antiozonants, for safe performance. Thus, simply removing the chemical of concern, 6PPD, from motor vehicle tires without replacement is not an option.

 $<sup>^{10}</sup>$  We also note that the Washington hazard criteria are essentially guidelines for use in the evaluation of alternatives and not standards for compliance (*e.g.*, like US EPA water quality criteria).

## 4 Scoping, Identification of Possible Alternatives and Relevant Factors

## 4.1 Purpose and Approach for this Stage 1 AA

As conceived by Gradient and the Consortium, the goal of a preliminary (Stage 1) AA under the SCP program is to answer the following question: Do potentially safer, functionally acceptable, and technically feasible alternatives to the Priority Product exist that should be given a more in-depth consideration to determine if they qualify as acceptable alternatives? If the answer to this question is yes, then a 2<sup>nd</sup> Stage AA is appropriate, where factors such as economic feasibility are considered. If the answer is no, then an Abridged AA is required. The aim of Stage 1 is not to definitively identify a final alternative but rather to identify apparently acceptable candidate alternatives for further study. We believe this philosophy is in accord with the SCP regulations (CalDTSC, 2013a).

## 4.2 Alternatives Under the SCP Regulation

The SCP AA process requires responsible entities to identify and consider alternatives to 6PPD that meet the definition of "alternative" under section 69501.1 and potentially meet the Priority Product's requirements as outlined in section 3 of this AA (CalDTSC, 2013b SCPR section 69505.5(b)). To create the list of candidate alternatives, the Safer Consumer Products regulation requires responsible entities to "evaluate available information that identifies existing possibly viable alternatives for consideration in the AA" (CalDTSC, 2013b SCPR section 69505.5(b)).

An alternative may include any of the following:

- Removal of a Chemical of Concern from a Priority Product, with or without the use of one or more replacement chemicals.
- Reformulation or redesign of a Priority Product and/or manufacturing process to eliminate or reduce the concentration of a Chemical of Concern in the Priority Product.
- Redesign of a Priority Product and/or manufacturing process to reduce or restrict potential exposures to a Chemical of Concern in the Priority Product.
- Any other change to a Priority Product or a manufacturing process that reduces the potential adverse impacts or potential exposures associated with the Chemical of Concern in the Priority Product, or the potential adverse waste and end-of-life effects associated with the Priority Product that also meets the Priority Products function.

## 4.3 Inclusion of Performance as a Consideration in the Stage 1 AA

The SCP regulations do not list performance as a required consideration (*i.e.*, relevant factors) for a Stage 1 AA but rather include it as a required consideration for Stage 2. However, the SCP regulations (§ 69505.5 (e)) permit for the Consortium to include additional factors that they deem relevant to the AA at their

discretion (CalDTSC, 2013a). The Consortium maintains that performance is a critical element of an AA at the initial stage because an alternative that has unacceptable performance (*i.e.*, performs poorly compared to the current product by a reasonable metric related to consumer safety or expectations) may impact tire safety, longevity, or health or environmental impacts, and will not be a viable alternative and should therefore not be considered further. Thus, performance is included as a relevant factor in this Stage 1 AA report.

## 4.4 Scoping: Alternatives Outside the Scope of This AA Report

The first element of an AA involves scoping, or determining the range of alternatives that will and will not be considered in the AA. Certain alternatives are being excluded from further analysis because they are too preliminary in their stage of development to meet the likely implementation schedule under the SCP regulation. These alternatives also have very limited data with which they can be evaluated and would require a revision of federal safety regulations which is something beyond the scope of the SCP program. Some alternatives also likely pose the same issues as current tires (*i.e.*, by using rubber tread) and so would not constitute an alternative with a different hazard profile. Three possible technologies are not being considered as part of this AA: 1) non-pneumatic tires, including Shape Memory Alloy Radial Technology (SMART) tires; 2) electrostatic particle collectors; and 3) modified EPDM or halobutyl elastomers.

## 4.4.1 Non-Pneumatic Tires

As mentioned in the DTSC Technical Report, some companies are working towards non-pneumatic, or airless motor vehicle tires. These transmit the vertical load and tractive forces from the roadway to the vehicle and generate the tractive forces that provide the directional control of the vehicle without the containment of any gas or fluid for providing these functions. Examples of prototype non-pneumatic tires are shown below (Figure 4.1).



**Figure 4.1 Examples of Non Pneumatic Tire Products**. Sources (left to right): Michelin, 2023; Bridgestone Corp., 2013; Goodyear, 2023.

As of early 2024, there are no commercially available non-pneumatic motor vehicle tires being sold in the United States, and the FMVSS currently require all new motor vehicles to be equipped with pneumatic tires. Market introduction of a non-pneumatic tire is expected within the next several years, but the FMVSS and multiple state laws would need to be amended before this can occur. Widespread adoption of non-pneumatic tires over a broad range of tire sizes, load capacities, and speed ratings is anticipated to be at least several additional years into the future. Non-pneumatic tires are incompatible with many of the current industrial processes utilized by the tire industry. Significant investment, potentially including new production facilities, will likely be required to mass-produce these types of tires.

Non-pneumatic tires for off-road applications such as agricultural equipment, construction equipment, and utility terrain vehicles (UTVs) are commercially available; however, tires for these applications are designed for significantly different operating conditions than motor vehicle tires (*e.g.*, lower speeds) and are not subject to FMVSS, as NHTSA's authority only extends to vehicles and vehicle components for on-highway use.

Although non-pneumatic tires rely on a different mechanism to support a vehicle load *versus* conventional motor vehicle tires, many of the components in non-pneumatic tires, including the tread, are composed of rubber compounds that contain 6PPD to protect the compounds from environmental degradation and fatigue. Accordingly, 6PPD would still be present in non-pneumatic tires. Also, due to the tread compounds used, treadwear rates of any future non-pneumatic tires may not be significantly different than conventional pneumatic tires of the same size, load capacity, and speed rating. While the total amount of 6PPD in such tires may be reduced (due to the lack of a sidewall), 6PPD would still be present in the portion of the product contacting the roadway. It is also unclear whether the concentration of 6PPD in the tread rubber would have to be different (*e.g.*, higher) in these tires to compensate for the reduced reservoir created by a lack of sidewall.

The DTSC report mentions nickel-titanium spring tires from the SMART Tire company as a possible alternative. The SMART tire is another concept for a non-pneumatic tire. At this time, the tire is not in production (the company website suggests it would be available for bicycle tires in 2023 but no other information is given) and the manufacturer indicates they are looking for investors. The company website indicates that the product for vehicle applications will contain a rubber tread compound, so it is not known if it will contain 6PPD. As with the other non-pneumatic tires discussed above, the concentration of 6PPD in the tread is not known.

Due to the unknown timing of widespread availability for motor vehicle use and their current materials of construction, non-pneumatic, or airless tires, are deemed to be outside the scope for this report.

## 4.4.2 Electrostatic Particle Collectors

Tyre Collective is a company that is working on an electrostatic collection device for TRWP (The Tyre Collective Ltd., 2024). This device is placed behind the tire on the vehicle and is not an innovation that can be applied to a tire. This technology is in the proof-of-concept phase and the most recent trial showed a 20% collection efficiency by mass. A 20% reduction in TRWP release may not be sufficient to meet the SCP requirement of a "material difference". It could also impose an additional burden on consumers if the collector has to be replaced or emptied periodically in a repair shop to function. The tire industry also has no authority to require additional products beyond tires to be installed on new vehicles or retrofitted to existing vehicles. This would be the choice and responsibility of vehicle manufacturers, who are not subject to the AA requirement. It is unknown if such a device would impact aerodynamics, fuel efficiency or vehicle clearance. Therefore, electrostatic particle collectors are also deemed to be outside the scope of this report.

## 4.4.3 Modified EPDM or halobutyl rubber to reduce 6PPD concentrations in sidewall

Consortium members also identified several approaches for reducing the concentration of 6PPD in tires, which constitute a potentially viable approach under the SCP regulations. The alternatives involve using (1) a modified EPDM to formulate the sidewall rubber (two approaches) or (2) bromobutyl rubber to formulate the tread. Both EPDM and bromobutyl rubber are inherently resistant to ozone attack. While these materials normally cannot be used in tire sidewall and tread due to chemical incompatibility (*e.g.*, bromobutyl rubber is used exclusively in the inner liner), the Consortium did discover patents which

describe ways to use these elastomers (with modifications) to formulate sidewall (EPDM) and tread (bromobutyl) rubber.

There are two reported EPDM applications. The first involves EPDM modified with Nchlorothiosulfonamides (Hopper, 1976). In this case, the structure of the EPDM has been modified to put a chlorine atom along the backbone of the EPDM, which is subject to thermal degradation to produce hydrochloric acid (HCl). It is well known that chlorinated polymers can undergo thermal degradation to produce HCl (Krongauz et al., 2011). Tire manufacturers avoid introducing chlorinated materials into the production process because HCl is known to corrode production equipment and would also likely require changes to air emissions permits if chlorinated compounds are being generated. Further, this application is only relevant for replacing 6PPD in sidewalls and thus would not affect the composition of tread material, which is the source of TRWP. The extent to which the tire sidewall contributes to the potential migration of 6PPD from tires into the environment (via blooming and washing off rather than from TRWP) is unknown and unquantified. Loss of 6PPD/6PPDQ from tread will occur predominantly while the vehicle is moving (by generation of TRWP), whereas sidewall loss of 6PPD/6PPDQ (by being washed off the sidewall) would occur predominantly during the parked stage of vehicle use (since, at least for passenger vehicles, the vast majority of time is spent parked). Since 6PPD will still be used in other parts of the tire, the potential benefit of this alternative is unclear. The Consortium is aware of one patent that considered the potential use of modified EPDM in tread (Sandstrom, 1992), but this still required the use of antidegradants such as 6PPD.

The second EPDM approach for sidewalls involves blends of conventional EPDM and natural rubber in an attempt to make ozone resistant sidewalls (Tinker and Jones, 1998). A special mixing protocol was required and mixing conditions were specified to maximize the probability of success. Despite these enhancements, the tires experienced structural failures during product testing. In addition, the mixing protocol used by the authors could not be effectively used in a tire plant because of "rework" issues. "Rework" is a recycling process where scrap material from processing operations is put back into fresh mixes of rubber. Because this study required the use of virgin ingredients, rework was not possible (to determine the effect of rework on a compound, a great deal of mixing is required so that the final levels of added material becomes stabilized). The potential for rework is critical because it involves a substantial amount of compound, and elimination of the use of rework would generate substantial production waste, require higher volumes of raw ingredients, with associated lifecycle impacts, and make tires uneconomical to consumers. In addition, as noted above, this approach is only relevant to sidewalls which would not address concerns with 6PPD in TRWP.

The bromobutyl rubber patent describes a formulation requiring an organosilane cross linking agent and "effective amounts of processing aids" whose nature is not stated. Like EPDM, the use of bromobutyl rubber would also significantly impact the possibility of rework, potentially creating more waste material. For use of bromobutyl rubber in tread, although the ozone resistance of the bromobutyl rubber is mentioned, antioxidants and antiozonants are still included as ingredients in the patents so the degree of 6PPD reduction is not clear. In addition, if other rubber (requiring 6PPD) is used for the sidewall and other tire components, the 6PPD in the rest of the tire could still migrate to the tread and be put into the environment *via* TRWP. Finally, it is commonly known that butyl elastomers have very high hysteresis relative to natural rubber, polybutadiene and SBR elastomers. This high hysteresis would result in higher operating temperatures as well as higher rolling resistance, which could result in failure to meet tire endurance performance requirements specified in US FMVSS 119 and 139. In addition, higher hysteresis results in lower fuel mileage for internal combustion engines and reduced range for electric vehicles..

Thus, because the potential benefit appears questionable and there are many unknowns regarding the feasibility of implementation, these options are not evaluated further in the AA.

## 4.4.4 Waxes and Coatings

As mentioned in the DTSC Product-Chemical Profile (CalDTSC, 2022), waxes and coatings could provide a physical barrier to prevent ozone attack. However, this approach only works under static conditions (*i.e.*, when tires are not in motion) and "face significant challenges under dynamic conditions – that is, during a tire's intended use on a motor vehicle" (CalDTSC, 2022).

Blends of microcrystalline and paraffin waxes are currently used as processing aids in production of tire rubber to provide lubrication, mold release, and improve ozone resistance (Akrochem, 2002). However, waxes are always used in combination with 6PPD in tires. When used alone, waxes can only protect tires from ozone under static conditions. While cars are in motion (*i.e.*, dynamic condition), the wax film breaks, exposes the underlying rubber to ozone attack, and causes earlier failure (Akrochem, 2002). Similarly, coatings wear off while cars are in motion and prematurely expose the underlying rubber to ozone attack. DTSC's Product-Chemical Profile (CalDTSC, 2022) noted that waxes and coatings are "likely infeasible means of protecting the tire tread." Because sidewalls also flex during movement and can experience scuffing and other environmental damage, coatings and waxes are also not a good solution for protecting non-tread parts of the tire from ozone.

## 4.5 Possible Alternatives to 6PPD in Motor Vehicle Tires

## 4.5.1 Approach for Identification of Alternatives

Once the scope of the AA has been identified, the next critical step is to gather information on possible alternatives. To conduct an informative AA, one needs to consider not only those products made by the Consortium members but also other products that are available as these may be possible alternatives. As required under the SCP regulation, the Consortium must consider any candidate alternatives that are posted on the Department's website (CalDTSC, 2024). DTSC indicates that additional information sources which should be considered are journals, articles, books, references, handbooks, encyclopedias, patents, and internal company files. To identify possible alternatives, the Consortium conducted literature searches, surveyed members for literature sources, and surveyed expert industry consultants for additional literature sources. The following literature sources were evaluated by the Consortium to identify candidate alternatives:

- Technical journals, including, but not limited to
  - Rubber Division of the American Chemical Society Rubber Chemistry and Technology
  - Rubber World
- Trade media, including, but not limited to
  - Tire Business
  - o Rubber and Plastics News
  - Tire Review
  - International Tire Technology Magazine
- Reference literature and books, including, but not limited to
  - "Ozonation of Organic and Polymer Compounds" by Gennady E. Zaikov and Slavcho K. Rakovsky
  - "Ozone Risk Communication and Management" by Edward Calabrese, Charles Gilbert, and Barbara Beck
  - Vanderbilt Rubber Handbook, 14<sup>th</sup> edition

- "Blends of Natural Rubber: Novel Techniques for blending with Specialty Polymers," edited by A. J. Tinker and K. P. Jones, Chapman & Hall, London, 1998.
- Rubber Compounding Chemistry and Applications, 2nd Ed., Brendon Rogers, CRC Press, 2016 p. 419-459
- Online reference material, including, but not limited to
  - PubChem
  - Chemical supplier websites
- Government materials, including, but not limited to
  - Chemical Profile for Motor Vehicle Tires Containing N-(1,3-Dimethylbutyl)-N'-phenylp-phenylenediamine (6PPD) (CalDTSC, 2022)
  - Washington Department of Ecology Technical Memo (WA Ecology, 2021)
- Patent searches, including, but not limited to
  - United States Patent and Trademark Office (USPTO)
  - Google patents
  - Consultation with university and government researchers, including but not limited to
    - o University of California, Berkley
    - University of Washington
    - Washington State University
    - o University of Massachusetts-Lowell
    - US Geological Survey (USGS)

In addition, USTMA prepared a survey concerning knowledge of research on alternatives that was sent to all Consortium members. The survey asked questions about alternatives currently under investigation as well as alternative antidegradants that had been evaluated in the past and found unsatisfactory. It also asked about company awareness of research or ideas being put forth by other entities. Responses to the survey were aggregated *via* a third party law firm to ensure confidentiality. A copy of the survey is included as Appendix D.

Over 70 possible alternative antiozonants were identified through this process (see Appendix F). The identified chemicals were screened for compound effectiveness against ozone. Once the information was collected, the chemicals were tabulated along with their related scientific information from the literature review in a spreadsheet and scored as to their perceived feasibility to function as an antiozonant in tires. The following scoring system was used:

#### **Compound Effectiveness Against Ozone:**

- 1 = There is existing ozone data that indicate the chemical does not work in tire rubber
- 2 = Feasibility data are lacking but the chemical structure indicates is it unlikely to work in tires as an antiozonant
- 3 = Feasibility data are lacking but the chemical structure is promising as an antiozonant
- 4 = There are some positive ozone data indicating effective performance in tires but data are limited

Based on this scoring approach, chemicals with compound effectiveness against ozone scores of 3 or 4 in Appendix F were evaluated further in this preliminary (Stage 1) AA. However, all possible alternatives suggested by DTSC in the Priority Product profile (CalDTSC, 2022) and all possible alternatives listed on DTSC's website (CalDTSC, 2024), including but not limited to those possible alternatives suggested by the Washington State Department of Ecology (WA Ecology, 2021) were retained for this Preliminary AA, irrespective of their scores for effectiveness against ozone. Although the search was broad enough to include combinations of antidegradants that could be used to replace 6PPD, relatively few antidegradants mixtures were identified. This is likely because investigating binary or greater mixtures of antidegradants

introduces a high degree of complexity into testing (*i.e.*, requiring tests of multiple concentration combinations).

Overall, a total of 43<sup>11</sup> possible alternatives were evaluated further in this Stage 1 AA. A number of the possible alternatives suggested by DTSC or references cited by DTSC (*e.g.*, modified EPDM, non-pneumatic tires, coating, wax) are discussed in more detail in Section 4.4, but not evaluated further for hazard, exposure, and performance in this Stage 1 AA.

## 4.5.2 Possible Alternatives Identified

## 4.5.2.1 Possible Alternative PPDs

Based on the review process outlined above, the Consortium identified 19 potential 6PPD analogs that could potentially serve as substitutes for 6PPD. Note that these are not demonstrated to be actual alternatives to 6PPD but rather are chemicals that merit evaluation as shown in the AA to see if there is sufficient information indicating they could be possible alternatives to 6PPD. The order of the chemicals listed in Table 4.1 below does not indicate a relative level of priority.

Chemical Name	Acronym	CAS
N-(1,4-Dimethylpentyl)-N'-phenyl-p-phenylenediamine	7PPD	3081-01-4
N-Isopropyl-N'-phenyl-p-phenylenediamine	IPPD	101-72-4
N-Cyclohexyl-N'-phenyl-p-phenylenediamine	CPPD	101-87-1
N,N'-Diphenyl-p-phenylenediamine	DPPD	74-31-7
N,N'-Bis(1,4-dimethylpentyl)-p-phenylenediamine	77PD	3081-14-9
4,4'-Dioctyldiphenylamine	DOPD	101-67-7
N,N'-Ditolyl-p-phenylenediamine	Commercial DTPD	68953-84-4
N,N'-Dicyclohexyl-p-phenylenediamine	CCPD	4175-38-6
Diaryl-p-phenylene diamine	DAPD	68953-84-4
N,N'-Di-2-naphthyl-p-phenylenediamine	DNPDA	93-46-9
N' -Phenyl-N-Fluorenyl-Para-Phenylenediamine	NA	No CAS
N-(p-Phenylthiomethylphenyl)-N'-(1,3-dimethylbutyl)-p-phenylenediamine	NA	No CAS
4-(2,5-Dimethyl-1H-pyrrol-1-yl)-N-phenylaniline	NA	No CAS
N,N - (Ethane-1,2-diyl ) bis (N-phenylbenzene-1 4-diamine or similar chemical 1-N-[2-(4-anilinoanilino)ethyl]-4-N-phenylbenzene-1,4-diamine	NA	No CAS
4-N-(2,3-Dimethylphenyl)-1-N-phenylbenzene-1,4-diamine- R1 and R2 are methyl	NA	No CAS
RU997, Irgazone 997 Reaction product of N-phenyl-N'-(1,3-dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether	NA	No CAS
4-[4-(4-Methylpentan-2-ylamino)anilino]phenol	NA	No CAS
Representative example from class (4-((4- (dimethylamino)phenyl)amino)phenol )	NA	6358-22-1
N-1-Methylheptyl-N'-phenyl-p-phenylenediamine	8PPD or UOP 688	15233-47-3

Table 4.1 Possible PPD Derived Alternatives Meriting Further Study in Stage 1 AA

Notes:

CAS = Chemical Abstracts Service Number; NA = Not Available; PPD = Paraphenylene Diamine.

<sup>&</sup>lt;sup>11</sup> While waxes and coatings were carried from Appendix F into the Stage 1 AA because they were suggested by DTSC, they were not evaluated further for hazards, exposure, and performance because they are non-viable alternatives when tires are in motion. Thus, they did not contribute to the count of possible alternatives evaluated in this Stage 1 AA. See Section 4.4 for more discussions on wax and coating.

Among these are PPDs that have been discussed in various documents as possible 6PPD replacements (*e.g.*, 7PPD, IPPD, 77PD, CPPD) and which are used to some extent commercially. The list also contains less well-known analogs, many of which lack CAS registry numbers. Lacking CAS numbers makes identifying toxicological or chemical data for these analogs challenging because all can be described by various chemicals names.

#### 4.5.2.2 Possible non-PPD chemical alternatives

The Consortium's search process also identified 24 non-PPD possible alternatives as shown below (Table 4.2). Again, these are not demonstrated to be alternatives to 6PPD but rather are chemicals that merit evaluation in the AA to see if there is sufficient information indicating they could be potential alternatives to 6PPD. The order of the chemicals listed below does not indicate a relative level of priority.

Chemical Name	Acronym	CAS
N-1,3-Dimethylbutyl-N'-phenyl quinone diimine	6QDI	52870-46-9
Polymerized 2,2,4-trimethyl-1,2-dihydroquinoline	TMQ Oligomer	26780-96-1
Nickel dibutyldithiocarbamate	NBC	13927-77-0
Ethoxyquin	NA	91-53-2
Dilauryl thiodipropionate	NA	123-28-4
N,N-Diethyl-2,2,4-trimethyl-1H-quinolin-6-amine (R= N(C2H5)2	NA	No CAS
Mixed xylene diamines N,N'-Dibenzyl-p-xylene-alpha,alpha'-diamine-	NA	25790-41-4
2,4,6-tris-(N-1,4-Dimethylpentyl-para-phenylenediamino)-1,3,5-triazine	TAPDT	121246-28-4
N-Phenyl-1-naphthylamine	NA	90-30-2
N-Phenyl-2-naphthylamine	NA	135-88-6
[2-Methyl-4,6-bis((octylthio)methyl)phenol (Irganox 1520) <sup>1</sup>	NA	110553-27-0
Specialized graphene <sup>2</sup>	NA	1034343-98-0
1,1' -Pentamethylenebis(2,2-di-n butylhydrazine)	NA	No CAS
lpha- C-4- Hydroxy- 3,5- dimethylphenyl- N-isopropyl combined with 2,2'-	NA	77-62-3
Methylenebis[6-(1-methylcyclohexyl)-p-cresol]		
N-(4-Methylpentan-2-yl)-10H-phenothiazin-3-amine	NA	No CAS
7-(4-Methylpentan-2-ylamino)-2,3,4,10-tetrahydro-1H-acridin-9-one	NA	No CAS
2-Cyclohexyl-N-(4-methylpentan-2-yl)-1H-indol-5-amine	NA	No CAS
4-(1H-Indol-2-yl)-N-(4-methylpentan-2-yl)aniline	NA	No CAS
lpha- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	NA	No CAS
Amine functionalized lignin	NA	No CAS
Rambutan peel extract	NA	No CAS
Octyl gallate <sup>3</sup>	NA	1034-01-1
Nano calcium carbonate surface modified by gallic acid	NA	No CAS
Specialized carbon nanotube mixture <sup>4</sup>	NA	No CAS

Table 4.2 Possible Non-PPD Alternatives Meriting Further Study in Stage 1 AA

Notes:

6QDI = N-1,3-dimethylbutyl-N'-phenyl quinone diimine; AA = Alternative Analysis; CAS = Chemical Abstracts Service Number; NA = Not Available; NBC = Nickel dibutyldithiocarbamate; PPD = Paraphenylene Diamine; TAPDT = 2,4,6-tris-(N-1,4-dimethylpentyl-para-phenylenediamino)-1,3,5 triazine; TMQ = Poly(1,2-dihydro-2,2,4-trimethyl-quinoline).

(1) A potential alternative is Irganox 1520 CAS 110553-27-0 blended with Vulcazon AFS. However, according to the source patent (Dall'Abaco *et al.*, 2018), the best ratio is where 100% of the blend is Irganox 1520. Additionally, no hazard data were located for Vulcazon AFS. Thus, only Irganox 1520 CAS 110553-27-0 was evaluated further in this AA.

(2) The materials referred to as graphene in this report are graphene-based materials (sometimes referred to as a graphene nano-platelet) with a surface area not greater than 180 m2/g, and a carbon content greater than 99% and an oxygen content less than 1%. The lateral particle size of these materials is between 100 nm and 5  $\mu$ m.

(3) Octyl gallate was investigated instead of propyl, butyl or pentyl gallates. Propyl gallate has been tested as an antiozonant for non-rubber applications. Propyl gallate, however, is expected to be less suitable for rubber than other gallate esters with longer carbon chains. Propyl gallate melts at 150°C, which is the temperature at which rubber is mixed. Natural rubber compounds are sometimes mixed at a lower temperature. Unless propyl gallate completely melts and is dispersed in the compound, it will not have an opportunity to function as an antiozonant. Octyl gallate is a much better choice for this analysis because it melts at approximately 100°C and is sure to melt during mixing. Butyl gallate melts at 144°C so it may be acceptable, but octyl gallate has been used as a food additive, is more readily available and has more hazard information.

(4) An example of specialized carbon nanotube mixture is MOLECULAR REBAR® carbon nanotubes. The composition of MOLECULAR REBAR® carbon nanotubes is not known to the consortium members. However, according to a patent by Molecular Rebar Design on mixtures of discrete carbon nanotubes (Peddini, 2021), the mixture is a dried liquid concentrate of carbon nanotubes and various rubber processing oils (*i.e.*, tris (2-ethylhexyl) trimellitate oil, TP-95, HyPrene 100 naphthenic oil, castor oil, carnauba wax, curing co-agents, and/or sebacates), in which carbon nanotubes may make up anywhere from 5 to 50% of the dried liquid concentrate. Additionally, according to a 2016 SDS from Molecular Rebar Design on MR 1420X DLC (naphthenic oil-MR dry liquid concentrate), this product is 15-20% of functionalized multiwalled carbon nanotubes (CAS # not yet assigned) and 75-85% of naphthenic oil CAS # 64742-52-5 (Molecular Rebar Design, 2015).

Notably these include a number of possible alternatives (*e.g.*, ethoxyquin, dilauryl thiodipropionate) that likely cannot replicate the functions of 6PPD but were included in the list because they were identified by DTSC or Washington Ecology as possible alternatives. As with the possible PPD alternatives, a number of these chemicals also lack CAS registry numbers making it difficult to identify potential toxicity and chemistry data.

## 4.6 Relevant Factors

According to the SCP regulations, each alternative must be reviewed to determine whether its use would lead to a material difference relative to the existing chemical (here 6PPD) in the various relevant factors listed in the regulation. We have considered the possibly relevant factors listed in Tables 3-1A and 3-2B of the DTSC AA Guide (2017) (which are consistent with those listed in the SCP regulation, § 69505.5 (c) (CalDTSC, 2013a). Our review occurred in stages. For some factors (*e.g.*, molecular weight) it was readily apparent that these would not be material differentiators among the different products under review. For others (notably the various toxicities specified in the SCP regulation [CalDTSC, 2013a]), we had to first tabulate data for the possible alternatives to understand if these factors differed materially among the products (the results of the tabulation are discussed in Section 5). Based on our current knowledge of the properties of the different alternatives we have identified, we have determined which factors make a material difference among the priority product and any possible alternatives such that it would inform the decision of the Stage 1 AA. The conclusions we have reached are provided in Table 4.4.

## 4.6.1 Information on Sales of the Priority Product in California

The SCP regulations require that product sales information be included in the AA. As indicated in Table 4.3 below, an estimated 33,332,000 passenger car/light duty truck tires were shipped to California in 2022 and an estimated 3,160,000 heavy duty truck and bus tires were shipped to California in that year, for an estimated total number of tires shipped to California in 2022 at 36,492,000 units. Further details regarding the derivation of this number are provided in Appendix E.

Vehicle Category	USTMA Tire Shipments in 2022			
	U.S.	CA (est.)		
Passenger/Light Duty Truck	298,847,000	33,332,000		
Heavy Duty Truck/Bus	33,139,000	3,160,000		
Total	331,986,000	36,492,000		

Table 4.3	Estimated Annual	Shipments of	the Priority	Product in California
	EStimated Annual	Simplification of	the interiority	

Notes:

CA = California; est. = Estimated; U.S. = United States.

## 4.6.2 Relevant Exposure Pathways

We have considered the exposure pathway-related factors listed in Table 3-2C of the DTSC AA Guide (2017) (which are consistent with those listed in the SCP regulation, § 69505.5 (c) (3)) (CalDTSC, 2013a). Based on our current knowledge of the properties of the different possible alternatives we have identified, we have determined which exposure pathway-related factors make a material difference among the priority product and any possible alternatives. The conclusions we have reached are described in Table 4.5.

## 4.6.3 Conceptual Model for Product Life Cycle

Figures 4.2 and 4.3 show the conceptual exposure models for the life cycle of the Priority Products (*i.e.*, 6PPD-containing tires), and the potential non-6PPD-based tires, respectively. Both types of products are assumed to have the same life cycle stages, although end of life process may vary depending on the alternative.

Across the various life cycle stages (*e.g.*, raw materials extraction, manufacturing, distribution, use, disposal) of the Priority Products, exposure to 6PPD (or its possible alternatives) for raw material production workers, tire manufacturing workers, consumers, or the general public *via* the inhalation and/or dermal contact exposure routes is possible. For the chemical production and tire manufacturing workers, the main routes of concern are inhalation and dermal, although engineering controls and PPEs are used for some chemical production and tire manufacturing activities. For both new and retreaded tires, there is potential exposure (inhalation mostly) during tire buffing and retread casing preparation, however, engineering controls are in place for workers.

During the use phase of the life cycle, there are some emerging exposure pathways relating potential inhalation exposures to airborne TRWP for the general public (Cao *et al.*, 2022; Johannessen *et al.*, 2022). More data are needed in order to confirm these findings.

With respect to fish consumption, existing studies have reported the presence of 6PPDQ or 6PPDQ metabolite only in parts of fish that are not commonly consumed such as the brain (Hiki *et al.*, 2022 223-10360; Liao *et al.*, 2024 224-6089), bile (Montgomery *et al.*, 2023 224-6594), and gills (Hiki *et al.*, 2022 223-10360). Data are unavailable to assess the presence of 6PPDQ in other parts of fish that are more commonly consumed (*e.g.*, muscle) and if there are species related differences in 6PPDQ uptake.

In California, the current industry metrics for end-of-life tire processing estimate that 45%, 16%, and 35% of end-of-life tires enter into landfills, tire derived fuel, and recycling (*e.g.*, retreaded as tires for buses and

heavy-duty trucks, crumb rubber), respectively, in 2021 (CalRecycle, 2023). As for environmental exposures relating to the end-of-life processing of 6PPD-containing tires, more data are needed in order to assess potential concerns because migration of 6PPD, and the degradation product 6PPDQ, from recycled tires in products such as crumb rubber is not well characterized.

End of life processes are not known for possible alternatives to 6PPD. It is possible that motor vehicle tires containing possible alternative(s) to 6PPD will have similar end of life processes like landfill, reuse, crumb rubber, *etc*.

## 4.7 Life Cycle Segments

Consistent with the requirements of the Stage 1 AA, we approached life cycle considerations from the perspective of what is readily known or understood about the possible alternatives without engaging in extensive analysis. The Stage 2 AA would involve a more detailed effort at substantiating and potentially quantifying life cycle differences among the different products under review. The information presented below is intended to be complimentary to the life cycle information provided in Tables 4.4 and 4.5. We note that several life cycle assessments (LCAs) have been developed for tires (BLIC, 2001; Piotrowska *et al.*, 2019; Dong *et al.*, 2021; Michelin, 2021); however, none of these are granular enough to discuss the life cycle impacts of using 6PPD in tires. Thus, these LCAs provide very limited information for evaluating the life cycle impacts of tires containing 6PPD *versus* those that contain a possible alternative.

## 4.7.1 Raw Materials Extraction

To better understand the impacts of 6PPD and the possible alternatives, we compiled information on chemical production for the 6PPD and the other alternatives by consulting the National Library of Medicines PubChem database, which provides data on chemical manufacturing processes including the raw ingredients (Table 4.6). We used the database to sequentially trace back the processes and ingredients used to produce 6PPD and the possible alternatives until we reached an apparent starting material (e.g., simple hydrocarbons that are extracted from fossil fuels or mineral salts obtained from mining). Information was available for most but not all possible alternatives; for some specific PPDs information was lacking but could be inferred from information provided for other members of this family. Based on our review, it is expected that 6PPD and most of its possible alternatives will ultimately be produced from fossil fuel sources or from mining activities (e.g., metal catalysts, sulfur). Carbon nanotubes are primarily produced by vapor deposition using carbon rich gases (*i.e.*, methane or ethylene) obtained from fossil fuels, although use of carbon dioxide (e.g., from carbon capture) is a possible although not widespread technology. Commercially available graphene products are produced by chemical treatment of mined graphite. Octyl gallate can be obtained from plant sources although it is likely that chemical synthesis will be required to meet demand. Both amine functionalized lignin and rambutan peel extract involve agriculturally produced materials. It is unclear whether current agricultural production of these materials is sufficient to support the antioxidant/antiozonant needs of tire production, especially for rambutan peel extract. If the active ingredients in rambutan peel extract are produced synthetically, they may well involve fossil fuel precursors.

While information on raw materials extraction is limited, we believe it indicates it is unlikely that there will be material differences among the possible alternatives, as all involve inputs involving fossil fuels and/or mined materials of various types (*e.g.*, metals). All of the alternatives will have a potential for worker exposure during the raw materials extraction phase; the extent and potential hazards of the exposure could differ but is unlikely to be materially relevant for those involving fossil fuel sources. For other possible alternatives, there is uncertainty whether chemical synthesis from fossil fuel precursors will be required to

meet demand, if chemical synthesis is required, then exposure potential during this life cycle stage will likely be the same. Thus, the overall impact on raw materials extraction is unlikely to be materially different for any of the possible alternatives under consideration based on information that is currently available.

## 4.7.2 Resource Inputs and Other Resource Consumption

This aspect of the life cycle describes raw ingredients and energy required to produce the antidegradant as well as other materials that might be impacted by the production processes (*e.g.*, water used for cooling). This information would likely be found in LCAs and those available for tires do quantify resource inputs (water, energy) for tire manufacturing. However, as noted above, there is no LCA available that quantifies the aspects of the life cycle specific to 6PPD and thus quantitative comparisons to the impacts of the possible alternatives is not feasible. None of the alternatives under consideration would require a wholesale change in the resource inputs to components which make up the majority of a tire; it is expected that tires will still be comprised of elastomers, fabrics, steel, silica, carbon black, etc. 6PPD (or a possible alternative) comprises only a small percentage of the mass of a tire, so even if the resource inputs required for an alternative were to be significantly different from 6PPD, the overall impact on the resource inputs for the tire would be small. It seems likely that production of alternate PPDs would involve similar resource inputs and consumption as 6PPD. Many of the non-PPD inputs also involve some of the same raw ingredients (e.g., aniline) so their production pathways should not be materially different in terms of their impact on resources. Some of the possible alternatives are at least partially mineral in nature (e.g., graphene, carbon nanotubes, nickel dibutyldithiocarbamate) and so their processes are likely to be different but data are lacking to allow for comparisons to 6PPD. Lignin is a waste product of wood product manufacture, but the production of amine functionalized lignin would introduce energy and chemical inputs. Rambutan peel extract similarly involves a waste product as a starting ingredient with further chemical processing. Octyl gallate can be obtained from plant materials but how this could be done at the scale needed to meet tire production is unknown. Details about production processes are limited based on internet searches for their producers or the processes involved. It is also conceivable that some possible alternatives would be produced at different locations which could have different impacts in terms of raw material and chemical intermediate transportation (both in terms of distance and transportation mode). However, data are lacking to assess such effects and moreover, transportation networks would likely change significantly to increase efficiency due to the large volume of antidegradant involved. Similarly, with data lacking on the particular production processes required for many of the non-PPD alternatives, it is not possible to know how exposure potential might vary during this life cycle phase. Exposure potential for the different PPD alternatives is likely to be similar to that of 6PPD unless different process chemistry is required. Overall, whether resource inputs and consumption during production would be materially different between 6PPD and the possible alternatives is unknown at this time.

#### 4.7.3 Intermediate Materials Processes

Intermediate materials processes refers to chemicals that are produced in the synthesis pathways of 6PPD and its possible alternatives. If an alternative involves a particularly hazardous (or non-hazardous) component during its production, that could constitute a material difference relative to 6PPD. On the other hand, an alternative that eliminates a hazardous chemical used during the production process would be preferable. Review of information on the chemical precursors of the functional ingredients in the Priority Product and possible alternatives (as summarized in Table 4.6) suggests that the PPD-related alternatives have essentially equivalent intermediate materials and processes as 6PPD. The non-PPD materials (*e.g.*, TMQ, ethoxyquin, NBC, DLTDP, graphene, carbon nanotubes) involve different chemistries but all appear to involve industrial chemicals with some significant hazard. The hazards of graphene and carbon nanotubes depend on the structure (*i.e.*, thickness and aspect ratio) and stage of the production process.

As shown in Table 4.6, amine functionalized lignin may involve some hazardous chemistries in lignin extraction (*e.g.*, methanol) or amine functionalization but details are lacking. Rambutan peel extract may similarly involve hazardous chemicals in extraction (*e.g.*, ethanol, methanol); less hazardous extraction solvents (*e.g.*, hydrochloric acid) appear possible but all extraction process data appears related to small scale production; the chemistries appropriate for large scale production are unclear. From what is known regarding production of octyl gallate, chemical derived from natural sources (*e.g.*, fatty acid esters) does not appear to involve particularly hazardous chemistries although production from fossil fuel sources does involve some hazardous reagents (nitrous acid and p-cresol). Overall, it appears unlikely that intermediate materials processes are materially different among 6PPD and the various possible alternatives both in terms of potential risks for exposure and potential impacts on the environment.

## 4.7.4 Manufacture

We interpret this life cycle stage as relating to the manufacturing of the tire itself (*i.e.*, the priority product) as earlier steps in the life cycle (e.g., manufacturing of tire ingredients) are discussed in sections 4.7.2 and 4.7.3). The basic process of tire manufacturing described earlier (*i.e.*, mixing the individual compound, assembling each component, and then building the tire from the inner side outwards in a tire assembly machine) will likely remain the same for the foreseeable future. As indicated in Tables 5.1 to 5.3, all the possible alternatives pose some hazards which could be relevant for workers during exposure. 6PPD and a number of the possible alternatives (DTPD, CCPD, ethoxyquin, Irgazone 997, N-phenyl naphthalenes, and octyl gallate) are category 1 skin sensitizers; as already mentioned, this could expose workers who do not wear appropriate PPE (in violation of employer instructions). Other possible alternatives (graphene, carbon nanotubes, DLTP, NBC) do not have this hazard and would present less risk. Many of the latter do have other hazards (e.g., NBC is a carcinogen, ethoxyquin has systemic toxicity) so this may not constitute a material difference. In addition, there are many toxicity data gaps for a number of the possible alternatives so the potential hazards for workers from these chemicals is hard to judge.<sup>12</sup> Whether existing methods for controlling worker exposure during production can also be employed for the alternatives is also unknown. Overall, 6PPD and all of the possible alternatives have some hazardous properties involved in their manufacture (Table 4.6) These range from aniline and its precursors for the PPD alternatives to nickel, quinoline, ethoxyaniline, and various mineral acids. It should be noted that 6PPD (or a possible alternative) comprises a small percentage of the mass of the tire and it is assumed that all other ingredients (e.g., elastomers, fabric belts, steel, carbon black, silica, other additives) will largely remain the same and contribute the same level of hazard and potential for exposure. The extent to which some additives may change with a new antidegradant is also not currently known. Moreover, tires will still require vulcanization which constitutes the major source of energy required during the production process. Thus overall, changing the antidegradant may not have a material difference on the impacts of the manufacturing stage but data are too limited to be certain.

#### 4.7.5 Packaging

There appears to be no material difference among the priority product and the possible alternatives under consideration in terms of the type of packaging that would be used. Tires are shipped in shipping containers, on pallets or individually, depending on the needs of the commercial customer. Any alternative would likely have the same general weight and dimensions as an existing tire so the method of packaging would not be expected to be different. Regarding the antidegradant itself, it is also anticipated that a chemical

<sup>&</sup>lt;sup>12</sup> It should be noted that for chemicals to be used in large volumes such as would be required for an antidegradant in tires, these toxicity data gaps would have to be filled under various global chemical registration programs. This would allow for a better understanding of these hazards; however, the data are not currently available.

alternative to 6PPD would likely have a similar volume and density as 6PPD, therefore no significant change in raw material packaging would occur as a result of substitution of an alternative, unless higher quantities were required to offset any difference in mobility or reactivity *vs.* 6PPD. In terms of exposure potential, 6PPD has very low volatility and would not pose a potential for exposure during the packaging stage. Potential worker exposure should be similarly low for all possible alternatives in the packaging stage; alternatives that have excessive volatility or otherwise migrate out of the tire into the workspace would defeat the purpose of having a long lasting antidegradant in the tire. Overall, there is no expectation that switching to an alternative product would require either more or less packaging.

#### 4.7.6 Transportation/Distribution

Transportation/distribution of the priority product and/or chemical of concern is considered to be an insignificant pathway in terms of 6PPD migration to the environment because the tires on the vehicles used for this purpose generate a very small fraction of the total TRWP generated on an annual basis in California and, as a result, pose minimal potential for significant exposure. Additionally, the tires being transported do not emit the chemical of concern during transport for distribution or sale. There is no evidence that transportation impacts (e.g., from chemical suppliers to formulators; from formulators to retail outlets) would be different among the priority product or any of the possible alternatives under consideration. For example, 6PPD and its possible alternatives do not constitute the bulk of the product weight, so shipping tonnage would not be expected to be different. Depending on the production of the alternative (e.g., at locations more or less distant from tire manufacturing facilities) there could be an impact on the transportation/distribution portion of the life cycle. However, no data are available to assess this potential impact since an alternative has not been definitively identified. It is also likely that cost would incentivize minimizing transportation of raw ingredients to tire production facilities so transportation impacts (e.g., CO<sub>2</sub> emissions, TRWP generated from transport trucks) from the antidegradant production facility to tire production locations may not significantly change but this is unclear at the present time. In terms of potential for exposure during the transport phase, as noted above, an alternative that had a greater potential to volatilize or otherwise be emitted from the tire than 6PPD would likely not be a suitable replacement.

#### 4.7.7 Use

Use is one area where the possible alternatives may have materially relevant differences to 6PPD, because the use phase of the product involves generation of TRWP and washing off of the antidegradant present on the tire surface. In addition, if a new alternative changes the abrasion rate of a tire, that will have an impact on TRWP generation. If the alternative also has differential hazards *versus* 6PPD (as some appear to do) this could be a materially relevant difference. If alternatives can reduce the release of 6PPD (or another chemical with similar toxicity to susceptible species) to the environment, then there would be a material difference in this parameter. Beyond the reported effect of 6PPD on certain salmon species, there are other aspects of the use phase which will need to be considered. One important consideration would be whether an alternative results in greater or lesser tire wear, potentially resulting in different environmental impact. However, TRWP generation rate likely depends more on driving conditions (*e.g.*, vehicle, load, speed, drive cycle, road surface type) than the antidegradant. Any alternative that could be implemented in a reasonable timeframe would still be rubber-based. Given the potential for a reduction in impact of the antidegradant, this does appear to be a factor that will be materially relevant to alternative selection.

The use phase may also consider storage of the tires when not in use (e.g., in an attached garage), in the vehicle trunk (e.g., spare tires), or in a distribution facility. Again, the low volatility of 6PPD does not indicate significant exposure during storage and an alternative that is more volatile or otherwise migrates

out of the tire during storage would not satisfy the goals of having a long-term reservoir of antidegradant in the tire.

## 4.7.8 Operation and Maintenance

Tires require little maintenance while in actual use. The maintenance that is required (*e.g.*, maintaining tire pressure, tire rotation, balancing) would not be expected to change with any of the possible alternatives. This phase of the product lifecycle is also not associated with release of 6PPD from the tire and exposure of any potential receptor. Consortium members cannot conceive of a mechanism by which this would be different for any of the possible alternatives given that the basic nature of a tire will remain the same. In terms of exposure potential, all antidegradants will be required to be present on the surface of the tire to act against ozone. Thus, dermal contact (*e.g.*, during tire rotation or changing) may be possible but would occur for all antidegradants to some degree (the extent is unknown). Overall, operation and maintenance is unlikely to be a relevant factor among 6PPD and the possible alternatives.

## 4.7.9 Waste Generation and Management

We interpret this stage of the lifecycle to refer to production waste during tire production. TRWP are a wear product generated during the use phase but the impact of these is discussed in Section 4.7.7. Manufacturers cycle excess formulated compound back into the manufacturing process (called rework); this substantially minimizes waste generated during tire production. The ability to continue the rework process is critical for minimizing production waste and will need to be considered for any alternative. For example, as noted, earlier Consortium members' experience suggests it may be difficult to use modified EPDM or bromobutyl rubber as rework in production due to issues of behavior with other tire additives (*e.g.*, carbon black) and is one of the reasons why they were not included in the AA. Any alternative antidegradant that impacts processing time or temperature stability could significantly impact the potential for rework. The extent to which any of the alternatives affect rework is not known at this time. It is also unclear to what extent worker exposure to possible alternatives from production waste management could differ. Although many of the physico-chemical properties of the alternatives would need to be similar to be effective in a tire, how the waste would be managed during production is unknown. Overall, this stage of the life cycle has the potential to be substantially different among the possible alternatives.

## 4.7.10 Reuse and Recycling

As shown in the conceptual model for tires (Figures 4.1 and 4.2), about half of passenger tires in California are reused or recycled at the end of their useful life. Significant portions of the spent tire stream is re-used as fill material (road paving or crumb rubber infill), as structural materials (*i.e.*, sea walls) or burned for energy (*e.g.*, in cement kilns). When burned for energy in a cement kiln or incinerator, 6PPD is likely degraded to  $CO_2$  and other gases; a typical operating temperature for a cement kiln is well over 1000°C and tire pyrolysis is reported to occur at temperatures ranging from 300°C to 700°C (Zerin *et al.*, 2023). The flashpoint of 6PPD is reported to be 202°C (see Appendix C), meaning that 6PPD would be consumed during the thermal processing of tires. US EPA (2016) noted "Laboratory testing of a rotary kiln incinerator simulator (RKIS) indicated that efficient combustion of supplemental TDF [tire-derived fuel] can destroy many volatile and semivolatile air contaminants. However, it is not likely that a solid fuel combustor without add-on particulate controls could satisfy air emission regulatory requirements in the US."

One particular consideration for truck and bus tires is retread. An alternative that interferes with the retreading process could substantially increase tire waste because new truck and bus tires would need to be

purchased and discarded more frequently. Depending on the possible alternative there could also be an increased or reduced potential for exposure of retread facility workers. During retreading, the old worn tread is "buffed" away to expose the tire casing. The tire tread is then rebuilt and the tire is cured in an analogous process to original tire manufacturing. Exposure may occur during these processes. It is not clear how implementation of possible alternatives might affect the retread processes and potential exposure of retread workers. It is assumed that major effects on the retread process would not be tolerable to manufacturers but how retread might differ with another antidegradant remains to be investigated. Overall, the reuse and recycling stage of the life cycle has the potential to be substantially different among the possible alternatives.

## 4.7.11 End-of-life Disposal

According to CalRecycle, in 2021, 45% of the passenger tires waste stream was landfilled, 16% was used as tire derived fuel, and 35% was recycled, for example as crumb or ground rubber or retreading for bus or heavy-duty truck tires, *etc.* (CalRecycle, 2023). Note that use as tire-derived fuel and use as crumb rubber are classified as reuse and are discussed in Section 4.7.10. If tires with alternative antidegradants have a different lifespan this could impact the amount of post-manufacturing tire waste generated and could exceed reuse and recycling capacity. It is also possible that chemicals can leach out of tires and impact the environment from improperly designed or operated disposal facilities. Overall, the end-of-life stage of the life cycle has the potential to be substantially different among the possible alternatives.

## 5.1 Hazard

#### 5.1.1 Hazard Evaluation Approach

According to the SCP Regulation (CalDTSC, 2013a), a hazard evaluation comparing 6PPD and possible alternatives must include hazard endpoints from the Green Chemistry Hazard Traits defined in the California Code Of Regulations, Title 22, Division 4.5, Chapter 54.<sup>13</sup> Gradient organized the human health hazard endpoints into two groupings (*i.e.*, Group A and Group B). Group A hazard endpoints have corresponding GHS hazard endpoints (*e.g.*, acute toxicity, carcinogenicity, mutagenicity), which allowed for transfer of existing hazard assignments according to each chemical's ECHA Registration, Evaluation, Authorisation, and Restriction of Chemicals (REACH) dossiers and GreenScreen assessments, if available. Some additional hazard endpoints, not included among the Green Chemistry Hazard Traits but necessary for our scoring system, were also added to the evaluation of Group A endpoints (*i.e.*, systemic organ toxicity, dermal and respiratory sensitization). The latter two endpoints were also evaluated as part of the SCP hazard trait of immunotoxicity.

Group B hazard endpoints are those that do not have corresponding GHS hazard endpoints (*e.g.*, ototoxicity, hematotoxicity, immunotoxicity, cardiovascular toxicity). To a large extent, these types of specific toxicity are subsumed in the larger category of "systemic toxicity," which *is* addressed in ECHA dossiers and GreenScreen assessments under Systemic Target Organ Toxicity – Repeated Exposure.<sup>14</sup> Nonetheless, to comply with the SCP regulations, we addressed these remaining health endpoints (*i.e.*, discussed herein after as "Group B" endpoints) by reviewing ECHA dossiers, supplemented by GreenScreen assessments, for data on these particular health effects.

#### 5.1.1.1 Hazard of Constituents *versus* Risk of Final Products

It is important to stress that while individual chemical-specific hazards are presented for possible alternatives in this section, chemical-specific hazards do not necessarily reflect the hazards of the actual final product (*i.e.*, a tire). Thus, when reviewing chemical-specific hazard data obtained from testing of individual product ingredients (such as anti-degradants), the indication of a hazard does not necessarily equate to an actual human or ecological health risk caused by using the chemical in the tire.

<sup>&</sup>lt;sup>13</sup> This evaluation is based on a literature review of the available studies and did not involve an independent verification of the results of any study.

<sup>14</sup> Two of the discretionary endpoints added to Group A, namely skin sensitization and respiratory sensitization, partially fit under the Group B endpoints (dermal toxicity, respiratory toxicity and immunotoxicity) but do not fully address all forms of toxicity that might fit under these categories. As noted above, these group B endpoints are not fully addressed by GHS hazard criteria. To comply with the SCP regulations, where one of these Group A endpoints is considered relevant, the corresponding Group B endpoint is considered relevant.

## 5.1.1.2 Group A Endpoints

For the Group A human health hazard endpoints (*e.g.*, carcinogenicity, mutagenicity, reproductive and developmental toxicity) (summarized in Table 5.1), we reviewed the hazard properties of 6PPD and the possible alternatives for hazard properties using mainly ECHA REACH dossiers (ECHA, 2024) and existing GreenScreen assessments, if available.

There are a number of additional potential hazard concerns required by the SCP regulations that are not classified in the ECHA dossiers. These were addressed as follows:

- Endocrine Disruption and California Proposition 65. The European Union's (EU) Endocrine Disruptor Priority List and the California Proposition 65 list were used to inform these endpoints (UL LLC, 2023).
- **Terrestrial Toxicity.** Pharos (Healthy Building Network, 2019) was used to inform this endpoint.
- **Bioaccumulative Potential.** Chemicals are considered bioaccumulative if the bioconcentration factor (BCF) is >1,000 according to California Code of Regulations, according to title 22, Division 4.5, Chapter 54, Article 5 (CalOEHHA, 2012).
- **Persistence.** Based on the Globally Harmonized System of Classification and Labelling of Chemicals (GHS) (UN, 2019), possible alternatives are considered persistent if 0 to <20% of the chemical degrades within 28 days, inherently biodegradable if 20 to <60% of the chemical degrades within 28 days, and readily biodegradable if 60 to 100% of the chemical degrades within 28 days.
- Global Warming Potential (GWP). We compared the possible alternatives to the greenhouse gases listed in Table 8.a.1 of the "Fifth Assessment Report of the Intergovernmental Panel on Climate Change (IPCC)" (IPCC, 2013). Chemicals that are neither gases nor chlorinated/fluorinated were considered to have negligible GWP.
- Ozone-Depleting Potential (ODP). US EPA's list of ozone-depleting substances (US EPA, 2018b) was used to evaluate this endpoint. Chemicals that are neither gases nor chlorinated/brominated were considered to have negligible ODP.
- Clean Air Act VOC Contributing to Smog Formation. We assessed whether each possible alternative is a volatile organic compound (VOC); a chemical was considered to be a VOC if it had a vapor pressure equal to or greater than 0.1 mm mercury (Hg) based on criteria in CARB (2009). Additionally, we noted whether the chemical is listed as a substance exempted under 40 CFR § 51.100 (as per CARB, 2009).

To meet the SCP regulations' requirement for an easily understood matrix of hazards, we adapted the colorcoding system used by various hazard evaluation tools, such as the GreenScreen hazard evaluation system (Clean Production Action, 2019). This employs a red/orange/yellow/green "heat map"-type color coding to allow the reader to easily compare the hazards of different chemicals at a high level. In addition, we added light grey shading to the endpoints for which no data were found (*i.e.*, data gaps). It should be noted that data gaps do not indicate a lack of toxicity; they merely indicate that no information was found.

## 5.1.1.3 Group B Endpoints

For group B endpoints (*e.g.*, ototoxicity, cardiovascular toxicity), we qualitatively summarized the reported findings concerning these adverse effects or the lack of relevant adverse effects, as well as any data gaps (Table 5.2). In conducting our review, we focused primarily on repeated-dose studies, because these typically have the most detailed evaluation of potential health effects, whereas acute dosing studies often

only examine a limited number of health effects using gross measures (*e.g.*, clinical signs, organ weight changes). This is a qualitative approach, but we believe that the alternative approach (creating an arbitrary and novel GHS-like scoring rubric for all of the additional SCP hazard endpoints that lack recognized classifications like the GHS) would be unreasonably burdensome and problematic, because, as noted above, many of these health effects are already addressed in the larger category of systemic toxicity. As noted earlier, when a discretionary Group A endpoint (*e.g.*, respiratory or dermal sensitization) was considered to be materially different among the possible alternatives, the corresponding Group B endpoint (immunotoxicity) were also considered relevant. Finally, for some Group B endpoints we took the following approaches for certain Group B endpoints:

- **Epigenetic Toxicity.** We noted from our review whether possible alternatives were or were not genotoxic. Genotoxicity generally implies changes in the DNA sequence, which is outside the scope of epigenetic toxicity, but genotoxicity also implies a potential for interaction with DNA, so it is evaluated given that more direct data on epigenetic effects are lacking. In addition, we looked for other relevant information in our data sources regarding other types of DNA activity (*i.e.*, altered methylation).
- **Reactive in Biological Systems.** In their Priority Product profile, DTSC cited this as a factor of concern. We are unable to make a determination about this relevant factor for a number of possible alternatives due to lack of hazard data. For 6PPD, DTSC listed 6PPDQ as reactive in biological systems in the Priority Product Profile (CalDTSC, 2022).
- **Immunotoxicity.** We included respiratory and dermal sensitization as relevant under this endpoint, as these are immune system-mediated effects.

Again, it should be noted that the chemical-specific hazards presented in Tables 5.1-5.4 for the possible alternatives do not represent the potential hazards or risk of finished tires. As part of the tire production process, the individual tire ingredients are chemically altered *via* various processes (*e.g.*, cross-linking, degradation) (Bebb, 1976) that make it difficult to quantitatively extrapolate from the risk of the individual ingredients tested as single substances to that of the finished product in a reliable manner. In addition, the finished tire limits the ability of 6PPD to migrate (*i.e.*, relative to the exposed surface area of the tire, concentration and accessibility of 6PPD is low).

## 5.1.1.4 Salmonid Acute Toxicity

Ecotoxicity data evaluated in Group A were obtained from ECHA dossiers which primarily relied on Organisation for Economic Co-operation and Development (OECD) Test Guidelines (TGs). Typical OECD TG test species of fish include zebrafish (*Danio rerio*), fathead minnow (*Pimephales promelas*), carp (*Cyprinus carpio*), Japanese medaka (*Oryzias latipes*), guppy (*Poecilia reticulata*), bluegill (*Lepomis macrochirus*), rainbow trout (*Oncorhynchus mykiss*), three-spined stickleback (*Gasterosteus aculeatus*), sheepshead minnow (*Cyprinodon variegatus*), European sea bass (*Dicentrarchus labrax*), and red sea bream (*Pagrus major*) (see for example, OECD TG No. 203 and No. 210). Coho salmon are not an OECD TG test species. Because DTSC listed 6PPD due to effects on coho salmon and certain other salmon species, we also gathered available published acute toxicity data in salmonids more broadly.

For each possible alternative, we conducted a comprehensive literature search to identify relevant lethal concentration 50 (LC<sub>50</sub>) data for the parent chemical and quinone product if applicable (*i.e.*, phenylene diamines). We focused specifically on data reported in the peer-reviewed literature and as summarized in scientific and regulatory agency reports (such as DTSC and WA DOE) for organisms in the Salmonidae family (*e.g.*, genuses Salmo, Salvelinus, Oncorhynchus), based on phylogenetic relationship to coho

salmon. Species in genus Salmo include Atlantic salmon (*Salmo salra*) and brown trout (*Salmo trutta*). Species in genus Salvelinus include Arctic char (*Salvelinus alpinus*), brook trout (*Salvelinus fontinalis*), and white-spotted char (*Salvelinus leucomaenis pluvius*). Species in genus Oncorhynchus (other than coho salmon, *Oncorhynchus kisutch*) include Chinook salmon (*Oncorhynchus tshawytscha*) and rainbow trout (*Oncorhynchus mykiss*). We conducted searches of the scientific literature in the PubMed, Scopus, and US EPA ECOTOX databases, agency reports (*e.g.*, DTSC, WA DOE), and, as mentioned, ECHA REACH dossiers. For PubMed<sup>15</sup> and Scopus<sup>16</sup> literature searches, we searched the common name and CASRN of each chemical in a search string to identify aquatic toxicity data in salmonids. In order to provide the most conservative evaluation in the current AA, we identified the lowest reported LC<sub>50</sub>s for each possible alternative for both the parent chemical itself and its quinone product if applicable and available. Other studies may have reported higher LC<sub>50</sub> values. We also reported the duration of exposure (*e.g.*, 24 hrs, 96 hrs), experimental flow conditions (*e.g.*, flow-through, static) and whether the exposure concentrations were nominal or verified by experimental measurement. Results of the salmonid acute toxicity evaluation are presented in Table 5.4.

## 5.1.1.5 USGS studies of alternatives involving cell lines

Although the SCP program does not require generation of additional data, under sponsorship from USTMA, the US Geological Survey (USGS) have conducted initial studies on a subset of PPD-related possible alternatives, specifically, CCPD, 77PD, and DPPD, with 6PPD studied for comparison. These three phenylene diamines were selected because they have slightly different chemical structures (alkyl alkyl or aryl aryl substitutions) compared to 6PPD (alkyl aryl substitution), or they were selected because they were mentioned as possible alternatives by documents published by California DTSC (CalDTSC, 2022) and/or Washington Department of Ecology (WA Ecology, 2021). The USGS authors had previously published a study demonstrating that the *in vitro*. Pacific salmon cell system was able to partially replicate the speciesspecific differences in susceptibility to 6PPDQ reported in whole animal studies (Greer et al., 2023a). The current studies used cell lines derived from two Pacific salmon species, coho and Chinook. The goal of these cell-based studies was to determine whether the selected alternatives produced toxicity in salmonid cells in a manner similar to 6PPD. Given the difficulty in conducting screening studies in whole salmon (e.g., time, cost, availability, low throughput) and the need for new approach methodologies that limit the use of live animal screening, these studies were also intended as proof-of-concept investigations to determine whether a cell-based screening approach would provide a rational basis for identifying a subset of possible alternatives to 6PPD.

The studies conducted in collaboration with the USGS focused on the selected possible alternatives as well as their ozonation products. Rather than isolated chemical solutions, the studies involved extracts of tire rubber that had been formulated with the different antiozonants at a standard concentration. Strips of the rubber were subjected to ozone at three different concentrations (0, 10, and 40 parts per hundred million (pphm)). Strips were then eluted in 100% ethanol overnight. A portion of each extract was sent to an

<sup>&</sup>lt;sup>15</sup> As an example PubMed search, the search string for IPPD was as follows: (101-72-4[EC/RN Number]) OR ("N-isopropyl-N'phenyl-p-phenylenediamine"[tiab:~0] OR IPPD) AND (oncorhynchus OR ecotox\*[Title/Abstract] OR aquatic[Title/Abstract] OR ecolog\*[Title/Abstract] OR ecosystem\*[Title/Abstract] OR fish[Title/Abstract] OR salmon[Title/Abstract] OR trout[Title/Abstract] OR "marine organisms"[Title/Abstract] OR "aquatic organisms"[Title/Abstract] OR "freshwater"[Title/Abstract] OR wildlife[Title/Abstract] OR fauna[Title/Abstract] OR "Ecotoxicology"[Mesh] OR "Aquatic Organisms"[Mesh] OR "Ecosystem"[Mesh] OR "Fishes"[Mesh] OR "Animals, Wild"[Mesh] OR "Ecology"[Mesh] OR "Salmon"[MeSH] OR "Trout"[Mesh]) NOT (zebrafish OR Human OR mice).

<sup>&</sup>lt;sup>16</sup> As an example Scopus search, the search string for IPPD was as follows: (CASREGNUMBER (101-72-4) OR TITLE-ABS-KEY ("IPPD" OR "N-isopropyl-N'-phenyl-p-phenylenediamine")) AND TITLE-ABS-KEY (ecotox\* OR aquatic OR fish OR salmon OR trout OR "marine organisms" OR "aquatic organisms" OR "freshwater" OR wildlife OR invertebrates OR oncorhynchus) AND NOT TITLE-ABS-KEY (zebrafish OR human OR mice).

analytical laboratory for quantification of parent compounds and associated quinones. The rubber extracts were then tested at different concentrations in both the coho-derived cell line (CSE-119) and Chinook cell line (CHSE-214). Because the research is being conducted under a Cooperative Research and Development Agreement (CRADA) with USGS, which intends to publish the results in a peer reviewed journal, full details of the methods cannot be provided here but may be able to be shared with DTSC by USTMA as confidential business information. A brief, preliminary overview of the results of these studies is provided below in Section 5.1.3.5.

## 5.1.1.6 Transformation Products

As required by the SCP regulations, we also identified the main transformation products of the possible alternatives and reviewed their chemical properties and potential toxicity (Table 5.10). We identified potential transformation pathways and products mainly *via* ECHA REACH dossiers (ECHA, 2024). We then reported the classified GHS hazards of the potential transformation products *via* the ECHA dossiers of the transformation products. Additionally, we noted if any transformation product is on the EU Persistent, Bioaccumulative, and Toxic (PBT) list, the California Toxic Air Contaminant (TAC) list, and/or the California Proposition 65 list, using Underwriters Laboratories Inc.'s (UL) List of Lists (LOLI) (UL LLC, 2023). In Table 5.10, possible alternatives that were present on any of these lists or which had significant toxicity under GHS classification (*e.g.*, carcinogenicity category 1, acute or chronic toxicity category 1, mutagen) were flagged with orange shading.

Prior to reliance on information in ECHA dossiers, we evaluated the potential to conduct analyses with the Open (Quantitative) Structure-Activity/Property Relationship App (OPERA) and the OECD QSAR Toolbox. Overall, neither program yielded useful information and in fact, while OPERA can predict physical-chemical properties, environmental fate parameters, and toxicity endpoints (Mansouri *et al.*, 2018), it does not have a module to predict transformation products. Our methodology and findings are discussed briefly below.

Regarding hydrolysis, we used the OECD QSAR Toolbox hydrolysis simulator at pH neutral, acidic, and basic. Under these conditions, none of the hydrolysis simulators predicted transformation products for any PPD-based alternatives, most likely because the p-phenylenediamine moiety (present in 6PPD and similar PPDs) may not be included in the software's hydrolysis model training set. In contrast, for two non-PPD alternatives, 6QDI and dilauryl dithiopropionate, hydrolysis transformation products could be modeled based on hydrolysis at their imine and ester bonds, respectively. The modeled hydrolysis products were identical to those reported in the ECHA REACH dossiers for the possible alternatives (described below).

Regarding ozonation and oxidation, we looked for tools that could address these transformation pathways. We did not find programs that could model transformation products through the ozonation pathway. The QSAR Toolbox does not contain an ozonation simulator, but does contain an autoxidation simulator. Autoxidation is a free radical reaction of a chemical with molecular oxygen that results in the formation of oxidation products of 6PPD (OECD, 2017). The QSAR Toolbox autoxidation simulator did not predict 6PPDQ as one of the oxidation products. We note that although 6PPD can react with ozone or potentially ozone-related secondary oxidants (*e.g.*, hydroxyl radical), no significant degradation of 6PPD in zero-grade (ozone-free) air was observed experimentally after 6 hours of exposure, as reported by Hu *et al.* (2022).

As mentioned, due to the lack of utility using QSAR modeling approaches to evaluate transformation products of the alternatives, we identified potential transformation pathways and products mainly *via* ECHA REACH dossiers (ECHA, 2024). We did not conduct a search of peer reviewed literature for transformation products of the alternatives but note that Cao *et al.* (2022) reported formation of respective quinone products of 6PPD, CPPD, DPPD, DTPD, and IPPD in environmental samples in Hong Kong

(*e.g.*, urban runoff, roadside soils, and air particles). In addition, Johannessen *et al.* (2022) reported various transformation products of 6PPD in passive air samples from 18 cities (*i.e.*, 6PPDQ, 6PPD TP181, 6PPD TP198, 6PPD TP212, 6PPD TP256, 6PPD TP274, 6PPD TP282, 6PPD TP298, and 6PPD TP373).

ECHA REACH dossiers provide some information on potential transformation products but are incomplete in terms of the range of possible transformation products covered. Data on potential transformation products was available for 6PPD and six of the 43 possible alternatives (Table 5.10). The six possible alternatives were: five PPD derived chemicals (7PPD, 77PD, IPPD, commercial DTPD, and DAPD) and the non-PPD alternative 6QDI. These six possible alternatives had hydrolysis studies reported in the ECHA dossier. The 37 remaining possible alternatives did not have ECHA REACH dossiers or their respective dossiers did not identify transformation products in their hydrolysis studies.

Among the six possible alternatives with data in ECHA REACH dossiers, 11 potential hydrolysis products were identified. According to ECHA, most PPDs have similar transformation pathways for hydrolysis where the first step is the cleavage of the alkyl chain resulting in a phenolic chemical (*e.g.*, 4-hydroxydiphenylamine for 6PPD, 7PPD, and IPPD) and an alkylamine (ECHA, 2024). The structure of the alkylamine will vary due to PPDs having different alkyl chain lengths. For example, for 6PPD the alkylamine formed is 1,3-dimethylbutylamine, whereas for 7PPD the alkylamine formed is 1,4-dimethylpentylamine. Most PPDs are currently understood to form a quinone-imine transformation product which may be from the oxidation of the phenolic chemical. The secondary hydrolysis products are usually p-benzoquinone and p-hydroquinone, and in some cases, an amine due to the cleavage of the second amino group. For PPDs linked to phenyl groups (*e.g.*, 6PPD, 7PPD), the amine formed is aniline. Most PPDs are expected to follow a similar transformation product in any of the ECHA dossiers of any of the examined chemicals.

## 5.1.2 Hazard Scoring Approach

We quantitatively scored hazards of 6PPD and possible alternatives using an adaptation of the Chemical Scoring Index (CSI). The CSI is a largely GHS-based tool for ranking the hazards of chemicals in oil and gas products (Verslycke *et al.*, 2014). The CSI has been used in prior AAs that have been accepted by DTSC. The CSI considers not only the hazard but also the percentage of each chemical in the product formulation. These two pieces of information are combined using a scoring matrix to arrive at a total hazard score for the chemicals in the product. The original form of the CSI is heavily focused on acute toxicity hazards and did not have all the endpoints required under the SCP regulations (Verslycke *et al.*, 2014), so some modifications to the CSI were required for this assessment. The modifications to the original CSI approach consisted of the following, and are also described in Tables 5.5-5.7:

No data substances. Gradient did not attempt to score chemicals that have no data, since doing so would result in low scores, which could be interpreted as less hazardous compared to chemicals with higher scores based on data. We also did not use predictive toxicity modeling software to fill endpoints such as aquatic toxicity or dermal sensitization because doing so would unfairly give data-poor compounds a scoring advantage compared to data-rich compounds. The majority of the data-gap endpoints cannot be modeled, thus data-gap penalty scores, which are always lower than the score of the most severe classification, would be applied. For example, DOPD CAS 101-67-7 is a complete data-gap chemical. If we were to score DOPD using the data-gap penalty scores, its total score would be 220, which would make DOPD appear as a less toxic alternative than 6PPD, a data-rich chemical, that has a total score of 275. However, DOPD should not be considered a less toxic alternative to 6PPD because DOPD lacks toxicity information.

- **Mixtures**. One of the possible alternatives being evaluated in this Stage 1 AA is a mixture of nitrones as a class and Lowinox WSP. While hazard data was identified for Lowinox WSP, no hazard data was identified for nitrones as a class. *If* hazard data were identified for nitrones as a class, then we would have used GHS mixture classification rules (UN, 2019) to classify the mixture based on hazards and percentage of the individual components. Note, it is also important to know if individual components in a mixture would react in a way resulting in a set of hazards that do not reflect the hazards of the individual components (*e.g.*, monomers and additives resulting in a polymer). It is not known what type of reaction nitrones as a class and Lowinox WSP would have and whether this would have an effect on the hazards of the mixture.
- Assigning a penalty for endpoint-specific data gaps for data-poor chemicals. The original CSI approach does not penalize data gaps on an *endpoint-by-endpoint* basis. It only penalized a product if <30% of its composition is accounted for by components with no data, with a maximum penalty score of 100 for the environmental categories, 100 for the human health categories, and 50 for the physical categories (if ≥30% of a product's composition is accounted for by components with no data, it would not be evaluated [see above]). Thus, the CSI lacks granularity in terms of how many or which health endpoints have missing data. For this AA, we added endpoint by endpoint penalty scores for data gaps, which is more conservative than the CSI's approach. These data gap scores were assigned based on hazard severity (*i.e.*, the maximum carcinogenicity and mutagenicity data gaps are scored 50 *versus* 10 for endocrine disruption). Also, in general, data gap penalty scores are lower than the Category 1 hazard scores for the same endpoint, and data gap penalty scores generally decrease with decreasing chemical concentrations, except for some categories of particular concern (*e.g.*, Category 1 carcinogens).
- Chronic aquatic toxicity. The CSI does not have scores for chronic aquatic toxicity; thus, the CSI's scoring system for acute aquatic toxicity was used.
- **Terrestrial toxicity and GWP.** The CSI does not have scores for terrestrial toxicity or GWP; thus, scores for these metrics were created.
- Mutagenicity, reproductive/developmental toxicity, and systemic toxicity single- and repeated-dose toxicity. Under the original CSI approach, scores did not differ between the GHS subcategories for mutagenicity, reproductive/developmental toxicity, and systemic toxicity single- and repeated-dose toxicity. To provide more granularity in the scoring, for this AA, we adopted the maximum CSI score for Category 1 for all of the abovementioned hazard endpoints, but scaled down to a lower score for subsequent subcategories (approximately 50% of the Category 1 score for Category 2, and so on). This approach is in line with the spirit of the GHS and CSI. Note that the CSI implemented lower scores for Categories 2 and below for carcinogenicity, corrosivity, and acute mammalian and aquatic toxicity, but not for the four abovementioned endpoints.
- Endocrine. We moved the endocrine hazard endpoint from ecological toxicity to human health toxicity. Additionally, we used a score of 25, instead of the original 50 in the CSI, for endocrine disruptors, because the EU's Endocrine Disruptor Priority List, which we used for this assessment, is a listing of chemicals with endocrine *concern* that should be explored *via* testing, rather than a list based on studies showing actual effects. In contrast, the maximum score for mutagenicity is 50 and is based on positive findings of a mutagenic effect.
- Skin and respiratory sensitization. We created separate skin and respiratory sensitization categories from the original CSI's "sensitizer" category. Additionally, we used a maximum score of 50, instead of the original 25 in the CSI, for skin and respiratory sensitization. This is because the original CSI approach was developed for oil and gas applications, in which sensitization was less of an issue.

- **VOCs contributing to tropospheric ozone formation.** We used a maximum score of 75, instead of the original 50 in the CSI, for this endpoint. Because smog formation is a particular concern for California cities, we increased the maximum score for this endpoint.
- Eye and skin irritation. We created separate categories for eye and skin irritation from the "irritant" category in the original CSI to be more consistent with the required SCP regulations' toxicity categories. We assigned a maximum data gap penalty score of 25 for products in which components with no data account for more than 30% of the composition, matching the score of 25 for Category 1 skin or eye irritants, because these are common hazards.

When the original CSI approach provided numerical scoring values for an endpoint, we used those scores, other than the abovementioned deviations for endocrine disruption, skin and respiratory sensitization, and VOCs contribution to tropospheric ozone formation. When scores for endpoints were created, we employed scores that were consistent with similar endpoints (*e.g.*, we used the same scoring used for "irritation" in the original CSI approach for the new eye and skin irritation scores). In our scoring approach, we did not score Group B endpoints (Table 5.2) or acute toxicity in salmonids, because any adverse effects that rise to the level of GHS classification would already be captured under the single target organ toxicity – repeated exposure endpoint and acute aquatic toxicity, respectively, and we wanted to avoid "double counting." We also did not attempt to score chemicals that have no data, since doing so would result in low scores, which could be interpreted as less hazardous compared to chemicals with higher scores based on data.

Lastly, Greenscreen assessments classified 6PPD and several possible alternatives as respiratory sensitizers based on dermal sensitization hazard, a respiratory sensitization structural alert (phenylenediamine alert from OECD QSAR Toolbox), and/or professional judgement (ToxServices, 2021a,b,c,f,g,h). Gradient listed ToxService's hazard assignments in the relevant hazard table (Table 5.1 Group A); however, Gradient did not score the endpoint based on respiratory sensitization assignment. Instead, a data gap score was assigned because there are very few recognized respiratory sensitizers relative to the large number of skin sensitizers (Kimber *et al.*, 2018; North *et al.*, 2016).

According to the United Kingdom government agency, Health and Safety Executive, there are only approximately 45 substances or chemical groups that are recognized respiratory sensitizers, mostly made up of enzymes, dusts, and low-molecule-weight chemicals (United Kingdom, Health and Safety Executive, 2021). Comparatively, there are many thousands of known or suspected dermal sensitizers (Kimber *et al.*, 2018). Additionally, many chemical allergens are exclusively dermal sensitizers (Kimber *et al.*, 2018). In other words, a substance's ability to elicit a dermal sensitization response is not a good predictor of its ability to elicit a respiratory sensitization response.

It is also important to recognize that the mechanisms in which dermal sensitization occurs (*i.e.*, adverse outcome pathway) is well understood, however, very little is known about the adverse outcome pathway leading to respiratory sensitization (North *et al.*, 2016). Respiratory sensitizers tend to induce a predominantly T helper cell type 2 (Th2) response involving IgE antibodies, whereas dermal sensitizers tend to induce a predominantly T helper cell type 1 (Th1) response (North *et al.*, 2016).

## 5.1.3 Hazard Scoring Results

### 5.1.3.1 Hazards of 6PPD and Possible Alternatives

Tables 5.1 and 5.3 summarize hazard scoring for the human health (Group A endpoints only), and environmental and physical evaluation parameters, respectively. The total hazard scores for each possible alternative are summarized in Table 5.8.

When interpreting the hazard scores, the higher the score, the greater the potential concern. However, while we quantitatively scored hazards of 6PPD and possible alternatives, it should be stressed that the hazard scores should be treated as approximations of hazards (*i.e.*, ballparks) because of the underlying uncertainties. A score of 100 would be considered less toxic than a score of 400, however, a score of 275 should be considered more or less the same compared to a score of 300.

6PPD and all 43 possible alternatives exhibited a lack of global warming or ozone depletion potential, contribution to smog formation, and flammability. That is, all 43 possible alternatives were essentially equal to 6PPD with respect to physical hazards and received scores of 0. Accordingly, total hazard scores were only affected by the human health and environmental hazard parameters. For example, 6PPD received a total hazard score of 275 based on scores of 125, 150, and 0 for human health, environmental, and physical hazards, respectively.

Human health and environmental hazard data were available for 19 of the 43 possible alternatives. Total hazard scores for these 19 alternatives ranged from 40 to 325. For a number of alternatives, no hazard data were available to develop a hazard score (*i.e.*, rambutan peel extract, amine functionalized lignin, DNPDA). The only possible alternative that received a total hazard score higher than 6PPD was 6QDI, a non-PPD possible alternative, with a score of 325. 6QDI was concluded to have many data gaps and relied on surrogate data from 6PPD for most hazard endpoints, and thus the higher hazard score for 6QDI was due to data gap penalties (*i.e.*, penalties of 25 each for data gaps for carcinogenicity and germ cell mutagenicity). Therefore, 6QDI may not necessarily need to be excluded from consideration as a possible alternative until relevant hazard endpoints are experimentally determined for 6QDI itself (rather than reliance on 6PPD as a surrogate).

Graphene (CAS 1034343-98-0), it is an engineered nanomaterial made completely from carbon. Not all graphenes are the same. The form of graphene evaluated in this report, due to positive performance, is a specialized form of graphene nanoplatelets. Even within this category of graphene, there could be differences (e.g., size, number of layers, surface area, surface chemistry) that could contribute to differences in toxicity (Fadeel et al., 2018; Achawi, et al., 2021). Additionally, from a worker safety perspective, graphene nanomaterials must be handled with proper engineering controls and PPE to limit inhalation exposure (The Graphene Council and Barkan, 2023). The same increased level of engineering controls and PPE would be required for carbon nanotubes as well. The form of graphene (CAS 1034343-98-0) reported in the toxicology studies of the ECHA dossier (reported in Tables 5.1, 5.2, 5.3, and 5.8 of this report) consists of sets of graphene nanoplatelets (The Graphene Council and Barkan, 2023; ECHA, 2024). It is not known if the toxicology studies cited in the ECHA dossier would apply to the form of graphene under study in this AA based on size and shape. However, a patent on one form of this material, Prophene<sup>TM</sup> by Akron Polymer Solutions, described three commercial grades of Prophene<sup>TM</sup>, which are various sizes and shapes of graphene without other modification<sup>17</sup> (Paschall *et al.*, 2023). Note, the manufacturer of Prophene<sup>TM</sup>, Akron Polymer Solutions, is not listed as one of the joint registrants of graphene in the ECHA dossier (ECHA, 2024).

No hazard information were identified for specialized carbon nanotube mixtures. While there are ECHA dossiers for various forms of carbon nanotubes (*i.e.*, branched and cross-linked multi-walled carbon nanotubes EC number 951-407-3, no CAS number; single wall carbon nanotubes EC number 943-098-9, no CAS number), the hazard information reported in these dossiers was not used to inform the hazards of specialized carbon nanotube mixtures. One example of specialized carbon nanotube mixture is MOLECULAR REBAR<sup>®</sup> carbon nanotubes. According to a patent by Molecular Rebar Design on mixtures

<sup>&</sup>lt;sup>17</sup> Grade PS50 with particle, sheet, or plate sizes of 50 nm to 5 microns; Grade PS100 with sheet or plate sizes 100 nm to 5 microns and increasing conductivity; and Grade PS150 with sheet or plate sizes of 150 nm to 10 microns.

of discrete carbon nanotubes (Peddini, 2021), the invention in question is dried liquid concentrates of carbon nanotubes and various rubber processing oils (*i.e.*, tris (2-ethylhexyl) trimellitate oil, TP-95, HyPrene 100 naphthenic oil, castor oil, carnauba wax, curing co-agents, and/or sebacates), in which carbon nanotubes may make up anywhere from 5 to 50% of the dried liquid concentrate. Furthermore, the carbon nanotubes described in the patent may be single, double, or multi-wall and they may or may not be oxidized on the interior and/or exterior surface. Additionally, according to a 2016 SDS from Molecular Rebar Design on MR 1420X DLC (naphthenic oil-MR dry liquid concentrate), this product is 15-20% of functionalized multiwalled carbon nanotubes (CAS # not yet assigned) and 75-85% of naphthenic oil CAS # 64742-52-5 (Molecular Rebar Design, 2015). It seems likely that the MOLECULAR REBAR<sup>®</sup> carbon nanotubes with the performance data (as described later in Table 5.13) is a version of the dried liquid concentrate described in the above-mentioned patent or SDS. Since carbon nanotubes may make up anywhere from 5 to 50% of the dried being potentially toxic rubber processing oils (*e.g.*, naphthenic oil), the ECHA dossier hazard profiles of pure carbon nanotubes were not used to inform the hazards of specialized carbon nanotube mixtures.

## 5.1.3.2 Group B Human Health Hazard Endpoints

Table 5.2 summarizes results for each possible alternative with respect to Group B human health hazard endpoints (*i.e.*, those that do not have corresponding GHS hazard endpoints such as ototoxicity or cardiovascular toxicity). As discussed, quantitative hazard scoring was not performed for Group B hazard endpoints and these endpoints are subsumed in the larger category of "systemic target organ toxicity" addressed in ECHA dossiers and GreenScreen assessments. Nevertheless, we qualitatively summarized Group B hazard information for each possible alternative in accordance with SCP guidelines.

Liver effects (*e.g.*, changes in liver organ weight and changes in liver functions) and hematological changes (*e.g.*, macrocystic anemia) were observed for 6PPD and some of the possible alternatives. However, the effects were either considered adaptive and/or not clinically significant by the respective ECHA dossier registrants (ECHA, 2024). Kidney effects, such as increased kidney organ weight and histopathological changes, were found for some of the possible alternatives; however, none of the respective ECHA dossier registrants considered the effects to classifiable either (ECHA, 2024). In addition, many of the possible alternatives and 6PPD are dermal sensitizers, but this information was already captured under Group A endpoints.

### 5.1.3.3 Salmonid Acute Toxicity – Parent Chemicals

Table 5.4 summarizes salmonid acute toxicity data (*i.e.*,  $LC_{50}$ ) for the possible alternative parent chemical. As mentioned, we identified  $LC_{50}$ s for the parent chemicals themselves, and for their quinone products when appropriate and available. This section discusses results for the parent chemicals. For 6PPD, the lowest reported  $LC_{50}$  in coho salmon was 250 µg/L in juveniles exposed for 24 hrs (Tian *et al.*, 2021), however, the lowest reported  $LC_{50}$  in a salmonid was 140 µg/L in rainbow trout exposed for 96 hrs (Monsanto Co., 1977, as cited in the EcoTox database).

Eight of the 43 possible alternatives (as the parent chemicals) had  $LC_{50}$  data available in a salmonid. The eight possible alternatives were four PPD derived chemicals (77PD, commercial DTPD, CCPD, and DAPD) and four non-PPD chemicals (6QDI, NBC, ethoxyquin, and N-Phenyl-1-naphthylamine). The lowest reported  $LC_{50}$ s of the nine possible alternatives ranged from 24 to >100,000 µg/L. The  $LC_{50}$ s were all determined in rainbow trout, with the exception of 77PD for which an  $LC_{50}$  of 24 µg/L was determined in juvenile coho salmon exposed for 96 hrs (Chapelet *et al.*, 2023). Two of the eight possible alternatives had LC<sub>50</sub>s lower than 6PPD: 77PD and CCPD, for which LC<sub>50</sub>s of 24 and 130 µg/L were reported after 96 hr exposures. As discussed, for 77PD the LC<sub>50</sub> data was reported in coho salmon (Chapelet *et al.*, 2023). For CCPD, the available LC<sub>50</sub> value was based on the LC<sub>50</sub> for 44PD as a surrogate (*i.e.*, 130 µg/L as reported by Dionne [1995 as cited in ECHA, 2024]). In contrast, the remaining six possible alternatives had LC<sub>50</sub>s higher than 6PPD, ranging from 440 to >100,000 µg/L. Two of these six chemicals had LC<sub>50</sub>s at least two orders of magnitude greater than that of 6PPD, the non-PPD chemicals ethoxyquin and NBC for which LC<sub>50</sub>s of 18,000 and >100,000 µg/L, respectively, were reported in rainbow trout after 96 hr exposures.

Overall, among the 43 possible alternatives,  $LC_{50}$  data in coho salmon was only available for 77PD. As mentioned, all other  $LC_{50}$ s were reported in rainbow trout, and although the majority were on the same order of magnitude as 6PPD, two non-PPD chemicals had  $LC_{50}$ s at least two orders of magnitude greater than that of 6PPD.

Ultimately, as discussed, changes to individual tire chemical components occur during manufacturing (Bebb, 1976). Thus, any potential toxicity hazards for individual tire components (to salmonids or otherwsie) cannot be directly extrapolated to potential hazards or risk associated with their presence in a final vehicle tire product.

## 5.1.3.4 Salmonid Acute Toxicity – Quinone Products

Table 5.4 summarizes the lowest reported salmonid acute toxicity data (*i.e.*,  $LC_{50}$ ) for 6PPDQ and quinone products of the potential PPD derived alternatives. For 6PPDQ, the lowest reported  $LC_{50}$  in coho salmon was 0.041 µg/L as measured in juveniles (alevins) exposed for 24 hrs (Lo *et al.*, 2023). In addition, Nair *et al.* (2023) reported an  $LC_{50}$  of 0.64 µg/L in juvenile rainbow trout exposed to 6PPDQ for 96 hrs. By contrast,  $LC_{50}$  values reported in all other Oncorhynchus species were three to four orders of magnitude greater than in coho salmon, ranging from >1.3-2.4 ug/L in juvenile chum salmon (McIntyre *et al.*, 2021) to 67 ug/L in juvenile Chinook salmon (Lo *et al.*, 2023).

For potential PPD derived alternatives,  $LC_{50}$  data for a quinone product was only available for five possible alternatives, 77PDQ, CPPDQ, DPPDQ, DTPDQ, and IPPDQ. Chapelet *et al.* (2023) reported an  $LC_{50} > 226 \mu g/L$  in juvenile coho salmon exposed to 77PDQ for 96 hrs. In addition, Nair *et al.* (2023) reported  $LC_{50}s > 50 \mu g/L$  (the highest tested concentrations) in juvenile rainbow exposed to 77PDQ, CPPDQ, DPPDQ, DPPDQ, DTPDQ, and IPPDQ for 96 hrs.

Overall, data regarding potential acute toxicity of the quinone products of the potential PPD derived alternatives is limited only to two studies involving 77PDQ (Chapelet *et al.*, 2023; Nair *et al.*, 2023), and CPPDQ, DPPDQ, DTPDQ, and IPPDQ (Nair *et al.*, 2023). These studies suggest there may be lower acute toxicity of 77PDQ, CPPDQ, DPPDQ, DTPDQ, and IPPDQ relative to 6PPDQ, however, these results will require validation by other laboratories and further studies. Hence, there is inadequate evidence to assess the potential acute toxicity hazard of quinone products of the potential PPD derived alternatives in salmonids.

### 5.1.3.5 USGS Predecisional Summary

Rubber samples containing 6PPD or one of three potential alternatives (77PD, CCPD, or DPPD) were used to address potential toxicity using cell Pacific cell-line toxicity assays established in Greer *et al.* (2023a). Studies conducted by USGS demonstrated the utility of cell-line-based approaches for initial screening of potential 6PPD alternatives. Ozonation of rubber containing 6PPD led to significant toxicity for coho cells in comparison with the Chinook salmon line. All rubber samples reacted with ozone resulting in production

of quinone transformation products. Representatives of all three classes of PPDs (77PD, CCPD, or DPPD) and their quinones showed differential toxicity relative to 6PPD, providing a preliminary indication that not all PPDs pose the same degree of hazard to coho as 6PPD.

Although preliminary, the Consortium believes this supports the consideration of other PPDs as part of the Stage 2 AA. As noted above, the USGS work is being conducted under a CRADA, and under the terms of that agreement, the results cannot be publicly released prior to publication by USGS. A confidential version of the results is being provided to DTSC separately as confidential business information. It is hoped that publication will allow for public release of the information during Stage 2.

## 5.1.3.6 Hazards of environmental degradation products

As shown in Table 5.10, ECHA dossier hazard data were available for 6PPDQ and four of the 11 potential hydrolysis products identified: 1) p-benzoquinone, 2) p-hydroquinone, 3) aniline, and 4) 1-methyl-propylamine. None of the potential breakdown products are listed on the EU PBT list. In addition, of the potential breakdown products, only aniline is present on the California Proposition 65. All of the identified breakdown products are present on the California TAC list, except for 6PPDQ and 1-methyl-propylamine.

All 4 of the hydrolysis breakdown products mentioned above are classified under GHS as Category 1 for acute aquatic toxicity, however, the classifications are not necessarily based on evidence in salmonids. Two of the breakdown products (p-hydroquinone and aniline) were also classified as Category 2 for carcinogenicity. A number of the breakdown products are also classified as skin sensitizers.

Overall, several of the possible PPD based alternatives (*i.e.*, 7PPD, 77PD, and 6QDI) have breakdown products in common with 6PPD (*i.e.*, aniline and p-benzoquinone) and thus have the potential for similar health hazards (*i.e.*, shaded orange in Table 5.10). For the majority of possible alternatives, breakdown products were not described in the ECHA dossiers and no conclusions about their potential hazards can be reached. This was notably the case for many of the possible non-PPD based alternatives (*e.g.*, NBC, DLTP, ethoxyquin, graphene). This constitutes an important source of uncertainty in the AA. Further research into the potential breakdown products of these possible alternatives is required.

It should also be noted that any chemical-hazards summarized above based on studies of pure chemical do not necessarily reflect actual hazards or risks associated with vehicle tires because the extent to which the possible alternative antidegradant will be released from the TRWP and be subject to breakdown is currently unknown.

# 5.2 Performance

As noted earlier, the Consortium members chose to include performance as a consideration in assessing alternatives in Stage 1. As discussed previously, demonstrating an alternative anti-degradant formulation is safe and effective in a tire is a lengthy process and it makes sense to focus the evaluation and available resources on the promising candidates as soon as practical. With 43 possible alternatives under consideration in Stage 1, evaluating what is known about their performance is a reasonable consideration for prioritization.

We investigated whether there were any available models to predict the behavior of antidegradants in tires (e.g., migration, ozone protection). Several models of migration of chemicals in polymers were identified (e.g., US EPA, 1990) but these were concerned with migration in simple polymers (e.g., polypropylene) rather than the complex matrix of rubber with all of the various additives and layers. These were not found

to be useful for assessing migration of antidegradants through tire rubber. A recently published paper discusses modeling to predict the interactions of PPDs with ozone (Rossomme *et al.*, 2023). The authors' conclusions were that the ability to form a quinone was essential to the effectiveness of the PPD as an antiozonant. While useful for understanding PPD reaction chemistry, this model was not considered useful for evaluating non-PPD alternatives and has also not been verified using actual testing. Overall, we did not locate any models which could help inform the possible properties of an antidegradant in tires.

Lacking suitable models, we relied on the experience and knowledge of a working group of technical advisors, composed of 11 tire industry representatives with many decades of experience in tire development and manufacturing. In the working group member's combined experience, the best antiozonant has the para phenylene diamine structure (an aromatic ring with two nitrogens in the para positions) which allows for formation of a quinone. The value of the diamines in the para position has long been recognized by others (*e.g.*, Cox, 1958). In addition, antiozonants that have been successfully used in rubber in the past contain at least one nitrogen—for example, ethoxyquin. Nitrones (an N-oxide of an imine) are another class of chemicals that show some promise based on published studies. Prediction of reactivity against ozone is difficult, however, and members primarily relied on published screening level studies as described below. In terms of potential migration of antidegradants in rubber, factors working group members considered were the size and polarity of the molecule, the potential interaction with the filler (silica or carbon black), the polymer type, and the operating temperature (see Choi, 2001; PPG Industries *et al.*, 2020). Unfortunately, with multiple factors affecting migration, there are no clear "yes/no" criteria that can be specified.

To organize performance information available for the possible alternatives we considered three groupings: information available from studies published pre-2020 (the date the Tian *et al.* 2021 study was published on-line); data from testing on a select set of PPD alternatives conducted by Flexsys in conjunction with the USGS toxicity studies; and data from performance studies published post 2020. Each of these is discussed below.

### 5.2.1 Performance Data from Studies Pre-2020

Table 5.11 summarizes data for 29 possible alternatives that were published in patents, journal publications or other sources prior to 2020. This table lists the class of compound, the chemical name and CAS registry number, the results and interpretation of the screening-level performance test. The references for where the test information can be found are also included.

Preliminary performance was assigned to one of four categories:

- Some promising ozone data (colored green). These possible alternatives had positive data regarding ozone performance in screening level tests (*i.e.*, bench scale testing, not in finished tires).
- Limited ozone data (also colored green). These possible alternatives had some positive data for ozone performance in screening level tests but there were some concerns about the study, often a lack of an appropriate control or a lack of information about controls.
- **Insufficient data, no ozone data (colored yellow).** The chemical lacks any data related to performance as an antiozonant. Note that a chemical having positive data as an antioxidant does not indicate the chemical can perform as an antiozonant (Akrochem, 2010).

• **Poor ozone performance (colored red)**. For these possible alternatives, the available data indicate the chemical does not perform as an antiozonant or performs poorly compared to 6PPD.

Those possible alternatives that have either some promising ozone data or limited ozone data are eligible to be included in the Stage 2 AA if they also have hazard information.

Note that in a few cases, literature sources indicated promising ozone performance, but Consortium members had actually tested the alternative and found it did not perform well against ozone. Consortium members conducted these tests in rubber compounds representative of those commonly used in tires and under laboratory conditions used to validate rubber compounds for use in tires, including critical aging tests that were not part of the literature sources. In such cases, the member experience was considered more definitive, and the materials were not considered to perform well against ozone.

In some cases, performance testing was pertained to static ozone resistance rather than dynamic ozone resistance. While not definitive, a chemical that is effective in static ozone testing may not perform well in a dynamic ozone test. Chemicals that have good dynamic ozone performance are assumed to have good static ozone performance.

Another potential concern is the migration potential of the alternative in tire rubber. As noted earlier, additives that migrate too slow or too fast relative to 6PPD will not provide stable ozone protection. However, no possible alternatives were eliminated solely based on concerns about migration potential.

It should also be noted that in these older studies, not all comparisons of performance were made against 6PPD. For example, some were compared to ethoxyquin or other PPDs, and in such cases, Consortium technical experts inferred whether this indicated similar or better performance compared to 6PPD was possible (generally we leaned towards including rather than excluding such compounds from further study). In addition, because the data comes from different sources, testing conditions may not be consistent across the chemicals evaluated, making interpretation challenging.

Class of Compound	Chemicals	CAS	Performance Test Results	Interpretation of Performance Data	Reference
Phenylene Diamine	N-(1,4-dimethylpentyl)-N'- phenyl-p-phenylenediamine (7PPD)	3081-01-4	Comparable to 6PPD in Tier 2 type dynamic ozone test in SBR	Some promising ozone data	G. Wilder, US 3,839,275 "Preserving rubber with N-(1,4 dimethylamyl) - N'-para-phenylenediamine"
Phenylene Diamine	N-isopropyl-N'-phenyl-p- phenylenediamine (IPPD)	101-72-4	Comparable to 6PPD in Tier 2 type dynamic ozone test in SBR	Some promising ozone data	G. Wilder, US 3,839,275 "Preserving rubber with N-(1,4 dimethylamyl) - N'-para-phenylenediamine"
Phenylene Diamine	N-cyclohexyl-N'-phenyl-p- phenylenediamine (CPPD)	101-87-1	5% improvement over "commercial control" in Tier 2 type dynamic ozone test in NR; 17% improvement in SBR. Control not identified	Limited ozone data	US 3,511,805, M. Kosmin <i>et al.,</i> "Rubber preserved with alicyclicmethyl phenylenediamines"
Phenylene Diamine related	N-1,3-dimethylbutyl-N'- phenyl quinone diimine (6QDI)	52870-46-9	Shows approximately the same stabilizing effect as 6PPD in outdoor aging studies - Tier 2 testing but ozone level not provided	Poor ozone performance – Produces 6PPD when mixed with rubber	F. Ignatz-Hoover <i>et al.,</i> "Chemical additives migration in rubber" Rubber Chemistry and Technology (2003) 76 (3): 747–768
Dihydroquinoline	Polymerized 2,2,4-trimethyl- 1,2-dihydroquinoline (TMQ)	26780-96-1	60% of activity of ethoxyquin in Tier 2 type testing	Poor ozone performance, not as active as ethoxyquin which is much less active than 6PPD	H. Kilbourne, "Chemical inhibition of ozone degradation of SBR", Rubber Chemistry and Technology (1959) 32 (4): 1155–1163
Diphenyl amine	4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	15% better ozone resistance than ethoxyquin and 15% better than DTPD in a Tier 1 type test	Limited ozone data. However, migration rate is unsuitable based on expert judgment	H. W. Kilbourne, <i>et al.</i> "Chemical inhibition of ozone degradation of SBR", Rubber Chemistry and Technology, Vol. 32, p. 1155 (1959).
Phenylene diamine	N,N'-Ditolyl-p- phenylenediamine (Commercial DTPD)	68953-84-4	Equivalent to Ethoxyquin in a Tier 1 type test	Poor ozone performance based on 2023 Flexsys study (Table 5.12)	H. W. Kilbourne, <i>et al.</i> "Chemical inhibition of ozone degradation of SBR", Rubber Chemistry and Technology, Vol. 32, p. 1155 (1959).
Phenylene diamine	N,N'-Di-2-naphthyl-p- phenylenediamine (DNPDA)	93-46-9	No data - Listed only as an antioxidant in two separate articles. No ozone data.	Insufficient data, no ozone data	J. Amberlang, <i>et al.</i> , "Antioxidants and Antiozonants for General Purpose Elastomers" Rubber Chemistry and Technology (1963) 36 (5): 1497–1541.

## Table 5.11 Performance Data on Possible Alternatives From Sources Prior to 2020

Class of Compound	Chemicals	CAS	Performance Test Results	Interpretation of Performance Data	Reference
Metal dithiocarbamate	Nickel dibutyldithiocarbamate (NBC)	13927-77-0	Shown to be an antiozonant but cures too fast, which affects compound processability	Poor ozone performance	C. Pinazzi <i>et al.</i> , "Protection of natural rubber against atmospheric agents. I. The effects of nickel dibutyldithiocarbamate alone and in combination with protective agents. "Rubber Chemistry and Technology (1955) 28 (2): 438–456
Dihydroquinoline	Ethoxyquin	91-53-2	Early antiozonant used in tires but not as effective as CCPD	Poor ozone performance	H. W. Kilbourne, <i>et al.</i> "Chemical inhibition of ozone degradation of SBR", Rubber Chemistry and Technology, Vol. 32, p. 1155 (1959).
Sulfur compound	Dilauryl thiodipropionate	123-28-4	No Data. No reference to the material as an antiozonant	Insufficient data. No ozone data and not expected to be an antiozonant based on structural properties	Not available
Phenylene Diamine	4-N-(2,3-dimethylphenyl)-1- N-phenylbenzene-1,4- diamine- R1 and R2 are methyl	No CAS	No Data. No reference to the material as an antiozonant	Insufficient data, no ozone data	Not available
Phenylene Diamine	N' -Phenyl-N-Fluorenyl-Para- Phenylenediamine	No CAS	Shown to be equivalent to 77PD in SBR/NR in static ozone testing	Some promising ozone data	J. Hunt, US 3,625,913 "N'-Alkyl and N'-Aryl-N-Fluorenyl-p- phenylene=diamines as antiozonants in natural and synthetic diene rubbers"
Phenylene Diamine	N-(p- phenylthiomethylphenyl)-N'- (1,3-dimethylbutyl)-p- phenylenediamine	No CAS	Material provided good antiozonant protection but is expected to be too slow to migrate for long term protection based on expert judgment	Limited ozone data – but is expected to be too slow to migrate for long term protection based on expert judgment	J. Kuczkowski, US 4,124,565, "N,N' - DISUBSTITUTED-P- PHENYLENEDIAMINES"

Class of Compound	Chemicals	CAS	Performance Test Results	Interpretation of Performance Data	Reference
Phenylene Diamine	4-(2,5-dimethyl-1H-pyrrol-1- yl)-N-phenylaniline	No CAS	Original physical properties are equal to that of 6PPD and ozone testing shows excellent performance	Poor ozone performance – Although this specific patent shows positive data, several members have tested the molecule and found that the protection is insufficient	Youngju Kim (김영주), KR20090100673A, "Tire sidewall rubber composition" Patent
Phenylene Diamine	N,N - (ethane-1,2-diyl ) bis (N-phenylbenzene-1 4- diamine [example chemical from patent]	No CAS	Tier 1 testing showed antiozonant activity is as good as 6PPD and Tier 2 testing showed better antifatigue activity	Limited ozone data – but molecular weight of compounds probably too high to effectively migrate based on expert judgment	M. Boone <i>et al.,</i> US Patent US 10,428,009 B2 METHODS OF MAKING COMPOUNDS AND MIXTURES HAVING ANTIDEGRADANT AND ANTIFATIGUE EFFICACY 2019
Phenylene Diamine (Kruger)	RU997, Irgazone 997 (Reaction product of N- phenyl-N'-(1,3- dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether)	444992-04-5	No ozone data. Shown to migrate much slower than 6PPD	Insufficient data, no ozone data and shown to migrate much slower than 6PPD	R. H. Kruger, C. Boissiere, K. Klein- Hartwig & H. J. Kretzschmar (2005) "New phenylenediamine antiozonants for commodities based on natural and synthetic rubber", Food Additives and Contaminants, 22:10, 968-974
Dihydroquinoline	N,N-diethyl-2,2,4-trimethyl- 1H-quinolin-6-amine (R= N(C2H5)2	No CAS	Shown to be a better antiozonant than ethoxyquin in Tier 2 lab tests	Some promising ozone data, but heavy staining may be an issue based on expert judgment	D. Beaver, et al. US Patent 2,713,047 6-diethylamino-1,2- dihydroquinolines
Hindered amine	N,N'-Dibenzyl-p-xylene- alpha,alpha'-diamine-	25790-41-4	Approximately 400% better than IPPD in time to cracking in NR and SBR compounds-Tier 2 testing but non- black compound	Some promising ozone data	E. Masatomo, <i>et al.</i> , US 3,634,316 "Sulfur vulcanizable natural and synthetic rubbery polymers containing xylene diamines as antiozonants"

Class of Compound	Chemicals	CAS	Performance Test Results	Interpretation of Performance Data	Reference
Triazine	2,4,6-tris-(N-1,4- dimethylpentyl-para- phenylenediamino)-1,3,5 triazine (Durazone 37 or TAPDT)	121246-28-4	Good solubility in NR but limited solubility in BR and SBR. Works as antiozonant at low levels in sidewall with phenolic resin as well as Durazone. No comparison to 6PPD. In EPDM sidewall ozone protection is equivalent to sample with no antiozonant. Several of the members tested this compound and it had poor performance.	Poor ozone performance	M. Pender, US 8,329,788 B2 "Tire having enhanced ozone resistance"
Phenylnaphthyl amines	N-Phenyl-1-naphthylamine	90-30-2	Listed as an antioxidant – poor ozone performance in chloroprene	Poor ozone performance	R. Murray, Factors Influencing the Ozone Resistance of Neoprene Vulcanizates under Flexure, RCT (1959) 32 (4):1117
Phenylnaphthyl amines	N-Phenyl-2-naphthylamine	135-88-6	Listed only as an antioxidant; no information regarding antiozonant potential	Insufficient data, no ozone data	Ambelang, J. C., <i>et al.</i> "Antioxidants and antiozonants for general purpose elastomers." Rubber Chemistry and Technology 36.5 (1963): 1497-1541.
Phenol	[2-Methyl-4,6- bis((octylthio)methyl)phenol (Irganox 1520)	110553-27-0	Samples containing very high levels of phenolic compound had good dynamic ozone performance in sidewall. At these levels, probably causes oxidation based on expert judgment.	Limited ozone data	D Dall'abaco, V. Formaggio, <i>et al.,</i> WO 2018/163041 A2, "TYRE FOR VEHICLE WHEELS "
Hydrazine	1,1' -Pentamethylenebis(2,2- Di-n- Butylhydrazine)	No CAS	Shown to be an antiozonant in dynamic testing of rubber but not compared to conventional antiozonants.	Limited ozone data	H. Stewart, US 3,157,616, Antiozonant rubber compositions containing alkylene bis-hydrazines
Nitrone + Phenolic AO	α- C-4- hydroxy- 3,5- dimethylphenyl - N-isopropyl and Lowinox WSP	Nitrone as a class, no CAS and Lowinox WSP - 77-62-3	Nitrone plus a phenolic antioxidant provided superior static ozone resistance to IPPD.	Some promising ozone data	G. Scott, UK Patent application 2137619 A-1984, "Nitrone compounds and stabilised rubber compositions containing them"

Class of Compound	Chemicals	CAS	Performance Test Results	Interpretation of Performance Data	Reference
Nitrone	α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	No CAS	Compared to IPPD - reasonable ozone performance, some antifatigue activity, synergistic with phenolic antioxidants	Some promising ozone data	G. Scott, L. Nethsinghe, 1984 UK patent application 2137619A for "Nitrone compounds and stabilised rubber compositions containing them"
Gallate related	Octyl gallate	1034-01-1	Compound is an antioxidant. A similar compound, propyl gallate, is used in food applications and is shown to be active against ozone in protecting biological systems, but no data in tires or tire compounding was found.	Limited ozone data	Pauls, K.P. and Thompson, J.E. (1982) Effects of Cytokinins and Antioxidants on the Susceptibility of Membranes to Ozone Damage. Plant and Cell Physiology, 23, 821-832.
Gallate related	Nano calcium carbonate surface modified by gallic acid	No CAS	Using Irganox 1010 as the control, the compound showed improved static ozone resistance. However, 6PPD would be a better control, since Irganox 1010 is not an antiozonant. Additionally, no dynamic ozone data are available.	Some promising ozone data	Poompradub, Sirilux <i>et al.</i> "Improving oxidation stability and mechanical properties of natural rubber vulcanizates filled with calcium carbonate modified by gallic acid." Polymer Bulletin 66 (2011): 965-977.

Notes:

BR = Butadiene Rubber; CAS = Chemical Abstracts Service Number; EPDM = Ethylene Propylene Diene Monomer; NR = Natural Rubber; SBR = Styrene Butadiene Rubber.

## 5.2.2 Performance Testing at Flexsys of Possible Alternatives also Tested by USGS

As noted earlier, several PPDs (*i.e.*, CCPD, 77PD, DTPD, and 7PPD with 6PPD studied for comparison) were formulated into a model sidewall rubber compound, which was extracted and tested by USGS for potential toxicity using cell-based methods. To understand the potential performance of these formulated rubber materials, samples of the rubber were also tested for cure and dynamic ozone performance. Samples were exposed to ozone at concentrations of 10 pphm and 40 pphm for 24, 48, and 96 hours at 15% strain. Another series of samples were exposed to 40 pphm of ozone for 96 hours at 15% strain. Ozone performance was evaluated visually by estimating the size and number of cracks under the different conditions. The results of these tests are shown in Table 5.12. The same scoring system as described in Section 5.2.1 was used, although the yellow/insufficient data category did not apply. Two compounds showed adequate ozone performance; CCPD and 7PPD. Only 7PPD had performance equivalent to 6PPD; CCPD appeared to be less effective at higher ozone concentrations but showed at least some antiozonant activity that warranted further study. CCPD, DPPD and DTPD all failed the dynamic ozone testing (*e.g.*, had many more cracks than 6PPD). Table 5.12 also lists a few earlier studies from Table 5.11 that corroborated the results of the Flexsys testing.

Chemical	CAS	Test Results	Interpretation of Performance Data	References
N,N'-Bis(1,4- dimethylpentyl)-p- phenylenediamine (77PD)	3081-14-9	Both samples failed in max ozone exposure where neither of two 6PPD samples failed	Poor ozone performance	Antiozonant study conducted at Flexsys in 2023
N,N'-diphenyl-p- phenylenediamine (DPPD)	74-31-7	Many more cracks than 6PPD at all ozone exposure conditions	Poor ozone performance	Antiozonant study conducted at Flexsys in 2023
N,N'-Dicyclohexyl- p- phenylenediamine (CCPD)	4175-38-6	Appearance was good at lower ozone exposure but at maximum ozone exposure only one of the two samples survived	Some promising ozone performance	Antiozonant study conducted at Flexsys in 2023
N,N'-Ditolyl-p- phenylenediamine (Commercial DTPD or DAPD)	68953-84-4	Both test samples failed at maximum ozone concentration.	Poor ozone performance	Antiozonant study conducted at Flexsys in 2023
N-(1,4- dimethylpentyl)-N'- phenyl-p- phenylenediamine (7PPD)	3081-01-4	Both test samples looked equivalent to 6PPD at all ozone levels.	Some promising ozone performance	Antiozonant study conducted at Flexsys in 2023

Table 5.12 Performance Testing at Flexsys of Possible Alternative Also Tested by	y USGS
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Notes:

CAS = Chemical Abstracts Service Number; USGS = United States Geological Survey.

### 5.2.3 Recent Performance Data at Other Laboratories (Post-2020)

Table 5.13 lists the performance results for a number of chemicals with published performance data since 2020. The same scoring system as described in Section 5.2.1 was used to evaluate the results. Several of the chemicals listed were tested with clear comparisons to 6PPD in terms of ozone protection, although again these may not involve tire compounds but other types of rubber where 6PPD is used (*e.g.*, nitrile

rubber). Nine chemicals were found to have at least some promising performance data indicating they would be appropriate for further study. The others were antioxidants but lacked data indicating antiozonant performance.

Class of Compound	Chemical	CAS	Results from Non- standard Tests/Calculation	Interpretation of Performance Data	References
Phenylene Diamine	N,N - (ethane- 1,2-diyl) bis (N- phenylbenzene- 1 4-diamine [example chemical from patent]	No CAS	Material is more effective in ozone protection of liquid nitrile rubber than 6PPD	Some promising ozone data	M. Boone <i>et al.</i> , EP3394028, "Compounds with antidegradant and antifatigue efficacy and compositions including said compounds"
Phenylene Diamine	4-[4-(4- Methylpentan-2- ylamino)anilino] phenol	No CAS	Material has equivalent ozone protection to 6PPD in natural rubber/carbon black compounds	Some promising ozone data	X. Yang, WO 2022/146441, "Rubber composition with longer lasting antiozonation"
Inorganic	Specialized graphene	1034343 -98-0	In sidewall compounds, it may be possible to reduce 6PPD if graphene is added to the rubber compound. However, Consortium members noted migration and diffusion across other tire components would need to be considered in assessing potential impacts.	Some promising ozone data	Doug Paschall <i>et al.,</i> "Tire Compounding with Prophene (sidewall)" Paper presented at Rubber Division Technical Meeting April 2022
Phenothiazine	N-(4- methylpentan-2- yl)-10H- phenothiazin-3- amine	No CAS	No ozone data but based on calculations the authors predict good ozone performance. It is an effective antioxidant	Some promising ozone data	C. Recker et al.,WO202206900 1A1, "Phenothiazine compound, its preparation and use in rubber blends and vehicle tires, as ageing protectant, antioxidant, antiozonant and colorant"

Table 5.13 Performance Data for Possible Alternative from 2020 to Jan 2024

Class of Compound	Chemical	CAS	Results from Non- standard Tests/Calculation	Interpretation of Performance Data	References
Amine	7-(4- Methylpentan-2- ylamino)- 2,3,4,10- tetrahydro-1H- acridin-9-one	No CAS	Compound has better oxidation onset temperature than 6PPD, but no ozone data.	Insufficient data, no ozone data	A. Jacob <i>et al.</i> , "Compound, rubber blend containing the compound, vehicle tire comprising the rubber blend in at least one component, process for producing the compound, and use of the compound as an ageing protectant and/or antiozonant and/or dye", WO 2023001338 A1
Amine	2-Cyclohexyl-N- (4- methylpentan-2- yl)-1H-indol-5- amine	No CAS	Compound has better oxidation onset temperature than 6PPD, but no ozone data.	Insufficient data, no ozone data	A. Jacob <i>et al.</i> , "Compound, rubber blend containing the compound, vehicle tire comprising the rubber blend in at least one component, process for producing the compound, and use of the compound as an ageing protectant and/or antiozonant and/or dye", WO 2023001339 A1
Amine	4-(1H-Indol-2- yl)-N-(4- methylpentan-2- yl)aniline	No CAS	Compound has better oxidation onset temperature than 6PPD, but no ozone data.	Insufficient data, no ozone data	A. Jacob <i>et al.</i> , "Compound, rubber mixture containing the compound, vehicle tire which has at least one component comprising the rubber mixture, process for preparing the compound, and use of the compound as an aging protection agent and/or antiozonant and/or colorant", WO2023001340A1

Phenylene diamine         Representative example from class (4-(4-, (dimethylamino) phenyl)amino) heno)         6358- 22-1         Good static and dynamic ozone resistance in natural rubber black compound.         Some promising ozone data         X Yang and J. Arnold, World Patent WO Ozol2/146441 A1 "Rubber Composition with Longer Lasting Antiozonation"           Polymeric amine ed lignin         Amine functionalized lignin         No CAS         Ozone testing was static, but comparable to 6PPD. Fatigue was similar to 6PPD. Since there is no blooming or reservoir, it is unlikely to provide long term protection         Some promising J. Chung, U. Hwang, J. Kim, N. Kim, J. J. Cho, B. Lee, I. Park, J. Subr, D. Nam, J. Jong, S. Kim, J. Cho, B. Lee, I. Park, J. Subr, D. Nam, J. Jong, S. Kim, J. Cho, B. Lee, I. Park, J. Subr, D. Nam, mantioxidant for rubber compounds" ACS Sustainable Chemistry and Engineering 2023, 11 (6), 2303-2313           Gallate related         Rambutan peel extract         No CAS         Ozone testing showed resistance to 6PPD         Some promising ozone data         Sukatta U, Prospects for rambutan peel extract a natural antioxidant for vulcanized natural antioxidant on the aging properties of vulcanized natural <th>Class of Compound</th> <th>Chemical</th> <th>CAS</th> <th>Results from Non- standard Tests/Calculation</th> <th>Interpretation of Performance Data</th> <th>References</th>	Class of Compound	Chemical	CAS	Results from Non- standard Tests/Calculation	Interpretation of Performance Data	References
class (4-((4- (dimethylamino)) phenyl)amino)) henol)natural rubber black compound.ozone data2022/146441 A1 "Rubber Composition with Longer Lasting Antizoanation"Polymeric amine functionalized ligninAmine functionalized ligninNo CASOzone testing was static, but comparable to 6PPD. Fatigue was similar to 6PPD. Since there is no blooming or reservoir, it is unlikely to provide long term protectionSome promising ozone dataJ. Chung, U. Hwang, J. Kim, N. Kim, J. Nam, J. Jung, S. Kim, J. Cho, B. Lee, I. Park, J. Suhr, D. Nam, "Amine- functionalized lignin as an eco-friendly antioxidant for rubber compounds" ACS Sustainable Chemistry and Engineering 2023, 11 (6), 2303-2313Gallate relatedRambutan peel extractNo CASOzone testing showed comparable crack resistance to 6PPDSome promising ozone dataSustata U, Rugthaworn P, Seangyen W, Tantaterdtam R, Smitthipong W, Chollakup R, Prospects for rambutan peel extract as natural antioxidant on the aging properties of vulcanized natural rubber. SPE Polymers. 2021/219:199-209, https://doi.org/10. L		-				-
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Class of Compound	Chemical	CAS	Results from Non- standard Tests/Calculation	Interpretation of Performance Data	References
Phenylene diamine	N-1- Methylheptyl-N'- phenylenediami ne (8PPD or UOP 688)	15233- 47-3	No data but expected to perform similar to 7PPD. Reported as a commercial antiozonant in the 1970s.	Limited ozone data	A. Jacob, US Patent Application 20230312874 for: "Compound, rubber mixture containing the compound, vehicle tire which has at least one component comprising the rubber mixture, method for producing the compound, and use of the compound as an aging protection agent and/or antioxidant agent and/or antiozonant and/or dye"
Nanoscale carbon material	Specialized carbon nanotube mixture	No CAS	Preliminary static ozone data is positive. It is not known if the compound would work alone as an antiozonant in dynamic ozone tests over a prolonged period. May need to be used in conjunction with another antidegradant. Additionally, material is highly reinforcing and difficult to mix. Major adjustments required for use.	Some promising ozone data	Molecular Rebar Design, May 22, 2024. "Successful results: Molecular Rebar Rubber Compounds Eliminating Need for 6PPD"

Notes:

CAS = Chemical Abstracts Service Number; SPE = Society of Plastics Engineers.

### 5.2.4 Future Testing Required

In the sections above, possible alternatives were identified that produced positive results in initial screening tests of ozone protection. It must be stressed that such tests are the first step in evaluating an alternative and they cannot be assumed to indicate actual effectiveness in a manufactured tire. To place the above results in context, we provide below a discussion of further performance testing that may be required prior to selecting a final alternative.

As mentioned in Section 3.4, all new passenger, light truck, truck and bus, trailer, and motorcycle tires sold in the United States must meet rigorous FMVSS (49 CFR Part 571). However, well before a possible alternative is incorporated into a tire, it must pass many feasibility tests to assess its suitability in a tire.

To assess the feasibility of any new chemical or material in tires, chemical manufacturers, researchers, and tire manufacturers may conduct the Tier 1a screening tests listed in Table 5.14 below. Tire manufacturers may conduct the Tier 1b laboratory-scale screening tests noted in Table 5.14 to evaluate the performance of the candidate chemicals or materials in green (non-cured) and cured rubber compounds. A primary purpose of the Tier 1 laboratory-scale tests is to disqualify chemicals or materials that do not perform as required. Once a candidate chemical or material achieves the required outcome in laboratory tests, a tire manufacturer then typically conducts pilot, or intermediate-scale (Tier 2) tests, involving tens to a few hundred kilograms of rubber compounds. A limited number of test tires may be produced using rubber compounds from the pilot-scale mixing tests and used for preliminary tire testing before resources are committed to conducting manufacturing-scale tests. Tier 2 tire compound testing usually requires several iterations to determine if acceptable properties can be obtained. Only once the material achieves the required outcome in Tier 2 tests, will the new material be evaluated in factory-scale processing trials, followed by tire builds, and finally long-term tire testing (Tier 3). These factory-scale tests are also used as a means to ensure consistent batch-to-batch properties of the rubber compounds, consistent industrial performance.

In order to assess the performance feasibility of 6PPD alternatives in tires, Consortium members would take the same approach as described above. Table 5.14 below is a non-exhaustive listing of tests which may be conducted by chemical manufacturers, researchers, and/or tire producers to screen and ultimately test candidate alternative antidegradants. Tire manufacturers will rely on properties listed in the Tier 1b, 2, and 3 sections of Table 5.14 and may also conduct additional testing, beyond what is listed in this table.

All tires sold in the US are required to comply with the requirements in all applicable FMVSS, so no possible alternatives would advance on to long-term tire testing if Tier 1 and 2 testing results are not favorable. In addition, regardless of how well a 6PPD alternative performs in laboratory-scale and pilot-scale testing, the performance of the alternative in a long-term, field, tire testing is the deciding factor regarding the suitability of the material for safe commercial scale use.

In addition to performance testing, toxicity testing would also be required to demonstrate that a selected alternative does not cause toxicity to coho salmon (or potentially other salmon species) and to support chemical registration in various jurisdictions if such registration has not already been established.

Tiered	Approach	Test Legally Required in the US?	Properties Tested	Description	Test Method
	Laboratory Testing for	<b>No.</b> Chemical manufacturers, researchers, and/or	Reaction with Ozone	Test measures the ability to protect polymer in solution. Quick and easy to run.	Layer, R. 1966. Rubber Chemistry and Technology 39(5):1584-1592
Tier 1a	Tier Screening of Chemicals for	Consortium members may utilize Tier 1a tests to screen chemicals before moving on to Tier 1b tests.	Migration	Tests ability of chemical to migrate to the surface	Ignatz-Hoover, F. 2003. Rubber Chemistry and Technology 76(3):747-768 (and references therein)
			Viscosity (Processability)	Rheological properties of green (uncured) rubber compounds	ASTM D6146
			Cure/Reversion	Speed of vulcanization / indicator of potential for reversion in a cured compound	ASTM D5289 ASTM D2084
	Laboratory (Small-Scale)		Stress-Strain	Mechanical properties of compound	ASTM D412-A
Tion		<b>No.</b> Consortium members utilize	Tear Strength	Ability to remove tire from mold	Die B Tear Strength: ASTM D624
Tier 1b	Testing of Candidate Alternatives in Green and	Tier 1b tests to screen	Ozone: Static	Ozone resistance	ASTM D1149
10	Cured Rubber Compounds	chemicals before moving on	Ozone: Dynamic	Ozone resistance in service	ASTM D1149
	Curea Rubber Compounds	to Tier 2 tests.	Ozone: Intermittent Dynamic/Static	Best overall test – Reflects all states of tire	ISO 1431-2012
			Fatigue to failure	Effect of flexing on compound life	ASTM D4482
			Wire adhesion testing (belt and body ply compounds)	Adhesion of steel reinforcement to rubber compounds	ASTM D2229
			Viscoelastic Properties	Tire performance predictors (traction & rolling resistance)	ASTM D5992

 Table 5.14 Non-Exhaustive List of Performance Testing for Candidate Antidegradant Chemicals or Materials in Tires

Tiered Approach		Legally Required to Pass in the US?	Properties Tested	Description	Testing Method
			Initial evaluation of performance in industrial processes		
	Pilot-Scale Testing of Tire Compounds <sup>1</sup> / Testing of Tires <sup>2</sup>		Green & cured properties listed in Tier 1b		
Tier 2		No, but all manufacturers would need alternatives to demonstrate acceptable performance for all Tier 2 tests before moving onto to Tier 3 tests.	Wear Test (Tread Compounds)	Tread lifetime	Various: LAT 100, ISO 23233:2009; DIN ASTM D5963
			Aged Endurance		Methods set by individual Consortium members
			High Speed Performance	Machine (drum) testing of tires	SAE J1561
			Rolling Resistance		ISO 17025, 28580 SAE J1269, J1270, J2452
			Traction	Wet & dry traction (and	Wet: ASTM F1649
				perhaps snow for some applications)	Dry: ASTM F1650 Snow: ASTM F2493
	Manufacturing-Scale Testing of Tire Compounds <sup>1</sup> / Testing of Tires <sup>2</sup>	Yes	Processability	Consistency of handling of	ASTM D1646
				rubber compound on tire plant	
				equipment	
			All green & cured rubber properties listed in Tier 1b.		
Tier 3			Endurance	Evaluates tire's ability to perform over extended lab test, including low pressure conditions for FMVSS No. 139 tires	FMVSS 119/139 (tire type dependent)
			High Speed Performance	Evaluates tire performance at high test speeds	FMVSS 119/139 (tire type dependent)
			Bead Unseat	Evaluates tire resistance to force applied to sidewall under lab test conditions	FMVSS 139 (passenger/some LT)
			Tire Strength	Evaluates performance of tire under plunger force applied to tread	FMVSS 119/139

Tiered App	proach	Legally Required to Pass in the US?	Properties Tested	Description	Testing Method
			Field performance	Long-term evaluation of tires	Methods set by
				on vehicles in a limited-scale,	individual Consortium
				monitored, evaluation	members

Notes:

6PPD = N-(1,3-Dimethylbutyl)-N'-phenyl-p-phenylenediamine; ASTM = American Society for Testing and Materials; FMVSS = Federal Motor Vehicle Safety Standards; ISO = International Organization for Standardization; US = United States; UTQGS = Uniform Tire Quality Grading Standards.

Sources: FMVSS 139 (https://www.nhtsa.gov/sites/nhtsa.gov/files/tp-139-02.pdf); UTQG (https://www.ecfr.gov/current/title-49/subtitle-B/chapter-V/part-575/subpart-B/section-575.104).

(1) Several of these tests are run on both the original sample and aged samples. There are a number of additional tests that are run on specific compounds. For example, ply-to-ply adhesion strength, wire adhesion to wirecoat compound (wire coverage and pullout force), filler performance on cure (agglomeration), and other tests.

(2) In addition to these tests, each individual company will run proprietary tests to evaluate noise, handling, wet performance, wear, cut/chip resistance, performance on ice, performance in snow, and ride comfort.

# 5.3 Relative Exposure Potential

## 5.3.1 Relative Exposure Potential of 6PPD and Possible Alternatives

For Table 5.9, we consulted experimental, modelled, and estimated data from a variety of sources, including study reports, mainly from ECHA REACH dossiers (ECHA, 2024) and US EPA's EPI Suite software (US EPA, 2019a). In Table 5.9, all experimental values are bolded to differentiate between experimental and modeled or estimated data. Similar to the hazard information, there are many data gaps regarding information on the physical-chemical properties of the possible alternatives' ingredients, particularly those that do not have ECHA REACH dossiers or are polymers, mixtures, or unknown or variable compositions, complex reaction products, and biological materials (UVCBs). Polymers, mixtures and UVCBs cannot be modeled in programs such as EPI Suite, due to a lack of a Simplified Molecular Input Line Entry System (SMILES) and a reliable underlying dataset. In addition, Gradient did not color-code this table, because no color-coding was provided by the various data sources and because it would be difficult to assign relative preferences for many of the relevant factors.

Many of the physical-chemical parameters are not materially relevant given how vehicle tires are used (*e.g.*, melting and boiling point are not relevant because the TRWP are not likely to reach temperatures in which the chemical state of 6PPD in the tire will be altered, and volatility is addressed *via* the vapor pressure parameter). For those endpoints that would be materially relevant for vehicle tires (*e.g.*, log octanol-water partition coefficient [K<sub>ow</sub>], log organic carbon partition coefficient [K<sub>oc</sub>], vapor pressure, and water solubility), we used the following criteria from US EPA's "Interpretive Assistance Document for Assessment of Discrete Organic Chemicals" (US EPA, 2013) for the evaluation of possible alternatives' exposure potential in air, water, soil, sediment, and groundwater *via* soil and sediment:

- Vapor Pressure Estimated by MPBPWIN:
  - $\geq 10^{-4}$  = Chemical mostly in the vapor (gas) phase.
  - $10^{-5}$  to  $10^{-7}$  = Chemical in the vapor and particulate phase.
  - $\leq 10^{-8}$  = Chemical mostly in the solid phase.
  - For chemicals with a vapor pressure  $< 10^{-6}$ , there is low concern for inhalation exposure.
- Water Solubility (mg/L) Estimated by WSKOWWIN:
  - >10,000 = Very soluble.
  - >1,000-10,000 = Soluble.
  - >100-1,000 = Moderate solubility.
  - >0.1-100 = Slightly soluble.
  - <0.1 = Negligible solubility.
- Log K<sub>ow</sub> Estimated by KOWWIN:
  - <1 = Highly soluble in water (hydrophilic).
  - >4 = Not very soluble in water (hydrophobic).
  - >8 = Not readily bioavailable.

• >10 = Not bioavailable – difficult to measure experimentally.

#### Log K<sub>oc</sub> – Estimated by PCKOCWIN:

- >4.5 = Very strong sorption to soil and sediment; negligible migration potential to groundwater.
- 3.5-4.4 = Strong sorption to soil and sediment; negligible to slow migration potential to groundwater.
- 2.5-3.4 = Moderate sorption to soil and sediment; slow migration potential to groundwater.
- 1.5-2.4 = Low sorption to soil and sediment; moderate migration potential to groundwater.
- <1.5 = Negligible sorption to soil and sediment; rapid migration potential to groundwater.

The physical-chemical data in Table 5.9 were examined in the context of the abovementioned US EPA criteria to look for differences among the possible alternatives. The results of the comparison are included in Table 5.9.

For 6PPD, low exposure potential *via* air is expected (based on a value of  $4.93 \times 10^{-6}$  mm Hg at 25°C). Low exposure potential *via* air is also expected for the majority of alternatives. Twenty-four chemicals had vapor pressures at least one order of magnitude lower than 6PPD (ranging from  $1.57 \times 10^{-19}$  to  $5.47 \times 10^{-7}$  mm Hg at 25°C), and eight chemicals were on the same order of magnitude as 6PPD (ranging from  $1.24 \times 10^{-6}$  to  $8.22 \times 10^{-6}$  mm Hg at 25°C). No vapor pressure was found for graphene. However, vapor pressure for graphene would be negligible since graphene is an inorganic with melting point above 4,000°C. In contrast, two of the 11 remaining chemicals may have some air exposure potential due to vapor pressures greater than 0.0001 mm Hg (the non-PPD alternatives NBC and ethoxyquin).

Regarding exposure potential via water for 6PPD, we identified a water solubility of 2.83 mg/L at 25°C and log K<sub>ow</sub> of 4.68 at 20°C, suggesting that 6PPD is relatively insoluble in water and hydrophobic. The majority of possible alternatives are expected to have similar exposure potential via water. In addition, 34 of the 43 possible alternatives had log K<sub>ow</sub> values greater than 3.5 (ranging from 3.66 to 11.9), suggesting they are relatively hydrophobic. Based on water solubility greater than 100 mg/L, only two possible alternatives, the possible non-PPD alternatives 1,1'-pentamethylenebis(2,2-di-n-butylhydrazine) and representative example from class (4-((4-(dimethylamino)phenyl)amino)phenol), are expected to be soluble or moderately soluble in water, respectively. However, all of the possible alternatives have log Kow values greater than 1, suggesting that none of the alternatives may be considered hydrophilic. For example, 1,1'pentamethylenebis(2,2-di-n-butylhydrazine) representative example and from class (4-((4-(dimethylamino)phenyl)amino)phenol) have log K<sub>ow</sub> values of 2.64 and 5.57, respectively.

Regarding exposure potential *via* sediment for 6PPD, we identified a log  $K_{oc}$  of 4.363 suggesting that the candidate chemical may sorb strongly to soil and sediment and have negligible to slow potential for migration to groundwater. All alternatives had log  $K_{oc} \ge 3$  (ranging from 2.992 to 11.407), suggesting at least moderate potential to sorb to soil and sediment and slow migration potential to groundwater. No log  $K_{oc}$  was identified for graphene.

Some of the possible alternatives have substantially less water solubility than 6PPD (*e.g.*, DOPD, DLTP, RU997, and TAPDT) which could affect their environmental partitioning. Similarly, some have

substantially higher log K<sub>ow</sub> values (an indication of partitioning into organic materials) than 6PPD (*e.g.*, DLTP, Ru997/Irgazone 997 blend, TAPDT, and DOPD) which again could result in different environmental behavior. Some also have substantially different vapor pressures (some higher, some lower) which could affect workplace exposures. While this evaluation provided some insight into the ingredient-level exposure potential of the possible alternatives, ideally, we would compare the product-level exposure data, because the ingredients are meant to react and create a structure that is distinctly different from the individual ingredients. Unfortunately, no product-level exposure information is available at this time for any alternative.

#### 5.3.2 Relative Exposure Potential of Potential Breakdown Products

As summarized in Table 5.10, 6PPDQ did not have an ECHA dossier and therefore did not report information regarding physical-chemical parameters of interest (*i.e.*, vapor pressure, water solubility, log K<sub>ow</sub>, and log K<sub>oc</sub>). Data regarding relevant physical-chemical parameters was available for seven of the 11 potential hydrolysis products (4-hydroxydiphenylamine, 1,3-dimethylbutylamine, p-benzoquinone, p-hydroquinone, aniline, 6QDI, and 1-methyl-propylamine). In general, these hydrolysis products may have some exposure potential *via* air and water, but low to moderate exposure potential *via* sediment.

Briefly, based on reported vapor pressures ranging from  $7.5 \times 10^{-6}$  to 0.305 mm Hg at 25°C, the hydrolysis products may have potential to be in the vapor phase and therefore may exhibit some exposure potential *via* air. Based on water solubility ranging from 7.9 to 7,200 mg/L, the hydrolysis products are generally expected to be slightly soluble to soluble in water. Only one of the hydrolysis products, the non-PPD 6QDI (a reported hydrolysis product of 6PPD), has a log K<sub>ow</sub> greater than 4, suggesting relative hydrophobicity. In contrast, log K<sub>ow</sub> of the other hydrolysis products range from 0.1 to 2.82, suggesting relative hydrophilicity. Finally, available log K<sub>oc</sub> values for the hydrolysis products range from 1.57 to 2.6, suggesting low to moderate potential to sorb to soil and sediment, and slow to moderate migration potential to groundwater.

As discussed, while this evaluation provided some insight into the exposure potential of the breakdown products (*via* hydrolysis only) of possible alternatives that could be used as ingredients, ideally, we would compare breakdown products resulting from completed vehicle tire. No such exposure information is available at this time.

## 6.1 Selecting Possible Alternatives to the Priority Product

Using the data gathered related to hazard exposure potential and performance, as discussed in Section 5, we then used the information in aggregate to draw conclusions about whether there were possible alternatives that should be considered in greater depth in a Stage 2 AA. There are various methods for selecting alternatives in an AA, ranging from purely qualitative and narrative approaches to sophisticated approaches such as multi-criteria decision analysis (MCDA) (Beaudrie *et al.*, 2020). Although we used a modified CSI approach to score toxicological hazards (with the exception of group B endpoints and salmonid toxicity), we were unable to develop quantitative scores for relative exposure potential or performance. Relative exposure potential was scored as better, similar, or worse than 6PPD. Performance was scored as whether the available information suggested a chemical might be suitable for further testing (*i.e.*, the data were suggestive of ozone reactivity but not informative of performance. Thus, our approach was both quantitative (with the modified CSI scoring) and qualitative (for group B endpoints, salmonid toxicity, relative exposure potential and performance). Given this combination of quantitative and qualitative information, a flexible narrative approach is the best method for considering the data and reaching a conclusion about a possible alternative's merits.

In our narrative analysis we considered the following:

- Whether the hazard score was preferable, similar, or less preferable than 6PPD. Preferable and less preferable were based on a chemical being 30 percent above or below the 6PPD CSI scores for human health, environmental hazards, and physical hazards. While this information was considered, we did not eliminate any possible alternatives from Stage 2 based on hazard score *alone*, except in the case of complete hazard data gaps. When there was absolutely no hazard data available from the sources we considered, we judged the barriers to obtaining regulatory approval for the chemical would be too high and that moving forward without such data could possibly lead to a regrettable substitution.
- Whether any of the Group B endpoints suggested a form of toxicity not found for 6PPD or more substantial than that reported for 6PPD. We found no convincing evidence of such a difference, so this did not figure in the consideration.
- Whether data on toxicity to salmonids suggested lower toxicity than 6PPD and 6PPDQ. If an alternative had an LC<sub>50</sub> value more than one order of magnitude greater or lower than that reported for 6PPD and 6PPDQ that was considered to be an important difference.
- Whether key data on exposure potential (*e.g.*, water solubility, vapor pressure, log K<sub>ow</sub>, log K<sub>oc</sub>) suggested differential migration in the environment and therefore different exposure potential relative to 6PPD. This was a difficult consideration to assess because various potential exposure pathways are being considered, and a factor that was beneficial for one pathway could be detrimental to another (*e.g.*, reduced water solubility could mean less exposure *via* water but more exposure *via* sediment). While this information was also considered, we did not eliminate possible alternatives from Stage 2 based on relative exposure potential *alone*.

• Whether preliminary data from bench scale tests regarding performance of the chemical as an antidegradant (particularly protection against ozone) suggested the possible alternative to be eligible for inclusion in Stage 2 AA. In addition, when there were concerns about potential for migration in tire rubber, these were noted but were not a factor in ruling out a possible alternative.

Regarding the relevant factors listed in Tables 4.4 and 4.5 and rated as "yes" or "potentially" as to their relevance for reaching an AA conclusion, most of the health hazard and environmental hazard (e.g., aquatic toxicity, effects on aquatic organisms, persistence) factors are already covered by the CSI hazard scoring approach and the salmonid data described above. Among the Group B human health endpoints, there were no differences among alternatives (considering data gaps and study limitations) that appeared to constitute a "material" difference. Among the environmental fate/physical and chemical properties described in Table 4.4, the most critical (e.g., water solubility, vapor pressure, lipid, and carbon solubility) are also covered in the decision-making process, as noted above. Other potentially relevant factors that are physical/chemical properties – such as molecular weight, boiling point, etc. – are related to these critical properties (e.g., chemicals with high boiling points also have high vapor pressures) and so were not explicitly considered separately. Flash point, a factor added by the Consortium members due to concerns about safety during manufacture, was judged to be only potentially relevant for those alternatives that lacked GHS flammability classification (which is based on flashpoint). A number of life cycle stages (e.g., use, waste generation and management, reuse/recycling, and end-of-life) were all scored as potentially relevant; more information about how the possible alternatives would affect these stages of the tire life cycle is needed in order to determine their potential impact.

Overall, the decision to retain a possible alternative for further consideration in Stage 2 was based primarily on a chemical having sufficient hazard data and if the possible alternative also had some promising or limited ozone performance data. Whether a chemical and its quinone transformation product had lower toxicity to salmonids was considered but was not a determinant for exclusion because data exist for so few chemicals. As noted above, relative exposure potential also appeared to not be clearly differentiated among alternatives. Some alternatives appeared more likely to migrate through certain environmental media more than others, but whether this results in a significant difference in risk to all receptors of concern was not clear.

# 6.2 Possible Alternatives to Priority Product to Consider in Stage 2

The results of the Stage 1 AA are summarized in Table 5.15 where existing data on hazard, relative exposure potential, and performance are shown for each possible alternative. It should be noted that performance data are limited to simple screens where such information is available. As detailed earlier in this Stage 1 AA report, much more extensive testing would be required before any actual alternative could be implemented.

Of the 43 possible alternatives considered in the Stage 1 AA, only a few had both hazard data and screening level performance data suggesting a potential to perform in tires as an antiozonant. The chemicals that met these criteria and would be further evaluated in Stage 2 are the following:

**7PPD.** This chemical has similar overall hazard score relative to 6PPD. It also has similar exposure potential. In screening level tests conducted by Flexsys it showed effective performance against ozone.

**IPPD.** This chemical has similar overall hazard score relative to 6PPD (slightly lower environmental hazard). Results of one study suggest the quinone metabolite is less toxic than

6PPDQ in rainbow trout. It is slightly more likely to migrate to water. In a screening level test reported in a 1970 patent, IPPD showed effective performance against ozone; however, more data are required about its long-term protection ability in tires.

**77PD.** This chemical has somewhat lower hazard scores than 6PPD. One study (Chapelet *et al.*, 2023) showed that while the parent chemical is somewhat more toxic to coho salmon (and other species) than 6PPD, the quinone metabolite is far less toxic to coho salmon than 6PPDQ, meaning that the overall impact on coho would likely be less. In addition, based on limited data from cell-based studies conducted by USGS, it appears that the parent chemical and its corresponding quinone may have a slightly improved toxicity profile for coho salmon relative to 6PPD, a finding the requires confirmation. 77PD has similar exposure potential relative to 6PPD. In screening level tests conducted by Flexsys it showed effective performance against ozone.

**CCPD.** This chemical has similar overall hazard to 6PPD and, based on limited data from cellbased studies conducted by USGS, it appears that the parent chemical and its corresponding quinone may have an improved toxicity profile for coho salmon relative to 6PPD; a finding that requires confirmation. It also has similar exposure potential. In screening level tests conducted by Flexsys it showed effective performance against ozone.

**Specialized graphene.** Graphene nanoplatelets (as discussed previously) are graphene-based materials with a surface area not greater than  $180 \text{ m}^2/\text{g}$ , and a carbon content greater than 99% and an oxygen content less than 1%. The lateral particle size of these materials is between 100 nm and 5 µm. This particular material has superior hazard scores relative to 6PPD, although, as noted in Section 5.1.3.1, potential differences in the structure of the graphene tested in performance studies and the form of graphene reported in ECHA dossiers may be important to consider in Stage 2 since differences in size, number of layers, surface area, and/or surface chemistry could contribute to differences in exposure and toxicity (Fadeel *et al.*, 2018; Achawi, *et al.*, 2021). There are no data indicating its toxicity to salmon and while potential toxicity seems unlikely, it should be verified. Graphene is likely to remain part of the rubber matrix, and it is non-volatile and non-water soluble. Since graphene does not migrate in rubber, 6PPD or another antiozonant will need to be used in conjunction with graphene in tires. However, use of graphene could constitute a potential method for reducing 6PPD concentrations without compromising important criteria such as the potential for rubber rework during the tire manufacturing process.

**Octyl gallate.** Octyl gallate can be synthesized from natural sources and/or fossil fuel-based sources. It has a lower hazard score and similar relative exposure potential compared to 6PPD. While no ozone performance was found in tire or tire compounding tests, a similar chemical, propyl gallate has been shown to be an antiozonant in biological systems.

**Irganox 1520.** Irganox 1520 is a phenolic antioxidant. It has a substantially lower hazard score compared to 6PPD. It is less water soluble, has lower vapor pressure and is more carbon- and fat-soluble compared to 6PPD. It has limited but promising performance data with respect to ozone protection.

# 6.3 Alternatives to be Eliminated from Further Consideration

As shown in Table 5.15, 36 of the 43 alternatives evaluated were eliminated from further consideration in Stage 2. Some alternatives were eliminated because they have so many data gaps in terms of toxicological hazard that they could not be confidently evaluated. This was the case for the following 24 alternatives:

- N-cyclohexyl-N'-phenyl-p-phenylenediamine (CPPD)
- 4,4'-Dioctyldiphenylamine (DOPD)
- N,N'-Di-2-naphthyl-p-phenylenediamine (DNPDA)
- N' -Phenyl-N-Fluorenyl-Para-Phenylenediamine
- N-(p-phenylthiomethylphenyl)-N'-(1,3-dimethylbutyl)-p-phenylenediamine
- 4-(2,5-dimethyl-1H-pyrrol-1-yl)-N-phenylaniline
- N,N (ethane-1,2-diyl ) bis (N-phenylbenzene-1 4-diamine [example chemical from patent]
- 4-N-(2,3-dimethylphenyl)-1-N-phenylbenzene-1,4-diamine- R1 and R2 are methyl
- 4-[4-(4-Methylpentan-2-ylamino)anilino]phenol
- Representative example from class (4-((4-(dimethylamino)phenyl)amino)phenol )
- N,N-diethyl-2,2,4-trimethyl-1H-quinolin-6-amine (R= N(C2H5)2
- N,N'-Dibenzyl-p-xylene-alpha,alpha'-diamine-
- 1,1' -Pentamethylenebis(2,2-Di-n- Butylhydrazine)
- α- C-4- hydroxy- 3,5- dimethylphenyl- N-isopropyl combined with 2,2'-Methylenebis[6-(1- methylcyclohexyl)-p-cresol]
- N-(4-methylpentan-2-yl)-10H-phenothiazin-3-amine
- 7-(4-methylpentan-2-ylamino)-2,3,4,10-tetrahydro-1H-acridin-9-one
- 2-cyclohexyl-N-(4-methylpentan-2-yl)-1H-indol-5-amine
- 4-(1H-indol-2-yl)-N-(4-methylpentan-2-yl)aniline
- α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone
- Amine functionalized lignin
- Rambutan peel extract
- N-1-Methylheptyl-N'-phenyl-p-phenylenediamine (8PPD or UOP 688)
- Nano calcium carbonate surface modified by gallic acid
- Specialized carbon nanotube mixture

As part of the Stage 2 AA process, we will revisit these to determine if reliable toxicological data are available. In particular, specialized carbon nanotube mixture, 8PPD, rambutan peel extract, and amine functionalized lignin are all complex mixtures with promising performance (chiefly in patents) but no available toxicity data. If such data are found or become available, they could also be considered in Stage 2.

Some possible alternatives that passed the initial screen described in Section 3 were subsequently eliminated from further evaluation in Stage 2 due to a lack of performance data or because available data indicated they would not perform well against ozone. Note that some of these eliminated chemistries were also dropped due to lack of toxicity data as noted above. These 18 eliminated possible alternatives due to performance data were:

- N,N'-diphenyl-p-phenylenediamine (DPPD)
- N-1,3-dimethyl butyl-N'-phenyl quinone diimine (6QDI)
- Polymerized 2,2,4-trimethyl-1,2-dihydroquinoline (TMQ)
- N,N'-Ditolyl-p-phenylenediamine (Commercial DTPD)
- Diaryl-p-phenylene diamine (DAPD)
- N,N'-Di-2-naphthyl-p-phenylenediamine (DNPDA)
- Nickel dibutyldithiocarbamate (NBC)
- Ethoxyquin
- Dilauryl thiodipropionate
- 4-(2,5-dimethyl-1H-pyrrol-1-yl)-N-phenylaniline

- 4-N-(2,3-dimethylphenyl)-1-N-phenylbenzene-1,4-diamine- R1 and R2 are methyl
- RU997, Irgazone 997 (Reaction product of N-phenyl-N'-(1,3dimethylbutyl)-p-phenylenediamine with an alkyl glycidylthioether)
- 2,4,6-tris-(N-1,4-dimethylpentyl-para-phenylenediamino)-1,3,5 triazine (Durazone 37 or TAPDT)
- N-Phenyl-1-naphthylamine
- N-Phenyl-2-naphthylamine
- 7-(4-methylpentan-2-ylamino)-2,3,4,10-tetrahydro-1H-acridin-9-one
- 2-cyclohexyl-N-(4-methylpentan-2-yl)-1H-indol-5-amine
- 4-(1H-indol-2-yl)-N-(4-methylpentan-2-yl)aniline

No possible alternatives were excluded based on relative exposure potential. Although some chemicals had different properties relevant for environmental mobility compared to 6PPD, it is not clear whether a chemical that is less water soluble but more fat soluble, or less water soluble but more volatile, would be a preferred alternative.

Investigation of possible alternatives to 6PPD is a highly active area of research. As noted above, one of the first tasks for the Stage 2 AA will be to review the recent scientific literature to determine if new data are available which could affect the decision to drop some alternatives from consideration.

# 6.4 Decision Concerning Abridged AA or Stage 2 AA

As noted in Section 6.1, a number of possible alternative antiozonants to replace 6PPD in motor vehicle tires were identified in the Stage 1 AA. Consequently, it is appropriate to carry these alternatives forward into the Stage 2 AA process where they will be examined in further detail consistent with the requirements of the SCP regulations.

# 7 Work Plan For Stage 2 AA

# 7.1 Tasks for Stage 2 AA and Final AA Report

The procedure for completing Stage 2 is outlined in the SCP regulations. The specific tasks we would undertake would be the following:

- Obtain DTSC approval on Stage 1 AA report (including resolving any comments received);
- Confirm that no new candidate alternatives have become available since the time of the Stage 1 submission; if these are available, we will determine their suitability for including in the analysis (*e.g.*, adding them to the Section 5 table to see if they are suitable);
- Update the hazard, performance, and exposure potential information for the 42 already identified possible alternatives based on new study data that becomes available after submission of the Stage 1 AA;
- Re-screen the available alternatives to understand which have sufficient data to support a Stage 2 AA;
- Revisit relevant factors for possible alternatives carried forward, as these could change;
- Perform a more in-depth evaluation of hazard and exposure potential (*e.g.*, looking more deeply into potential transformation products including their likelihood of being formed and their potential to migrate in the environment);
- Revisit the conceptual model to see if it requires revision for the revised set of possible alternatives;
- Update the literature search to be sure the most current information is available on product performance;
- Work with an economist to quantify the economic impacts of the priority product and possible alternatives;
- Use the sequential, simultaneous, or hybrid decision framework to evaluate possible alternatives and come to a decision;
- Prepare Stage 2 AA report;
- Include self-evaluation described in chapter 11 of DTSC's AA Guide (2017);
- Schedule a call with DTSC at the 6-month time point to discuss any issues that arise; and

At the end of the Stage 2 AA, we are optimistic that we will have identified one or more possible alternatives that hold promise to replace or materially reduce 6PPD in motor vehicle tires, subject to future performance testing to ensure comparable tire safety and performance. Additional toxicity testing may need to be performed to satisfy regulatory requirements and to fill important data gaps.

# 7.2 Proposed Stage 2 AA Completion Schedule

The SCP regulation requires an implementation schedule be submitted as part of the Stage 1 report. The following schedule is offered to comply with that requirement. Given the rapidly changing situation with research related to 6PPD alternatives, we reserve the right to adjust and modify this schedule as needs arise. We request a 1-year extension and have built that extension into the schedule to allow for additional preliminary performance testing of candidate alternatives in rubber and potential toxicity testing the Consortium envisions carrying out.

Action Item	Potential Completion Date	
	All times below are after DTSC	
	acceptance of Stage 1	
Update possible alternatives search	Weeks 1 to 8	
Revisit conceptual model		
Initiate more in-depth hazard and exposure factor review		
Develop preliminary performance testing plan (NOT REQUIRED UNDER SCP)		
Develop additional toxicological testing plan (USGS, also NOT REQUIRED UNDER SCP)		
Engage with economist, begin assessment of economic impacts	Weeks 3 to 12	
Meeting with DTSC to discuss issues expected in Stage 2	Week 4 to 6	
Preliminary performance testing begins	Week 8	
Additional tox testing begins	Week 8	
Update performance database, determine if newer data are available	Weeks 8 to 48	
Determine if newer hazard data on identified possible alternatives are available	Weeks 8 to 48	
Revisit relevant factors for Stage 2 in light of reduced possible alternative set	Week 8	
Initial data review/tabulation for hazard, exposure, performance, life cycle, and economics impact phase	Weeks 20 to 30	
Discuss progress/outstanding questions with DTSC	Week 30	
Explore decision frameworks	Weeks 30 to 32	
Preliminary performance testing results available	Week 80	
Evaluate preliminary performance testing results, follow up questions	Week 80	
Incorporate preliminary performance testing results into decision framework	Week 81	

Table 7.1	<b>Proposed Stage</b>	2 AA Completion	Schedule
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Action Item	Potential Completion Date	
Initial decision using appropriate decision framework	Week 82	
Internal review of initial decision, QC by larger group	Week 85	
Prepare final AA report	Weeks 86 to 92	
Report review by Working Group	Weeks 92 to 94	
Report review by full Consortium	Weeks 95 to 98	
Revise final AA report, final edits	Weeks 99 to 103	
Submit final AA report to DTSC	Week 104	

Notes:

AA = Alternatives Analysis; DTSC = Department of Toxic Substances Control; QC = Quality Control; SCP = Safer Consumer Products.. Indented entries address testing that is not required under the SCP AA process.

# 8 Uncertainty Analysis

A number of possible sources of uncertainty were encountered in the course of conducting this Stage 1 AA. The key sources are summarized below.

**Identification of Possible Alternatives.** Alternatives were identified based on patent searches, journal article searches, general Internet searches, and surveys and conversations with Consortium members. It is possible that other alternative formulations exist and were not identified, but this is considered unlikely given that the AA was directed by a large Consortium with great familiarity with the industry and that the searches involved the major sources of information available about tire manufacturing. For any alternative that could have been missed (*e.g.*, an obscure patent, possibly not in English), it is doubtful whether it would have data on chemical composition, hazards, exposure, and performance and yet be unknown to the Consortium. Lacking such data, it would not likely affect the conclusions of the AA. Thus, the conclusions of this Stage 1 AA would not change.

**Evaluation of Relevant Factors.** For some of the relevant factors, the existence of material differences or lack thereof is fairly apparent. For example, it is clear from chemical manufacturing data that HFC or other high potency global warming gases are not used in the manufacturing process. Similarly, we can be fairly confident that while neither 6PPD nor any of the alternatives are listed under Proposition 65, many of the alternatives do contain components in their lifecycle that are listed (*e.g.*, benzene, toluene, nickel). On the other hand, data for particular toxic modalities are lacking for many of the alternatives, many of which do not even have CAS numbers. In addition, while data are available on the physical chemical properties from data sources such as EPI Suite, the exposure potential of the alternatives when formulated as part of a tire may be different from that of the pure chemical.

**Hazard Evaluation.** To evaluate the hazards of the alternatives, we primarily relied on ECHA REACH dossiers and GreenScreens. These two sources sometimes differed in terms of their assignment of particular hazard scores and we typically used the more conservative score in our CSI scoring process. However, GreenScreens were only available for a small subset of alternatives so not all alternatives had the same level of data. Due to the large number of alternatives examined, we also did not conduct an exhaustive literature review on each chemical of interest. Had we done so, we may have uncovered additional hazard data that could conflict with the data in the aforementioned sources or that could fill in data gaps. This more detailed evaluation of health hazard data will be conducted during Stage 2 of the AA process. Moreover, as noted above, the composition information we had on some of the patent-identified alternatives is for example formulations, which may not reflect the composition of any actual commercial product.

**CSI Scoring for Human Health** *versus* **Aquatic Toxicity.** In our CSI scoring system, described in Tables 5.5 to 5.7, we followed the scoring used in prior AAs which weighted human health concerns more heavily than ecological effects. Because DTSC has listed the priority product largely due to its reported effects on certain fish species, acute aquatic toxicity could have been weighted more heavily in the CSI scoring. To explore the impact of this question we conducted a sensitivity analysis and increased the scoring for acute and chronic aquatic toxicity (*e.g.*, the CSI score for acute and chronic GHS category 1 was doubled from 25 to 50, the penalty for data gaps was increased from 10 to 25, *etc.*). Under this revised scheme, the *overall* hazard score for 6PPD increased by 18%. The maximum increase among the possible alternatives was 36 percent for octyl gallate and the minimum increase was 0 percent (*i.e.*, no change) for Irganox 1520, dilauryl thiodipropionate and nickel dibutyldithiocarbamate. These are relatively modest changes and would not alter the final conclusions; octyl gallate and Irganox 1520 would still proceed to Stage 2 and the

other two chemicals either have poor performance against ozone or lack ozone data. Despite the higher aquatic toxicity score, the human health toxicity score for octyl gallate is low enough that it still meets the screening criteria in Table 5.15.

**Evaluating Alternatives for Their Hazards** *versus* **6PPD.** In Table 5.15 we color coded the possible alternatives for their hazards relative to 6PPD. Whether an alternative had a hazard profile more desirable (green) or less desirable (red) than 6PPD was based on a CSI score more than 30 percent different from that of 6PPD. The 30 percent difference was arbitrary, although it is believed to be reasonable given the SCP requirement for alternatives that have a material difference from the priority product. To explore this further, we relaxed the color coding criteria to more or less than 20 percent different. A greater reduction in the evaluation criteria seemed unlikely to meet the requirement for "materially different". Under the less stringent evaluation scheme, several alternatives scored for overall hazard as being better than 6PPD *versus* being scored as similar to 6PPD. These were 77PD, commercial DTPD, DAPD, ethoxyquin, RU997, N-phenyl-1-naphthylamine, and N-phenyl-2-naphthylamine. No other changes in scoring were evident. These altered results would have no effect on the outcome of Stage 1 because 77PD was already selected to go onto stage 2 (it has promising performance data and its quinone appears much less toxic to coho salmon) whereas the others were rejected due to poor performance data (N-phenyl-2-naphthylamine was also dropped due to poor human health scores which also remains the case with the looser criteria).

**Rejection of Alternatives with Extensive Data Gaps for Toxicity.** Although we made reasonable efforts to evaluate the hazards of possible alternatives, for some chemicals the number of data gaps was so extensive that we felt we could not reliably evaluate their hazard in a way that would allow consistent comparison across possible alternatives. Read across approaches could have been applied but these involve some uncertainties. In addition, the fact that these chemicals lack substantial amounts of toxicological data suggests they would require a very substantial testing program before they could be adopted as a replacement for 6PPD, which could add multiple years to adoption of a potential alternative. Finally, many of the chemicals with very limited data for toxicity also lacked data for performance.

**Environmental Transformation Products.** Due to the release of antidegradant into the environment *via* TRWP, the potential for transformation or breakdown of the antidegradant into other chemicals is an important consideration in the AA. Unfortunately, information on potential environmental breakdown products for the possible alternatives is very limited. Aside from studies of 6PPD, we were unable to locate any studies examining the potential breakdown products of possible alternatives from reaction with ozone. For the PPDs, formation of quinones seems likely but the degree to which it occurs and the quinone persists in the environment is unknown. Information on ozone related breakdown products of the non-PPD alternatives was not located. We also could not find any modeling programs that would describe transformation with ozone. More generally, information on environmental breakdown products by other processes (*e.g.*, hydrolysis) was not present in the ECHA dossiers we consulted. A deeper examination using additional sources will be required in Stage 2 for the smaller number of alternatives considered viable based on hazard and initial screens of performance for the parent chemicals.

**Performance.** As discussed in Section 4, evaluation of the performance of an antidegradant in tires will involve a very large battery of tests, ranging from bench scale studies to field tests of manufactured tires placed on vehicles. The whole range of tests is likely to take several years to complete. Consequently, for this Stage 1 AA, we only had preliminary bench scale testing results available for a subset of possible alternatives. It is conceivable that alternatives that performed well in bench scale studies could fail to perform adequately in subsequent, more sophisticated tests. While this might not disqualify that alternative completely (because modifications may be possible to address the issues), it would impact the conclusions of the AA. In addition, the data we had available only covered a few of the alternatives. Although these were the ones that appeared most promising from a chemical structure basis, it is possible that if such data

were available for other alternatives, we could have reached different conclusions. It is hoped that additional data will be available for review in Stage 2.

**Cost.** We did not assess costs of the possible alternatives considered in Stage 1. Data on the bulk prices of specialty chemicals is particularly difficult to obtain. We did examine some chemical supplier websites (*e.g.*, Alibaba) but found that the ranges supplied from different suppliers were so broad that they were both of questionable accuracy and not likely to be useful. We did not consider cost of the possible alternatives in Stage 1. This will be done in Stage 2 where it is hoped that an economist on the team will provide guidance on this issue.

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**Tables and Figures** 

## Table 4.4 Consideration of Potentially Relevant Factors Identified in the SCP Regulations

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
Life Cycle Segments	Raw materials extraction	Unlikely for most alternatives, but data are limited. Potentially for amine treated lignin and rambutan peel extract.	While there are multiple LCAs for tires that characterize resource inputs such as energy, water, and other material requirements (Michelin, 2021; Dong <i>et al.</i> , 2021; Piotrowska <i>et al.</i> , 2019), there are no LCAs which focus specifically on the impacts associated with having 6PPD in tires nor is there an LCA available for any of the possible alternatives if present in a tire. A review of general information on chemical production for the functional ingredients in the products (Table 4.6) suggests that, qualitatively, there are unlikely to be material differences between them, as most involve inputs involving fossil fuels and/or mined materials of various types ( <i>e.g.</i> , metals, calcium carbonate). There are multiple methods for producing graphene and carbon nanotubes but the most common involves deposition of methane (typically fossil fuel derived) onto a substrate. For amine functionalized lignin, the base material, lignin, is a byproduct of wood processing so might involve no new materials extraction, assuming lignin from wood processing could meet the needs of tire production could have adverse impacts on land use. The same could be said of octyl gallate which can be derived from fatty acids ( <i>e.g.</i> , coconut fatty acid), although it can also be synthesized from fossil fuelbased precursors. If the active antiozonant(s) are identified these could be produced industrially, likely from fossil fuelbased materials.
	Resource inputs and other resource consumption	Unclear	We define this life cycle segment as involving the resources used and consumed to produce the ingredients used as input for tire manufacturing, but not tire manufacturing itself, which is addressed below. There is no LCA available that specifically addresses the resource inputs or consumption associated with producing 6PPD or the possible alternatives for use in tires. Data on the inputs (other than chemical inputs discussed in Table 4.6) required to produce 6PPD and the possible alternatives are lacking. Production of graphene appears to be a highly energy intensive process ( <i>i.e.</i> , high temperatures) but again, no data are available to allow comparison to other possible alternatives. Information is lacking regarding production of amine functionalized lignin or rambutan peel extract in terms of the reagents or other inputs required because these specific materials are listed in patents and do not appear to be produced at scale commercially (although for rambutan peel extract the processing could be limited). It is also conceivable that some possible alternatives would be produced at different locations which could have different impacts in terms of raw material and chemical intermediate transportation (both in terms of distance and transportation mode). However, data are lacking to assess such effects; it is unclear how production of an alternative antiozonant could change in terms of suppliers and their locations in order to meet demand and moreover, transportation networks would likely change significantly to increase efficiency due to the large volume of antidegradant involved. Analyzing the potential suppliers and their geographic locations for the possible alternatives under consideration was considered outside the scope of the Stage 1 AA.
	Intermediate materials production processes	Unlikely, but data are limited	A review of information on the chemical precursors of the functional ingredients in the Priority Product and possible alternatives (as summarized in Table 4.6) suggests that the PPD-related alternatives have

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
			essentially equivalent intermediate materials and processes as 6PPD. Most of the non-PPD materials ( <i>e.g.</i> , TMQ, ethoxyquin, NBC, DLTDP, graphene) involve different chemistries but all involve industrial chemicals with significant hazard ( <i>e.g.</i> , nickel for NBC, ethoxyquin and DLTDP involve various acids). The details (and therefore hazards) of producing amine functionalized lignin are unknown as this material is proposed in a patent. Various internet searches for companies producing amine functionalized lignin extraction such as formaldehyde or methanol (see Table 4.6). Likewise, some hazardous chemicals may be involved in production of rambutan peel extract ( <i>e.g.</i> , methanol, hydrochloric acid) although details related to large scale extraction are lacking. The active ingredients in rambutan peel extract may also need to be produced synthetically and various searches for rambutan peel extract synthesis did not uncover any data for this. The hazards of graphene depend on the structure ( <i>i.e.</i> , thickness) and stage of the production process.
	Product manufacture	Possible, but data are limited	As noted above, this lifecycle segment pertains to the hazards and impacts of producing the Priority Product and possible alternatives ( <i>i.e.</i> , tires). As indicated in Tables 5.1 to 5.3, all the antidegradants pose some hazards which could be relevant for workers during exposure. It should be noted that the antidegradant comprises a small percentage of the mass of the tire and it is assumed that all other ingredients ( <i>e.g.</i> , rubbers, fabric belts, steel, carbon black, silica, other additives) will largely remain the same. The extent to which some additives may change with a new antidegradant is not currently known. Moreover, tires will still require vulcanization which constitutes the major source of energy required during the production process. Thus overall, it appears that changing the antidegradant will not have a material difference on the impacts of the manufacturing stage but data are too limited to be certain. How the manufacturing process would change with a new antidegradant is unknown because it may require changes in tire formulation or production. We are unaware of any studies of potential worker exposure to any of the chemicals under study. This will need to be further explored in Stage 2.
	Packaging	Unlikely	This lifecycle segment pertains to the hazards and impacts of packaging the Priority Product and possible alternatives ( <i>i.e.</i> , tires). Tires are packaged as either single units or on pallets. A change in the antidegradant appears unlikely to affect how the priority product is packaged.
	Distribution	Unclear	This lifecycle segment pertains to the hazards and impacts of transporting the Priority Product and possible alternatives ( <i>i.e.</i> , tires) to various sales sites. It is unlikely that the location of tire factories would change significantly if one of the possible alternatives was implemented in lieu of the priority product, so transportation impacts (CO <sub>2</sub> emissions, road wear particles) from the production facility to sales locations should not change.
	Use	Likely	If alternatives can eliminate or reduce the release of 6PPD or another chemical with similar toxicity to susceptible species to the environment, then there would be a material difference in this life cycle segment. Beyond the reported effect of 6PPD on certain salmon species, there are other aspects of the use phase which will need to be considered. One important consideration would be whether an alternative results in

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
			greater or lesser tire wear, potentially resulting in different environmental impact. However, tire wear particle generation likely depends more on driving conditions ( <i>i.e.</i> , speed, road surface type), than the antidegradant. Any alternative that could be implemented in a reasonable timeframe would still be rubber- based. The use phase may include storage of the tires when not in use ( <i>e.g.</i> , in an attached garage, in the vehicle trunk (for passenger tires), or in a distribution facility.) The low volatility of 6PPD does not indicate significant exposure during storage and an alternative that is more volatile or otherwise migrates out of the tire during storage would not satisfy the goals of having a long-term reservoir of antidegradant in the tire. Thus, it appears unlikely there would be a material difference among an alternative in terms of the storage aspect of tire use but data are limited.
	Operation and maintenance	Unlikely	Tires require little maintenance while in actual use. The maintenance that is required ( <i>e.g.</i> , maintaining proper inflation, periodic tire rotation and balancing) would not be expected to change with any of the possible alternatives.
	Waste generation and management	Potentially	This lifecycle segment pertains to manufacturing of the tire. During manufacturing manufacturers cycle excess formulated compound back into the manufacturing process (called rework); this substantially minimizes waste generated during tire production. The ability to continue the rework process is critical for minimizing production waste and will need to be considered for any of the possible alternatives. An alternative antidegradant that impacts processing time or temperature stability could significantly impact the potential for rework.
	Reuse and recycling	Potentially	A significant portion of tires are re-used as fill material or burned for energy ( <i>e.g.</i> , in cement kilns) (See Figures 4.1 and 4.2). If possible alternatives alter this situation, there could be a material difference in terms of waste minimization potential. For example, NBC contains the carcinogen nickel which could impact air emissions from cement kilns or use as fill in artificial turf. Possible alternatives that have different environmental mobility ( <i>e.g.</i> , greater water solubility) might also lead to a material difference in impact during reuse. One particular consideration for truck and bus tires is retread. An alternative that interferes with the retreading process could substantially increase tire waste because whole truck and bus tires would need to be purchased and discarded more frequently.
	End-of-life disposal	Potentially	About half of passenger tires in California are recycled to energy ( <i>i.e.</i> , tire derived fuel) or other uses ( <i>e.g.</i> , crumb rubber infill) (CalRecycle, 2023). If tires with alternative antidegradants have a different lifespan this could impact the amount of post-manufacturing tire waste generated and could exceed reuse and recycling capacity. However, consumers will only accept a limited decrease in product lifespan and so an alternative that substantially reduces product lifespan will be rejected.
Adverse Air Quality Impacts**	Would the product bring any changes to emissions of California Toxic Air Contaminants ( <i>e.g.</i> , benzene, Cr[VI])?	Unclear	Based on a review of the California Toxic Air Contaminant list (CARB, 2020), neither 6PPD nor any of the possible alternatives are Toxic Air Contaminants. As shown in Table 4.6, nearly all of the possible alternatives have chemicals in their production stream ( <i>e.g.</i> , benzene, nickel) that are Toxic Air

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
			Contaminants but the extent to which emissions would increase from increased production of the alternatives is not known.
	CO <sub>2</sub> emissions	Unclear	LCA exist for tires that describes carbon dioxide (CO <sub>2</sub> ) emissions (Dong <i>et al.</i> , 2021; Piotrowska <i>et al.</i> , 2019). However, details related to the contribution of 6PPD or any possible alternative to overall product CO <sub>2</sub> emissions are lacking. Thus, no data on this factor are available for a comparison.
	HFC emissions	No	As indicated in Table 4.6, HFCs do not appear to be used in production of 6PPD or any of the possible alternatives.
	Methane emissions	Unlikely, but data are limited	Methane is released during fossil fuel extraction and most of the possible alternatives have fossil fuels (oil, natural gas) as part of their production life cycle. Graphene can be produced using methane <i>via</i> vapor deposition. Methane is known to be an input in production of NBC but is presumably consumed and not released to the environment. It is unknown whether methane is associated with the production of amine functionalized lignin or rambutan peel extract, but this seems unlikely.
	Nitrogen fluoride emissions Perfluorocarbon emissions Sulfur hexafluoride emissions	No	Based on the available production process information for 6PPD and the possible alternatives (Table 4.6), emissions of these chemicals are not expected to be part of the life cycle of the Priority Product or any of the possible alternatives. However, the available data on production process are limited.
	Other global warming gas emissions	No	All the ingredients of the Priority Product and possible alternatives are produced (or harvested in the case of rambutan peel) industrially so each involves some CO <sub>2</sub> emissions. No other greenhouse gases are known to be produced in the product life cycles. Methane is known to be an input in production of NBC but is presumably consumed and not released to the environment.
	Particulate matter emissions	Potentially, in part	Use of tires produces TRWP. TRWP will still be produced if any of the possible alternatives are used in lieu of the Priority Product. Whether the absolute amount of TRWP generated will change is not known although since particles are generated by tires gripping the road and this is an important factor for safety, a large change in particle number seems unlikely. That being said, the reported impact of TRWP, given the appropriate conditions, on certain sensitive species ( <i>e.g.</i> , juvenile coho salmon, as reported in laboratory studies) potentially would be reduced if 6PPD is replaced with an alternative without such a reported effect on sensitive species. However, other potential impacts of TRWP ( <i>e.g.</i> , other additives) would likely be the same. Thus, the relevance is considered to be "potentially, in part" because, while the emissions themselves are unlikely to change significantly, the impact could be materially different.
	Nitrogen oxide emissions Ozone-depleting substances emissions Sulfur dioxide emissions	No	Based on the known production process for 6PPD and the possible alternatives (Table 4.6), emissions of these chemicals are not expected to be part of the life cycle of the Priority Product or possible alternatives. None of the functional ingredients in the Priority Products and alternatives are ozone-depleting substances. A few of the possible alternatives (DOPD, DLTP) have production processes that involve sulfuric acid, which may be produced using sulfur dioxide, but this is not the only method of production and release of sulfur dioxide may not be part of the production process.

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
	Would the product bring any changes to emissions of compounds that might lead to tropospheric ozone production?	Unclear	Tropospheric ozone is formed by the reaction of solar energy with hydrocarbons and nitrogen oxides. 6PPD and all of the possible alternatives have hydrocarbons in their production chain, and several have nitrogen compounds (nitric acid). Whether compounds contributing to tropospheric ozone formation are emitted during production and whether production occurs in California is unclear.
Adverse Ecological Impacts*	Would the product, its constituents, or its likely breakdown products have any acute or chronic toxicity to impact aquatic, avian, or terrestrial animal or plant organisms or microbes?	Yes	Recent laboratory studies have reported that 6PPD's transformation product 6PPDQ, in certain concentrations, exhibits acute toxicity to certain fish species such as coho salmon. Although toxicity of quinone products of the alternatives is not known with certainty, preliminary data suggests that the quinones of certain other PPDs do not possess the same degree of toxicity to such species. These data were collected in <i>in vitro</i> systems, however and would need to be verified. In terms of other environmental breakdown products of 6PPD and the possible alternatives, data are limited. To the extent that information was found in the ECHA dossiers we examine, as shown in Table 5.10, 6PPD and many possible alternatives have similar breakdown products which do have acute aquatic toxicity (GHS category 1), including p- benzoquinone, p-hydroquinone, and aniline. Data on environmental transformation products for many of the possible alternatives (TMQ, ethoxyquin, DLPTP) are lacking. Graphene would presumably have no transformation products in the environment but could release nano materials which could have an environmental impact.
	Would the product bring changes in population size, reduction in biodiversity, or changes in ecological communities?	Yes	Recent laboratory studies have reported that 6PPD's transformation product 6PPDQ can cause symptoms associated with pre-spawn mortality in certain salmonid species which, among other factors, has the potential to impact the population size of this species. Moreover, reductions in the population size of these salmon species could affect biodiversity. As noted earlier, there is preliminary information suggesting that other PPD antidegradants do not possess this property. It appears unlikely that non-PPD alternatives ( <i>e.g.</i> , TMQ, ethoxyquin) can produce these or equivalent effects although this has not been studied.
	Would the product bring changes to the abilities of an endangered or threatened species to survive or reproduce?	Yes	See discussion above.
	Would the product bring changes to deterioration or the loss of environmentally sensitive habitats?	Unclear	6PPD is not known to directly cause habitat deterioration or loss. Along with other factors, laboratory studies have indicated that 6PPDQ can potentially impact populations of coho salmon and other sensitive species. The salmon spawning process is known to be important in transporting nutrients to riverine ecosystems. Thus, while 6PPD and 6PPDQ are not known to directly cause habitat loss or deterioration, there may be an indirect effect. It is unknown whether any of the possible alternatives could impact sensitive habitats since they are not currently used in a manner similar to 6PPD. Alternatives that do not cause pre-spawn mortality of coho salmon or other sensitive species (or which are less potent in this regard) would seem to be preferred alternatives with respect to this relevant factor. If additional agricultural land is required to produce lignin or rambutan peel extract (assuming these are not synthesized chemically), there could be associated habitat loss.

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
	Would the product bring changes that contribute to or cause vegetation contamination or damage?	Unclear	6PPD is not known to directly cause habitat deterioration or loss. This is not stated to be a property of the possible alternatives but these have not been used in as widespread a product as motor vehicle tires. If additional agricultural land is required to produce lignin or rambutan peel extract (assuming these are not synthesized chemically), there could be associated habitat loss.
	Would it bring adverse effects on environments that have been designated as impaired by a California State or federal regulatory agency?	Unclear	6PPD is not known to directly cause habitat deterioration or loss. Along with other factors, laboratory studies have indicated that 6PPDQ can potentially impact populations of coho salmon and other sensitive species. The salmon spawning process is known to be important in transporting nutrients to riverine ecosystems. Thus, while 6PPD and 6PPDQ are not known to directly cause habitat loss or deterioration, there may be an indirect effect. It is unknown whether any of the possible alternatives could impact sensitive habitats since they are not currently used in a manner similar to 6PPD. Alternatives that do not cause pre-spawn mortality of coho salmon or other sensitive species (or which are less potent in this regard) would seem to be preferred alternatives with respect to this relevant factor. If additional agricultural land is required to produce lignin or rambutan peel extract (assuming these are not synthesized chemically), there could be associated habitat loss.
	Would it result in biological or chemical contamination of soils?	Unclear	6PPD does not appear to affect soils during its use in tires; tire wear particles generally enter the aquatic environment or remain on roadways. The extent to which 6PPD/6PPDQ can migrate to soils from use of tire rubber as fill ( <i>e.g.</i> , in artificial turf) is not well studied. Any alternative would presumably have the same potential if tires containing it were used in a similar manner. As noted earlier, any alternative to 6PPD would have to migrate through tire rubber which would seem to indicate a similar ability to migrate from rubber infill to soil.
	<ul> <li>Any other adverse effects, as defined in Section 69401.2(a) (CalDTSC, 2012a), for environmental hazard traits and endpoints specified in Article 4 of Chapter 54, as follows:</li> <li>Domesticated animal toxicity</li> <li>Eutrophication</li> <li>Impairment of waste management organisms</li> <li>Loss of genetic diversity (including biodiversity)</li> <li>Phytotoxicity</li> </ul>	Potentially, in part	Because of the reported phenomenon of pre-spawn mortality (Scholz <i>et al.</i> , 2011), 6PPDQ has the potential to affect coho reproductive success. Certain studies suggest it may affect coho development (Greer <i>et al.</i> , 2023b). DTSC's Product-Chemical Profile for 6PPD in motor vehicle tires (CalDTSC, 2022) also states that published studies also suggest 6PPDQ could have an impact on the following concerns: genetic diversity, wildlife development, reproduction, survival and growth. Preliminary <i>in vitro</i> data suggest some of the possible alternatives do not have this effect. However, data are incomplete on the potential for these possible alternatives to have other adverse ecological impacts. Literature searches with "6PPD" and any of the following: "eutrophication", "domestic animals", "waste management", and "phytotoxicity" did not indicate studies reporting an effect of 6PPD on these concerns. Whether possible alternatives could have such concerns will be investigated as part of Stage 2 with the refined alternative list.
	<ul> <li>Wildlife developmental impairment</li> </ul>		

Category	Factor that Is Relevant if Materially Different Between	Relevant?*	Basis
	Priority Product and Alternatives		
	<ul> <li>Wildlife growth impairment</li> </ul>		
	<ul> <li>Wildlife reproductive impairment</li> </ul>		
	<ul> <li>Wildlife survival impairment</li> </ul>		
	Evidence for environmental hazard traits ( <i>i.e.</i> , from standard aquatic and terrestrial toxicity testing, research-based investigations, mechanistic evidence from cell-based or whole organism-based assays showing perturbations of known physiological, biochemical, or other pathways, or evidence from quantitative structure activity		
Adverse Soil Quality Impacts*	relationship programs). Would the product impact soil compaction or other soil structure changes? Would the product impact soil erosion? Would the product cause loss of organic matter in soil? Would the product cause soil sealing?	No	Given the primary use of the product (driving on engineered roadways) it would not be expected that effects on the soil physical characteristics listed would be materially different among possible alternatives.
	Would the product be expected to directly enter the municipal storm sewer systems ( <i>e.g.</i> , car wash detergents)?	No	6PPD and its possible alternatives would all be expected to be present in tire road wear particles and enter stormwater sewers during precipitation events. Since 6PPD and all of the alternatives are similar in this regard, the factor would not be materially different and is therefore considered not relevant.
	Would the product bring any increase in biological oxygen demand within the water system?	No	Neither 6PPD nor any of the possible alternatives have been demonstrated to have the potential to affect the growth of biological organisms in a way that would alter biological oxygen demand.
	Would the product bring any increase in chemical oxygen demand within the water system?	Unlikely	As antioxidants, it can be expected that all alternatives would have the <i>potential</i> to affect chemical oxygen demand. Whether such effects are materially relevant (given the relatively low concentrations of these chemicals detected in surface water bodies) has not been investigated. It seems logical to expect that other chemicals, present at higher concentrations, would be more likely to impact chemical oxygen demand.
	Would the product bring any increase in the temperature of water systems?	No	Neither 6PPD nor any of the possible alternatives have the potential to alter the temperature of water systems. Key determinants of water temperature in streams are the source of incoming waters, the speed and depth of the water, and the degree of vegetative cover affecting sunlight. None of these would be affected by 6PPD or its possible alternatives.
	Would the product bring any increase in total dissolved solids in water systems?	Potentially	Total dissolved solids (TDS) indicates the amount of inorganic and organic chemicals present as molecular, ionized, or colloidal particles in water. Some of the possible alternatives are inorganic and the alternatives

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
			vary in their water solubility. It is possible that this property could differ among the possible alternatives although whether it would be materially different is unclear.
	Increase in California CWA priority pollutants Increase in California CWA pollutants	No	Based on a review of the California Clean Water Act (CWA) Hazardous Substances, Priority Pollutants, and Toxic Pollutants lists (UL LLC, 2020) neither 6PPD nor any of its possible alternatives are present on these lists. The same applies to the main transformation products, to the extent that these are known.
Adverse Water Quality Impacts*	Increase in chemicals with drinking water MCLs	No	Based on a review of the relevant regulations (22 CCR § 64431, § 64444, and § 64449) neither 6PPD nor any of its possible alternatives have drinking water MCLs. The same applies to the main transformation products of 6PPD and the possible alternatives, to the extent that such transformation products are known.
	Increase in chemicals with drinking water notification levels	No	Based on a review of the California guidance (CalSWRCB, 2020) neither 6PPD nor any of its possible alternatives have drinking water notification levels. The same applies to the main transformation products of 6PPD and the possible alternatives to the extent that such are known.
	Increase in chemicals with drinking water public health goals	No	Based on a review of the relevant regulation (CalOEHHA, 2019b) neither 6PPD nor its possible alternatives have drinking water public health goals (PHG).
	Exceedance of a standard relating to the protection of the environment	No	The consortium members are not aware of any material difference between 6PPD and any of the possible alternatives with respect to this relevant factor.
Public Health Impacts*§	Acute mammalian toxicity [Not included as a SCP hazard trait but included at preparer's discretion]	No	As shown in Table 5.1, there are slight differences in acute toxicity across the possible alternatives. Some possible alternatives have a higher acute toxicity than 6PPD (GHS category 3 <i>versus</i> 4) and many ( <i>e.g.</i> , 7PPD, DPPD, 6QDI, Durazone 37, graphene) are not classified under GHS nor classified for acute mammalian toxicity. These differences are not likely to have a significant bearing on the choice of alternative ( <i>i.e.</i> , there are no GHS acute toxicity category 1 or 2 alternatives) There are also many alternatives with data gaps for this factor.
	Carcinogenicity	Potentially	As shown in Table 5.1, 6PPD is GHS not classified for carcinogenicity. A number of possible alternatives are similarly not classified (7PPD, DTPD, DLTP, Durazone 37). Several possible alternatives ( <i>i.e.</i> , TMQ oligomer, N-phenyl-2-naphthylamine, NBC) are classified as GHS category 2 for carcinogenicity. There are many possible alternatives with data gaps for this factor.
	Developmental toxicity Reproductive toxicity	Yes	As shown in Table 5.1, 6PPD is has a high hazard classification (category 1B) for reproductive and developmental toxicity. A number of PPDs have similar classification (although some of these are based on using 6PPD as a surrogate). A few possible alternatives have low potential for reproductive and developmental toxicity ( <i>i.e.</i> , GHS not classified). There are many possible alternatives with data gaps for this factor.
	Cardiovascular toxicity	Potentially	As shown in Table 5.2, 6PPDQ has been suggested to affect the cardiovascular system in coho salmon by affecting vascular permeability, however, there are many data gaps and no relevant observations have been made in mammalian species. A few possible alternatives ( <i>e.g.</i> , NBC, DLTP) have been reported to affect the cardiovascular system.

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
	Dermatotoxicity (including skin sensitization)	Yes	As shown in Table 5.1, 6PPD is not classified for skin irritation or corrosion ( <i>i.e.</i> , low hazard). Most possible alternatives are either similarly not classified or have a data gap for this factor. Two possible alternatives score worse than 6PPD for this factor (CCPD and N-phenyl-2-naphthylamine). However, 6PPD and several other PPDs ( <i>e.g.</i> , 7PPD, Durazone 37) are classified as category 1 skin sensitizers (Table 5.1). Several non-PPD alternatives are GHS not classified for skin sensitization (TMQ oligomer, NBC, graphene, Irganox 1620). There are many chemicals with data gaps.
	Ocular toxicity	Yes	As shown in Table 5.1, 6PPD is not classified for eye irritation (ECHA) or classified as category 3 (Ecology GreenScreen). A number of possible alternatives are GHS category 2 for eye irritation ( <i>e.g.</i> , 6QDI, CCPD, NBC, N-phenyl-2-naphthylamine). A number of possible alternatives ( <i>e.g.</i> , 77PD, DTPD, TMQ oligomer, Durazone 37, graphene) score more favorably. There are many possible alternatives with data gaps for this factor.
	Organ toxicity [Not included as a SCP hazard trait but included at preparer's discretion]	Yes	As shown in Table 5.1, 6PPD is classified as category 2, moderate hazard, for repeated dose systemic toxicity (which encompasses organ toxicity) as evaluated by ToxServices. However, in the EU ECHA dossier, it is "not classified." A number of the 6PPD alternatives (7PPD, NBC, ethoxyquin, N-phenyl-2-naphthylamine) have similar or worse scores. A number are GHS not classified for this factor and so appears to have lower hazard than 6PPD ( <i>e.g.</i> , DPPD, DTPD, DLTP, Durazone 37, graphene). There are a number of data gaps.
	Endocrine toxicity	Potentially	As shown in Table 5.1, the picture is mixed. None of the possible alternatives are listed as endocrine disrupters in the EU. 6PPD and a few possible alternatives (7PPD, IPPD, 6QDI and ethoxyquin) were listed as having moderate evidence of endocrine activity in the Ecology GreenScreens. Many possible alternatives have data gaps for this factor.
	Epigenetic toxicity	No	Epigenetic toxicity refers to the ability to alter gene expression without necessarily changing gene structure. In reviewing toxicity summaries for the possible alternatives, we found no indication that any of the possible alternatives exerted epigenetic effects. However, such effects are not often studied.
	Genotoxicity	Potentially	As shown in Table 5.1, 6PPD is not classified for genotoxicity/mutagenicity. All possible alternatives but one are either similarly not classified or have a data gap for this factor. DPPD is reported to be GHS category 2 (moderate hazard) for this factor.
	Hematotoxicity	Potentially	As shown in Table 5.2, 6PPD has been found to exert hematotoxicity at high doses in rats in some studies. A number of other PPDs (IPPD, DPPD, DAPD) and non PPDs (NBC, ethoxyquin, DLTP) have also shown hematotoxic effects. Data for a number of chemicals ( <i>e.g.</i> , TMQ, graphene) are lacking.
	Hepatotoxicity and digestive system toxicity	Potentially	As shown in Table 5.2, 6PPD has given some evidence of adverse effects in the liver in rodents. A few of the possible alternatives ( <i>e.g.</i> , polymerized TMQ, ethoxyquin) have also shown adverse effects on the liver. Note that effects on liver weight alone are not considered adverse but may be considered adaptive.

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
	Immunotoxicity	Yes	Other than being a dermal sensitizer, 6PPD is not known to be an immunotoxicant. Data gaps exist for most of the possible alternatives; however, most of the phenylene diamine possible alternatives are dermal sensitizers. It should be noted that dermal sensitization is also considered under another relevant factor in this table ( <i>i.e.</i> , dermatotoxicity). A few (NBC, DAPD) have reports of changes in the weight or size of immune system related organs but no more direct indications of adverse effects.
	Musculoskeletal toxicity	Potentially	6PPD is not noted to cause musculoskeletal toxicity. No relevant information was found for any of the possible alternatives except NBC, which was noted to cause degeneration of skeletal muscle in rats. Effects on skeletal development during gestation are addressed under Developmental Toxicity.
	Nephrotoxicity and other toxicity to the urinary system	Potentially	6PPD is not reported to cause adverse effects on the kidney. Most possible alternatives similarly were not reported to cause adverse kidney effects. Ethoxyquin and N-phenyl-1-naphthylamine have been reported to cause kidney degeneration in rats.
	Neurodevelopmental toxicity Neurotoxicity	Potentially	6PPD is not reported to cause adverse neurological effects either in adults or juvenile animals. Most possible alternatives have data gaps for this factor. NBC and ethoxyquin and are reported to have caused adverse neurological effects in experimental animals.
	Ototoxicity	No	No data were located which indicated ototoxic (hearing related) effects for 6PPD or any of the possible alternatives.
	Reactivity in biological systems	Potentially	DTSC has stated that 6PPDQ is reactive in biological systems as part of the Agency's Priority Product Profile. We believe that the effects of 6PPDQ reported by some researchers are more appropriately and precisely described by other relevant factors ( <i>e.g.</i> , those related to aquatic toxicity). That being said, preliminary data discussed in section 5 suggests that at least some other PPDs and their quinone transformation products may have differential toxicity to coho salmon. While all molecules are reactive at some dose in biological systems, it appears likely that the toxicity will differ. More data are needed to fully evaluate this endpoint.
	Respiratory toxicity (including respiratory sensitization)	Yes	6PPD and several other PPDs are classified in the Ecology GreenScreens as respiratory sensitizers based primarily on a structural alert (Table 5.1). There are many chemicals with data gaps, although a few possible alternatives are not classified for respiratory sensitization ( <i>i.e.</i> , low hazard) such as Durazone 37. No data were located which indicated any other adverse respiratory effects for 6PPD or any of the possible alternatives. Potential effects of 6PPDQ on respiration in coho salmon are noted in CalDTSC's Priority Product Profile (CalDTSC, 2022).
	Evidence for other toxicological hazard traits	No	In our review of hazard data, we did not discover evidence for other toxicological hazard traits.
	Exceedance of an enforceable California or federal standard related to public health	No	To the best of our knowledge, use of the Priority Product or any of the possible alternatives will not involve intentional exceedance of such a standard, other than the ones already addressed elsewhere in this table.

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
Waste and End-of-Life Effects*	Would the product bring any change to the volume or mass of the waste materials and byproducts generated during the life cycle?	Unclear	Because the possible alternatives are all replacements for 6PPD in vehicle tires and represent only a small fraction of the mass of a tire, a materially relevant impact on the amount of waste or byproducts produced is not foreseen. However, as noted above, tire manufacturers rely on reusing some production waste (rework) in the tire manufacturing process to minimize waste generation. An alternative that alters that ability could result in more waste generation but the extent to which that might occur is not currently known.
	Would the product need any special handling to mitigate adverse impacts resulting from the waste materials generated during the life cycle?	Unclear	At this time, it is not expected that there will be any different requirements for waste material handling from any of the possible alternatives. However, as noted above, their production process (certainly at scale) is not well understood. This will be explored in the Stage 2 AA.
	Effects on solid waste or wastewater disposal or treatment	Unclear	At this time, it is not expected that there will be any different requirements for waste material handling from any of the possible alternatives as it related to waste disposal. However, as noted above, their production process (certainly at scale) is not well understood. This will be explored in the Stage 2 AA.
	Effects on discharge(s) or disposal(s) to storm drains or sewers adversely affecting wastewater or storm water treatment facilities	Unclear	At this time, it is not expected that there will be any different impacts on wastewater or stormwater treatment facilities. It is expected that all production facilities operate with water discharge permits which might need to be modified to address production of tires with a new antidegradant.
	Release to the environment	Unclear	Consortium members are not aware of any other factors associated with releases to the environment that would be associated with any of the possible alternatives.
Environmental Fate*	Aerobic and anaerobic half-lives of the product, its constituents, or its likely breakdown products	Unclear	No information on the aerobic or anaerobic half-lives could be found in EpiSuite or ECHA dossiers for 6PPD and the possible alternatives. Although we are aware of specific studies related to the aerobic and anaerobic half-lives of 6PPDQ, we are unaware of any data that exists for the potential alternatives, making comparisons impossible. This relates to the factor of persistence, discussed below.
	Aqueous hydrolysis half-life of the product, its constituents, or its likely breakdown products	Unclear	No information on hydrolysis rate constant could be found in EpiSuite or ECHA dossiers.
	Atmospheric oxidation rate	Unclear	No information on hydrolysis rate constant could be found in EpiSuite or ECHA dossiers. However, see Table 5.9 regarding the environmental half-life in air which is influenced by atmospheric oxidation rate. Although there are differences among 6PPD and possible alternatives for this property, they are not considered materially relevant since the primary concern is migration of 6PPD to surface water. All values are less than 1.0. 6PPD and most possible alternatives have very low vapor pressures suggesting limited impacts on air. While 6PPD associated with dust has been measured in air it is unclear how this would be affected by atmospheric processes since the chemical is contained within the particle substrate.
	Bioaccumulation of the product, its constituents, or its likely breakdown products	Yes	As shown in Figure 5.11, neither 6PPD nor any of the possible alternatives are listed as persistent, bioaccumulative or toxic. 6PPD and the possible alternatives do vary substantially in terms of the bioaccumulation potential. For example, as shown in table 5.3, while 6PPD, DPPD, 7PPD, DTPD are considered bioaccumulative based on California criteria, other possible alternatives (IPPD, TMQ oligomer,

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
			ethoxyquin, and DLTP) are not. There are a number of chemicals with data gaps. Given that 6PPD and 6PPDQ are of concern for toxicity to certain fish species, bioaccumulation potential should be considered a relevant criterion if there is a difference among possible alternatives.
	Mobility in environmental media	Yes	See below regarding water solubility, lipid solubility, log k <sub>ow</sub> , <i>etc</i> . There is a difference in ability to move through environmental media across the possible alternatives. Given that the key concerns regarding 6PPD are effects on certain fish species, environmental mobility is a relevant factor if there are differences among the possible alternatives.
	Persistence	Yes	6PPD and the PPD related alternatives are either considered persistent or have data gaps. NBC, TMQ oligomer, and ethoxyquin are considered persistent although DLTP is not.
	Photodegradation	Unclear	Data for the photolysis rate constant is lacking for most of the possible alternatives. Moreover, given the low vapor pressure of the possible alternatives, it does not appear that free chemicals will be present in the air. Whether photolysis is a relevant mechanism for chemicals in tire wear particles is unknown.
Materials and Resource Consumption	Impacts on consumption of renewable resources, including energy and raw materials, throughout the product life cycle	Potentially	Very few inputs to the production of 6PPD and its possible alternatives as renewable resources. Tires do contain some renewable resource content (natural rubber) and whether this would change with different antidegradants (particularly non PPD antiozonants) is unclear. It is unknown how renewable energy requirements might differ for the possible alternatives. For electric vehicles, which have an increasing share of the vehicle market in California, consumption of renewable energy is important, and this could be affected by rolling resistance and vehicle energy efficiency per mile traveled. This could be affected by the possible alternatives if they affect rolling resistance.
	Impacts on consumption of non-renewable resources, including petroleum, coal, metals, minerals, and other finite resources, throughout the product life cycle	Potentially	As shown in Table 4.5, all of the possible alternatives appear to have some fossil fuel or otherwise non- renewable ingredients. Whether there is a quantitative difference among possible alternatives that is materially different is unclear. As noted above, possible alternatives that could affect the rolling resistance could impact consumption of non-renewable energy sources.
Physicochemical Properties*	Do the product or the alternatives exhibit oxidizing properties that facilitate combustion?	No	6PPD and all of the possible alternatives are being used as antidegradants, and ideally have both antioxidant and antiozonant properties. Therefore, the alternatives would likely inhibit rather than facilitate combustion.
	Do the product or the alternatives exhibit explosivity?	Unlikely	Based on the available ECHA REACH dossiers of the chemicals, none of the products exhibit this property (ECHA, 2020). However, for a number of possible alternatives, information on explosivity is not available.
	Do the product or the alternatives exhibit flammability?	Unlikely	Based on the available ECHA REACH dossiers of the chemicals, none of the products exhibit this property (ECHA, 2020). However, for a number of possible alternatives, information on flammability is not available.
	Do the product and alternatives have different physical states?	Potentially	Some alternatives, like 6PPD, are solids whereas others are viscous liquids. However, this difference is not expected to be materially relevant to the impact of the chemical on humans or the environment. It could be an important difference in terms of technical feasibility because current tire production processes are designed to work with solid antidegradant.

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
	Molecular weight	Yes	As shown in Table 5.9, the molecular weights range broadly across the possible alternatives. Molecular weight influences migration of the molecule in tire rubber and is thus an important factor in performance.
	Density	Unclear	Density information is lacking for many possible alternatives. Most have densities around 1 g/mL. Density could be important in terms of technical feasibility of incorporating the chemical into rubber compound.
	Vapor pressure	Potentially	As would be expected for compounds with high molecular weights, 6PPD and the possible alternatives all have low vapor pressures (maximum 1.56 E-3 mmHg at 25°C). Vapor pressure would not be expected to constitute a materially relevant difference in terms of performance or environmental impact. In terms of technical feasibility, vapor pressure is important because the antidegradant has to survive the high temperatures of the tire manufacturing process without excessive loss.
	Melting point	Potentially	As noted above, some of the possible alternatives are solids (like 6PPD) and some are liquids. The physical state could impact the technical feasibility of incorporation into rubber compound but should not affect environmental impacts.
	Boiling point	Potentially	As shown in Table 5.9, all of the possible alternatives have high boiling points (the lowest, ethoxyquin is 123-125°C). Boiling point would not be expected to be a materially relevant difference in terms of either technical feasibility or environmental impact.
	Flash point	Potentially	Flash point is not one of the SCP mandated relevant factors. This property is materially relevant however, because it relates to manufacturing safety concerns with high temperature processes. This is captured in the flammability endpoint included in Table 5.3, where a number of possible alternatives have data gaps.
	Water solubility	Yes	As shown in Table 5.9, water solubility among the various possible alternatives varies by more than 10 orders of magnitude. Water solubility will have a significant bearing on movement of the chemical from tire wear particles through the environment.
	Lipid solubility	Yes	As shown in Table 5.9, the log K <sub>ow</sub> (an indicator of lipid solubility) varies substantially among the possible
	Octanol-water partition coefficient (log K <sub>ow</sub> )	Yes	alternatives. Lipid solubility will have a significant impact on where in the environment the chemical will partition.
	Organic carbon partition coefficient (K <sub>oc</sub> )	Yes	As shown in Table 5.9, the $K_{oc}$ varies substantially among the possible alternatives. $K_{oc}$ will have a significant impact on where in the environment the chemical will partition ( <i>e.g.</i> , water versus sediment).
	Sorption coefficient for soil and sediment	Unclear	No data could specifically be located for this factor but see the discussion above for $K_{oc}$ .
	Octanol-air partition coefficient (K <sub>oa</sub> )	Unclear	No data on this parameter could be located. See below regarding Henry's Law constant.
	Diffusivity in air and water	Potentially	As shown in Table 5.9, this parameter varied relatively modestly across the possible alternatives and is therefore unlikely to result in a material difference among them.
	Henry's Law constant	Yes	This parameter measures the tendency of a chemical to partition between water <i>versus</i> air. This parameter varies substantially among the possible alternatives and would be important in assessing environmental mobility of the chemical. Its relevance is somewhat linked to both water solubility and vapor pressure.

Category	Factor that Is Relevant if Materially Different Between Priority Product and Alternatives	Relevant?*	Basis
	Redox potential	Unclear	As shown in Table 5.9, redox potential data are available for 6PPD and a few of the possible alternatives. The values are all fairly similar, so this is not expected to make a material difference among possible alternatives where such information is available. Information is lacking, however, for many of the possible alternatives.
	Photolysis rates	Unclear	As shown in Table 5.9, there is a significant difference in photolysis potential among 6PPD and the possible alternatives. However, all have very low vapor pressures and would not be expected to be present in the atmosphere to an appreciable extent. However, all might be present in airborne dust where the impact of photolysis is not known.
	Hydrolysis rates	Unclear	As shown in Table 5.9, no data on hydrolysis rate constants for 6PPD nor any of the possible alternatives could be located.
	Dissociation constants	Unclear	Dissociation constant data were only located for a few of the possible alternatives. The available data suggest a limited potential for dissociation and therefore this factor is not believed to be materially relevant.
	Reactivity, including electrophilicity	Yes	The products are all chemically reactive as part of their function. To perform adequately, they must be chemically reactive, and the degree of reactivity is important to their antidegradant function in the tire.
Product Function and Performance*	Are there material differences in terms of the useful life of the product?	Potentially	It is possible that a 6PPD alternative that is not as effective could result in a shorter useful life for the product. This would have negative impacts on raw materials and energy consumption during production, waste generation, consumer costs and potentially consumer safety (if consumers replace tires less frequently due to cost). Thus, there are very strong incentives not to accept alternatives that result in a short product life.
	Are there material differences in terms of the function and performance of the product?	Potentially	It is the position of the consortium members that tires must perform in a manner that is safe and consistent with both federal regulations and company product stewardship requirements. An alternative that does not have comparable function and performance as 6PPD would not be acceptable.
	Are there material differences in terms of the functional acceptability of the product?	No	It is the position of the consortium members that any 6PPD alternative must function comparably to 6PPD as an antidegradant.
	Are there material differences in terms of the technical feasibility of the product?	Potentially	There is no alternative that will be a "drop in" replacement for 6PPD. Most alternatives will require modification of the tire formulation and/or the tire production process ( <i>e.g.</i> , balancing processing time <i>vs</i> curing rate or scorch time). The extent of modification required will likely vary considerably among the possible alternatives and will have to be investigated through production process research.
Economic Impacts*	Will the product and its alternatives have a different cost to consumers or other users?	Potentially	It is possible that the use of a possible 6PPD alternative in tires could have some effect on the cost of the product. There will likely be substantial costs for new production equipment as well as product testing required by law or company product stewardship specifications which would have to be reflected in the cost of the product. Since many of the possible alternatives are not commercially produced at scale, it is unclear how large that difference in cost could be. This will be addressed more fully in Stage 2.

#### Notes:

6PPD = N-(1,3-Dimethylbutyl)-N'-phenyl-p- phenylenediamine; 6PPDQ = N-(1,3-Dimethylbutyl)-N'-phenyl-p- phenylenediamine quinone; 6QDI = N-1,3-Dimethyl Butyl-N'-Phenyl Quinone Diamine; 7PPD = N-(1,4-Dimethylpentyl)-N'-phenyl-p-phenylenediamine; AA = Alternatives Analysis; CalDTSC = California Department of Toxic Substances Control; CARB = California Air Resources Board; CO<sub>2</sub> = Carbon Dioxide; DAPD = Diaryl-p-phenylene diamine; DLTDP = Dilauryl thiodipropionate; DLTP = Dilauryl thiodipropionate; DOPD = 4,4'-Dioctyldiphenylamine; DPPD = N,N'-Diphenyl-p-phenylenediamine; ECHA = European Chemicals Agency; EU = European Union; GHS = Globally Harmonized System; HFC = Hydrofluorocarbon; IPPD = N-Isopropyl-N'-phenyl-p-phenylenediamine; LCA = Life Cycle Assessment; MCL = Maximum Contaminant Level; NBC = Nickel dibutyldithiocarbamate; PPD = Paraphenylene Diamine; RE = Responsible Entity; REACH = Registration, Authorisation, and Restriction of Chemicals; SCP = Safer Consumer Products; TMQ = Poly(1,2-dihydro-2,2,4-trimethyl-quinoline); TRWP = Tire and Road Wear Particles; US EPA = United Sates Environmental Protection Agency; VOC = Volatile Organic Compound.

\* Whether a factor was considered relevant to drawing a distinction between possible alternatives and candidate chemical in the Priority Product was described using five terms. The terms "yes" and "no" were used when the available data indicated a clear material difference would be expected or is obvious. The terms "potentially" and "unlikely" were used when the available data were too limited to be fully confident on whether a material difference can be expected but the available data leaned towards one direction (*i.e.*, towards either yes or no but not reliably). Finally, the term "unclear" was used when data were not available to give an indication of whether a factor would be relevant or not.

\*\* For theses relevant factors, we consider the impacts of the Priority Product and possible alternatives but not of chemicals upstream in their product lifecycle.

<sup>§</sup> Public health impacts relate to 6PPD and possible alternatives. Hazards of 6PPDQ are not included because data for this chemical are limited and data for the quinone products of possible alternatives are unavailable, making comparison impossible. Safer Consumer Products (SCP) regulations: CalDTSC (2013) (22 CCR § 69505.5).

Category	Element	Relevant?*	Basis
Chemical Quantity Information	Would the alternative change the quantities of the chemical(s) of concern or other replacement chemicals necessary to manufacture the	Unclear	For possible alternatives to 6PPD as an antidegradant, data are not yet available to determine how the quantity required per tire might change. It is expected that the
	product?		amount of antiozonant will not change substantially ( <i>e.g.,</i> by several fold) but this
	Would the alternative change the quantities of the chemical(s) of concern or other replacement chemicals placed into the stream of commerce in California?	Unclear	remains to be seen.
Market Presence of Product	Would the alternative change statewide sales of the product by volume?	Potentially	Since all possible alternatives are replacements for 6PPD, tire sales volume and number of units should not change unless costs significantly increase or unless the use of the alternative decreases the life of the tire ( <i>i.e.</i> , increasing the wear or aging rate), which would require more frequent replacement. Also, an alternative that affects retread
	Would the alternative change statewide sales of the product by number of units?		potential could require larger sales of new truck and bus tires.
	Would the alternative change the intended product use(s), and types and age groups of targeted customer base(s)?	No	Tires would be used in the same manner and by the same type of individuals. Tires are not used in different ways by different segments of the population.
Occurrence or Potential	Will there be a difference in occurrence or potential occurrence of	Potentially	If the selected alternative is a PPD, it may not affect exposure to candidate chemicals
Occurrence of Exposure	exposure to Candidate Chemicals in the product?		(while still having an improved hazard profile for fish species of concern). If another, non-PPD alternative is selected, this could change the potential for candidate chemical exposure during certain phases of the life cycle, although details are lacking about how these possible alternatives would be handled and managed.
Household and Workplace Presence	Will the product be used in the home?	Potentially	Tires may be stored in the home, but it is abrasion of tires by roads which results in the release. It is possible that volatilization of the antidegradant could result in migration to air. Although DTSC has indicated that the vapor pressure of 6PPD is negligible (DTSC, 2023), some of the alternatives may have relatively greater volatility. Thus, storage in the home ( <i>e.g.</i> , the garage) could potentially be a relevant factor for some of the alternatives.
	Will the product be used in the workplace?	Yes	Tires are used on vehicles used during work ( <i>e.g.</i> , cars used for work, and trucks and buses). However, that use involves minimal exposure for the worker because the tires are not located where the work is located. In some cases, tires are used in tire warehouses and dealerships where this is potential for exposure. The extent of exposure is unknown.
Potential Exposure	Are there differences in the manufacturing, use, storage, transportation, waste, or end-of-life management of the product and alternatives?	Unclear	Details about how tires could be manufactured with the selected alternative(s) are lacking. It is assumed some differences in manufacturing would be required but the degree of change is unknown at this time. Refer to Table 4.6 for the lifecycle step-by-step consideration.

## Table 4.5 Life Cycle Elements Considered in Evaluating Potential Exposures

Category	Element	Relevant?*	Basis
	Is the product manufactured, stored, or transported through California but not used in California?	No	The product is used in California.
	Is the product an intermediate product used to manufacture an exempted product?	No	No, 6PPD is used to manufacture tires which are not an exempted product.
	Does the product have household use?	No	No, the product is not used inside the home. Tires may be stored in the home ( <i>e.g.</i> , garage) but it is abrasion of tires by roads which results in the release. Thus, even if stored in the home there is minimal potential for exposure.
	Does the product have recreational use?	No	The priority product is not intended for recreational use. At end of life, the product may be used as crumb rubber fill in recreational fields.
	Are there sensitive subpopulations that use the product and alternatives?	Yes	Sensitive populations include workers, sensitized individuals, children, the elderly, and pregnant women. These individuals do use vehicles containing tires in a manner similar to the general population. However, their use does not result in particular exposure. If the use of spent tires as crumb rubber fill is considered, there may be particular types of exposure to certain populations. However, it is not clear if use of spent tires for this purpose falls within the designation of the priority product or whether such materials would need to be designated as their own priority product.
	Is the product used in homes?	No	No, the product is not used inside the home. Tires may be stored in the home ( <i>e.g.</i> , garage) but it is abrasion of tires by roads which results in the release. Thus, even if stored in the home there is minimal potential for exposure.
	Is the product used in schools?	No	Tires are not intended for use inside schools. School buses have tires, but this use does not appear related to the intention of this relevant factor.
	Is the product used in workplaces?	Yes	Tires are used on vehicles used during work ( <i>e.g.</i> , cars used for work, and trucks and buses). However, that use involves minimal exposure for the worker because the tires are not located where the work is located. In some cases, tires are used in tire warehouses and dealerships where this is potential for exposure. The extent of exposure is unknown.
	Is the product used in other unusual locations?	Yes	Spent tires are used in various applications including energy production, retaining walls, marine applications, and as noted above, as an artificial turf component.
	Is there a difference in the frequency, extent, level, and duration of exposure potential for the product and its alternatives during use?	Unclear	Tires will be used in the same manner regardless of whether 6PPD or an alternative is used. If the possible alternative lacks 6PPD/6PPDQ toxicity to certain fish species, the impact of the exposure will be different.
	Is there a difference in the frequency, extent, level, and duration of exposure potential for the product and its alternatives at end-of-life?	Unclear	At this time, it is unclear what the end-of-life implications of 6PPD alternatives are for the end-of-life stage. This will be further explored in Stage 2.
Potential Exposure	Is there a difference in how the candidate chemical is contained within the product and its alternatives?	Potentially	It is anticipated that most possible alternatives would be non-bound in the polymer structure so as to be able to migrate through the tire as needed. Graphene is one exception as it will not migrate through the rubber matrix.

Category	Element	Relevant?*	Basis
	Is there a difference in terms of engineering and administrative	Unclear	To date, we have not identified any engineering or administrative controls involved in
	controls to reduce exposure among the product and its alternatives?		any of the possible alternatives being evaluated.
	Is there a difference in the potential of the candidate chemical and	Yes	As noted in Table 5.9, the possible alternatives have different physical and chemical
	degradation products to release into, accumulate in, and persist in the		properties that could impact their ability to be released into, accumulate in, and persist
	environment?		in the environment.

Notes:

6PPD = N-(1,3-Dimethylbutyl)-N'-phenyl-p- phenylenediamine; DTSC = Department of Toxic Substances Control; PPD = Paraphenylene Diamine.

\* Whether a factor was considered relevant to drawing a distinction between possible alternatives and candidate chemical in the Priority Product was described using five terms. The terms "yes" and "no" were used when the available data indicated a clear material difference would be expected or is obvious. The terms "potentially" and "unlikely" were used when the available data were too limited to be fully confident on whether a material difference can be expected but the available data leaned towards one direction (*i.e.*, towards either yes or no but not reliably). Finally, the term "unclear" was used when data were not available to give an indication of whether a factor would be relevant or not.

### Gradient

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
6PPD	793-24-8	6PPD is produced <i>via</i> the reduction of either P-nitro- or P-nitrosodiphenylamine to form P- aminodiphenylamine, followed by reaction with methyl isobutyl ketone (MIBK) and hydrogenation over a catalyst (PubChem). The former is produced by the by reaction of an alkyl ether of 4-nitrosophenol with aniline (PubChem). MIBK is derived from acetone <i>via</i> the intermediate mesityl oxide. The key ingredients are therefore 4-nitrosophenol, aniline, and acetone. Acetone is produced in multiple ways including oxidation of cumene; dehydrogenation or oxidation of isopropyl alcohol with metallic catalyst; vapor-phase oxidation of butane; by- product of synthetic glycerol production. Aniline is produced from nitrobenzene, which is itself produced by reaction of benzene with forms of nitric acid. Nitrosophenol is produced by the reaction of nitrous acid on phenol; phenol itself is produced by oxidation of cumene. The base ingredients are therefore hydrocarbons such as cumene, benzene, and butane as well as various catalysts and reagents ( <i>e.g.</i> , acids).	Yes. Cumene, benzene, butane, and subsequent chemicals in the production chain are all likely derived from fossil fuel sources.	acetone aniline benzene butane cumene glycerol isopropyl alcohol mesityl oxide metallic catalysts methyl isobutyl ketone P-aminodiphenylamine nitric acid nitrobenzene P-nitrodiphenylamine P-nitrosodiphenylamine 4-nitrosophenol nitrous acid
7PPD (N-(1,4- dimethylpentyl)-N'-phenyl- p-phenylenediamin)	3081-01-4	No PubChem data on manufacturing but is likely to be similar to 6PPD with the use of different isomers. Likely to have the same impact on raw material requirements.	Yes, expected to be similar to 6PPD.	phenol Yes, expected to be similar to 6PPD.
IPPD (N-isopropyl-N'- phenyl-p- phenylenediamine)	101-72-4	Reaction of p-chloronitrobenzene with aniline to yield p-nitrodiphenylamine which is reductively alkylated with acetone over a nickel/chromium catalyst. Chloronitrobenzenes are produced by nitration of chlorobenzene which is itself produced by reaction of benzene with gaseous chlorine in the presence of a catalyst. Aniline is produced from nitrobenzene which is itself produced by reaction of benzene with forms of nitric acid. In both cases, benzene is derived from fossil fuel sources. Acetone is produced in multiple ways including oxidation of cumene; dehydrogenation or oxidation of isopropyl alcohol with metallic catalyst; vapor-phase oxidation of butane; by-product of synthetic glycerol production. The base ingredients are therefore benzene, cumene, nitric acid, an unstated catalyst, chlorine, and isopropyl alcohol.	and subsequent chemicals in the production chain are all	acetone aniline benzene butane chlorine chlorobenzenes chloronitrobenzenes cumene glycerol isopropyl alcohol nickel/chromium catalyst nitric acid nitrobenzene P-nitrodiphenylamine
CPPD (N-cyclohexyl-N'- phenyl-p- phenylenediamine)	101-87-1	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
DPPD (N,N'-diphenyl-p- phenylenediamine)	74-31-7	Prepared by condensing hydroquinone or p-aminophenol with aniline. Hydroquinone is produced <i>via</i> multiple pathways: <i>via</i> hydroxylation of phenol with hydrogen peroxide, <i>via</i> oxidation of aniline with manganese dioxide, or from derivatization of benzene and cumene. Phenol is typically produced via oxidation of benzene or toluene. Aniline is produced from nitrobenzene which is itself produced by reaction of benzene with forms of nitric acid.	Yes. Cumene, benzene, and toluene are all likely derived from fossil fuel sources.	aniline p-aminophenol benzene cumene hydroquinone hydrogen peroxide manganese dioxide nitric acid nitrobenzene phenol toluene
6QDI (N-1,3-dimethyl butyl- N'-phenyl quinone diamine)	52870-46-9	6QDI is the oxidized form of 6PPD (where the nitrogen atoms around the central phenyl ring have double bonds instead of single bond plus a hydrogen atom). It therefore would have the same production process as 6PPD.	Yes, expected to be the same as 6PPD.	Expected to be the same as 6PPD.
TMQ (polymerized 2,2,4- trimethyl-1,2- dihydroquinoline)	147-47-7	Reaction of quinoline and acetone. Quinoline is produced <i>via</i> reaction of aniline with glycerol and nitrobenzene in presence of sulfuric acid. Acetone is produced in multiple ways including oxidation of cumene; dehydrogenation or oxidation of isopropyl alcohol with metallic catalyst; vapor-phase oxidation of butane; and by-product of synthetic glycerol production. Aniline is produced from nitrobenzene which is itself produced by reaction of benzene with forms of nitric acid. Nitrosophenol is produced by the reaction of nitrous acid on phenol; phenol itself is produced by oxidation of cumene. Glycerol is derived from plant matter while nitrobenzene (as noted above) is derived from benzene and nitric acid. The base ingredients are therefore hydrocarbons such as cumene, benzene, and butane as well as various catalysts and reagents ( <i>e.g.</i> , acids).	Yes. Cumene, benzene and butane are all likely derived from fossil fuel sources.	acetone aniline benzene butane cumene glycerol isopropyl alcohol nitric acid nitrobenzene nitrosophenol nitrous acid phenol quinoline sulfuric acid various catalysts
77PD (N,N'-Bis(1,4- dimethylpentyl)-p- phenylenediamine)	3081-14-9	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	Self-condensation of p-octylaniline in the presence of mineral acids such as hydrochloric acid. By analogy with ethylaniline, octyl aniline is likely produced by heating aniline and octanol with sulfuric acid, with subsequent distillation. Aniline is produced from nitrobenzene which is itself produced by reaction of benzene with forms of nitric acid. Hydrochloric acid is produced by reaction of sodium chloride and sulfuric acid or sulfur dioxide or as a by-product of the synthesis of chlorinated hydrocarbons.	Yes. Benzene and other chemicals produced from fossil fuel materials.	aniline benzene chlorinated hydrocarbons hydrochloric acid nitric acid nitrobenzene octanol p-octylaniline sulfuric acid sulfur dioxide

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
CCPD (N,N'-dicyclohexyl-p- phenylenediamine)	4175-38-6	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
DAPD (diaryl-p-phenylene diamine)	68953-84-4	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
DNPDA (N,N'-Di-2-naphthyl- p-phenylenediamine)	93-46-9	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
NBC (Nickel dibutyldithiocarbamate)	13927-77-0	Reaction of aqueous solutions of sodium dibutyldithiocarbamate and nickel chloride, acetate, or sulfate. Sodium dibutyldithiocarbamate is produced by reaction of carbon disulfide with dibutylamine in the presence of aqueous sodium hydroxide. Carbon disulfide is produced by combining sulfur with charcoal or methane. Dibutylamine is produced <i>via</i> reaction of ammonia with butyl bromide or chloride which, in turn, are produced by reacting the halogen gas with natural gas fractions. Sodium hydroxide is produced commercially <i>via</i> electrolysis of sodium chloride. Nickel chloride/acetate/sulfate are produced by reacting nickel ores ( <i>e.g.</i> , nickel oxide) with the requisite acid.	Yes. There are hydrocarbons and other chemicals produced from fossil fuel materials.	acids ammonia butyl bromide or chloride carbon disulfide charcoal dibutylamine methane nickel chloride, acetate, or sulfate nickel ores (e.g., nickel oxide) sodium chloride sodium dibutyldithiocarbamate sodium hydroxide sulfur
Ethoxyquin	91-53-2	Reaction of acetone with p-phenetidine and iodine at 120-130 C. Distillation to isolate ethoxyquin. Acetone is produced in multiple ways including oxidation of cumene; dehydrogenation or oxidation of isopropyl alcohol with metallic catalyst; vapor-phase oxidation of butane; by- product of synthetic glycerol production. p-Phenetidine (ethoxylaniline). No data in PubChem but presumably involves oxidation of aniline. Iodine is obtained from natural sources (brine) often with the use of sulfuric acid and chlorine.	Yes. There are hydrocarbons and other chemicals produced from fossil fuel materials.	acetone aniline butane chlorine cumene ethoxyquin glycerol iodine isopropyl alcohol metallic catalyst p-phenetidine (ethoxylaniline) sulfuric acid

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
Dilauryl thiodipropionate	123-28-4	Reaction of thiodipropionitrile (TDPN) with lauryl alcohol using acid catalysts (hydrochloric acid and sulfuric acid). Lauryl alcohol produced by addition of ethylene to triethylaluminum or hydrogenation of methyl laurate. Also produced <i>via</i> reduction of esters of lauric acid with sodium and absolute alcohol or by reduction of coconut-oil fatty acids. Ethylene is obtained from fossil fuel sources. Hydrochloric acid is produced by reaction of sodium chloride and sulfuric acid or sulfur dioxide or as a by-product of the synthesis of chlorinated hydrocarbons. Sulfuric acid is produced by reacting sulfur dioxide with oxygen or nitric oxide	Yes. There are hydrocarbons and other chemicals produced from fossil fuel materials.	chlorinated hydrocarbons coconut fatty acids ethanol ethylene hydrochloric acid lauric acid lauryl alcohol methyl laurate nitric oxide oxygen sodium chloride sulfur dioxide sulfuric acid thiodipropionitrile (TDPN) triethylaluminum
DTPD (N,N'-Ditolyl-p- phenylenediamine)	68953-84-4	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
1,4-Benzenediamine, N,N'- bis(2-methylphenyl)	15017-02-4	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
N' -Phenyl.N-Fluorenyl- Para-Phenylenediamine	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
N-(p- phenylthiomethylphenyl)- N'-(1,3 dimethyl-butyl)-p- phenylenediamine	None Provided	No PubChem data on manufacturing. However, it contains the 6PPD structure bonded to thioanisole between the terminal benzene moiety of 6PPD and the terminal methyl moiety of thioanisole. It is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle from the production of 6PPD. Additionally, phenyl mercaptan and an alkylating agent are used to manufacture thioanisole (US Patent 4124646A) so it is likely they are also part of the production process. Thus, aside from reacting phenyl mercaptan with an alkylating agent, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD Phenyl Mercaptan
4-(2,5-dimethyl-1H-pyrrol- 1-yl)-N-phenylaniline	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
N,N - (ethane-1,2-diyl) bis (N-phenylbenzene-1 4- diamine; or similar chemical 1-N-[2-(4- anilinoanilino)ethyl]-4-N- phenylbenzene-1,4- diamine	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
4-N-(2,3-dimethylphenyl)-1- N-phenylbenzene-1,4- diamine- R1 and R2 are methyl	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
RU997 Irgazone 997	444992-04-5	No PubChem data on manufacturing. However, CAS 444992-04-05 is a reaction product of N- phenyl-N'-(1,3dimethylbutyl)-p-phenylenediamine with an alkyl glycidylthioether. N-phenyl-N'- (1,3 dimethylbutyl)-p-penylenediamine is a PPD family molecule, so it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have a similar production process as 6PPD.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD glycidylthioether
4-[4-(4-Methylpentan-2- ylamino)anilino]phenol	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
This is a class of compounds. Reference uses case where R1 and R2 are methyl; n,p and q are zero and m=1 and is in the para position. Reference compound is CAS 6358-22-1	6358-22-1	This class of compounds is produced by reaction of N,N-Dimethyl-p-phenylenediamine and phenol. N,N-Dimethyl-p-phenylenediamine is produced <i>via</i> the reduction of p-nitrosodimethylaniline with zinc dust and hydrochloric acid, and the oxidation of cumene. P-nitrosodimethylaniline is produced <i>via</i> reduction of nitrous acid with dimethylaniline. Dimethylaniline is produced from aniline and methanol under pressure in the presence of acidic catalysts. Therefore, the key ingredients are aniline, methanol, and cumene.	Yes, cumene, aniline, and methanol are likely derived from fossil fuel remnants.	Aniline Cumene Dimethylaniline Methanol Nitrous Acid
N,N-diethyl-2,2,4-trimethyl- 1H-quinolin-6-amine (R= N(C2H5)2	None Provided	No PubChem data on manufacturing. However, as the same class of chemical as N,N-Dimethyl-p- phenylenediamine it is likely that chemicals such as dimethylaniline, nitrous acid, aniline, methanol, and cumene are all parts of the production cycle. Thus, it appears likely to have similar production processes as other compounds in the same family as N,N-Dimethyl-p- phenylenediamine.	Yes, cumene, aniline, and methanol are likely derived from fossil fuel remnants.	Expected to be similar to Phenol, 4-[[4- (dimethylamino)phenyl]a mino]-
Mixed xylene diamines N,N'-Dibenzyl-p-xylene- alpha,alpha'-diamine-	25790-41-4	No PubChem data on manufacturing. However, as the same class of chemical as benzylamine it is likely that chemicals such as chlorine, toluene, and ammonia are all parts of the production cycle. Thus, it appears likely to have similar production processes as other compounds in the Benzylamine family. Benzylamine is produced by the reaction of benzyl chloride with ammonia in aqueous solution. Benzyl chloride is produced by passing chlorine over boiling toluene, and then washing with water.	Yes. Toluene is produced during petroleum refining operations.	Ammonia Benzylamine Benzyl chloride Chlorine Toluene

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
2,4,6-tris-(N-1,4- dimethylpentyl-para- phenylenediamino)- 1,3,5triazine, TAPDT	121246-28-4	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
N-Phenyl-1-naphthylamine	90-30-2	Reaction of alpha-napthylamine (CAS 134-32-7) with aniline, and purified <i>via</i> distillation. Aniline is produced from nitrobenzene which is itself produced by reaction of benzene with forms of nitric acid. Alpha-napthylamine is produced from either reduction by catalytic hydrogenation with a nickel catalyst or being reduced with iron in hydrochloric acid. 1- nitronapthalene in turn is prepared by the action of acids (nitric and sulfuric) on ground napthalene. The key ingredients are therefore benzene, nitric acid, napthalene, nitric acid, and hydrochloric acid.	Yes. Napthalene is derived from coal tar.	Aniline Benzene Hydrochloric Acid Metallic Catalyst Napthalene α-Napthylamine Nitric Acid Nitrobenzene 1-Nitronapthalene Sulfuric Acid
N-Phenyl-2-naphthylamine	135-88-6	<ul> <li>Produced either by heating 2-naphthol with aniline hydrochloride, or condensation of 2-naphthol and aniline in the presence of a catalyst.</li> <li>2-Napthol is produced by caustic fusion of napthalene-2-sulfonic acid. No PubChem data is available on the production of napthlene-2-sulfonic acid; however, it is likely this process involves napthalene and sulfonic acid. Aniline is produced from nitrobenzene which is itself produced by reaction of benzene with forms of nitric acid. Aniline hydrochloride is produced from a reaction of aniline and hydrochloride. The key ingredients are therefore benzene, aniline, hydrochloride, napthalene, sulfonic acid, and nitric acid.</li> </ul>	Yes. Napthalene is derived from coal tar.	Aniline Aniline Hydrochloride Benzene Benzenamine Hydrochloride Napthalene 2-Naphthol Nitric Acid Nitrobenzene Sulfonic Acid
Irganox 1520	110553-27-0	The only available PubChem manufacturing data is a TSCA Commercial Activity Status for Phenol. CAS 110553-27-0 can thus be defined as a phenol family compound. It is likely that chemicals such as cumene and oxygen are parts of the production cycle. Thus, it is likely that production involves oxidation of cumene with oxygen to cumene hydroperoxide, and cleavage of cumene hydroperoxide in an acidic medium to phenol and acetone. The compound, additionally, has two long alkyl sulfur chains. However, there is no manufacturing information available for these structures on PubChem.	Yes. Cumene is likely derived from fossil fuel sources.	Cumene Oxygen
Graphene	1034343-98-0	Produced from organic materials that are rich in carbon, such as coal, lignite, wood, nut shells, peat, pitches, and cokes. Manufacturing is done through either thermal activation or chemical activation.	Some carbon sources are fossil fuel based.	Carbon

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
1,1' - Pentamethylenebis(2,2-Di- n- Butylhydrazine)	None Provided	No PubChem data on manufacturing. However, 1,1' - pentamethylenebis(2,2-Di-n- Butylhydrazine) is a hydrazine, so it is likely that chemicals such as chlorine, sodium hydroxide, and ammonia are parts of the production cycle. The production of hydrazine involves the reaction of sodium hypochlorite and ammonia to yield chloramine and sodium hydroxide, followed by the reaction of chloramine, ammonia, and sodium hydroxide to yield hydrazine, sodium, chloride, and water. Sodium hypochlorite is produced when chlorine is added to a cold dilute solution of sodium hydroxide.	Hydrazine is not fossil fuel based, but compound has large alkyl groups that are likely fossil fuel based.	Ammonia Chloramine Chlorine Hydrazine Sodium Hydroxide Sodium Hypochlorite
α- C-4- hydroxy- 3,5- dimethylphenyl (Lowinox WSP - 77-62-3). No number for nitrone - N-isopropyl nitrone and Lowinox WSP	77-62-3 No CAS for N- isopropyl nitrone	No PubChem data on manufacturing. However, CAS 77-62-3 is a phenol family compound, so it is likely that chemicals such as cumene and oxygen are parts of the production cycle. Thus, it is likely that production involves oxidation of cumene with oxygen to cumene hydroperoxide, and cleavage of cumene hydroperoxide in an acidic medium to phenol and acetone. No PubChem manufacturing information is available for N-isopropyl nitrone. However, N- isopropyl nitrone is likely a formaldehyde family compound because Nitrone (75-17-2) is defined as a formaldehyde family compound by the TSCA. It is then likely that chemicals such as methanol are a part of the production process.	Yes. Cumene is likely derived from fossil fuel sources.	Cumene Formaldehyde Methanol Oxygen
N-(4-methylpentan-2-yl)- 10H-phenothiazin-3-amine	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
7-(4-methylpentan-2- ylamino)-2,3,4,10- tetrahydro-1H-acridin-9- one	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
2-cyclohexyl-N-(4- methylpentan-2-yl)-1H- indol-5-amine	None Provided	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD
4-(1H-indol-2-yl)-N-(4- methylpentan-2-yl)aniline	None Provided	No PubChem data on manufacturing. However as an aniline-family molecule it is likely that chemicals such as nitrobenzene and benzene are both parts of the production cycle. It is likely that the nitration of benzene with mixed acid, and the hydrogenation of nitrobenzene are both parts of the production process. As this chemical has the same MIBK derivative tail structure as 6PPD it is likely that acetone is also a key component of the manufacturing process.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
α- C-4- Hydroxy- 3,5- dimethylphenyl- N-tert. butyl nitrone	None Provided	No PubChem manufacturing information is available for $\alpha$ - C-4- hydroxy- 3,5- dimethyl phenyl- N-tert. butyl nitrone. However, this chemical is likely a formaldehyde family compound because nitrone (CAS 75-17-2) is defined as a formaldehyde family compound by the TSCA. According to the patent related to identification of this possible alternative (see Appendix D), the chemical was produced by reaction of N-tert-butyl-hydroxylamine and 3,5-dimethyl-4-hydroxybenzaldehyde in ethanol. No information could be located in PubChem for these two reagents but benzaldehydes in general are reported to be ultimately derived from toluene.	Yes. Toluene is likely derived from fossil fuel sources.	t-Butyl hydroxylamine Dimethyl benzaldehyde Ethanol Toluene
Amine functionalized lignin	None Provided	Lignin is a waste product of wood processing. Lignin must be isolated from the wood pulp and this appears to be done using various solvents such as sodium sulfide, dioxane, or methanol, along with acid catalysts or formaldehyde (Bertella and Luterbacher, 2020). The specific isolation process yields different results and would need to be optimized for an antiozonant. Formaldehyde and urea or azides have been suggested as possible reagents for amination of lignin (Bertella and Luterbacher, 2020) although the specific of amine functionalization to produce anti-ozone activity are not known. It should also be noted that as lignin is a biological material, consistency in material over time is a concern.	Not if lignin is obtained from wood product production waste.	Azides Dioxane Formaldehyde Methanol Sodium sulfide Urea
Rambutan peel extract	None Provided	Rambutan peel extract is extracted from the peel of the tropic fruit, rambutan. Peels are washed, dried, and then extracted with various solvents such as water/ethanol or methanol. Hydrochloric acid and sodium hydroxide are also reported to be used during extraction (Zhang <i>et al.</i> , 2022). Purification of the crude extract has been conducted with purification resins, High-performance liquid chromatography (HPLC), <i>etc</i> . It seems likely that other methods would be required to produce extract at scale. Only one study is available on rambutan peel extract's potential antioxidant/antiozonant properties in tires (Sukatta <i>et al.</i> , 2021), in which the active ingredients responsible for potential antioxidant or antiozonant properties are not confidently identified. It is unclear if the existing volume of discarded rambutan peel would be able to produce enough extract to replace GPPD in tires. It also seems likely that synthesis of the active ingredients (once these are known) to replace GPPD in tires would likely involve use of fossil fuel based chemistry.	Not if produced solely from agricultural waste; yes if active ingredients are synthesized.	Ethanol Hydrochloric acid Methanol Possible fossil fuel based precursors Sodium hydroxide
N-1-Methylheptyl-N'- phenyl-p- phenylenediamine (8PPD)	15233-47-3	No PubChem data on manufacturing. However, as a PPD-family molecule it is likely that chemicals such as aniline, nitrobenzene, and benzene are all parts of the production cycle. Thus, it appears likely to have similar production processes as the other PPDs.	Yes, expected to be similar to 6PPD.	Expected to be similar to 6PPD

Chemical	CAS No.	Production Process	Precursor Fossil Fuel Based?	Possible Chemicals Involved Across Lifecycle
Octyl gallate	1034-01-1	By analogy with propyl gallate, possible production pathways for octyl gallate likely include esterification of gallic acid with octyl alcohol followed by separation <i>via</i> distillation. Octyl alcohol can be produced from natural sources (coconut oil and other oils or fatty acids) or from fossil fuel derived ethylene or 1,3-butadiene <i>via</i> polymerization and subsequent oxidation. Gallic acid (trihydroxy benzoic acid) is produced from chemical or enzymatic hydrolysis of tannins, a by-product of paper and pulp production. The structurally-related hydroxybenzoic acid is produced by oxidation of p-cresol, by carboxylation of potassium phenolate, and by interaction of 1-aminobenzoic acid and nitrous acid; all processes that are fossil fuel based. Whether similar processes could be commercially employed to produce gallic acid/trihydroxy benzoic acid is unclear.	Possibly if naturally derives sources are insufficient to meet demand.	Coconut oil Plant based oils Ethylene Butadidene Tannins p-Cresol (possibly) Potassium phenolate (possibly) Aminobenzoic acid (possibly)
Nano calcium carbonate surface modified by gallic acid	None Provided	Gallic acid modified calcium carbonate nanomaterials are produce by reaction of calcium carbonate nanoparticles with gallic acid. Calcium carbonate nanoparticles can be produced by mechanical processing of bulk calcium carbonate ( <i>e.g.</i> , crushing, milling, or grinding) or by controlled reactions of solutions of bicarbonate ion and calcium ( <i>e.g.</i> , sodium bicarbonate and sodium chloride) (Fadia <i>et al.</i> , 2021). The starting materials for these processes are all obtained by mining. Gallic acid (trihydroxy benzoic acid) is produced from chemical or enzymatic hydrolysis of tannins, a by-product of paper and pulp production. The structurally-related hydroxybenzoic acid is produced by oxidation of p-cresol, by carboxylation of potassium phenolate, and by interaction of 1-aminobenzoic acid and nitrous acid; all processes which are fossil fuel-based. Whether similar processes could be commercially employed to produce gallic acid/trihydroxy benzoic acid is unclear.	Possibly if naturally derived sources are insufficient to meet demand for gallic acid.	Calcium carbonate Sodium bicarbonate Sodium chloride Tannins p-Cresol (possibly) Potassium phenolate (possibly) Aminobenzoic acid (possibly)
Specialized carbon nanotube mixture	None Provided	Carbon nanotubes are produced by various methods, the most common being chemical vapor deposition (CVD) (Mubarak <i>et al.</i> , 2014). All production requires a source of carbon, in the case of CVD, gases such as methane, carbon monoxide or ethylene are used (Mubarak <i>et al.</i> , 2014; Zhuo <i>et al.</i> , 2018). These are typically obtained from fossil fuels although non-fossil fuel sources ( <i>e.g.</i> , biogas) have been discussed. Whether these non-fossil fuel sources can meet production needs is unclear. The production process also involves metal catalysts.	Possibly if naturally derived carbon sources are insufficient to meet needs or are less economical.	Ethylene Methane Carbon monoxide Various metal catalysts

CAS No. = Chemical Abstracts Service Number; HPLC = High-Performance Liquid Chromatography; MIBK = Methyl isobutyl Ketone; PPD = Paraphenylene Diamines; TSCA = Toxic Substances Control Act.

Table 5.1 Chemical	Table 5.1 Chemical-Specific Human Health Hazards (Group A Endpoints)									Group A Endpoints						
Class of Compound	Chemical	CAS	Score <sup>1</sup>	Reference	Acute Mammalian Toxicity	Carcinogenicity	Eye Irritation/ Corrosion	Skin Irritation/ Corrosion (Dermatotoxicity)	Germ Cell Mutagenicity	Target Organ Toxicity – Single Exposure	Target Organ Toxicity – Repeated Exposure	Reproductive/ Developmental Toxicity	Sensitizer – Respiration <sup>2</sup>	Sensitizer – Skin	Endocrine Disruptor	California Proposition 65
Current Priority Prod	ict Chemical of Concern															
Phenylene Diamine related	N-(1,3-dimethylbutyl)-N'-phenyl-p- phenylenediamine (6PPD)	793-24-8		ECHA, 2024; ToxServices, 2021a; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European	Oral: Cat. 4; Dermal: Not Classified; Inhalation: DG (ECHA, 2024; ToxServices, 2021a)	Not Classified	Not Classified (ECHA, 2024): Cat. 3 (ToxServices, 2021a)	Not Classified	Not Classified	Not Classified (ECHA. 2024); DG (ToxServices, 2021a)	Not Classified (ECHA, 2024): Cat. 2 (ToxServices, 2021a)	Cat. 1B (ECHA, 2024)	DG (ECHA, 2024): Cat. 1B based on skin sensitization and respiratory structural alert (ToxServices, 2021a)	<u>Cat. 1 (ECHA, 2024);</u> <u>Cat. 1A (ToxServices,</u> <u>2021a)</u>	Not Listed (EU): Moderate based on altered female pubertal development in rats (ToxServices, 2021a)	Not Listed
				Communities, 2001												
Potential Alternative			125		5	0	5	0	0	0	10	50	5	25	25	0
Phenylene Diamine	N-(1,4-dimethylpentyl)-N'-phenyl-p-	3081-01-4		ECHA, 2024;	Oral: Not Classified	Not Classified	Not Classified (ECHA,	Not Classified	Not Classified	Not Classified	Not Classified (ECHA,	Cat. 1B (ECHA, 2024;	DG (ECHA, 2024); Cat. 1B	Cat. 1 (ECHA, 2024);	Not Listed (EU); Moderate based	Not Listed
	phenylenediamine (7PPD)			ToxServices, 2021b; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Dermal: Not Classified; Inhalation: DG (ECHA, 2024; ToxServices, 2021b)		2024): Cat. 3 (ToxServices, 2021b)				2024); Cat. 2 (ToxServices, 2021b)	ToxServices, 2021b)	based on skin sensitization and respiratory structural alert (ToxServices, 2021b)	Cat. 1A (ToxServices, 2021b):	on altered female pubertal development in rats for surrogate 6PPD CAS 793-24-8 (ToxServices, 2021b)	
Phenylene Diamine	N-isopropyl-N'-phenyl-p-phenylenediamine (IPPD)	101-72-4	120	ECHA, 2024; ToxServices, 2021c; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	0 Oral: Cat. 4; Dermal: Not Classified; Inhalation: DG (ECHA, 2024; ToxServices, 2021c)	0 <u>DG (ECHA, 2024); Not</u> <u>classified based on</u> <u>surrogate 6PPD</u> <u>(ToxServices, 2021c)</u>	5 <u>Not Classified (ECHA, 2024); Cat. 2B</u> ( <u>ToxServices, 2021c)</u>	0 Not Classified	0 Not Classified	0 Not Classified (ECHA, 2024); DG (ToxServices, 2021c)	10 Not Classified (ECHA, 2024); Cat. 2 (ToxServices, 2021c)	50 Not Classified (ECHA, 2024); Cat. 1B based on 6PPD (ToxServices, 2021c)	5 <u>DG (ECHA, 2024); Cat. 1B</u> <u>based on skin sensitization</u> <u>and respiratory structural</u> <u>alert (ToxServices, 2021c)</u>	25 <u>Cat. 1 (ECHA, 2024);</u> <u>Cat. 1A (ToxServices,</u> <u>2021c)</u>	25 <u>Not Listed (EU); Moderate based</u> <u>on altered female pubertal</u> <u>development in rats for surrogate</u> <u>6PPD CAS 793-24-8. In silico</u> <u>modeling reported IPPD to be a</u> <u>potential endocrine receptor</u> <u>modulator (ToxServices, 2021c)</u>	0 Not Listed
			150		5	25	5	0	0	0	10	50	5	25	25	0
Phenylene Diamine	N-cyclohexyl-N'-phenyl-p- phenylenediamine (CPPD)	101-87-1		CalOEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though very little data are available for this chemical	Not Listed
			Not assigned based on complete data gap													
Phenylene Diamine	N,N'-diphenyl-p-phenylenediamine (DPPD)	74-31-7		ECHA, 2024; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Oral: Not Classified; Dermal: Not Classified; Inhalation: Not Classified	Not Classified	Not Classified	Not Classified	Cat. 2	Not Classified	Not Classified	Cat. 2	DG	Cat. 1	Not Listed (EU)	Not Listed
Phenylene Diamine	N-1,3-dimethyl butyl-N'-phenyl quinone	52870-46-9	80	ECHA, 2024;	0 Oral: Not Classified;	0 DG (ECHA, 2024); Not	0 Cat. 2	0 Not Classified	25 DG (ECHA, 2024);	0 DG (ECHA, 2024); Cat. 3	0	25 DG (ECHA, 2024): Cat.	5 DG	25 Not classified (ECHA,	0 Not Listed (EU); Moderate based	0 Not Listed
related	diimine (6QDI)	52070-40-5		ToxServices, 2021d; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Dermal: Not Classified; Inhalation: DG	classified based on surrogate 6PPD (ToxServices, 2021d)			Not classified based on surrogate 6PPD (ToxServices, 2021d)	(ToxServices, 2021d	based on surrogate 6PPD (ToxServices, 2021d)	1B based on 6PPD (ToxServices, 2021d)		2024); Cat. 1A based on surrogate 6PPD (ToxServices, 2021d)	on altered female pubertal development in rats for surrogate 6PPD (ToxServices, 2021d)	
Dihydroquinoline	Polymerized 2,2,4-trimethyl-1,2- dihydroquinoline (TMQ Oligomer)	26780-96-1	175	ECHA, 2024; ToxServices, 2021e; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	0 Oral: Cat. 4 (based on surrogate ethoxyquin CAS 91-53-2) (ToxServices, 2021e); Not classified (ECHA, 2024); Dermal: Not classified (based on surrogate ethoxyquin CAS 91-53-2), Not classified (ECHA, 2024); Inhalation: Not classified (based on surrogate ethoxyquin CAS 91-53-2); DG (ECHA, 2024)	25 <u>Cat. 2 (ToxServices,</u> 2021e): Not classified (ECHA, 2024)	5 Not classified	0 Not classified	25 Not classified	5 Not classified (ECHA, 2024); Cat. 1 based on surrogate ethoxyquin CAS 91-53-2 even though the more similar surrogate TMDHQ, oligomers (26780-96-1) was not a hazard for both oral and dermal (ToxServices, 2021e)	10 <u>Not classified (ECHA,</u> <u>2024): Cat. 1</u> (ToxServices, 2021e)	50 <u>Not classified (ECHA,</u> 2024): Cat. 2 based on a surrogate TMDHQ, oligomers (26780-96- 1) and surrogate ethoxyquin CAS 91-53- 2 (ToxServices, 2021e)	5 DG (ECHA, 2024); Not classified (ToxServices, 2021e)	25 Not classified based on surrogate TMDHQ, oligomers (26780-96-1) (ToxServices, 2021e); Not classified (ECHA, 2024)	25 Not Listed (EU); DG (ToxServices, 2021d)	0 Not Listed
Phonylono Diamina	N N' Pic(1.4 dimethylpontyl) p	3081-14-9	170	ECHA, 2024;	5 Oral: Cat. 4;	75 Not Classified	0 Not Classified	0 Not Classified	0 Not Classified	25 Not Classified	25 Not Classified (ECHA,	25 Not Classified (ECHA,	5 DC (ECHA 2024): Cat 1P	0 Cat. 1	10 Not Listed (EU); DG (ToxServices,	0 Not Listed
Phenylene Diamine	N,N'-Bis(1,4-dimethylpentyl)-p- phenylenediamine (77PD)	5061-14-9		Communities, 2001	Dermal: Not Classified; Inhalation: DG						2024): Cat. 2 (ToxServices, 2021f)	2024): Cat. 2 for Developmental toxicity (ToxServices, 2021f)	DG (ECHA, 2024); Cat. 1B based on skin sensitization and respiratory structural alert (ToxServices, 2021f)		2021f)	
Diphenyl amine	4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	80	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	5 DG	0 DG	0 DG	0 DG	0 DG	0 DG	10 DG	25 DG	5 DG	25 DG	10 Not Listed (EU) though no data are available for this chemical	0 Not Listed
			Not assigned based on complete data gap													
Phenylene diamine	N,N'-Ditolyl-p-phenylenediamine (Commercial DTPD)	68953-84-4		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Oral: Not Classified; Dermal: Not Classified; Inhalation: Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Cat. 2	DG	Cat. 1B	Not Listed (EU)	Not Listed
Phenylene Diamine	N,N'-Dicyclohexyl-p-phenylenediamine	4175-38-6	55	ToxServices, 2021g;	0 Oral: Cat. 3 (based on surrogate 44PD)	0 Not classified based	0 Cat. 2A based on	0 Cat. 1 based on surrogate	0 Not classified	DG	0 Cat. 1 based on	25 Cat. 2 based on	5 Cat 1B based on surrogate	25 Cat. 1A based on	0 Not Listed (EU); DG (ToxServices,	0 Not Listed
	(CCPD)			Cal OEHHA, 2021g, Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Dermal: Cat. 3 (based on surrogate 44PD) Inhalation: DG		surrogate 44PD	44PD			surrogate 44PD	surrogate 44PD	44PD's dermal sensitization data and respiratory sensitization structural alert (ToxServices, 2021g)	surrogate 44PD	2021g)	
			115		10	0	5	5	0	5	25	25	5	25	10	0

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Group A Endpoints												,				
Class of Compound	Chemical	CAS	Score <sup>1</sup>	Reference	Acute Mammalian Toxicity	Carcinogenicity	Eye Irritation/ Corrosion	Skin Irritation/ Corrosion (Dermatotoxicity)	Germ Cell Mutagenicity	Target Organ Toxicity – Single Exposure	Target Organ Toxicity – Repeated Exposure	Reproductive/ Developmental Toxicity	Sensitizer – Respiration <sup>2</sup>	Sensitizer – Skin	Endocrine Disruptor	California Proposition 65
Phenylene diamine	Diaryl-p-phenylene diamine (DAPD is a class, main commercial DAPD is DTPD CASRN 68953-84-4)	68953-84-4		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Oral: Not Classified; Dermal: Not Classified; Inhalation: Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Cat. 2	DG	Cat. 1B	Not Listed (EU)	Not Listed
Phenylene diamine	N,N'-Di-2-naphthyl-p-phenylenediamine (DNPDA, CASRN 93-46-9)	93-46-9	55	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	0 DG	DG	0 DG	0 DG	0 DG	DG	DG	25 DG	5 DG	25 DG	0 Not Listed (EU) though no data are available for this chemical	0 Not Listed
Metal dithiocarbamate	Nickel dibutyldithiocarbamate (NBC) <sup>3</sup>	13927-77-0	Not assigned based on complete data gap	ECHA, 2024; ToxServices, 2021h; Cal OEHHA, 2023; Danish EPA, 2023; Commission of the European Communities, 2001	Oral: Not Classified; Dermal: Not Classified; Inhalation: Not Classified	<u>Cat. 2 (ECHA, 2024);</u> <u>Cat. 1 (ToxServices, 2021h)</u>	Cat. 2 (ECHA, 2024); Cat. 2A (ToxServices, 2021h)	Not Classified	Not Classified	Not Classified	DG (ECHA, 2024); Cat. 1 (ToxServices, 2021h)	Not Classified (ECHA, 2024); Cat. 2 Repro (ToxServices, 2021h)	DG (ECHA, 2024); Cat. 1B based on professional judgement even though compound is not a dermal sensitizer and does not trigger structural alerts (ToxServices, 2021h)	Not Classified	Not Listed (EU); DG (ToxServices, 2021h)	Carcinogen as Nickel Compound
Dihydroquinoline	Ethoxyquin	91-53-2	170	ECHA, 2024; ToxServices, 2021i; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	0 <u>ECHA:</u> <u>Oral: Cat. 4;</u> <u>Dermal: Not Classified:</u> <u>Inhalation: DG</u> <u>ToxServices:</u> <u>Oral: Cat. 4;</u> <u>Dermal: Cat. 4;</u> <u>Inhalation: Cat. 4</u>	100 Not Classified	5 Not Classified	0 Not Classified	0 Not Classified	0 <u>Not Classified (ECHA, 2024);</u> . <u>Cat. 1 (ToxServices, 2021i)</u>	25 <u>Not Classified (ECHA,</u> <u>2024); Cat. 1</u> <u>(ToxServices, 2021i)</u>	25 Not Classified (ECHA, 2024); Cat. 2 (ToxServices, 2021i)	5 DG	0 <u>Not Classified (ECHA, 2024); Cat. 1B</u> (ToxServices, 2021i)	10 Not Listed (EU); Moderate based on "antiandrogenic effects exhibited in an in vitro screening, study of 200 pesticides, and some positive high throughput in vitro screening assays for estrogen receptor, androgen receptor, steroidogenesis, and thyroid receptor activities" and TEDX listing (ToxServices, 2021)	75 (see note 2) Not Listed
Sulfur compound	Dilauryl thiodipropionate	123-28-4	135	ECHA, 2024; ToxServices, 2021j; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	5 Oral: Not Classified; Dermal: Not Classified; Inhalation: DG	0 DG (ECHA, 2024); Not classified based on modeling (ToxServices 2021j)	0 Not Classified	0 Not Classified	0 Not Classified	25 Not Classified	25 Not Classified	25 Not Classified	5 DG (ECHA, 2024); Not classified based on negative skin sensitization data (ToxServices, 2021j)	25 Not Classified	25 Not Listed (EU); DG (ToxServices, 2021j)	0 Not Listed
Phenylene Diamine	N' -Phenyl.N-Fluorenyl-Para- Phenylenediamine	No CAS	40 Not assigned based on	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	0 DG	25 DG	0 DG	0 DG	0 DG	0 DG	DG	0 DG	5 DG	0 DG	10 Not Listed (EU) though no data are available for this chemical	0 Not Listed
Phenylene Diamine	N-(p-phenylthiomethylphenyl)-N'-(1,3 dimethyl-butyl)-p-phenylenediamine	No CAS	complete data gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
Phenylene Diamine	4-(2,5-dimethyl-1H-pyrrol-1-yl)-N- phenylaniline	No CAS	complete data gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
Phenylene Diamine	N,N - (ethane-1,2-diyl ) bis (N- phenylbenzene-1 4-diamine [example chemical from patent]	No CAS	complete data gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
Phenylene Diamine	4-N-(2,3-dimethylphenyl)-1-N- phenylbenzene-1,4-diamine- R1 and R2 are methyl	No CAS	complete data gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
Phenylene Diamine (Kruger)	RU997, Irgazone 997 (Reaction product of N-phenyl-N'-{1,3dimethylbutyl}-p- phenylenediamine with an alkyl glycidylthioether)	444992-04-5	Not assigned based on complete data gap	NZ Environmental Risk Management Authority, 2005; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	Irgazone 997: Skin sensitizer Cat. 1	Not Listed (EU) though no data are available for this chemical	Not Listed
Phenylene Diamine	4-[4-(4-Methylpentan-2- ylamino)anilino]phenol	No CAS	130	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	5 25 DG	5 5 DG	5 DG	25 DG	5 DG	5 DG	25 DG	5 DG	25 DG	Not Listed (EU) though no data are available for this chemical	Not Listed
Phenylene diamine	Representative example from class (4-((4- (dimethylamino)phenyl)amino)phenol)	6358-22-1	Not assigned based on complete data gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on complete data gap													

							-	-	-	Group A Endpoints						
Class of Compound	Chemical	CAS	Score <sup>1</sup>	Reference	Acute Mammalian Toxicity	Carcinogenicity	Eye Irritation/ Corrosion	Skin Irritation/ Corrosion (Dermatotoxicity)	Germ Cell Mutagenicity	Target Organ Toxicity – Single Exposure	Target Organ Toxicity – Repeated Exposure	Reproductive/ Developmental Toxicity	Sensitizer – Respiration <sup>2</sup>	Sensitizer – Skin	Endocrine Disruptor	California Proposition 65
Dihydroquinoline	N,N-diethyl-2,2,4-trimethyl-1H-quinolin-6- amine (R = N(C2H5)2	No CAS	Not assigned based on	Cal OEHHA, 2023; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			complete data gap													
Hindered amine	N,N'-Dibenzyl-p-xylene-alpha,alpha'- diamine-	25790-41-4	Not assigned based on	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			complete data gap													
Triazine	2,4,6-tris-(N-1,4-dimethylpentyl-para- phenylenediamino)-1,3,5triazine (Durazone 37 or TAPDT)	121246-28-4		ECHA, 2024; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Oral: Not Classified; Dermal: Not Classified; Inhalation: Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	DG	Not Classified	Cat. 1B	Not Listed (EU)	Not Listed
Phonyloaphthyl	N-Phenyl-1-naphthylamine	90-30-2	50	ECHA, 2024;	0 Oral: Cat. 4;	0 Not Classified	0 Not Classified	0 Not Classified	0 Not Classified	0 Not Classified	0 Cat. 2	25 Not Classified	0 DG	25 Cat. 1B	0 Not Listed (EU)	0 Not Listed
Phenylnaphthyl amines	м-ғленун-т-парпскуюнние	90-30-2		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities. 2001	Dermal: Not Classified; Inhalation: DG											
Phenylnaphthyl	N-Phenyl-2-naphthylamine	135-88-6	45	ECHA, 2024;	5 Oral: DG;	0 Cat. 2	0 Cat. 2	0 Cat. 2	0 DG	0 DG	10 DG	0 DG	5 DG	25 Cat. 1	0 Not Listed (EU)	0 Not Listed
amines				Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Dermal: DG; Inhalation: DG		ut 2							cut I		Not Listed
			180		5	75	5	5	25	5	5	25	5	25	0	0
Phenol	[2-Methyl-4,6-bis((octylthio)methyl)phenol (Irganox 1520) <sup>4</sup>	110553-27-0		ECHA, 2024; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	Oral: Not Classified; Dermal: Not Classified; Inhalation: DG	DG	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	DG	Not Classified	Not Listed (EU)	Not Listed
Inergonia	Crankana	1034343-98-0	30	ECHA, 2024;	0 Oral: Not Classified;	25	0	0 Not Classified	0 Not Classified	0 Not Classified	0 Not Classified	0	5	0	0	0 Not Listed
Inorganic	Graphene	1034343-30-0		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities 2001	Dermal: DG; Inhalation: Not Classified	DG	Not Classified	Not Classified				DG	DG	Not Classified	Not Listed (EU)	
Hydrazine	1,1' -Pentamethylenebis(2,2-Di-n-	No CAS	55	Cal OEHHA, 2024;	0 DG	25 DG	0 DG	0 DG	0 DG	0 DG	0 DG	25 DG	5 DG	0 DG	0 Not Listed (EU) though no data	0 Not Listed
	Butylhydrazine)		Not assigned based on	Danish EPA, 2023; Commission of the European Communities, 2001											are available for this chemical	
			complete data gap													
Nitrone + Phenolic AO	α- C-4- hydroxy- 3,5- dimethylphenyl - N-isopropyl and Lowinox WSP	Nitrone as a class, no CAS and Lowinox WSP - 77-62-3		ECHA, 2024; Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities. 2001	Oral: Not Classified; Dermal: Not Classified; Inhalation: Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	Not Classified	DG	Not Classified	Not Listed (EU)	Not Listed
Phenothiazine	N-(4-methylpentan-2-yl)-10H-phenothiazin-	No CAS	5	Cal OEHHA, 2024;	0 DG	0 DG	0 DG	0 DG	0 DG	0 DG	0 DG	0 DG	5 DG	0 DG	0 Not Listed (EU) though no data	0 Not Listed
	3-amine		Not assigned based on	Communities, 2001											are available for this chemical	
Amine	7-(4-methylpentan-2-ylamino)-2,3,4,10- tetrahydro-1H-acridin-9-one	No CAS	complete data gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on complete data gap													

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										Group A Endpoints						
Class of Compound	Chemical	CAS	Score <sup>1</sup>	Reference	Acute Mammalian Toxicity	Carcinogenicity	Eye Irritation/ Corrosion	Skin Irritation/ Corrosion (Dermatotoxicity)	Germ Cell Mutagenicity	Target Organ Toxicity – Single Exposure	Target Organ Toxicity – Repeated Exposure	Reproductive/ Developmental Toxicity	Sensitizer – Respiration <sup>2</sup>	Sensitizer – Skin	Endocrine Disruptor	California Proposition 65
Amine	2-cyclohexyl-N-(4-methylpentan-2-yl)-1H- indol-5-amine	No CAS		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on complete data gap													
Amine	4-(1H-indol-2-yl)-N-(4-methylpentan-2- yl)aniline	None provided		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on complete data gap													
Nitrone	α- C-4- Hydroxy- 3,5- dimethylphenyl-N- tert. butyl nitrone	No CAS		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on complete data gap													
Polymeric amine functionalized lignin	Amine functionalized lignin	No CAS		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on													
Gallate related	Rambutan peel extract	No CAS	complete data gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on													
Gallate related	Octyl gallate	1034-01-1	complete data gap	EFSA, 2015, ECHA, 2024 Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European	Oral: Cat. 4; Dermal: DG Inhalation: DG	Not Classified	Cat. 1 (based on propyl gallate)	Not Classified (based on propyl gallate)	on propyl gallate)		Not Classified (based on propyl gallate)	Not Classified	DG	Cat. 1	Not Listed (EU)	Not Listed
Gallate related	Nano calcium carbonate surface modified	No CAS	40	Cal OEHHA, 2024;	5 DG	DG	0 5 DG	0 DG	0 DG	0 DG	0 DG	0 DG	5 DG	25 DG	0 Not Listed (EU) though no data	Not Listed
	by gallic acid			Danish EPA, 2023; Commission of the European Communities, 2001											are available for this chemical	
			Not assigned based on complete data gap													
Phenylene Diamine	N-1-Methylheptyl-N'-phenyl-p- phenylenediamine (8PPD or UOP 688)	15233-47-3		Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on complete data gap													
Inorganic	Specialized carbon nanotube mixture	No CAS	Comprete trata gap	Cal OEHHA, 2024; Danish EPA, 2023; Commission of the European Communities, 2001	DG	DG	DG	DG	DG	DG	DG	DG	DG	DG	Not Listed (EU) though no data are available for this chemical	Not Listed
			Not assigned based on													
Notes:			complete data gap		1											

Notes: CAS = Chemical Abstracts Service Number; Cat. = Category; CLP = Classification, Labelling, and Packaging Regulation; DG = Data Gap; ECHA = European Chemicals Agency; EU = European Union; GHS = Globally Harmonized System of Classification and Labelling of Chemicals. (1) This is assuming that the concentration of the chemical in the tire is <10%.

(2) ToxService's hazard assignments, but did not score the endpoint based on respirator sensitization assignment. Instead, a data gap score was assigned. See report Section 5.1.2 for more details.

(3) NBC CAS 13927-77-0 is classified Cat. 1 for carcinogenicity under ToxServices GS, 2021 and Cat. 2 under ECHA dossier. It is also on California Prop 65 as a carcinogen. The highest score was counted (i.e., Cat. 1) and counted only once.

(4) A potential alternative is Irganox 1520 CAS 110553-27-0 blended with Vulcazon AFS. However, according to the source patent (Pirelli Tyre S.P.A, 2018), the best ratio is where 100% of blend is Irganox 1520. Additionally, no data were located for Vulcazon AFS. Thus, data and scoring is 100% based on Irganox 1520 CAS 110553-27-0. Legend for Group A Hazards:

 Cat. 1
 Category 1 is most hazardous classification for all endpoints. For a minority of endpoints (*i.e.*, acute mammalian and chronic aquatic toxicity), Category 4 is the least hazardous.

 Cat. 1
 Category 1 is most hazardous classification for all endpoints. For a minority of endpoints (*i.e.*, acute mammalian and chronic aquatic toxicity), Category 4 is the least hazardous.

 Cat. 2
 For the rest of the endpoints, excluding physical endpoints, Category 2 is the least hazardous. "Not Classified" indicates no hazard according to endpoint-specific GHS criteria.

 Cat. 3
 Texts are underlined when information from difference sources result in different classifications. When there are different classifications for an endpoint, color and scoring are based on more conservative classifications.

 Not Classified/Not Listed
 Specific color coding upging by health endpoint according to GroepScreen Chamical Hazard Criteria Section V. Annor 1 (Clean Production Action 2019). In apparature of indicate indicates in the indicates i

Specific color-coding varies by health endpoint according to GreenScreen Chemical Hazard Criteria Section V - Annex 1 (Clean Production Action, 2018). In general, reds indicate wery high and high hazards, yellow indicate moderate hazards, and green indicate no/low hazards. Data gaps are gray.

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Table 5.2 Chemical-Specific Human Health Ha	azards (Group B En	dpoints) <sup>1</sup>	1										
Chemical	CAS	Reference	Respiratory Toxicity	Cardiovascular Toxicity	Epigenetic Toxicity	Hematotoxicity	Reactive in Biological Systems	Group B Endpoints Hepatotoxicity and Digestive System Toxicity	Immunotoxicity	Musculoskeletal Toxicity	Nephrotoxicity	Neurotoxicity	Ototoxicity
Current Priority Product Chemical of Concern N-(1,3-dimethylbutyl)-N'-phenyl-p- phenylenediamine (6PPD) Possible Alternatives	793-24-8	ECHA, 2024; ToxServices, 2021a, DTSC, 2022	May be a respiratory toxicant in Coho salmon (DTSC, 2022).	May be a vascular toxicant in Coho salmon (DTSC, 2022).	Not genotoxic; no other relevant data found	May induce anemia in rats at high doses, but dossier did not classify (ECHA, 2024).	DTSC lists 6PPDQ as reactive in biological systems in the Priority Product Profile	Increase in liver weight with fatty changes and vacuolar liver degeneration in rats, but dossier did not classify (ECHA, 2022).	Dermal sensitizer	DG	No relevant adverse effects observed	No relevant adverse effects observed in a 2 year oral study in rats (ECHA, 2024)	DG
	2001 01 4	FCUA 2024				DC.	Unalaar	In second in lives weight in sets, however,	Dermal consitions based				
N-(1,4-dimethylpentyl)-N'-phenyl-p- phenylenediamine (7PPD)	3081-01-4	ECHA, 2024	DG	DG	Not genotoxic; no other relevant data found	DG	Unclear	Increase in liver weight in rats; however, to not considered adverse in the absence of other effects. Dossier also did not classify.	on similar chemicals and	DG	DG	DG	DG
N-isopropyl-N'-phenyl-p-phenylenediamine (IPPD)	101-72-4	ECHA, 2024	Respiratory irritation were seen in subchronic animal studies, but dossier did not classify.	DG	Not genotoxic; no other relevant data found	Treatment related changes in several hematology parameters found in rats, but dossier did not classify.	Unclear	Increase in liver weight in rats; however, not considered adverse in the absence of other effects. Dossier also did not classify.		Incomplete skeletal ossification in rat offsprings; however, maternal toxicity was observed at the same dose, but dossier did not classify.	Increased kidney weight; however, not considered adverse in the absence of other effects. Dossier also did not classify.	DG	DG
N-cyclohexyl-N'-phenyl-p-phenylenediamine (CPPD)	101-87-1	-	DG	DG	DG	DG	Unclear	DG	DG	DG	DG	DG	DG
N,N'-diphenyl-p-phenylenediamine (DPPD)	74-31-7	ECHA, 2024	DG	DG	Mutagenic in <i>in vitro</i> assays, no <i>in vivo</i> data	Treatment related changes in several hematology parameters in rats, but dossier did not classify.	Unclear	Decrease in liver weight in rats; however, not considered adverse in the absence of other effects. Dossier also did not classify.	Dermal sensitizer	DG	Increase kidney weight and incidence of calcification, but dossier did not classify.	DG	DG
N-1,3-dimethyl butyl-N'-phenyl quinone diimine	52870-46-9	-						DG for all endpoints.					
(6QDI)													
Polymerized 2,2,4-trimethyl-1,2- dihydroquinoline (TMQ)	26780-96-1	ECHA, 2024	DG	DG	Not genotoxic; no other relevant data found.	DG	DG	Depressed liver function, fatty liver, enlarged liver, gross nodules in the liver in rats but dossier did not classify.	Not a dermal sensitizer. No other relevant data found.	In rat offsprings, statistically significant increase in skeletal abnormalities found in the presence of maternal toxicity. Dossier also did not classify.	DG	DG	DG
N,N'-Bis(1,4-dimethylpentyl)-p- phenylenediamine (77PD)	3081-14-9	ECHA, 2024	DG	DG	Not genotoxic; no other relevant data found	No relevant adverse effects observed	Unclear	Increase in liver weight in rats; however, not considered adverse in the absence of other effects. Dossier also did not classify.	Dermal sensitizer	No treatment-related skeletal abnormalities found in rabbits or rats.	No relevant adverse effects observed	DG	DG
4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	-		I	I	1	L	DG for all endpoints.	1	1	1		
N,N'-Ditolyl-p-phenylenediamine (Commercial DTPD)	68953-84-4	ECHA, 2024	DG	DG	Not genotoxic; no other relevant data found	Macrocytic anemia found in rats, but effects were reversible. Dossier did not classify.	Unclear	Increase in liver weight in rats; however, not considered adverse in the absence of other effects. Dossier also did not classify.		DG	Increase in kidney weight in rats, however not considered adverse in the absence of other effects. Dossier also did not classify.	DG	DG
N,N'-Dicyclohexyl-p-phenylenediamine (CCPD)	4175-38-6	-	DG	DG	DG	DG	Unclear	DG	DG	DG	DG	DG	DG
Diaryl-p-phenylene diamine (DAPD)	68953-84-4	ECHA, 2024	DG	DG	Not genotoxic; no other relevant data found	Macrocytic anemia found in rats, but effects were reversible. Dossier did not classify.	Unclear	Increase in liver weight in rats; however, not considered adverse in the absence of other effects. Dossier also did not classify.		DG	Increase in kidney weight in rats; however, not considered adverse in the absence of other effects. Dossier also did not classify.	DG	DG
N,N'-Di-2-naphthyl-p-phenylenediamine (DNPDA)	93-46-9	-		<u> </u>	<u> </u>			DG for all endpoints.		<u> </u>			
Nickel dibutyldithiocarbamate (NBC)	13927-77-0	ECHA, 2024; ToxServices, 2021h	DG	Dose-dependent degeneration of the heart muscles in rats with presence of fibrotic areas. Dossier did not classify.	Not genotoxic; no other relevant data found	Treatment related changes in several hematology parameters in rats. Dossier did not classify.	DG	Decreased liver weights, dose- dependent hyperemia in the liver. Dossier did not classify.		Degeneration of skeletal muscle in rats accompanied by necrosis and mononuclear inflammation in high- dose males. Dossier did not classify.		DG	DG
Ethoxyquin	91-53-2	ECHA, 2024; ToxServices, 2021i	DG	DG	Not genotoxic; no other relevant data found	Treatment related changes in several hematology parameters in rats, but dossier did not classify.	DG	Adverse liver effects observed in rats and dogs, including hepatocellular necrosis, cytoplasmic vacuolation, and bile-duct hyperplasia. Dossier did not classify.	one animal study, but	DG	Dose-dependent nephropathy, regeneration of the tubular epithelium, renal tubular dilatation, and papillary necrosis in rats. Dossier did not classify.	Study authors of an acute inhalation study suggested potential neurotoxic effects based on tremors observed. Dossier did not classify.	DG
Dilauryl thiodipropionate	123-28-4	ECHA, 2024	DG	Inflammation of cardiac tissues at high dose, but ECHA dossier did not classify	Not genotoxic; no other relevant data found	No treatment-related adverse effects compared to historical control	DG	No relevant adverse effects observed	DG	No relevant adverse effects observed	DG	DG	DG
N' -Phenyl.N-Fluorenyl-Para-Phenylenediamine	No CAS	-						DG for all endpoints.	I				
N-(p-phenylthiomethylphenyl)-N'-(1,3 dimethyl- butyl)-p-phenylenediamine	No CAS	-						DG for all endpoints.					
4-(2,5-dimethyl-1H-pyrrol-1-yl)-N-phenylaniline	No CAS	-						DG for all endpoints.					
N,N - (ethane-1,2-diyl ) bis (N-phenylbenzene-1 4-diamine [example chemical from patent]	No CAS	-						DG for all endpoints.					

Chemical	CAS	Reference	Respiratory Toxicity	Cardiovascular Toxicity	Epigenetic Toxicity	Hematotoxicity	Reactive in Biological Systems	Hepatoto
4-N-{2,3-dimethylphenyl}-1-N-phenylbenzene- 1,4-diamine- R1 and R2 are methyl	No CAS	-						
RU997, Irgazone 997 (Reaction product of N- phenyl-N'-(1,3dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether)	444992-04-5	-						
4-[4-(4-Methylpentan-2-ylamino)anilino]phenol	No CAS	-						
Representative example from class (4-((4- (dimethylamino)phenyl)amino)phenol )	6358-22-1	-						
N,N-diethyl-2,2,4-trimethyl-1H-quinolin-6- amine (R= N(C2H5)2	No CAS	-						
N,N'-Dibenzyl-p-xylene-alpha,alpha'-diamine-	25790-41-4	-						
2,4,6-tris-(N-1,4-dimethylpentyl-para- phenylenediamino)-1,3,5triazine (Durazone 37 or TAPDT)	121246-28-4	-						
N-Phenyl-1-naphthylamine	90-30-2	ECHA, 2024	No reliable studies are located	DG	Not genotoxic; no other relevant data found	Hemolytic anemia found in rats. Dossier classified category 2.	DG	No re
N-Phenyl-2-naphthylamine	135-88-6	ECHA, 2024	DG	DG	DG	DG	DG	
[2-Methyl-4,6-bis((octylthio)methyl)phenol (Irganox 1520) <sup>2</sup>	110553-27-0	ECHA, 2024	DG	DG	Not genotoxic; no other relevant data found	No relevant adverse effects observed	DG	No relev
Graphene	1034343-98-0	ECHA, 2024	No relevant adverse effects observed	DG	Not genotoxic; no other relevant data found	DG	DG	
1,1' -Pentamethylenebis(2,2-Di-n- Butylhydrazine)	No CAS	-						
α- C-4- hydroxy- 3,5- dimethylphenyl - N-isopropyl and Lowinox WSP	Nitrone as a class, no CAS and Lowinox WSP - 77-62-3	ECHA, 2024	DG	No relevant adverse effects observed	Not genotoxic; no other relevant data found	No relevant adverse effects observed	DG	No relev
N-(4-methylpentan-2-yl)-10H-phenothiazin-3- amine	No CAS	-						
7-(4-methylpentan-2-ylamino)-2,3,4,10- tetrahydro-1H-acridin-9-one	No CAS	-						
2-cyclohexyl-N-(4-methylpentan-2-yl)-1H-indol- 5-amine	No CAS	-						
4-(1H-indol-2-yl)-N-(4-methylpentan-2-yl)aniline	No CAS	-						
α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	No CAS	-						
Amine functionalized lignin	No CAS	-						
Rambutan peel extract	No CAS	-						

	Group B Endpoints					
ological Is	Hepatotoxicity and Digestive System Toxicity	Immunotoxicity	Musculoskeletal Toxicity	Nephrotoxicity	Neurotoxicity	Ototoxicity
	DG for all endpoints.					
	DG for all endpoints.					
	DG for all endpoints.					
	DG for all endpoints.					
	DG for all endpoints.					
	DG for all endpoints.					
	DG for all endpoints.					
	No reliable studies are located	Dermal sensitizer	No relevant adverse effects observed	Degeneration/regeneration of the proximal tubules in male rats and centrilobular hypertrophy in female rats, but dossier did not classify.	No relevant adverse effects observed	DG
	DG	Dermal sensitizer	DG	DG	DG	DG
	No relevant adverse effects observed	Not a dermal sensitizer. No other relevant data found.	No relevant adverse effects observed	DG	DG	DG
		iound.				
	DG	Not a dermal sensitizer.	DG	DG	DG	DG
		No other relevant data found.				
	DG for all endpoints.					
	No relevant adverse effects observed	Not a dermal sensitizer. No other relevant data found.	DG	DG	No relevant adverse effects observed	DG
	DG for all endpoints.	I		<u> </u>		
	DG for all endpoints.					
	DG for all endpoints.					
	DG for all endpoints.					
	Do for all encodints.					
	DG for all endpoints.					
	DG for all endpoints.					
	DG for all endpoints.					

								Group B Endpoints					
Chemical	CAS	Reference	Respiratory Toxicity	Cardiovascular Toxicity	Epigenetic Toxicity	Hematotoxicity	Reactive in Biological Systems	Hepatotoxicity and Digestive System Toxicity	Immunotoxicity	Musculoskeletal Toxicity	Nephrotoxicity	Neurotoxicity	Ototoxicity
Octyl Gallate	1034-01-1	EFSA, 2015; ECHA, 2024	DG	DG	Not genotoxic; no other relevant data found	Based on propyl gallate, reduction of hemoglobin concentration, packed cell volume and red blood cell concentration content and the morphological changes in the spleen, all at a high dose. The dossier did not classify on this endpoint.	Unclear	Based on propyl gallate, no relevant adverse effects observed	Dermal sensitizer	Based on propyl gallate, no relevant adverse effects observed	Based on propyl gallate, no relevant adverse effects observed	Based on propyl gallate, no relevant adverse effects observed	DG
Nano calcium carbonate surface modified by gallic acid	No CAS	-						DG for all endpoints.					
N-1-Methylheptyl-N'-phenyl-p- phenylenediamine (8PPD or UOP 688)	15233-47-3	-						DG for all endpoints.					
Specialized carbon nanotube mixture	No CAS	-						DG for all endpoints.					

Notes: CAS = Chemical Abstracts Service Number; DG = Data Gap.

(1) This table presents the hazards of the individual product ingredients, which may not reflect the hazards of the actual final tire product when fully cured.
 (2) A potential alternative is Irganox 1520 CAS 110553-27-0 blended with Vulcazon AFS. However, according to the source patent (Pirelli Tyre S.P.A, 2018), the best ratio is where 100% of blend is Irganox 1520. Additionally, no data were located for Vulcazon AFS. Thus, data and scoring is 100% based on Irganox 1520 CAS 110553-27-0.

Legend for Group B Hazards: Potential Concern No Relevant Adverse Effects Observed DG

\\gra-bos-01\Projects\221077\_US\_Tire\_SCP\_AA\WorkingFiles\Report tables, figs, appendices\Hazard tables 5.1, 5.2, 5.3, 5.8\Table 5.2 HH (Group B)

## Table 5.3 Chemical-Specific Environmental and Physical Hazards

( )							Environmen	tal			Pl	nysical	
Chemical	CAS	Environmental Score <sup>1</sup>	Physical Score <sup>1</sup>	Reference	Aquatic Toxicity – Acute	Aquatic Toxicity – Chronic	Persistent	Bioaccumulation	Terrestrial Ecotoxicity (from Pharos Only)	Global Warming Potential	Ozone Depleting Potential	CAA VOC Contributing to Smog Formation	g Flammability
Current Priority Product Chem	nical of Concern							·					
N-(1,3-dimethylbutyl)-N'- phenyl-p-phenylenediamine (6PPD)	793-24-8			ECHA, 2024; ToxServices GreenScreen, 2021a; PubChem, 2023; US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1	soil is 1,800 hours (75 days)	ECHA (2024): BCF of 569 (QSAR) which would be considered not bioaccumulative under California Code of Regulations, according to title 22, Division 4.5, Chapter 54, Article 5. 6PPD hydrolyzes in water with half-life of 8 hours. Hydrolysis products 4-hydroxydiphenylamine (experimental BCF 3.3: 40 in Cyprinus carpio), n-phenyl-p-benzoquinone monoimine (experimental BCF in Cyprinus carpio is <1.2-23), and 1,3-dimethylbutylamine (experimental BCF in Cyprinus carpio is <1.7-17) are all not bioaccumulative (ECHA, 2024); ToxServices (2021a): Bioaccumulative based on measured BCFs of 1,500- 1,700 for the surrogate N-(1-methylheptyl)-N'-phenylbenzene-1,4-diamine (CAS 15233-47-3).	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		150	0		25	25	50	50	0	0	0	0	0
Possible Alternatives	2004.04.4	1	T	5014 2024	0.1.4	<b>C</b> 1 <b>A</b>			20	N	N		
N-(1,4-dimethylpentyl)-N'- phenyl-p-phenylenediamine (7PPD)	3081-01-4			ECHA, 2024; ToxServices GreenScreen, 2021b; US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1	ECHA (2024): Not readily biodegradable in water (0% degraded in 35 days). Half-life in aerobic soil is <45.6 hours (1.9 days). Strong absorption to soil. ToxServices (2021b): Persistent, modeled half-life in soil is 75 days	ECHA (2024): BCF of 1197 L/Kg (QSAR) which would be considered bioaccumulative under California Code of Regulations, according to title 22, Division 4.5, Chapter 54, Article 5. However, 7PPD hydrolyzes in water (pH 7, 20C) with half-life of 7 hours. Hydrolysis products of 7PPD, 4-anilinophenol (4-hydroxydiphenylamine) (experimental BCF 3.3-49) and its oxidized form N- Phenylphenyl-p-benzoquinone monoimine (BCF <1.2 - 23), are not bioaccumulative. ToxServices (2021b): Bioaccumulative based on measured BCFs of 1,500- 1,700 for the surrogate N-(1-methylheptyl)-N'-phenylbenzene-1,4-diamine (CAS 15233-47-3).	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		150	0		25	25	50	50	0	0	0	0	0
N-isopropyl-N'-phenyl-p- phenylenediamine (IPPD)	101-72-4			ECHA, 2024; ToxServices GreenScreen, 2021c; PubChem, 2023; US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1	ToxServices (2021c): Persistent, modeled half-life in soil is 75 days.	<ul> <li>ECHA (2024): BCF of 31.2 (QSAR) which is considered not bioaccumulative under California Code of Regulations, according to title 22, Division 4.5, Chapter 54, Article 5. IPPD hydrolyzes in water with half-life of 2.5 hours. Hydrolysis products of IPPD, 4-hydroxydiphenylamine (experimental BCF 3.3-49) and p-quinoneimine-N-phenyl (BCF &lt;1.2 - 23), are not bioaccumulative. ToxServices (2021c): Very low bioaccumulation based on measured log kow (2.77).</li> </ul>	DG	No	No	Not a VOC (low vapor pressure)	• Not Classified
		100	0		25	25	50	0	0	0	0	0	0
N-cyclohexyl-N'-phenyl-p- phenylenediamine (CPPD)	101-87-1			US EPA, 2023b; US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	
		Not assigned based on complete data gap	0							0	0	0	0
N,N'-diphenyl-p- phenylenediamine (DPPD)	74-31-7			ECHA, 2024; PubChem, 2023; US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1	Not readily biodegradable in water (0.2% degraded in 14 days). A dissipation half-life (DT50) of 187.46 years (QSAR).	BCF of 260-2,150 derived in Cyprinus carpio from an experimental OECD 305C study. Given the bioconcentration threshold (BCF/BAF = 1000) under California Code of Regulations, according to title 22, Division 4.5, Chapter 54, Article 5, DPPD's BCFs straddles both bioaccumulative and not bioaccumulative. DPPD is conservatively classified as bioaccumulative.	DG	No	No	Not a VOC (low vapor pressure)	• Not Classified
		150	0		25	25	50	50	0	0	0	0	0
N-1,3-dimethyl butyl-N'- phenyl quinone diimine (6QDI)	52870-46-9			ECHA, 2024; ToxServices GreenScreen, 2021d; US EPA, 2023b; US EPA, 2018; IPCC, 2013; ECHA Harmonized CLP, 2023	Cat. 1	Cat. 1	DG (ECHA, 2024); Persistent based on estimated half- life of 337.5 days in sediment, its predicted dominant environmental compartment (ToxServices, 2021d)	for the surrogate N-(1-methylheptyl)-N'-phenylbenzene-1,4-diamine (CAS 15233-47-3).	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		150	0		25	25	50	50	0	0	0	0	0
Polymerized 2,2,4-trimethyl- 1,2-dihydroquinoline (TMQ Oligomer)	147-47-7			ECHA, 2024; ToxServices GreenScreen, 2021e; US EPA, 2023b; US EPA, 2018; IPCC, 2013	Cat. 2 based on surrogate Ethoxyquin CAS 91-53-2 (ToxServices, 2021e); No data (ECHA, 2024)		Persistent based on modeled half-life in soil is 75 days (ToxServices, 2021e); No biodegradation of TMQ was observed in 28 days in an EU Method C.4-E test (ECHA, 2024).	Very low bioaccumulative potential based on measured log kow of 2.93 (ToxServices, 2021e); Not bioaccumulative based on BCF between 108 and 1300 in Cyprinus carpio based on OECD 305C test (ECHA, 2024).	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
1 1													

						I	Environmer	ntal	1		PI	hysical	
Chemical	CAS	Environmental Score <sup>1</sup>	Physical Score <sup>1</sup>	Reference	Aquatic Toxicity – Acute	Aquatic Toxicity – Chronic	Persistent	Bioaccumulation	Terrestrial Ecotoxicity (from Pharos Only)	Potential	Ozone Depleting Potential	CAA VOC Contributing to Smog Formation	Flammability
N,N'-Bis(1,4-dimethylpentyl)- p-phenylenediamine (77PD)	3081-14-9			ECHA, 2024; ToxServices GreenScreen, 2021f; US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1		<ul> <li>ECHA (2024): Not bioaccumulative. 77PD's half-life in water is 3.6 hours.</li> <li>The BCFs for the primary hydrolysis products of 77PD are all below California Code of Regulations title 22, Division 4.5, Chapter 54, Article 5, bioconcentration threshold of BCF/BAF = 1000.</li> <li>ToxServices (2021f): Bioaccumulative under GHS based on a modeled BAF of 614.6. However, this BAF is not considered bioaccumulative under California Code of Regulations title 22, Division 4.5, Chapter 54, Article 5.</li> </ul>	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		100	0		25	25	50	0	0	0	0	0	0
4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	Not assigned based on	0	US EPA, 2023b; US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No 0	No 0	Not a VOC (low vapor pressure)	DG 0
N,N'-Ditolyl-p- phenylenediamine (Commercial DTPD)	68953-84-4	complete data gap		ECHA, 2024; US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1	Persistent based on negligible biodegradation in multiple studies.	Bioaccumulative based on measured BCF of 2107 in rainbow trout.	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		150	0		25	25	50	50	0	0	0	0	0
N,N'-Dicyclohexyl-p- phenylenediamine (CCPD)	4175-38-6			ToxServices, 2021g; US EPA, 2023b; US EPA, 2018; IPCC, 2013	Cat. 1 based on surrogate	Cat. 1 based on surrogate 44PD	Persistent based on modeled half-life of 75 days in soil, which is the primary partitioning compartment.	Bioaccumulative based on modeled BAF of 1059.	DG	No	No	Not a VOC (low vapor pressure)	-
		150	0		25	25	50	50	0	0	0	0	0
Diaryl-p-phenylene diamine (DAPD is a class, main commercial DAPD is DTPD CASRN 68953-84-4)	68953-84-4			ECHA, 2024; US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1	Persistent based on negligible biodegradation in multiple studies.	Bioaccumulative based on measured BCF of 2107 in rainbow trout.	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		150	0		25	25	50	50	0	0	0	0	0
N,N'-Di-2-naphthyl-p- phenylenediamine (DNPDA)	93-46-9			US EPA, 2023b; US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG
		Not assigned based on complete data gap	0							0	0	0	0
Nickel dibutyldithiocarbamate (NBC)	13927-77-0			ECHA, 2024; ToxServices GreenScreen, 2021h; US EPA, 2018; IPCC, 2013	Not Classified	ECHA (2024): Chronic 4 ToxServices (2021h): Not classified	Persistent based on a degradation of 0% at day 28 in an OECD 301B test.	ECHA (2024): Not bioaccumulative based on a modeled BCF of 75.96 L/kg ToxServices (2021h): Not bioaccumulative. BCFs are between 100 and 500	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		50	0		0	0	50	0	0	0	0	0	0
Ethoxyquin	91-53-2			ECHA, 2024; ToxServices GreenScreen, 2021i; US EPA, 2018; IPCC, 2013	ECHA (2024): Not classified ToxServices (2021i): Cat. 2		ECHA (2024): Not readily biodegradable (QSAR) ToxServices (2021i): Not readily biodegradable and partitioning to the soil with a half- life of 75 days.	ECHA (2024): Not bioaccumulative based on a modeled BCF of 455.8 L/kg ToxServices (2021i): Very low for bioaccumulation based on a measured log Kow value of 3.39 at pH 7 and an estimated BCF value of 129.3	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
Dilauryl thiodipropionate	123-28-4	70	0	ECHA, 2024; ToxServices GreenScreen, 2021j; US EPA, 2018; IPCC, 2013	10 Not Classified	10 Not Classified	50 Readily biodegradable based on 82% degraded in 28 days in an OECD 301C study.	7.43 (ECHA, 2024) and modeled BAG of 1.078 (ToxServices, 2021j)	0 DG	0 No	0 No	0 Not a VOC (low vapor pressure)	
N' -Phenyl.N-Fluorenyl-Para-	No CAS	0	0	US EPA, 2018; IPCC,	0 DG	0 DG	0 DG	DG DG	0 DG	0 No	0 No	0 Not a VOC (low vapor	0 DG
Phenylenediamine		Not assigned based on	0	2013; US EPA EPISuite, 2019						0	0	pressure)	
N-(p- phenylthiomethylphenyl)-N'- (1,3 dimethyl-butyl)-p- phenylenediamine	No CAS	complete data gap		US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No		Not a VOC (low vapor pressure)	DG
4-(2,5-dimethyl-1H-pyrrol-1-	No CAS	Not assigned based on complete data gap	0	US EPA, 2018; IPCC,	DG	DG	DG	DG	DG	0 No	0 No	0 Not a VOC (low vapor	DG
yl)-N-phenylaniline		Not assigned based on	0	2013; US EPA EPISuite, 2019						0	0	pressure) 0	
		complete data gap											

							Environmer	ntal		Physical				
Chemical	CAS	Environmental Score <sup>1</sup>	Physical Score <sup>1</sup>	Reference	Aquatic Toxicity – Acute	Aquatic Toxicity – Chronic	Persistent	Bioaccumulation	Terrestrial Ecotoxicity (from Pharos Only)	Potential	Ozone Depleting Potential	CAA VOC Contributing to Smog Formation	Flammability	
N,N - (ethane-1,2-diyl ) bis (N- phenylbenzene-1 4-diamine or similar chemical 1-N-[2-(4- anilinoanilino)ethyl]-4-N- phenylbenzene-1,4-diamine	No CAS			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG	
		Not assigned based on complete data gap	0							0	0	0		
4-N-(2,3-dimethylphenyl)-1-N- phenylbenzene-1,4-diamine- R1 and R2 are methyl	No CAS			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG	
		Not assigned based on complete data gap	0							0	0	0		
RU997 Irgazone 997 Reaction product of N-phenyl-N'- (1,3dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether.	444992-04-5			NZ Environmental Risk Management Authority, 2005; US EPA, 2023b; US EPA EPISuite, 2019	Irgazone 997: Cat. 3	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG	
		65	0		5	10	25	25		0	0	0		
4-[4-(4-Methylpentan-2- ylamino)anilino]phenol	No CAS			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG	
		Not assigned based on complete data gap	0							0	0	0		
This is a class of compounds - Reference uses case where R1 and R2 are methyl; n,p and q are zero and m=1 and is in the para position . Representative example from class (4-((4- (dimethylamino)phenyl)amin o)phenol)	6358-22-1			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG	
		Not assigned based on complete data gap	0							0	0	0		
N,N-diethyl-2,2,4-trimethyl- 1H-quinolin-6-amine (R= N(C2H5)2	No CAS			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG	
		Not assigned based on complete data gap	0		20			20	50	0	0	0		
Mixed xylene diamines N,N'- Dibenzyl-p-xylene- alpha,alpha'-diamine- 25790- 41-4	25790-41-4			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019; US EPA, 2023b		DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	<b>CompTox</b> : Flashpoint: 94.3°C	
		Not assigned based on complete data gap	0						0	0	0	0	0	
2,4,6-tris-(N-1,4- dimethylpentyl-para- phenylenediamino)- 1,3,5triazine, TAPDT	121246-28-4			ECHA (2024); US EPA, 2018; IPCC, 2013	Cat. 1	Cat. 1	ECHA (2024): Not readily biodegradable based on 18% degradation in water at 28 days (OECD TG 301B)	ECHA (2024): Not bioaccumulative based on a modeled BCF average of 16 L/Kg (QSAR)	DG	No	No	Not a VOC (low vapor pressure)	ECHA (2024): Not Classified	
N-Phenyl-1-naphthylamine	90-30-2		0	ECHA (2024); US EPA, 2018; IPCC, 2013	25 Cat. 1	25 Cat. 1	50 ECHA (2024): Not readily biodegradable based on 0% degradation water at 28 days in an OECD TG 301C study (Kanne 1980 as cited in ECHA, 2024). Similar results were reported 1t 14 days in an OECD TG 301C study (MITI Database, 2002 as cited in ECHA, 2024). Inherent biodegradation observed in a CO2 evolution study following a US EPA recommendation, 50% degradation was reported within 5 days in sewage effluent and >75% and 100% degradation within 2 and 10 days, respectively in supplemented sewage sludge (Sikka <i>et al.</i> , 1981, as cited in ECHA, 2024). For lake water, 50% degradation after 5 days and > 90% degradation after 18 days were observed.		DG	0 No	No	O Not a VOC (low vapor pressure)	0 Not Classified	
N-Phenyl-2-naphthylamine	135-88-6	150	0	ECHA (2024); US EPA, 2018; IPCC, 2013	25 DG	25 Cat. 2	50 DG	50 Not bioaccumulative based Log Pow of 23C	0 DG	0 No	0 No	0 Not a VOC (low vapor pressure)	0 Not Classified	
		45	0	-	10	10	25	0	0	0	0	0	0	

			_		Environmental					P		nysical	
Chemical	CAS	Environmental Score <sup>1</sup>	Physical Score <sup>1</sup>	Reference	Aquatic Toxicity – Acute	Aquatic Toxicity – Chronic	Persistent	Bioaccumulation	Terrestrial Ecotoxicity (from Pharos Only)	Dotontial	Ozone Depleting Potential	CAA VOC Contributing to Smog Formation	Flammability
[2-Methyl-4,6- bis((octylthio)methyl)phenol (Irganox 1520) <sup>2</sup>	110553-27-0			ECHA (2024); US EPA, 2018; IPCC, 2013	Not Classified	Not Classified	ECHA (2024): Not readily biodegradable based on 4% degradation in water at 28 days (OECD TG 301B). ECHA (2024): Moderately biodegradable based on the DT50 = 0.33 - 0.43 days in water/sediment (OECD <u>TG 308).</u>	024): Moderately biodegradable based on = 0.33 - 0.43 days in water/sediment (OECD		No	No	Not a VOC (low vapor pressure)	Not Classified
		10	0	-	0	0	10	0	0	0	0	0	0
Graphene	1034343-98-0			ECHA (2024); US EPA, 2018; IPCC, 2013	Cat. 3	Cat. 3	Inorganic; In a ready biodegradability test (OECD 301D), graphene degraded 0% by % ThCOD in 28 days.	Inorganic, no BCF or Log kow data	DG	No	No	DG	Not Classified
1,1' -Pentamethylenebis(2,2- Di-n- Butylhydrazine)	No CAS	85	0	US EPA, 2018; IPCC, 2013	5 DG	5 DG	DG	25 DG	0 DG	0 No	0 No	0 Not a VOC (low vapor pressure)	0 DG
		Not assigned based on complete data gap	0	_				0		0	0	0	
α- C-4- hydroxy- 3,5- dimethylphenyl - N-isopropyl and Lowinox WSP CAS 77-62-3	Nitrone as a class, no CAS and Lowinox WSP - 77-62-3			ECHA (2024); US EPA, 2018; IPCC, 2013	Lowinox WSP: Not Classified	Lowinox WSP ECHA (2024): Cat. 4	Lowinox WSP ECHA (2024): Persistent based on results of an OECD 301B study. Degradation of Lowinox WSP was 10-12% at 28 days. The degradation did not reach criteria for ready biodegradability (60% COD within a 10 day window).	DG	DG	No	No	Not a VOC (low vapor pressure)	Not Classified
		75 for Lowinox WSP, complete data gap for nitrone as a class	0	_	0	0	50	25		0	0	0	0
N-(4-methylpentan-2-yl)-10H- phenothiazin-3-amine	- No CAS			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG
		Not assigned based on complete data gap	0	-						0	0	0	
7-(4-methylpentan-2- ylamino)-2,3,4,10-tetrahydro- 1H-acridin-9-one	No CAS			US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG
		Not assigned based on complete data gap	0							0	0	0	
2-cyclohexyl-N-(4- methylpentan-2-yl)-1H-indol- 5-amine	No CAS	Not assigned based on	0	US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No 0	No 0	Not a VOC (low vapor pressure) 0	DG
		complete data gap			20								
4-(1H-indol-2-yl)-N-(4- methylpentan-2-yl)aniline	No CAS	Not assigned based on	0	US EPA, 2018; IPCC, 2013; US EPA EPISuite, 2019	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure) 0	DG
α- C-4- Hydroxy- 3,5-	No CAS	complete data gap		US EPA, 2018; IPCC,	DG	DG	DG	DG	DG	No	No	DG	DG
dimethylphenyl-N-tert. butyl nitrone				2013	50					No			
		Not assigned based on complete data gap	0							0	0	0	
Amine functionalized lignin	No CAS			US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No	No	DG	DG
		Not assigned based on complete data gap	0	_						0	0	0	
Rambutan peel extract	No CAS			US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No	No	DG	DG
		Not assigned based on complete data gap	0							0	0	0	
Octyl gallate	1034-01-1			ECHA, 2024; US EPA, 2018; IPCC, 2013		Cat. 1 (based on propyl gallate)	Based on propyl gallate, persistent based on results of an OECD 301F study. Degradation of propyl gallate was 49.4% at 28 days. The degradation did not reach criteria for ready biodegradability (60% ThOD within a 10 day window).		DG	No	No	Not a VOC (low vapor pressure)	Not Classified (based on propyl gallate)
1		100	0	-	25	25	50	0		0	0	0	0

							Environmen	tal			Ρ	hysical	sical	
Chemical	CAS	Environmental Score <sup>1</sup>	Physical Score <sup>1</sup>	Reference	Aquatic Toxicity – Acute	Aquatic Toxicity – Chronic	Persistent	Bioaccumulation	Terrestrial Ecotoxicity (from Pharos Only)	Global Warming Potential	Ozone Depleting Potential	CAA VOC Contributing to Smog Formation	; Flammability	
Nano calcium carbonate surface modified by gallic acid	No CAS			US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No	No	DG	DG	
		Not assigned based on complete data gap	0	-						0	0	0		
N-1-Methylheptyl-N'-phenyl- p-phenylenediamine (8PPD or UOP 688)	15233-47-3			US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No	No	Not a VOC (low vapor pressure)	DG	
		Not assigned based on complete data gap	0	-						0	0	0		
Specialized carbon nanotube mixture	No CAS			US EPA, 2018; IPCC, 2013	DG	DG	DG	DG	DG	No	No	DG	DG	
		Not assigned based on complete data gap	0							0	0	0		

BCF = Bioconcentration Factor; CAA = Clean Air Act; CAS = Chemical Abstracts Service Number; COD = Chemical oxygen demand; DG = Data Gap; ECHA = European Chemicals; GWP = Global Warming Potential; K<sub>ow</sub> = Octanol-Water Partition Coefficient; OECD = The Organisation for Economic Co-operation and Development; ODP = Oxygen-Depleting Potential; TG = Test Guideline; ThCOD = Theoretical chemical oxygen demand; ThOD = Theoretical oxygen demand; TSCA = Toxic Substances Control Act; US EPA = United States Environmental Protection Agency; VOC = Volatile Organic Compound.

Ingredients are considered bioaccumulative if BCF is >1,000 according to California Code of Regulations, according to title 22, Division 4.5, Chapter 54, Article 5 (Cal OEHHA, 2012). US EPA's list of ozone-depleting substances (US EPA, 2018) was used to evaluate ODP. Pharos (Healthy Building Network, 2023) was used to inform terrestrial toxicity. GWP was evaluated using Table 8.a.1 of the IPCC 5<sup>th</sup> Technical Report (IPCC, 2013). VOCs were considered chemicals with vapor pressures equal to or greater than 0.1 mm mercury (Hg) at 20C based on criteria in CARB (2009). Additionally, we noted whether the chemical is listed as a substance exempted under 40 CFR § 51.100 (CARB, 2009). (1) This is assuming that the concentration of the chemical in the tire is <10%.

(2) A potential alternative is Irganox 1520 CAS 110553-27-0 blended with Vulcazon AFS. However, according to the source patent (Pirelli Tyre S.P.A, 2018), the best ratio is where 100% of blend is Irganox 1520. Additionally, no data were located for Vulcazon AFS. Thus, data and scoring is 100% based on Irganox 1520 CAS 110553-27-0.

### Legend:

Categories assigned according to ECHA dossiers (ECHA, 2024) and Pharos (Healthy Building Network, 2023). Texts are underlined when information from difference sources result in different classifications. When there are different classifications for an endpoint, color and scoring are based on more conservative classifications.

Specific color coding varies by endpoint according to GreenScreen Chemical Hazard Criteria Section V - Annex 1 (Clean Production Action, 2018). In general, reds indicate very high and high hazards, yellow indicate moderate hazards, and green indicate no/low hazards. Data gaps are gray.

Category 1	Category 1 is most hazardous classification for all endpoints. For a minority of endpoints (i.e., chronic aquatic toxicity), Category 4 is
Category 2	the least hazardous. For the rest of the endpoints, excluding physical endpoints, Category 2 is the least hazardous. Not classified
Category 3	indicated no hazard according to endpoint-specific GHS criteria.
Category 4	
Category 5	

# Table 5.4 Acute Toxicity Data in Salmonids Reported in Existing Scientific Literature

	CAS No.		Parent Compound		Quinone		
Chemical Name		Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	7
N-(1,3-dimethylbutyl)-N'-	793-24-8	Pink salmon (Oncorhynchus gorbuscha), juvenile			>12.8		
phenyl-p-			N/A	N/A	(static, measured)	48	Foldvik <i>et al.</i> , 2024
phenylenediamine (6PPD)		Chum salmon ( <i>Oncorhynchus keta</i> ), juvenile			>1.3-2.4		
					(static, estimated in 320 mg/L tire wear		
			N/A	N/A	particle leachate)	24	McIntyre <i>et al.</i> , 2021
		Coho salmon (Oncorhynchus kisutch ), juvenile	250		0.041		
			(flow-through, measured)	24	(static, measured)	24	Tian et al., 2021; Lo et al. , 2023
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile			>3.5		
			N/A	N/A	(static, measured)	96	Hiki and Yamamoto, 2022
		Rainbow trout (Oncorhynchus mykiss ), juvenile	140		0.64		Monsanto Co., 1977, as cited in EcoTox,
			(static, nominal)	96	(static, measured)	96	2023; Nair <i>et al.,</i> 2023
		Sockeye salmon (Oncorhynchus nerka)			>50		
			N/A	N/A	(flow-through, measured)	24	Greer <i>et al.</i> , 2023
		Chinook salmon (Oncorhynchus tshawytscha)			67		
			N/A	N/A	(flow and measurement not specified)	N/R	Lo <i>et al.</i> , 2023
		Atlantic salmon (Salmo salar)			>12.16		
			N/A	N/A	(static, measured)	48	Foldvik et al., 2022
		Brown Trout ( <i>Salmo trutta</i> )			>12.16		
			N/A	N/A	(static, measured)	48	Foldvik <i>et al.,</i> 2022
		Arctic char (Salvelinus alpinus )			>14.2		
			N/A	N/A	(static, measured)	96	Brinkmann et al., 2022
		Southern Asian dolly varden (Salvelinus curilus)			>3.8		
			N/A	N/A	(static, measured)	96	Hiki and Yamamoto, 2022
		Brook trout (Salvelinus fontinalis )			0.59		
			N/A	N/A	(static, measured)	24	Brinkmann et al., 2022
		White-spotted char (Salvelinus leucomaenis pluvius)			0.51		
			N/A	N/A	(static, measured)	24	Hiki and Yamamoto, 2022
N-(1,4-dimethylpentyl)-N'-	3081-01-4	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
phenyl-p-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
phenylenediamine (7PPD)		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone			
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations	
			μg/L	hr	μg/L	hr		
N-isopropyl-N'-phenyl-p-	101-72-4	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A		
phenylenediamine (IPPD)		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
					>50			
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	(static, measured)	96	Nair <i>et al.</i> , 2023	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A		
N-cyclohexyl-N'-phenyl-p-	101-87-1	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A		
phenylenediamine (CPPD)		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
					>50			
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	(static, measured)	96	Nair <i>et al.</i> , 2023	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A		
N,N'-diphenyl-p-	74-31-7	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A		
phenylenediamine (DPPD)		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
					>50			
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	(static, measured)	96	Nair <i>et al</i> ., 2023	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A		

			Parent Compound		Quinone			
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations	
			μg/L	hr	μg/L	hr	]	
N-1,3-dimethyl butyl-N'-	52870-46-9	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A		
phenyl quinone diimine		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
(6QDI)		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
		Landlocked masu salmon (Oncorhynchus masou masou), juvenile	N/A	N/A	N/A	N/A		
		Rainbow trout ( <i>Oncorhynchus mykiss</i> )	638 (semi-static, measurement not specified)	96	N/A	N/A	Flexsys, 2007, as cited in ECHA, 2024	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilu s)	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A		
Polymerized 2,2,4-	26780-96-1	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A		
trimethyl-1,2-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
dihydroquinoline (TMQ)		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A		
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A		
N,N'-Bis(1,4-	3081-14-9	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A		
dimethylpentyl)-p-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
phenylenediamine (77PD)		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	24 (flow-through, measured)	96	>226 (flow-through, measured)	96	Chapelet <i>et al.,</i> 2023	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A		
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A		

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
4,4'-Dioctyldiphenylamine	101-67-7	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
(DOPD)		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
N,N'-Ditolyl-p-	68953-84-4	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
phenylenediamine		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
(Commercial DTPD)		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
			480		>50		Unnamed study, 1997, as cited in ECHA,
		Rainbow trout (Oncorhynchus mykiss )	(flow-through, measured)	96	(static, measured)	96	2024; Nair <i>et al.</i> , 2023
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone			
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations	
			μg/L	hr	μg/L	hr		
N,N'-Dicyclohexyl-p-	4175-38-6	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A		
phenylenediamine (CCPD)		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
			130 (based on surrogate 44PD)					
		Rainbow trout (Oncorhynchus mykiss )	(static, nominal)	96	N/A	N/A	ToxServices, 2021	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A		
Diaryl-p-phenylene	68953-84-4	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A		
diamine (DAPD is a class,		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
main commercial DAPD is		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
DTPD CASRN 68953-84-4)		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
			>480				Dionne, 1995, as cited in ECHA, 2024	
		Rainbow trout (Oncorhynchus mykiss )	(flow-through, measured)	96	N/A	N/A	(Weight of evidence 001)	
		Sockeye salmon ( <i>Oncorhynchus nerka</i> )	N/A	N/A	N/A	N/A		
		Chinook salmon ( <i>Oncorhynchus tshawytscha</i> )	N/A	N/A	N/A	N/A		
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A		
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A		
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A		
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A		
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A		
N,N'-Di-2-naphthyl-p-	93-46-9	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A		
phenylenediamine		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A		
(DNPDA)		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A		
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A		
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A		
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A		
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A		
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A		
		Brown Trout (Salmo trutta )	N/A N/A	N/A	N/A	N/A N/A		
		Arctic char (Salvelinus alpinus )	N/A N/A	N/A	N/A	N/A N/A	-	
		Southern Asian dolly varden ( <i>Salvelinus curilus</i> )	N/A	N/A	N/A	N/A		
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A		
	1	White-spotted char (Salvelinus leucomaenis pluvius )	11/ <b>n</b>	11/7	11/A	11/7		

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	1
Nickel	13927-77-0	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
dibutyldithiocarbamate		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
(NBC)		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout ( <i>Oncorhynchus mykiss</i> )	>100,000 (static, measurement not specified)	96	N/A	N/A	Mayer and Ellersieck, 1986
		Sockeye salmon ( <i>Oncorhynchus nerka</i> )	N/A	N/A	N/A	N/A	
		Chinook salmon ( <i>Oncorhynchus tshawytscha</i> )	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
Ethoxyquin	91-53-2	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
, .		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
			18,000				Unnamed Study, 2007, as cited in ECHA,
		Rainbow trout (Oncorhynchus mykiss )	(flow-through, measured)	96	N/A	N/A	2024
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
Dilauryl thiodipropionate	123-28-4	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
1		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
N' -Phenyl.N-Fluorenyl-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
Para-Phenylenediamine		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
N-(p-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
phenylthiomethylphenyl)-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
N'-(1,3 dimethyl-butyl)-p-		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
phenylenediamine		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
. ,		Rainbow trout (Oncorhynchus mykiss)	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
1		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
4-(2,5-dimethyl-1H-pyrrol-	No CAS	Pink salmon ( <i>Oncorhynchus gorbuscha</i> ), juvenile	N/A	N/A	N/A	N/A	
1-yl)-N-phenylaniline		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
// F - /		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss)	N/A	N/A	N/A	N/A	
		Sockeye salmon ( <i>Oncorhynchus nerka</i> )	N/A	N/A	N/A	N/A	
		Chinook salmon ( <i>Oncorhynchus tshawytscha</i> )	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
N,N - (ethane-1,2-diyl ) bis	No CAS	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
(N-phenylbenzene-1 4-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
diamine [example chemical		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
from patent]		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss)	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
4-N-(2,3-dimethylphenyl)-1-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
N-phenylbenzene-1,4-		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
diamine- R1 and R2 are		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
methyl		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
,		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
RU997, Irgazone 997	444992-04-5	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
(Reaction product of N-		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
phenyl-N'-		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
(1,3dimethylbutyl)-p-		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
phenylenediamine with an		Rainbow trout (Oncorhynchus mykiss)	N/A	N/A	N/A	N/A	
alkyl glycidylthioether)		Sockeye salmon ( <i>Oncorhynchus nerka</i> )	N/A	N/A	N/A	N/A	
, , , , , , , , , , , , , , , , , , , ,		Chinook salmon ( <i>Oncorhynchus tshawytscha</i> )	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden ( <i>Salvelinus curilus</i> )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
4-[4-(4-Methylpentan-2-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
ylamino)anilino]phenol		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
Representative example	6358-22-1	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
from class (4-((4-		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
(dimethylamino)phenyl)am		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
ino)phenol		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
<i></i>		Rainbow trout (Oncorhynchus mykiss)	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus)	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
N,N-diethyl-2,2,4-trimethyl-	No CAS	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
1H-quinolin-6-amine (R=		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
N(C2H5)2		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss)	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	7
N,N'-Dibenzyl-p-xylene-	25790-41-4	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
alpha,alpha'-diamine-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	
2,4,6-tris-(N-1,4-	121246-28-4	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
dimethylpentyl-para-		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
phenylenediamino)-		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
1,3,5triazine (Durazone 37		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
or TAPDT)		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
N-Phenyl-1-naphthylamine	90-30-2	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
, , ,		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
			440	,	•	,	Unnamed Study, 1981, as cited in ECHA,
		Rainbow trout ( <i>Oncorhynchus mykiss</i> )	(semi-static, nominal)	96	N/A	N/A	2024
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden ( <i>Salvelinus curilus</i> )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	1

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
N-Phenyl-2-naphthylamine	135-88-6	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	
[2-Methyl-4,6-	110553-27-0	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
bis((octylthio)methyl)phen		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
ol (Irganox 1520)		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
1		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
Graphene	1034343-98-0	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon ( <i>Oncorhynchus kisutc</i> h), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon ( <i>Oncorhynchus masou masou</i> ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	CAS No. Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
1,1' -	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
Pentamethylenebis(2,2-Di-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
n- Butylhydrazine)		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
α- C-4- hydroxy- 3,5-	Nitrone as a	Pink salmon (Oncorhynchus gorbuscha ), juvenile	N/A	N/A	N/A	N/A	
dimethylphenyl	class, no CAS	Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
<ul> <li>N-isopropyl and Lowinox</li> </ul>	and Lowinox	Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
WSP	WSP - 77-62-3	Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
N-(4-methylpentan-2-yl)-	No CAS	Pink salmon ( <i>Oncorhynchus gorbuscha</i> ), juvenile	N/A	N/A	N/A	N/A	
10H-phenothiazin-3-amine		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon ( <i>Oncorhynchus masou masou</i> ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout ( <i>Oncorhynchus mykiss</i> )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon ( <i>Oncorhynchus tshawytscha</i> )	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)  Proven Trout (Salmo trutta)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta)	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden ( <i>Salvelinus curilus</i> )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A	
	1	White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
7-(4-methylpentan-2-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
ylamino)-2,3,4,10-		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
tetrahydro-1H-acridin-9-		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
one		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char ( <i>Salvelinus alpinu</i> s)	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
2-cyclohexyl-N-(4-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
methylpentan-2-yl)-1H-		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
indol-5-amine		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon ( <i>Oncorhynchus nerka</i> )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char ( <i>Salvelinus alpinus</i> )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	
4-(1H-indol-2-yl)-N-(4-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
methylpentan-2-yl)aniline		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon ( <i>Salmo salar</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
α- C-4- Hydroxy- 3,5-	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
dimethylphenyl- N-tert.		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
butyl nitrone		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	
Amine functionalized lignin	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	
Rambutan peel extract	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon ( <i>Oncorhynchus kisutch</i> ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon ( <i>Oncorhynchus tshawytscha</i> )	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

			Parent Compound		Quinone		
Chemical Name	CAS No.	Species	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
Octyl Gallate	1034-01-1	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytsch a)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
Nano calcium carbonate	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
surface modified by gallic		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
acid		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar)	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden ( <i>Salvelinus curilus</i> )	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	
N-1-Methylheptyl-N'-	15233-47-3	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
phenyl-p-		Chum salmon (Oncorhynchus keta ), juveniles	N/A	N/A	N/A	N/A	
phenylenediamine (8PPD		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
or UOP 688)		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon (Oncorhynchus nerka)	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout (Salmo trutta )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis )	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius)	N/A	N/A	N/A	N/A	

		Species	Parent Compound		Quinone		
Chemical Name	CAS No.		Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Acute LC <sub>50</sub> (Lowest Reported) (Flow Conditions, Exposure Measurement)	Duration of Exposure	Citations
			μg/L	hr	μg/L	hr	
Specialized carbon	No CAS	Pink salmon (Oncorhynchus gorbuscha), juvenile	N/A	N/A	N/A	N/A	
nanotube mixture		Chum salmon ( <i>Oncorhynchus keta</i> ), juveniles	N/A	N/A	N/A	N/A	
		Coho salmon (Oncorhynchus kisutch ), juvenile	N/A	N/A	N/A	N/A	
		Landlocked masu salmon (Oncorhynchus masou masou ), juvenile	N/A	N/A	N/A	N/A	
		Rainbow trout (Oncorhynchus mykiss )	N/A	N/A	N/A	N/A	
		Sockeye salmon ( <i>Oncorhynchus nerka</i> )	N/A	N/A	N/A	N/A	
		Chinook salmon (Oncorhynchus tshawytscha)	N/A	N/A	N/A	N/A	
		Atlantic salmon (Salmo salar )	N/A	N/A	N/A	N/A	
		Brown Trout ( <i>Salmo trutta</i> )	N/A	N/A	N/A	N/A	
		Arctic char (Salvelinus alpinus )	N/A	N/A	N/A	N/A	
		Southern Asian dolly varden (Salvelinus curilus)	N/A	N/A	N/A	N/A	
		Brook trout (Salvelinus fontinalis)	N/A	N/A	N/A	N/A	
		White-spotted char (Salvelinus leucomaenis pluvius )	N/A	N/A	N/A	N/A	

a.i. = Active Ingredient; CAS No. = Chemical Abstracts Service Number; hr = Hour; LC<sub>50</sub> = Median Lethal Concentration; N/A = Not Available; N/R = Not Reported.

Hazard Endpoint	Classification		Concentratio	on in Product	
	Classification	<10%	10-29%	30-59%	60-100%
Carcinogenicity	Category 1	100	100	100	100
	Category 2/Prop 65	75	75	75	75
	Data Gap	25	25	50	50
Acute Toxicity	Category 1	75	75	100	100
	Category 2	50	50	75	75
	Category 3	10	25	50	50
	Category 4	5	5	10	10
	Data Gap	5	10	25	50
Mutagenicity <sup>2</sup>	Category 1	50	50	50	50
	Category 2	25	25	25	25
	Data Gap	25	25	50	50
Reproductive Toxicity <sup>2</sup>	Category 1	50	50	50	50
· ,	Category 2/Prop 65	25	25	25	25
	Data Gap	25	25	50	50
Developmental Toxicity <sup>2</sup>	Category 1	50	50	50	50
	Category 2/Prop 65	25	25	25	25
	Data Gap	25	25	50	50
Endocrine <sup>3</sup>	EU Priority List or	25	25	25	25
	Endocrine concern				
	Data gap	10	10	10	10
Systemic Toxicity/Organ	Category 1	25	25	50	50
Toxicity – Single Dose <sup>2</sup>	Category 2	10	10	25	25
	Category 3	5	5	15	15
	Data Gap	5	10	25	50
Systemic Toxicity/Organ	Category 1	25	25	50	50
Toxicity – Repeated Dose <sup>2</sup>	Category 2	10	10	25	25
	Data Gap	5	10	25	50
Skin Sensitizer <sup>4</sup>	Category 1	25	25	50	50
	Data Gap	5	10	25	50
Respiratory Sensitizer <sup>4</sup>	Category 1	25	25	50	50
. ,	Data Gap	5	10	25	50
Eye Irritant <sup>5</sup>	Category 1	5	10	25	25
	Category 2	5	5	10	10
	Data Gap	5	10	25	25
Skin Irritant <sup>5</sup>	Category 1	5	10	25	25
	Category 2	5	5	10	10
	Data Gap	5	10	25	25
Not Required to Be Classified/Not Listed		0	0	0	0

Table 5.5 Scoring Matrix – Human Health Endpoints<sup>1</sup>

AA = Alternatives Analysis; CSI = Chemical Scoring Index; EU = European Union; GHS = Globally Harmonized System of Classification and Labelling of Chemicals; SCP = Safer Consumer Products.

(1) The original CSI approach did not evaluate products if "more than 30% of [the] product's composition is due to the contribution of components with 'No Data Available,'" with the idea that the product will be re-evaluated at a later time "when more information may be available" (Verslycke *et al.*, 2014). We did not follow this approach, because the SCP regulations do not require additional testing, and the timeframe for compliance would not allow for this. Additionally, the original CSI approach does not penalize data gaps on an *endpoint by endpoint* basis. This approach only penalizes a product if <30% of its composition is accounted for by components with data gaps (although the number of data gaps is immaterial), with a singular maximum penalty score of 100 for the environmental categories, 100 for the human health categories, and 50 for the physical

categories. If chemicals with data gaps account for  $\geq$ 30% of a product's composition, the product would be classified as "Do Not Evaluate." Thus, the CSI approach lacks granularity in terms of how many or which health endpoints have missing data. For this AA, we added endpoint by endpoint penalty scores for data gaps, which is more conservative than the CSI approach. These data gap scores were assigned based on hazard severity (*i.e.*, the maximum carcinogenicity and mutagenicity data gaps are scored 50 *versus* 10 for endocrine disruption). Also, in general, data gap penalty scores are lower than the Category 1 hazard scores for the same endpoint, and the data gap penalty scores generally decrease with decreasing chemical concentrations, except for some categories of particular concern (*e.g.*, Category 1 carcinogens).

(2) Under the original CSI approach, scores did not differ between these endpoints To provide more granularity in the scoring, for this AA, we adopted the maximum CSI score for Category 1 hazards for all of the abovementioned hazard endpoints. However, we scaled down to a lower score for subsequent subcategories (approximately 50% of the Category 1 score for Category 2 and so on). This approach is in line with the spirit of the GHS and CSI.

(3) Endocrine hazard was moved from ecological toxicity, under the CSI to human health toxicity, under this AA's approach. Additionally, we used a score of 25 instead of the original 50 in the CSI for endocrine disruptors, because the EU's Endocrine Disruptor Priority List, which we used in this assessment, is a listing of chemicals with potential endocrine risk that should be explored *via* testing, rather than a list based on studies showing actual effects. In contrast, the maximum score for mutagenicity is 50 and is based on positive findings of a mutagenic effect.

(4) We created separate skin and respiratory sensitization categories from the original CSI's "sensitizer" category, to be consistent with the SCP regulations' required toxicity categories. Additionally, we used a maximum score of 50 instead of the original 25 in the CSI for skin and respiratory sensitization. This is because the original CSI approach was developed for oil and gas applications, in which sensitization was less of an issue. Because sensitization is an important hazard for worker exposure, we increased the maximum score for these endpoints.

(5) We created separate categories for eye and skin irritation from the CSI's "irritant" category, to be more consistent with the SCP regulations' required toxicity categories. We assigned a maximum data gap score of 25 for products in which components with no data account for more than 30% of the composition, matching the score of 25 for Category 1 skin or eye irritants, because these are common hazards.

Useand Endnaint	Classification		Concentratio	on in Product	
Hazard Endpoint	Classification	<10%	10-29%	30-59%	60-100%
Acute Aquatic Toxicity	Category 1	25	50	75	100
	Category 2	10	25	50	75
	Category 3	5	10	25	50
	Data Gap	10	25	50	75
Chronic Aquatic Toxicity <sup>2</sup>	Category 1	25	50	75	100
	Category 2	10	25	50	75
	Category 3	5	10	25	50
	Data Gap	10	25	50	75
Terrestrial Toxicity <sup>3</sup>	Yes	25	50	75	100
	Data Gap	0	0	0	0
Bioaccumulative	Yes	50	50	50	50
	Data Gap	25	25	25	25
Persistent	Persistent	50	50	50	50
	Inherently Biodegradable	10	10	10	10
	Readily Biodegradable	0	0	0	0
	Data Gap	25	25	25	25
Not Required to Be Classified		0	0	0	0

Table 5.6 Scorir	g Matrix – Ecol	ogical Health	Endpoints <sup>1</sup>
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AA = Alternatives Assessment; CSI = Chemical Scoring Index; SCP = Safer Consumer Products.

(1) The original CSI approach did not evaluate products if "more than 30% of [the] product's composition is due to the contribution of components with 'No Data Available,'" with the idea that the product will be re-evaluated at a later time "when more information may be available" (Verslycke *et al.*, 2014). We did not follow this approach, because the SCP regulations do not require additional testing, and the timeframe for compliance would not allow for this. Additionally, the original CSI approach does not penalize data gaps on an *endpoint by endpoint* basis. This approach only penalizes a product if <30% of its composition is accounted for by components with data gaps (although the number of data gaps is immaterial), with a singular maximum penalty score of 100 for the environmental categories, 100 for the human health categories, and 50 for the physical categories. If chemicals with data gaps account for  $\geq$ 30% of a product's composition, the product would be classified as "Do Not Evaluate." Thus, the CSI approach lacks granularity in terms of how many or which health endpoints have missing data. For this Abridged AA, we added endpoint by endpoint penalty scores for data gaps, which is more conservative than the CSI approach. The data gap penalty scores are lower than the Category 1 hazard scores for the same endpoint, and the data gap penalty scores generally decrease with decreasing chemical concentrations, except for certain endpoints of particular concern (*i.e.*, persistent and bioaccumulative).

(2) The CSI does not have scores for chronic aquatic toxicity. Thus, the CSI's scores for acute aquatic toxicity were used.

(3) The CSI does not have scores for terrestrial toxicity. Thus, we created scores for this endpoint. However, because many chemicals lack data for this endpoint, the data gap penalty score was zero.

Uppord Fudgesigt	Classification	Concentration in Product			
Hazard Endpoint		<10%	10-29%	30-59%	60-100%
Ozone Depletion Potential	Yes	50	50	50	50
Direct Global Warming Contributor	Yes	10	25	50	75
Flammability (Liquid or Solid)	Category 1	25	50	75	100
	Category 2	10	25	50	75
	Category 3	5	10	25	50
VOC Contributing to Tropospheric	Yes	10	25	50	75
Ozone Formation <sup>2</sup>	Data Gap	5	10	25	25
"No" or Data Gap for Any Category		0	0	0	0
Besides VOC					

## Table 5.7 Scoring Matrix – Physical/Chemical Hazards<sup>1</sup>

Notes:

AA = Alternatives Assessment; CSI = Chemical Scoring Index; SCP = Safer Consumer Products; VOC = Volatile Organic Compound. (1) The original CSI approach did not evaluate products if "more than 30% of [the] product's composition is due to the contribution of components with 'No Data Available,'' with the idea that the product will be re-evaluated at a later time "when more information may be available" (Verslycke *et al.*, 2014). We did not follow this approach, because the SCP regulations do not require additional testing, and the timeframe for compliance would not allow for this. Additionally, the original CSI approach does not penalize data gaps on an *endpoint by endpoint* basis. This approach only penalizes a product if <30% of its composition is accounted for by components with data gaps (although the number of data gaps is immaterial), with a singular maximum penalty score of 100 for the environmental categories, 100 for the human health categories, and 50 for the physical categories. If chemicals with data gaps account for  $\geq$ 30% of a product's composition, the product would be classified as "Do Not Evaluate." Thus, the CSI approach lacks granularity in terms of how many or which health endpoints have missing data. For this AA, we added endpoint by endpoint penalty scores for data gaps, which is more conservative than the CSI approach. The data gap penalty scores are lower than the data-supported hazard scores for the same endpoint, and data gap penalty scores generally decrease with decreasing chemical concentrations.

(2) For this endpoint, we used a maximum score of 75 instead of the original maximum score of 50 in the CSI. Because VOCs' contribution to ozone formation is an important hazard for products that are used in urban areas, and because smog formation is a particular concern for California cities, we increased the maximum score for this endpoint.

### Table 5.8 Chemical-Specific Hazard Scoring Summary

Class of Compound	Chemical	CAS	Human Health Score	Environmental Score	Physical Score	Total Score <sup>1</sup>	Notes
Current Priority Product Ch	emical of Concern						
Phenylene Diamine related	N-(1,3-dimethylbutyl)-N'-phenyl-p-phenylenediamine (6PPD)	793-24-8	125	150	0	275	Data-rich. GreenScreen™ Benchmark 1 - Avoid - Chemical of High Concern
Potential Alternatives	•••••••••••••••••••••••••••••••••••••••				•		
Phenylene Diamine	N-(1,4-dimethylpentyl)-N'-phenyl-p-phenylenediamine (7PPD)	3081-01-4	120	150	0	270	Data-rich. Suspected PBT according to ECHA, evaluation in progress. GreenScreen Benchmark 1 - Avoid - Chemical of High Concern
Phenylene Diamine	N-isopropyl-N'-phenyl-p-phenylenediamine (IPPD)	101-72-4	150	100	0	250	Data-rich. GreenScreen™ assessment used surrogate 6PPD for some endpoints. GreenScreen™ Benchmark 1 - Avoid - Chemical of High Concern
Phenylene Diamine	N-cyclohexyl-N'-phenyl-p-phenylenediamine (CPPD)	101-87-1	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on mostly data gap	No hazard data
Phenylene Diamine	N,N'-diphenyl-p-phenylenediamine (DPPD)	74-31-7	80	150	0	230	Data-rich
Phenylene Diamine related	N-1,3-dimethyl butyl-N'-phenyl quinone diimine (6QDI)	52870-46-9	175	150	0	325	Many data gaps. GreenScreen™ Benchmark 1 - Avoid - Chemical of High Concern. GreenScreen™ assessment used surrogates, 6PPD, for most endpoints.
Dihydroquinoline	Polymerized 2,2,4-trimethyl-1,2-dihydroquinoline (TMQ)	26780-96-1	170	70	0	240	GreenScreen™ assessment used surrogate TMQ oligomer and ethoxyquin for most endpoints. GreenScreen™ Benchmark 2 - Use but search for safer substitutes. Classified as Cat. 1 for target organ toxicity, both single and repeated exposure.
Phenylene Diamine	N,N'-Bis(1,4-dimethylpentyl)-p-phenylenediamine (77PD)	3081-14-9	80	100	0	180	Data-rich. GreenScreen™ Benchmark 2 - Use but search for safer substitutes
Diphenyl amine	4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Phenylene diamine	N,N'-Ditolyl-p-phenylenediamine (Commercial DTPD)	68953-84-4	55	150	0	205	Data-rich. Suspected PBT according to ECHA, evaluation in progress. Proposed as reproductive/developmental toxicity Category 1B under EU harmonized classification, classification not adopted yet.
Phenylene Diamine	N,N'-Dicyclohexyl-p-phenylenediamine (CCPD)	4175-38-6	115	150	0	265	Many data gaps. GreenScreen™ Benchmark 1 - Avoid - Chemical of High Concern. GreenScreen™ assessment used surrogates, 44PD and 77PD, for most endpoints.
Phenylene diamine	Diaryl-p-phenylene diamine (DAPD is a class, main commercial DAPD is DTPD CASRN 68953-84-4)	68953-84-4	55	150	0	205	Data-rich. Suspected PBT according to ECHA, evaluation in progress. Proposed as reproductive/developmental toxicity Category 1B under EU harmonized classification, classification not adopted yet.
Phenylene diamine	N,N'-Di-2-naphthyl-p-phenylenediamine (DNPDA)	93-46-9	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Metal dithiocarbamate	Nickel dibutyldithiocarbamate (NBC)	13927-77-0	170	50	0	220	Data-rich. GreenScreen™ Benchmark 1 - Avoid - Chemical of High Concern. On many regulatory restriction lists due to nickel.

Class of Compound	Chemical	CAS	Human Health Score	Environmental Score	Physical Score	Total Score <sup>1</sup>	Notes
Dihydroquinoline	Ethoxyquin	91-53-2	135	70	0	205	Data-rich. GreenScreen™ Benchmark 2 - Use but search for safer substitute
Sulfur compound	Dilauryl thiodipropionate	123-28-4	40	0	0	40	Data-rich. GreenScreen™ Benchmark 3 <sub>dg</sub> - Use but still opportunity for improvement
Phenylene Diamine	N' -Phenyl.N-Fluorenyl-Para-Phenylenediamine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Phenylene Diamine	N-(p-phenylthiomethylphenyl)-N'-(1,3 dimethyl-butyl)- p-phenylenediamine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Phenylene Diamine	4-(2,5-dimethyl-1H-pyrrol-1-yl)-N-phenylaniline	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Phenylene Diamine	N,N - (ethane-1,2-diyl ) bis (N-phenylbenzene-1 4- diamine [example chemical from patent]	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Phenylene Diamine	4-N-(2,3-dimethylphenyl)-1-N-phenylbenzene-1,4- diamine- R1 and R2 are methyl	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Phenylene Diamine (Kruger)	RU997, Irgazone 997 (Reaction product of N-phenyl-N'- (1,3dimethylbutyl)-p-phenylenediamine with an alkyl glycidylthioether)	444992-04-5	130	65	0	195	Almost entirely data gap, other than skin sensitization and aquatic toxicity
Phenylene Diamine	4-[4-(4-Methylpentan-2-ylamino)anilino]phenol	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Phenylene diamine	Representative example from class (4-((4- (dimethylamino)phenyl)amino)phenol )	6358-22-1	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Dihydroquinoline	N,N-diethyl-2,2,4-trimethyl-1H-quinolin-6-amine (R= N(C2H5)2	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
lindered amine	N,N'-Dibenzyl-p-xylene-alpha,alpha'-diamine-	25790-41-4	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
riazine	2,4,6-tris-(N-1,4-dimethylpentyl-para- phenylenediamino)-1,3,5triazine (Durazone 37 or TAPDT)	121246-28-4	50	100	0	150	Data rich
Phenylnaphthyl amines	N-Phenyl-1-naphthylamine	90-30-2	45	150	0	195	Data rich
Phenylnaphthyl amines	N-Phenyl-2-naphthylamine	135-88-6	180	45	0	225	Data rich - Classified as a Cat. 2 carcinogen among other hazards

Class of Compound	Chemical	CAS	Human Health Score	Environmental Score	Physical Score	Total Score <sup>1</sup>	Notes
Phenol	[2-Methyl-4,6-bis((octylthio)methyl)phenol (Irganox 1520)	110553-27-0	30	10	0	40	Data rich
Inorganic	Graphene	1034343-98-0	55	85	0	140	Data rich. There are worker inhalation concerns due to nano particles
Hydrazine	1,1' -Pentamethylenebis(2,2-Di-n- Butylhydrazine)	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data
Nitrone + Phenolic AO	α- C-4- hydroxy- 3,5- dimethylphenyl - N-isopropyl and Lowinox WSP	Nitrone as a class, no CAS and Lowinox WSP - 77-62-3	5 (for Lowinox WSP, however, no information on nitrone as a class)	75 (for Lowinox WSP, however, no information on nitrone as a class)	0 (for Lowinox WSP, however, no information on nitrone as a class)	Not assigned based on lack of data for nitrone as a class	Data rich for Lowinox WSP - 77-62-3; however, no information on nitrone as a class
Phenothiazine	N-(4-methylpentan-2-yl)-10H-phenothiazin-3-amine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Amine	7-(4-methylpentan-2-ylamino)-2,3,4,10-tetrahydro-1H- acridin-9-one	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Amine	2-cyclohexyl-N-(4-methylpentan-2-yl)-1H-indol-5- amine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Amine	4-(1H-indol-2-yl)-N-(4-methylpentan-2-yl)aniline	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Nitrone	α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Polymeric amine functionalized lignin	Amine functionalized lignin	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Gallate related	Rambutan peel extract	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data

Class of Compound	Chemical	CAS	Human Health Score	Environmental Score	Physical Score	Total Score <sup>1</sup>	Notes
Gallate related	Octyl Gallate	1034-01-1	40	100	0	140	Some hazard data available. Propyl gallate was used as a read-across for some endpoints.
Gallate related	Nano calcium carbonate surface modified by gallic acid	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data. There are worker inhalation concerns due to nano particles.
Phenylene Diamine	N-1-Methylheptyl-N'-phenyl-p-phenylenediamine (8PPD or UOP 688)	15233-47-3	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data
Inorganic	Specialized carbon nanotube mixture	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No hazard data. There are worker inhalation concerns due to nano particles.

Notes:

CAS = Chemical Abstracts Service Number; DG = Data Gap; ECHA = European Chemicals Agency; EU = European Union; PBT = Persistent, Bioaccumulative, and Toxic.

The Chemical Scoring Index (CSI) approach (Verslycke *et al.*, 2014) was modified to provide quantitative hazard scores. The higher the score, the worse the hazard profile; however, these scores should only be used as approximations of hazards (*i.e.*, ball parks), due to the underlying uncertainties. (1) This is assuming that the concentration of the chemical in the tire is <10%.

Chemical	CAS No.	SMILES	Molecular Weight	Density at 20°C (g/cm³)	Log K <sub>ow</sub> at 20°C (Octanol-Water Partition Coefficient, Describes Lipid Solubility)	Log K <sub>oa</sub> at 25°C (Octanol-Air Partition Coefficient)	Log K <sub>oc</sub> (Organic Carbon Partition Coefficient, Describes Sorption in Soil and Sediment)	K <sub>H</sub> <sup>1</sup> (Henry's Law Constant at 25°C, atm-m <sup>3</sup> /mole)	Vapor Pressure (Saturated, mm Hg at 25°C)	Melting Point (°C at 1 atm)	Boiling Point <sup>2</sup> (°C at 1 atm)	Water Solubility (mg/L at 25°C)	y Physical State	Hydrolysis Rate Constant (L M <sup>-1</sup> s <sup>-1</sup> )	Dissociation Constant at 20°C	Photolysis Rate Constant (s <sup>-1</sup> )	Standard Reduction Potential (V)	Air Diffusion Coefficient (Diffusivity) (cm <sup>2</sup> /s) at 20°C and 1 atm	Water Diffusion Coefficient (Diffusivity) (cm <sup>2</sup> /s) at 20°C and 1 atm	Reactivity/ Electrophilicity Index	Environmental Half-life in Air (Days)	Environmental Half-life in Air (Hrs)
N-(1,3-dimethylbutyl)-N'-phenyl-p- phenylenediamine (6-PPD)	793-24-8	CC(C)CC(C)NC1=CC=C(C=C1)NC2=CC= CC=C2	268.41	0.995 at 50°C	4.68	11.542	4.363	3.36E-09	4.93E-06	121.5	369.67	2.83	Solid	NA	2.00E-07	1.63E-04	NA	4.22E-02	4.20E-06	NA	0.047	0.567
Possible alternatives								1						1				1				1
N-(1,4-dimethylpentyl)-N'-phenyl-p- phenylenediamine (7PPD)	3081-01-4	CC(C)CCC(C)NC1=CC=C(C=C1)NC2=CC =CC=C2	282.43	1.01	5.17	11.909	4.623	4.46E-09	2.11E-06	129.78	381.27	0.67	Liquid	NA	2.00E-07	1.63E-04	NA	4.10E-02	4.05E-06	NA	0.047	0.563
N-isopropyl-N'-phenyl-p-phenylenediamine (IPPD)	101-72-4	CC(C)NC1=CC=C(C=C1)NC2=CC=CC=C 2	226.32	1.04	3.28	10.51	3.636	1.44E-09	7.11E-05	74	161	58.071	Solid	NA	1.58E-07	1.58E-7 - 1.74E-7	NA	4.66E-02	4.76E-06	NA	0.049	0.588
N-cyclohexyl-N'-phenyl-p-phenylenediamine (CPPD)	101-87-1	C1CCC(CC1)NC2=CC=C(C=C2)NC3=CC =CC=C3	266.39	NA	4.64	11.858	3.854	1.48E-09	1.24E-06	134.41	388.82	6.6544	Solid	NA	NA	NA	NA	4.24E-02	4.37E-06	NA	0.046	0.548
N,N'-diphenyl-p-phenylenediamine (DPPD)	74-31-7	C1=CC=C(C=C1)NC2=CC=C(C=C2)NC3= CC=CC=C3	260.34	1.268	4.04	12.117	4.715	2.05E-10	6.35E-09	144	220-225	1.5867	Solid	NA	NA	NA	NA	4.32E-02	4.56E-06	NA	0.053	0.642
N-1,3-dimethyl butyl-N'-phenyl quinone diimine (6QDI)	52870-46-9	CC(C)CC(C)N=C1C=CC(=NC2=CC=CC=C 2)C=C1	266.39	NA	5.86	7.58	5.62	4.66E-04	4.85E-05	64.21	352.87	4.08E-02	Liquid	NA	NA	NA	NA	4.24E-02	4.21E-06	NA	0.088	1.055
Polymerized 2,2,4-trimethyl-1,2- dihydroquinoline (TMQ)	26780-96-1	Polymer	173.26	1.042 at 20°C	3.9 at 25°C	NA	NA	NA	< 4.8E-06	48	> 280	< 1.00E0	Solid	NA	Insignificant	NA	NA	5.29E-02	5.59E-06	NA	NA	2
N,N'-Bis(1,4-dimethylpentyl)-p- phenylenediamine (77PD)	3081-14-9	CC(C)CCC(C)NC1=CC=C(C=C1)NC(C)CC C(C)C	304.52	0.909	6.3	11.701	4.532	9.72E-08	8.22E-06	112.67	364.35	0.074747	Liquid	NA	3.16E-08	8.96E-05	NA	3.91E-02	3.68E-06	NA	0.085	1.021
4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	CCCCCCCCC1=CC=C(C=C1)NC2=CC=C( C=C2)CCCCCCCC	393.66	NA	11.26	13.819	7.016	6.76E-05	2.11E-09	184.89	477.99	3.94E-07	Solid	NA	NA	NA	NA	3.35E-02	3.20E-06	NA	0.049	0.586
N,N'-Ditolyl-p-phenylenediamine (Commercial DTPD)	68953-84-4	CC1=CC=C(C=C1)NC2=CC(=CC=C2)NC 3=CC=C(C=C3)C	288.4	1.2	3.93	13.12	5.126	2.49E-10	1.01E-07	158.74	421.38	1.56E-01	Solid	NA	3.98E-02	Not sure where to find this	NA	4.07E-02	4.21E-06	NA	0.053	0.641
N,N'-Dicyclohexyl-p-phenylenediamine (CCPD)	4175-38-6	C1CCC(CC1)NC2=CC=C(C=C2)NC3CCC CC3	272.44	NA	5.24	11.599	2.992	1.07E-08	2.24E-06	131.43	379.45	4.4631	Solid	NA	NA	NA	NA	4.17E-02	4.21E-06	NA	0.077	0.926
Diaryl-p-phenylene diamine (DAPD)	68953-84-4	CC1=CC=C(C=C1)NC2=CC(=CC=C2)NC 3=CC=C(C=C3)C	288.4	1.2	3.93	13.12	5.126	2.49E-10	1.01E-07	158.74	421.38	1.56E-01	Solid	NA	3.98E-02	Not sure where to find this	NA	4.07E-02	4.21E-06	NA	0.053	0.641
N,N'-Di-2-naphthyl-p-phenylenediamine (DNPDA)	93-46-9	C1=CC=C2C=C(C=CC2=C1)NC3=CC=C( C=C3)NC4=CC5=CC=CC=C5C=C4	360.46	NA	6.39	16.488	6.765	1.95E-12	1.43E-11	235	539.98	6.90E-04	Solid	NA	NA	NA	NA	3.60E-02	3.84E-06	NA	0.027	19.253
Nickel dibutyldithiocarbamate (NBC)	13927-77-0	CCCCN(CCCC)C(=S)[S- ].CCCCN(CCCC)C(=S)[S-].[Ni+2]	467.43	1.301	5.44	9.34E-06	6.254	NA	3.98E-04	9.02E+01	296	8.31E+01	Solid: Particulate/ powder	NA	NA	NA	NA	3.71E-02	3.30E-06	NA	0.058	0.691
Ethoxyquin	91-53-2	CCOC1=CC2=C(C=C1)NC(C=C2C)(C)C	217.31	1.031	3.87	8.875	3.104	2.42E-07	1.32E-04	104.68	123-125	20.093	Liquid	NA	2.75E-5 (at 22°C)	NA	NA	4.79E-02	4.89E-06	NA	0.083	1.001
Dilauryl thiodipropionate	123-28-4	0=c(0cccccccccccccccccccccccccccccccccc	514.85	1.04	11.79	15.576	6.973	5.73E-07	8.89E-09	40-42	519.29	5.15E-07	Solid: Flakes	NA	NA	NA	NA	3.06E-02	2.76E-06	NA	0.205	2.465
N' -Phenyl.N-Fluorenyl-Para- Phenylenediamine	No CAS	C1=C(NC(C=CC2)=CC=2)C=CC(=C1)NC( C(C1=CC2)=CC=2)C(=CC=C2)C1=C2	348.45	NA	5.41	15.53	6.318	1.85E-12	1.59E-10	216.27	507.87	5.59E-03	Solid	NA	NA	NA	NA	0.0367	3.90E-06	NA	0.051	0.607
N' -Phenyl.N-Fluorenyl-Para-	No CAS	C1C=CC=C(C=1)NC(=CC=C1NC(=CC=C	349.45	NA	6.06	15.65	6.522	6.36E-12	1.25E-10	217.84	511.22	1.46E-03	Solid	NA	NA	NA	NA	0.0367	3.90E-06	NA	0.053	0.64
Phenylenediamine	No CAS	2)C(CC3=CC4)=C2C3=CC=4)C=C1 CC(C)CC(C)NC1=CC=C(C=C1)NC2=CC=	390.59	NA	6.99	16.480	6.422	7.19E-12	3.74E-10	204.49	498.68	1.70E-03	Solid	NA	NA	NA	NA	3.52E-02	3.46E-06	NA	0.046	0.549
N-(p-phenylthiomethylphenyl)-N'-(1,3 dimethyl-butyl)-p-phenylenediamine	NO CAS	C(C=C2)CSC3=CC=CC=C3	350.55		0.55	10.400	0.422	7.152 12	5.742 10	204.45	450.00	1.762 05	Solid	115			NA.	5.522.02	5.462.00	10	0.040	0.545
4-(2,5-dimethyl-1H-pyrrol-1-yl)-N- phenylaniline	No CAS	CC1=CC=C(N1C2=CC=C(C=C2)NC3=CC =CC=C3)C	262.36	NA	5.04	14.535	4.631	7.83E-12	5.47E-07	147.48	397.43	7.19E-01	Solid	NA	NA	NA	NA	4.29E-02	4.46E-06	NA	0.053	0.641
N,N - (ethane-1,2-diyl ) bis (N-phenylbenzene 1 4-diamine or similar chemical 1-N-[2-(4- anilinoanilino)ethyl]-4-N-phenylbenzene-1,4- diamine		C1=CC=C(C=C1)NC2=CC=C(C=C2)NCC NC3=CC=C(C=C3)NC4=CC=CC=C4	394.52	NA	4.4	19.709	6.87	1.20E-17	8.35E-12	235.45	548.93	4.34E-02	Solid	NA	NA	NA	NA	3.51E-02	3.53E-06	NA	0.048	0.58
4-N-(2,3-dimethylphenyl)-1-N-phenylbenzene- 1,4-diamine- R1 and R2 are methyl	No CAS	CC1=C(C(=CC=C1)NC2=CC=C(C=C2)NC 3=CC=CC=C3)C	288.4	NA	5.13	13.122	5.143	2.49E-10	1.07E-07	158.74	421.38	1.56E-01	Solid	NA	NA	NA	NA	4.07E-02	4.21E-06	NA	0.053	0.641
RU997 Irgazone 997 Reaction product of N- phenyl-N'-(1,3dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether	444992-04-5	C1=CC(=CC=C1)NC1C=CC(=CC=1)N(CC (0)CSC(C)(C)CCCCCCCCCC)C(CC(C)C)C	526.87	NA	10.6	20.959	7.037	1.07E-12	1.16E-15	253.56	587.68	1.54E-05	Solid	NA	NA	NA	NA	2.98E-02	2.76E-06	NA	0.04	28.55
4-[4-(4-Methylpentan-2- ylamino)anilino]phenol	No CAS	CC(C)CC(C)NC1=CC=C(C=C1)NC2=CC= C(C=C2)O	284.4	NA	3.85	14.694	4.471	3.50E-13	6.70E-08	152.39	404.5	5.68E+00	Solid	NA	NA	NA	NA	4.18E-02	4.15E-06	NA	0.047	0.566

Chemical	CAS No.	SMILES	Molecular Weight	Density at 20°C (g/cm³)	Log K <sub>ow</sub> at 20°C (Octanol-Water Partition Coefficient, Describes Lipid Solubility)	Log K <sub>oa</sub> at 25°C (Octanol-Air Partition Coefficient)	Log K <sub>oc</sub> (Organic Carbon Partition Coefficient, Describes Sorption in Soil and Sediment)	K <sub>H</sub> <sup>1</sup> (Henry's Law Constant at 25°C, atm-m <sup>3</sup> /mole)	Vapor Pressure (Saturated, mm Hg at 25°C)	Melting Point (°C at 1 atm)	Boiling Point <sup>2</sup> (°C at 1 atm)	Water Solubility (mg/L at 25°C)	Physical State	Hydrolysis Rate Constant (L M <sup>-1</sup> s <sup>-1</sup> )	Dissociation Constant at 20°C	Photolysis Rate Constant (s <sup>-1</sup> )	Standard Reduction Potential (V)	Air Diffusion Coefficient (Diffusivity) (cm²/s) at 20°C and 1 atm	Water Diffusion Coefficient (Diffusivity) (cm <sup>2</sup> /s) at 20°C and 1 atm	Reactivity/ Electrophilicity Index	Environmental Half-life in Air (Days)	Environmer Half-life in / (Hrs)
This is a class of compounds. Reference uses case where R1 and R2 are methyl; n,p and q are zero and m=1 and is in the para position. Representative example from class (4-(4- dimethylamino)phenyl)amino)phenol)	6358-22-1	CN(C)C1=CC=C(C=C1)NC2=CC=C(C=C2 )O	228.3		2.64	12.788	3.28	1.74E-12	1.48E-06	132.28	361.66	3.74E+02	Solid	NA	NA	NA	NA	4.78E-02	4.92E-06	NA	0.053	0.633
N,N-diethyl-2,2,4-trimethyl-1H-quinolin-6- amine (R= N(C2H5)2	No CAS	CCN(CC)C1=CC2=C(C=C1)NC(C=C2C)(C )C	244.38	NA	4.46	9.912	3.329	8.64E-08	2.19E-05	123.66	341.46	5.98E+00	Solid	NA	NA	NA	NA	4.44E-02	4.43E-06	NA	0.035	25.243
Vixed xylene diamines N,N'-Dibenzyl-p- cylene-alpha,alpha'-diamine-	25790-41-4	Mixture																				
2,4,6-tris-(N-1,4-dimethylpentyl-para- yhenylenediamino)-1,3,5triazine, TAPDT	121246-28-4	CC(C)CCC(C)NC1=CC=C(C=C1)NC2=NC (=NC(=N2)NC3=CC=C(C=C3)NC(C)CCC( C)C)NC4=CC=C(C=C4)NC(C)CCC(C)C	694.03	1.07 at 20°C	11.9	30.903	11.407	2.43E-21	1.57E-19	54	225	6.94E-07	Solid	NA	NA	NA	NA	2.62E-02	2.40E-06	NA	0.038	27.144
N-Phenyl-1-naphthylamine	90-30-2	C1=CC=C(C=C1)NC2=CC=CC3=CC=CC= C32	219.29	1.16 at 20°C	4.28	9.576	4.473	1.03E-07	7.63E-06	62	335	3.96E+00	Solid	NA	1.17E-05	1.33E-04 - 2.02E-03	NA	4.66E-02	5.10E-06	NA	0.031	22.222
N-Phenyl-2-naphthylamine	135-88-6	C1=CC=C(C=C1)NC2=CC3=CC=CC=C3C =C2	219.29	1.242 at 25°C	4.38	9.756	4.464	1.03E-07	1.64E-06	108	395.5	1.26E+00	Solid	NA	NA	NA	NA	4.66E-02	5.10E-06	NA	0.031	22.222
Irganox 1520	110553-27-0	CCCCCCCCCCC1=CC(=C(C(=C1)C)O)CSC CCCCCCC	424.75	0.981 at 20°C	10.2	16.9	7.637	4.53E-09	2.13E-11	14	508.31	8.89E-06	Liquid	NA	2.51E-11	7.60E-11	NA	3.38E-02	3.13E-06	NA	0.143	1.721
Graphene	1034343-98-0	NA	NA	2.259 at 20°C	NA	NA	NA	NA	NA	>4000	NA	<3 mg/L	Solid: Powder	NA	NA	NA	NA	NA	NA	NA	NA	NA
1,1' -Pentamethylenebis(2,2-Di-n- Butylhydrazine)	No CAS	CCCCN(CCCC)NCCCCCNN(CCCC)CCCC	356.64	NA	5.57	14.49	5.724	2.94E-11	2.21E-07	151.05	411.96	3.86E+03	Solid	NA	NA	NA	NA	3.66E-02	3.27E-06	NA	0.052	0.625
α- C-4- hydroxy- 3,5- dimethylphenyl (Lowinox WSP). No number for nitrone- N- isopropyl nitrone and Lowinox WSP	Lowinox WSP - 77 62-3, No CAS for nitrone	- NA Mixture																				
N-(4-methylpentan-2-yl)-10H-phenothiazin-3- amine	No CAS	CC(C)CC(C)NC1=CC2=C(C=C1)NC3=CC =CC=C3S2	298.45	NA	5.76	14.2	4.615	8.89E-11	1.14E-07	170.85	414.55	6.37E-01	Solid	NA	NA	NA	NA	4.13E-02	4.18E-06	NA	0.047	0.567
7-(4-methylpentan-2-ylamino)-2,3,4,10- tetrahydro-1H-acridin-9-one	No CAS	CC(C)CC(C)NC1=CC2=C(C=C1)NC3=C(C 2=O)CCCC3	298.43	NA	5.2	14.358	3.496	1.70E-11	7.60E-08	173.7	420.42	2.17E+00	Solid	NA	NA	NA	NA	4.06E-02	4.10E-06	NA	0.09	1.084
2-cyclohexyl-N-(4-methylpentan-2-yl)-1H- indol-5-amine	No CAS	CC(C)CC(C)NC1=CC2=C(C=C1)NC(=C2) C3CCCCC3	298.48	NA	6.81	13.443	5.92	5.69E-09	1.11E-07	157.01	421.39	1.91E-01	Solid	NA	NA	NA	NA	3.96E-02	3.94E-06	NA	0.045	0.543
4-(1H-indol-2-yl)-N-(4-methylpentan-2- yl)aniline	No CAS	CC(C)CC(C)NC1=CC=C(C=C1)C2=CC3= CC=CC=C3N2	292.43	NA	5.76	13.81	5.398	2.18E-10	1.99E-08	172.04	444.85	1.72E-01	Solid	NA	NA	NA	NA	4.03E-02	4.08E-06	NA	0.042	29.969
α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	No CAS	C1(/C=[N+](\[O- ])C(C)(C)C)=CC(C)=C(C(C)=C1)O	221.3	NA	1.83	14.249	4.264	9.32E-15	1.02E-09	191.87	455.63	5.11E+01	Solid	NA	NA	NA	NA	4.87E-02	4.83E-06	NA	0.436	5.227
Amine functionalized lignin	No CAS	NA, polymer																				
Rambutan peel extract	No CAS	NA, complex mixture																				
Nano Calcium Carbonate Surface Modified by Gallic Acid	No CAS	NA																				
Octyl Gallate	1034-01-1	CCCCCCCCCCC(=O)C1=CC(=C(C(=C1)O) O)O	282.33	NA	3.66	17.6	4.00	2.84E-16	1.95E-08	101-104	421.61	2.00E+01	Solid	7.99E-03	NA	NA	NA	4.44E-02	4.25E-06	NA	0.108	1.293
N-1-Methylheptyl-N'-phenyl-p- phenylenediamine (8PPD or UOP 688)	15233-47-3	CCCCCCCC()NC1=CC=C(C=C1)NC2=CC =CC=C2	296.46	NA	5.74	12.36	4.96	5.92E-09	4.99E-07	145.77	399.87	1.57E-01	Solid	NA	NA	NA	NA	3.99E-02	3.92E-06	NA	0.047	0.56
Specialized carbon nanotube mixture	No CAS	NA, complex mixture																				
Notes: CAS No. = Chemical Abstracts Service Registry Num Bolded text indicates experimental values. Non-bo Data were mainly obtained from ECHA (2024) and ( (1) Reference hierarchy of Henry's Law Constant so (2) Boiling point: Preference was given to experim	ded text indicates r JS EPA's EPI Suite so ources in EPI Suite:	nodeled or calculated values. If available, e oftware (US EPA, 2019). (1) Vapor pressure/water solubility, if exper	xperimental melting p rimental data are avail	oint, boiling point, vap able; (2) Group; (3) Boi	or pressure, and water solubility valu nd.						nplex Reaction F	roducts and Biologi	cal Materials.									

### Table 5.10 Physical-Chemical Properties and Hazards of Transformation Products of 6PPD and Possible Alternative Chemicals

Chemical Name	CAS No.	SMILES	Transformation Product Name	Transformation Product CAS	Transformation Product SMILES	log K <sub>ow</sub>	log K <sub>oc</sub>	VP (saturated mn Hg at 25°C)	n Water Solubility (mg/L)	ECHA Dossier GHS Conclusion	EU Persistent Bioaccumulative and Toxic	California Toxic Air Contaminant	California Proposition 65
6PPD	793-24-8	CC(C)CC(C)NC1=CC=C(C=C1)NC2=CC=C2	6PPD-quinone	2754428-18-5	CC(C)CC(C)NC1=CC(=O)C(=C C1=O)NC2=CC=CC=C2		No informati	on in ECHA dossier		No ECHA dossier; however, existing aquatic data indicates acute toxicity to fish.	Not listed	Not listed	Not listed
			4-Hydroxydiphenylamine	122-37-2	C1=CC=C(C=C1)NC2=CC=C(C =C2)O	2.82 (From EPISUITE. Hansch, C et al. (1995))		o information in EC	HA dossier	No useful information in ECHA dossier	Not listed	Not listed	Not listed
			1,3-Dimethylbutylamine	108-09-8	CC(C)CC(C)N	No infor	mation in ECHA	dossier	7.9 (From EPISUITE. Yaws, CL <i>et al.</i> (2001))	No ECHA dossier	Not listed	Not listed	Not listed
			p-Benzoquinone	106-51-4	C1=CC(=O)C=CC1=O	Between 0.1 and 0.3 (at 23°C)	No information in ECHA dossier	2.93E-02	1.47E+04 (at 20°C)	Acute Tox. 3 (Oral), Aquatic Acute 1, Skin Irrit. 2, Eye Irrit. 2, STOT SE 3, Acute Tox. 3 (Inhalation), Aquatic Chronic 1, Muta. 2, Skin Sens. 1, Flam. Sol. 1	Not listed	Listed	Not listed
			p-Hydroquinone	123-31-9	C1=CC(=CC=C10)O	0.59 (at 20°C)	1.585	2.40E-05	7.20E+04 (at 25°C)	Aquatic Acute 1, Carc. 2, Muta. 2, Eye Dam. 1, Acute Tox. 4 (Oral), Skin Sens. 1, Aquatic Chronic 1, Skin Sens.	Not listed	Listed	Not listed
			Aniline	62-53-3	C1=CC=C(C=C1)N	0.91 (at 25°C)	2.6	3.05E-01	3.50E+04 (at 20°C)	Carc. 2, STOT RE 1, Acute Tox. 3 (Dermal), Muta. 2, Eye Dam. 1, Acute Tox. 3 (Inhalation), Acute Tox. 3 (Oral), Aquatic Acute 1, Aquatic Chronic 1, Skin Sens. 1B	Not listed	Listed	Listed (Cancer)
			N-Phenyl-p-benzoquinone-monoimine	2406-04-4	C1=CC=C(C=C1)N=C2C=CC(= O)C=C2	=	No informati	on in ECHA dossier		No ECHA dossier	Not listed	Not listed	Not listed
			N-(1,3-Dimethylbutyl)-N'-(phenyl)-1,4- benzoquinonediimine (6QDI)	52870-46-9		4.2 (at 25°C)	No information in ECHA dossier	7.50E-06	1.54E+01 (at 20°C)	6QDI is an oxidation product of 6PPD which is expected to form under hydrolytic conditions in the presence of oxygen. There is no information on ECHA about GHS conclusions for 6QDI.	Not listed	Not listed	Not listed
N-(1,4-dimethylpentyl)-N'- phenyl-p-phenylenediamine (7PPD)	3081-01-4	CC(C)CCC(C)NC1=CC=C(C=C1)NC2=CC=C2	4-Hydroxydiphenylamine	122-37-2	C1=CC=C(C=C1)NC2=CC=C(C =C2)O	2.82 (From EPISUITE. Hansch, C <i>et al.</i> (1995))	N	o information in EC	HA dossier	No useful information in ECHA dossier	Not listed	Not listed	Not listed
			N-Phenyl-p-benzoquinone-monoimine	2406-04-4	C1=CC=C(C=C1)N=C2C=CC(= 0)C=C2	=	No informati	on in ECHA dossier		No ECHA dossier	Not listed	Not listed	Not listed
			1,4-Dimethylpentylamine	28292-43-5	CC(C)CCC(C)N		No informati	on in ECHA dossier		No ECHA dossier	Not listed	Not listed	Not listed
			p-Benzoquinone (pH 4 and 7)	106-51-4	C1=CC(=O)C=CC1=O	Between 0.1 and 0.3 (at 23°C)	No information in ECHA dossier	2.93E-02	1.47E+04 (at 20°C)	Acute Tox. 3 (Oral), Aquatic Acute 1, Skin Irrit. 2, Eye Irrit. 2, STOT SE 3, Acute Tox. 3 (Inhalation), Aquatic Chronic 1, Muta. 2, Skin Sens. 1, Flam. Sol. 1	Not listed	Listed	Not listed
			p-Hydroquinone (pH 4)	123-31-9	C1=CC(=CC=C10)0	0.59 (at 20°C)	1.585	2.40E-05	7.20E+04 (at 25°C)	Aquatic Acute 1, Carc. 2, Muta. 2, Eye Dam. 1, Acute Tox. 4 (Oral), Skin Sens. 1, Aquatic Chronic 1, Skin Sens. 1B	Not listed	Listed	Not listed
			Aniline (pH 4, 7, and 9)	62-53-3	C1=CC=C(C=C1)N	0.91 (at 25°C)	2.6	3.05E-01	3.50E+04 (at 20°C)	Carc. 2, STOT RE 1, Acute Tox. 3 (Dermal), Muta. 2, Eye Dam. 1, Acute Tox. 3 (Inhalation), Acute Tox. 3 (Oral), Aquatic Acute 1, Aquatic Chronic 1, Skin Sens. 1B	Not listed	Listed	Listed (Cancer)
			4-(Phenylnitroso)benzen-1-olate (overoxidised 4-HDPA)	CAS not identified	NA		Unable to asse	ess due to lack of CA	AS	Unable to assess due to lack of CAS		CAS not identified	1
N-isopropyl-N'-phenyl-p- phenyldiamine (IPPD)	101-72-4	CC(C)NC1=CC=C(C=C1)NC2=CC=C2	4-Hydroxydiphenylamine	122-37-2	C1=CC=C(C=C1)NC2=CC=C(C =C2)O	2.82 (From EPISUITE. Hansch, C et al. (1995))		o information in EC	HA dossier	No useful information in ECHA dossier	Not listed	Not listed	Not listed
			Benzo-quinoneimine-N-phenyl	2406-04-4 (most				on in ECHA dossier		No ECHA dossier	Not listed	Not listed	Not listed
N-cyclohexyl-N'-phenyl-p- phenylenediamine (CPPD)	101-87-1	C1CCC(CC1)NC2=CC=C(C=C2)NC3=CC=CC=C3				ECHA dossier availab					Not listed	Not listed	Not listed
N,N'-diphenyl-p- phenylenediamine (DPPD)	74-31-7	C1=CC=C(C=C1)NC2=CC=C(C=C2)NC3=CC=CC=C3			ECHA dossi	er contains no inform	ormation on transformation products				Not listed	Not listed	Not listed

Chemical Name	CAS No.	SMILES	Transformation Product Name	Transformation Product CAS	Transformation Product SMILES	log K <sub>ow</sub>	log K <sub>oc</sub>	VP (saturated mm Hg at 25°C)	Water Solubility (mg/L)	ECHA Dossier GHS Conclusion	EU Persistent Bioaccumulative and Toxic	California Toxic Air Contaminant	California Proposition 65
N-1,3-dimethyl butyl-N'-phenyl quinone diimine (6QDI)	52870-46-9	CC(C)CC(C)N=C1C=CC(=NC2=CC=CC)C=C1	4-Hydroxydiphenylamine	122-37-2	C1=CC=C(C=C1)NC2=CC=C( =C2)O	C 2.82 (From EPISUITE. Hansch, C <i>et al.</i> (1995))	N	o information in EC	HA dossier	Acute Tox. 4, Skin Sens. 1, Aquatic Acute 1, Aquatic Chronic 1	Not listed	Not listed	Not listed
			p-Benzoquinone	106-51-4	C1=CC(=O)C=CC1=O	Between 0.1 and 0.3 (at 23°C)	No information in ECHA dossier	2.93E-02	1.47E+04 (at 20°C)	Acute Tox. 3 (Oral), Aquatic Acute 1, Skin Irrit. 2, Eye Irrit. 2, STOT SE 3, Acute Tox. 3 (Inhalation), Aquatic Chronic 1, Muta. 2, Skin Sens. 1, Flam. Sol. 1	Not listed	Listed	Not listed
			p-Hydroquinone	123-31-9	C1=CC(=CC=C10)0	0.59 (at 20°C)	1.585	2.40E-05	7.20E+04 (at 25°C)	Aquatic Acute 1, Carc. 2, Muta. 2, Eye Dam. 1, Acute Tox. 4 (Oral), Skin Sens. 1, Aquatic Chronic 1, Skin Sens. 1B	Not listed	Listed	Not listed
			Aniline	62-53-3	C1=CC=C(C=C1)N	0.91 (at 25°C)	2.6	3.05E-01	3.50E+04 (at 20°C)	Carc. 2, STOT RE 1, Acute Tox. 3 (Dermal), Muta. 2, Eye Dam. 1, Acute Tox. 3 (Inhalation), Acute Tox. 3 (Oral), Aquatic Acute 1, Aquatic Chronic 1, Skin Sens. 1B	Not listed	Listed	Listed (Cancer)
Polymerized 2,2,4-trimethyl- 1,2-dihydroquinoline (TMQ)	26780-96-1	CC1=CC(NC2=CC=C12)(C)C			ECHA dossi	er contains no informa	ation on transf	ormation products	1	-	Not listed	Not listed	Not listed
N,N'-Bis(1,4-dimethylpentyl)-p-	3081-14-9	CC(C)CCC(C)NC1=CC=C(C=C1)NC(C)CCC(C)C	Quinone-diimine	CAS not identified	NA		Unable to asse	ess due to lack of CA	AS	CAS not identified		CAS not identified	1
phenylenediamine (77PD)			N-1,4 Dimethyl- pentyl-p-phenol (4PD-OH)	CAS not identified	NA		Unable to asse	ess due to lack of CA	AS	CAS not identified		CAS not identified	
			1,4-Dimethylpentylamine p-Benzoquinone	28292-43-5 106-51-4	CC(C)CCC(C)N C1=CC(=O)C=CC1=O	- Between 0.1 and 0.3 (at 23°C)	- No information in ECHA dossier	- 2.93E-02	- 1.47E+04 (at 20°C)	No ECHA dossier Acute Tox. 3 (Oral), Aquatic Acute 1, Skin Irrit. 2, Eye Irrit. 2, STOT SE 3, Acute Tox. 3 (Inhalation), Aquatic Chronic 1, Muta. 2, Skin Sens. 1, Flam. Sol. 1	Not listed Not listed	Not listed Listed	Not listed Not listed
			p-Hydroquinone	123-31-9	C1=CC(=CC=C10)0	0.59 (at 20°C)	1.585	2.40E-05	7.20E+04 (at 25°C)	Aquatic Acute 1, Carc. 2, Muta. 2, Eye Dam. 1, Acute Tox. 4 (Oral), Skin Sens. 1, Aquatic Chronic 1, Skin Sens. 1B	Not listed	Listed	Not listed
4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	CCCCCCCCC1=CC=C(C=C1)NC2=CC=C(C=C2)CCCCCCCC			No	ECHA dossier availab	e to support e	valuation.		10	Not listed	Not listed	Not listed
N,N'-Ditoyl-p- phenylenediamine (Commercial DTPD)	68953-84-4	CC1=CC=C(C=C1)NC2=CC(=CC=C2)NC3=CC=C(C=C3)C	Hydroxydiphenylamine	122-37-2 (most likely	<pre>/) C1=CC=C(C=C1)NC2=CC=C(( =C2)O</pre>	2.82 (From EPISUITE. Hansch, C <i>et al.</i> (1995))		o information in EC	HA dossier	No useful information in ECHA dossier	Not listed	Not listed	Not listed
			Methyl hydroxydiphenylamine	CAS not identified	NA			ess due to lack of CA	AS	CAS not identified		CAS not identified	1
N,N'-Dicyclohexyl-p- phenylenediamine (CCPD)	4175-38-6	C1CCC(CC1)NC2=CC=C(C=C2)NC3CCCCC3		122 27 2 (most likely		ECHA dossier availab					Not listed	Not listed	Not listed
Diaryl-p-phenylene diamine (DAPD)	68953-84-4	CC1=CC=C(C=C1)NC2=CC(=CC=C2)NC3=CC=C(C=C3)C	Hyaroxyaipnenyiamine	122-37-2 (most likely	<pre>/) C1=CC=C(C=C1)NC2=CC=C(0 =C2)O</pre>	2.82 (From EPISUITE. Hansch, C <i>et al.</i> (1995))		o information in EC	HA dossier	No useful information in ECHA dossier	Not listed	Not listed	Not listed
	02.46.0		Methyl hydroxydiphenylamine	CAS not identified	NA			ess due to lack of CA	AS	No useful information in ECHA dossier	Netlisted	CAS not identified	
N,N'-Di-2-naphthyl-p- phenylenediamine (DNPDA)		C1=CC=C2C=C(C=CC2=C1)NC3=CC=C(C=C3)NC4=CC5= CC=CC=C5C=C4				ECHA dossier availab					Not listed	Not listed	Not listed
Nickel dibutyldithiocarbamate (NBC)	13927-77-0	CCCCN(CCCC)C(=S)[S-].CCCCN(CCCC)C(=S)[S-].[Ni+2]			ECHA dossi	er contains no informa	tion on transfo	ormation products.			Not listed	Not listed	Not listed
Ethoxyquin	91-53-2	CCOC1=CC2=C(C=C1)NC(C=C2C)(C)C			ECHA dossi	er contains no informa	tion on transfo	ormation products.			Not listed	Not listed	Not listed
Dilauryl thiodipropionate	123-28-4	0=c(0cccccccccccccccccccccccccccccccccc			ECHA dossi	er contains no informa	tion on transfo	ormation products.			Not listed	Not listed	Not listed
N' -Phenyl.N-Fluorenyl-Para- Phenylenediamine	No CAS	C1=C(NC(C=CC2)=CC=2)C=CC(=C1)NC(C(C1=CC2)=CC= 2)C(=CC=C2)C1=C2			No	ECHA dossier availab	e to support e	valuation.			Not listed	Not listed	Not listed
N-(p-phenylthiomethylphenyl)- N'-(1,3 dimethyl-butyl)-p- phenylenediamine	No CAS	CC(C)CC(C)NC1=CC=C(C=C1)NC2=CC=C(C=C2)CSC3=CC =CC=C3			No	ECHA dossier availabl	e to support e	valuation.			Not listed	Not listed	Not listed
4-(2,5-dimethyl-1H-pyrrol-1-yl) N-phenylaniline	No CAS	CC1=CC=C(N1C2=CC=C(C=C2)NC3=CC=CC=C3)C			Nc	ECHA dossier availab	e to support e	valuation.			Not listed	Not listed	Not listed
N,N - (ethane-1,2-diyl ) bis (N- phenylbenzene-1 4-diamine or similar chemical 1-N-[2-(4- anilinoanilino)ethyl]-4-N- phenylbenzene-1,4-diamine	No CAS	C1=CC=C(C=C1)NC2=CC=C(C=C2)NCCNC3=CC=C(C=C3) NC4=CC=CC=C4			Nc	ECHA dossier availabl	e to support e	valuation.			Not listed	Not listed	Not listed

Chemical Name	CAS No.	SMILES	Transformation Product Name	Transformation Product CAS	Transformation Product SMILES	log K <sub>ow</sub>	log K <sub>oc</sub>	VP (saturated mm Hg at 25°C)	Water Solubility (mg/L)	ECHA Dossier GHS Conclusion	EU Persistent Bioaccumulative and Toxic	California Toxic Air Contaminant	California Proposition 65
4-N-(2,3-dimethylphenyl)-1-N- phenylbenzene-1,4-diamine- R1 and R2 are methyl	No CAS	CC1=C(C(=CC=C1)NC2=CC=C(C=C2)NC3=CC=CC=C3)C			No	ECHA dossier availab	e to support ev	valuation.	· · · · ·		Not listed	Not listed	Not listed
RU997 Irgazone 997 Reaction product of N-phenyl-N'- (1,3dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether.	444992-04-5	C1=CC(=CC=C1)NC1C=CC(=CC=1)N(CC(0)CSC(C)(C)CCC CCCCCC)C(CC(C)C)C			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
4-[4-(4-Methylpentan-2- ylamino)anilino]phenol	No CAS	CC(C)CC(C)NC1=CC=C(C=C1)NC2=CC=C(C=C2)O			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
This is a class of compounds. Reference uses case where R1 and R2 are methyl; n,p and q are zero and m=1 and is in the para position. Representative example from class (4-((4- (dimethylamino)phenyl)amino) phenol)	6358-22-1	CN(C)C1=CC=C(C=C1)NC2=CC=C(C=C2)O			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
N,N-diethyl-2,2,4-trimethyl-1H- quinolin-6-amine (R= N(C2H5)2	No CAS	CCN(CC)C1=CC2=C(C=C1)NC(C=C2C)(C)C		No ECHA dossier available to support evaluation.								Not listed	Not listed
Mixed xylene diamines N,N'- Dibenzyl-p-xylene-alpha,alpha'- diamine-	25790-41-4	Mixture		No ECHA dossier available to support evaluation.								Not listed	Not listed
2,4,6-tris-(N-1,4- dimethylpentyl-para- phenylenediamino)- 1,3,5triazine, TAPDT		CC(C)CCC(C)NC1=CC=C(C=C1)NC2=NC(=NC(=N2)NC3= CC=C(C=C3)NC(C)CCC(C)C)NC4=CC=C(C=C4)NC(C)CCC( C)C		ECHA dossier contains no information on transformation products.									Not listed
N-Phenyl-1-naphthylamine	90-30-2	C1=CC=C(C=C1)NC2=CC=CC3=CC=CC32			ECHA dossie	r contains no informa	tion on transfo	rmation products.			Not listed	Not listed	Not listed
N-Phenyl-2-naphthylamine	135-88-6	C1=CC=C(C=C1)NC2=CC3=CC=C3C=C2			ECHA dossie	r contains no informa	tion on transfo	rmation products.			Not listed	Not listed	Not listed
Irganox 1520	110553-27-0	CCCCCCCCCCCC1=CC(=C(C(=C1)C)O)CSCCCCCCCC			ECHA dossie	r contains no informa	tion on transfo	rmation products.			Not listed	Not listed	Not listed
Graphene	1034343-98- 0	[C]				r contains no informa		-			Not listed	Not listed	Not listed
1,1' -Pentamethylenebis(2,2-Di- n- Butylhydrazine)	No CAS	CCCCN(CCCC)NCCCCCNN(CCCC)CCCC			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
α- C-4- hydroxy- 3,5- dimethylphenyl (Lowinox WSP). No number for nitrone - N-isopropyl nitrone and Lowinox WSP	Lowinox WSP - 77-62-3, No CAS for nitrone	Mixture			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
N-(4-methylpentan-2-yl)-10H- phenothiazin-3-amine	No CAS	CC(C)CC(C)NC1=CC2=C(C=C1)NC3=CC=CC=C3S2			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
7-(4-methylpentan-2-ylamino)- 2,3,4,10-tetrahydro-1H-acridin- 9-one	No CAS	CC(C)CC(C)NC1=CC2=C(C=C1)NC3=C(C2=O)CCCC3			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
2-cyclohexyl-N-(4- methylpentan-2-yl)-1H-indol-5- amine	No CAS	CC(C)CC(C)NC1=CC2=C(C=C1)NC(=C2)C3CCCCC3			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
4-(1H-indol-2-yl)-N-(4- methylpentan-2-yl)aniline	No CAS	CC(C)CC(C)NC1=CC=C(C=C1)C2=CC3=CC=CC=C3N2				ECHA dossier availab					Not listed	Not listed	Not listed
α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	No CAS	NA			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed
Amine functionalized lignin	No CAS	NA			No	ECHA dossier availab	e to support ev	valuation.			Not listed	Not listed	Not listed

Chemical Name	CAS No.	SMILES	Transformation Product Name	Transformation Product CAS	Transformation Product SMILES	log K <sub>ow</sub>	log K <sub>oc</sub>	VP (saturated mm Hg at 25°C)	Water Solubility (mg/L)	ECHA Dossier GHS Conclusion	EU Persistent Bioaccumulative and Toxic	California Toxic Air Contaminant	California Proposition 65
Rambutan peel extract	No CAS	NA			No	ECHA dossier availabl	e to support ev	valuation.		•	Not listed	Not listed	Not listed
Nano Calcium Carbonate Surface Modified by Gallic Acio	No CAS	NA			No	ECHA dossier availabl	e to support ev	valuation.			Not listed	Not listed	Not listed
Octyl Gallate	1034-01-1	CCCCCCCCCC(=0)C1=CC(=C(C(=C1)0)0)0			No	ECHA dossier availabl	e to support ev	valuation.			Not listed	Not listed	Not listed
8PPD	15233-47-3	CCCCCCC(C)NC1=CC=C(C=C1)NC2=CC=C2			No	ECHA dossier availabl	e to support ev	valuation.			Not listed	Not listed	Not listed
Specialized carbon nanotube mixture	No CAS	NA			No	ECHA dossier availabl	e to support ev	valuation.			Not listed	Not listed	Not listed

Notes:

Orange shading indicates a breakdown product of potential concern.

CAS No. = Chemical Abstracts Service Number; ECHA = European Chemicals Setem of Classification and Labelling of Chemicals; EU = European Union; K<sub>oc</sub> = Log Organic Carbon Partition Coefficient; NA = Not Available; SMILES = Simplified Molecular-Input Line-Entry System; VP = Vapor Pressure.

### Table 5.15 Stage 1 Alternatives Analysis Report Conclusions Based on Available Data

	,						Salmonid Toxicity			
Chemical	CAS	Human Health Score (Table 5.1)	Environmental Score (Table 5.3)	Physical Score (Table 5.3)	Total Score (Table 5.8)	Salmonid Toxicity Parent (ug/L) (Table 5.4)	Quinone/O <sub>3</sub> reaction product (ug/L) (Table 5.4)	Ingredient Exposure Potential (Table 5.9)	Summary of Existing Performance Data (Tables 5.11-5.13)	Conclusions
N-(1,3-dimethylbutyl)-N'-phenyl- p-phenylenediamine (6PPD)	793-24-8	125	150	0	275	140 (96 hr)	0.041 (24 hr)	-	NA	Candidate chemical in priority product
Possible Alternatives										
N-(1,4-dimethylpentyl)-N'- phenyl-p-phenylenediamine (7PPD)	3081-01-4	120	150	0	270	No data	No data	Similar environmental partitioning	Some promising performance data (Tables 5.11 and 5.12)	Include in Stage 2. Has hazard data and some promising performance data
N-isopropyl-N'-phenyl-p- phenylenediamine (IPPD)	101-72-4	150	100	0	250	No data	>50 (96 hr)	Slightly more likely to migrate in water	Some promising performance data (Tables 5.11)	Include in Stage 2. Has hazard data and some promising performance data
N-cyclohexyl-N'-phenyl-p- phenylenediamine (CPPD)	101-87-1	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	>50 (96 hr)	Slightly more likely to migrate in water	Limited data as an antiozonant (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards
N,N'-diphenyl-p- phenylenediamine (DPPD)	74-31-7	80	150	0	230	No data	>50 (96 hr)	Similar environmental partitioning	Poor performance (Table 5.12)	Drop from Stage 2 due to poor performance data
N-1,3-dimethyl butyl-N'-phenyl quinone diimine (6QDI)	52870-46-9	175	150	0	325	638 (96 hr)	No data	Substantially less water soluble and more fat soluble	Releases 6PPD when mixed in rubber (Table 5.11)	Drop from Stage 2. Releases 6PPD when mixed in rubber. Worse human health score than 6PPD
Polymerized 2,2,4-trimethyl-1,2- dihydroquinoline (TMQ)	26780-96-1	170	70	0	240	No data	No data	Similar environmental partitioning	Poor performance (Table 5.11)	Drop from Stage 2 due to poor performance data, worse human health score than 6PPD
N,N'-Bis(1,4-dimethylpentyl)-p- phenylenediamine (77PD)	3081-14-9	80	100	0	180	24 (96 hr)	>226 (96 hr)	Similar environmental partitioning	Some promising performance data (Tables 5.12)	Include in Stage 2. Has hazard data and some promising performance data
4,4'-Dioctyldiphenylamine (DOPD)	101-67-7	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Much less volatile and water soluble, much more carbon and fat soluble.	Limited data as an antiozonant (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards
N,N'-Ditolyl-p- phenylenediamine (Commercial DTPD)	68953-84-4	55	150	0	205	480 (96 hr)	>50 (96 hr)	Similar environmental partitioning	Poor performance (Table 5.12)	Drop from Stage 2 due to poor performance data
N,N'-Dicyclohexyl-p- phenylenediamine (CCPD)	4175-38-6	115	150	0	265	No data	No data	Similar environmental partitioning	Some promising performance data (Tables 5.12)	Include in Stage 2. Has hazard data and some promising performance data
Diaryl-p-phenylene diamine (DAPD)	68953-84-4	55	150	0	205	>480 (96 hr)	No data	Similar environmental partitioning	Poor performance (Table 5.12)	Drop from Stage 2 due to poor performance data
N,N'-Di-2-naphthyl-p- phenylenediamine (DNPDA)	93-46-9	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Much less volatile and water soluble, much more carbon and fat soluble.	Insufficient data, no ozone data (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards and no ozone data
Nickel dibutyldithiocarbamate (NBC)	13927-77-0	170	50	0	220	>100,000 (96 hr)	No data	Similar environmental partitioning	Poor performance (Table 5.11)	Drop from Stage 2 due to poor performance data, worse human health score than 6PPD
Ethoxyquin	91-53-2	135	70	0	205	18,000 (96 hr)	No data	More water soluble, higher vapor pressure and lower carbon and fat solubility	Poor performance (Table 5.11)	Drop from Stage 2 due to poor performance data
Dilauryl thiodipropionate	123-28-4	40	0	0	40	No data	No data	Less water soluble, low vapor pressure and more carbon and fat soluble	Insufficient data, no ozone data (Table 5.11)	Drop from Stage 2. No ozone data
N' -Phenyl.N-Fluorenyl-Para- Phenylenediamine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Less water soluble, low vapor pressure and more carbon and fat soluble	Some promising performance data (Tables 5.11)	Drop from Stage 2. Complete data gap regarding hazards
N-(p-phenylthiomethylphenyl)- N'-(1,3 dimethyl-butyl)-p- phenylenediamine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Less water soluble, low vapor pressure and more carbon and fat soluble	Limited data as an antiozonant (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards

						Salmonid Toxicity	Salmonid Toxicity			
Chemical	CAS	Human Health Score (Table 5.1)	Environmental Score (Table 5.3)	Physical Score (Table 5.3)	Total Score (Table 5.8)	Parent (ug/L) (Table 5.4)	Quinone/O <sub>3</sub> reaction product (ug/L) (Table 5.4)	Ingredient Exposure Potential (Table 5.9)	Summary of Existing Performance Data (Tables 5.11-5.13)	Conclusions
4-(2,5-dimethyl-1H-pyrrol-1-yl)- N-phenylaniline	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Similar environmental partitioning	Poor performance (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards and poor performance
N,N - (ethane-1,2-diyl ) bis (N- phenylbenzene-1 4-diamine [example chemical from patent]	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	More carbon soluble, lower vapor pressure, less water soluble	Some promising performance data (Tables 5.13)	Drop from Stage 2. Complete data gap regarding hazards
4-N-(2,3-dimethylphenyl)-1-N- phenylbenzene-1,4-diamine- R1 and R2 are methyl	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Similar environmental partitioning	Insufficient data, no ozone data (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards and no ozone data
RU997, Irgazone 997 (Reaction product of N-phenyl-N'- (1,3dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether)	444992-04-5	130	65	0	195	No data	No data	Less water soluble, low vapor pressure and more carbon and fat soluble	Insufficient data, no ozone data and shown to migrate too fast (Table 5.11)	Drop from Stage 2. No ozone data
4-[4-(4-Methylpentan-2- ylamino)anilino]phenol	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Similar environmental partitioning	Some promising performance data (Tables 5.13)	Drop from Stage 2. Complete data gap regarding hazards
Representative example from class (4-((4- (dimethylamino)phenyl)amino)p henol)	6358-22-1	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap		No data	No data	Similar environmental partitioning	Some promising performance data (Tables 5.13)	Drop from Stage 2. Complete data gap regarding hazards
N,N-diethyl-2,2,4-trimethyl-1H- quinolin-6-amine (R= N(C2H5)2	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Similar environmental partitioning	Some promising performance data (Tables 5.11)	Drop from Stage 2. Complete data gap regarding hazards
N,N'-Dibenzyl-p-xylene- alpha,alpha'-diamine-	25790-41-4	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Mixture, properties undetermined	Some promising performance data (Tables 5.11)	Drop from Stage 2. Complete data gap regarding hazards
2,4,6-tris-(N-1,4-dimethylpentyl- para-phenylenediamino)-1,3,5 triazine (Durazone 37 or TAPDT)	121246-28-4	50	100	0	150	No data	No data	Far less water soluble and lower vapor pressure and much more carbon and fat soluble	Poor performance (Table 5.11)	Drop from Stage 2 due to poor performance data
N-Phenyl-1-naphthylamine	90-30-2	45	150	0	195	440 (96 hr)	No data	Similar environmental partitioning	Poor performance (Table 5.11)	Drop from Stage 2 due to poor performance data
N-Phenyl-2-naphthylamine	135-88-6	180	20	0	200	No data	No data	Similar environmental partitioning	Insufficient data, no ozone data (Table 5.11)	Drop from Stage 2. No ozone data and worse human health score than 6PPD
[2-Methyl-4,6- bis((octylthio)methyl)phenol (Irganox 1520 CAS 110553-27-0)	110553-27-0	30	10	0	40	No data	No data	Less water soluble, low vapor pressure and more carbon and fat soluble	Limited data as an antiozonant (Table 5.11)	Include in Stage 2. Has hazard data and limited performance data as an antiozonant
Specialized graphene (e.g., Prophene™)	1034343-98-0	55	85	0	140	No data	No data	Organic carbon material, negligible vapor pressure (but may be suspended in air), similar water solubility	Some promising performance data (Tables 5.13)	Include in Stage 2. Has hazard data and some promising performance data
1,1' -Pentamethylenebis(2,2-Di- n- Butylhydrazine)	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap		No data	No data	Similar vapor pressure and carbon/fat solubility, greater water solubility	Limited data as an antiozonant (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards
α- C-4- hydroxy- 3,5- dimethylphenyl - N-isopropyl and Lowinox WSP CAS 77-62-3	Nitrone as a class, no CAS and Lowinox WSP - 77-62 3	however no	0 (for Lowinox WSP, however, no information on nitrone as a class)	0 (for Lowinox WSP, however, no information on nitrone as a class)	Not assigned based on lack of data for nitrone as a class	No data	No data	Mixture, properties for nitrone undetermined due to no CAS	Some promising performance data (Tables 5.11)	Drop from Stage 2. Complete data gap regarding hazards
N-(4-methylpentan-2-yl)-10H- phenothiazin-3-amine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Similar environmental partitioning	Some promising calculations (Tables 5.13)	Drop from Stage 2. Complete data gap regarding hazards

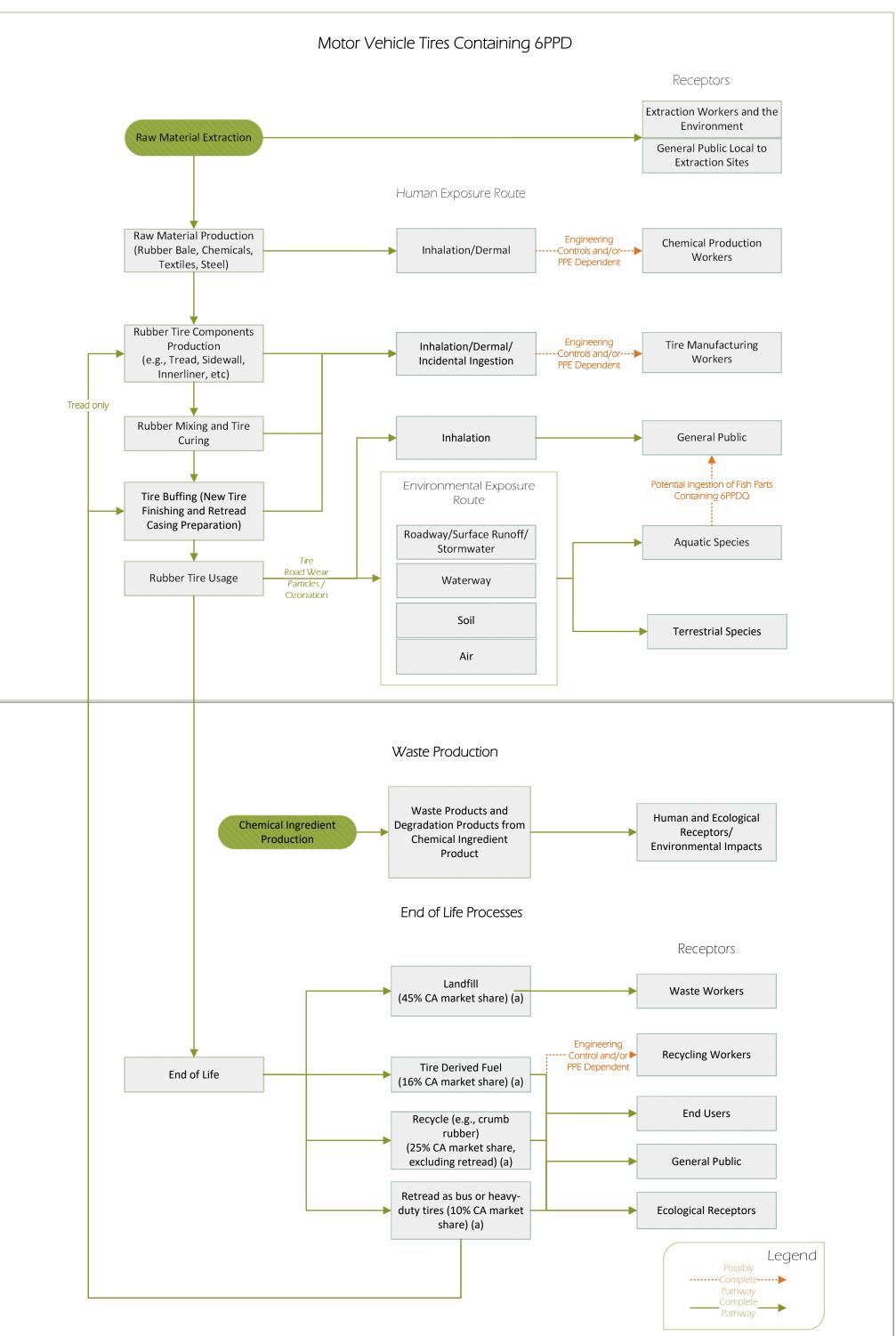
							Salmonid Toxicity			
Chemical	CAS	Human Health Score (Table 5.1)	Environmental Score (Table 5.3)	Physical Score (Table 5.3)	Total Score (Table 5.8)	Salmonid Toxicity Parent (ug/L) (Table 5.4)	Quinone/O <sub>3</sub> reaction product (ug/L) (Table 5.4)	Ingredient Exposure Potential (Table 5.9)	Summary of Existing Performance Data (Tables 5.11-5.13)	Conclusions
7-(4-methylpentan-2-ylamino)- 2,3,4,10-tetrahydro-1H-acridin-9- one	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Greater lipid solubility, similar vapor pressure, lower water solubility	Insufficient data, no ozone data (Table 5.13)	Drop from Stage 2. Complete data gap regarding hazards and no ozone data
2-cyclohexyl-N-(4-methylpentan- 2-yl)-1H-indol-5-amine	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Greater lipid solubility, similar vapor pressure, lower water solubility	Insufficient data, no ozone data (Table 5.13)	Drop from Stage 2. Complete data gap regarding hazards and no ozone data
4-(1H-indol-2-yl)-N-(4- methylpentan-2-yl)aniline	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Greater lipid solubility, similar vapor pressure, lower water solubility	Insufficient data, no ozone data (Table 5.13)	Drop from Stage 2. Complete data gap regarding hazards and no ozone data
α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Similar environmental partitioning	Some promising performance data (Tables 5.11)	Drop from Stage 2. Complete data gap regarding hazards
Amine functionalized lignin	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Polymer, properties undetermined	Some promising performance data (Tables 5.13)	Drop from Stage 2. Complete data gap regarding hazards
Rambutan peel extract	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Complex mixture, properties undetermined	Some promising performance data (Tables 5.13)	Drop from Stage 2. Complete data gap regarding hazards
Octyl gallate	1034-01-1	40	100	0	140	No data	No data	Similar environmental partitioning	Limited data as an antiozonant (Table 5.11)	Include in Stage 2. Has hazard data and limited performance data as an antiozonant
Nano calcium carbonate surface modified by gallic acid	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Complex chemical, properties undetermined	Some promising performance data (Tables 5.11)	Drop from Stage 2. Complete data gap regarding hazards
N-1-Methylheptyl-N'-phenyl-p- phenylenediamine (8PPD or UOP 688)	15233-47-3	Not assigned based on complete data gap	Not assigned based	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Similar environmental partitioning	Limited data as an antiozonant (Table 5.11)	Drop from Stage 2. Complete data gap regarding hazards
Specialized carbon nanotube mixtures	No CAS	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	Not assigned based on complete data gap	No data	No data	Complex chemical, properties undetermined	Some promising performance data (Tables 5.13)	Drop from Stage 2. Complete data gap regarding hazards

Notes:

CAS = Chemical Abstracts Service Number; NA = Not Available;  $O_3$  = Ozone.

Legend:

Substantially Higher than Priority Products			
Less desirable than Priority Product (for CSI score, worse by more than 30%)			
Similar to Priority Products			
Lower than Priority Products			
More desirable than Priority Product (for CSI score, better by more than 30%)			
Not Applicable: No Comparison Data Available			
Higher and Lower than Priority Products			



### Figure 4.2 Conceptual Exposure Model: Motor Vehicle Tires Containing 6PPD

(a) Of the 35% of recycled tires in California, 13% are recycled as crumb rubber, 10% are retreaded as tires for buses and heavy-duty trucks, 8% are sold as used tires, 1% are used as tire-derived aggregate, and

3% are recycled via other means. Source: CalRecycle, 2023

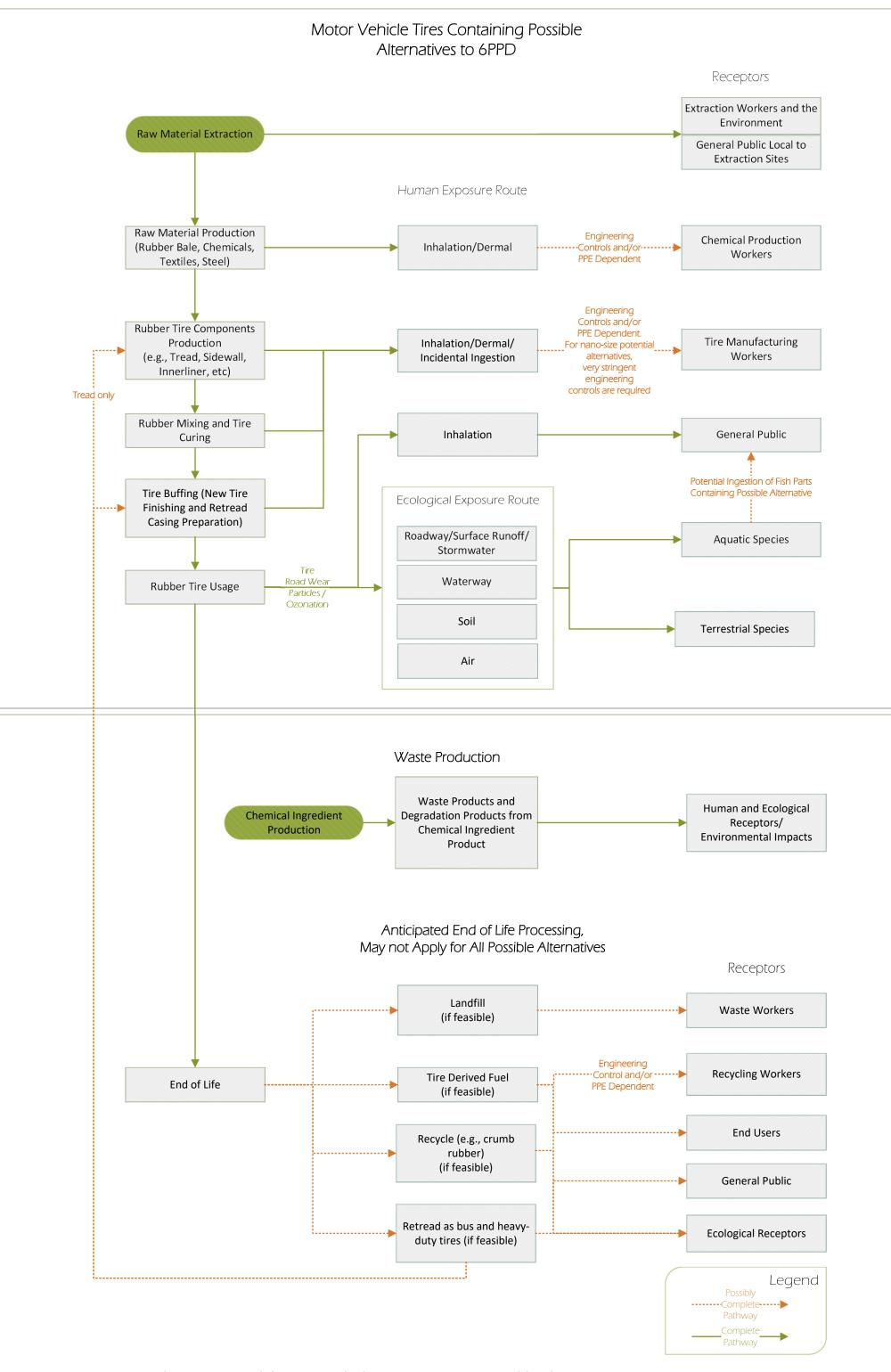


Figure 4.3 Conceptual Exposure Model: Motor Vehicle Tires Containing Possible Alternatives to 6PPD

# Appendix A

Glossary of Tire Related Terms

**Antidegradant**: Antidegradants are added to rubber compounds to protect tires from overly rapid deterioration by ozone, oxygen, fatigue, and heat. Antidegradants include both antioxidants and antiozonants. Antidegradants in tires must serve in two load performance conditions – static and dynamic operations modes - which describe when the tire is at rest or flexing under motion, respectively.

Antioxidant: A compound that helps to keep rubber from breaking down due to the effects of temperature and oxygen exposure.

Antiozonant: A compound that impedes the effects of exposure to ozone on the surface of the tire.

**Bead**: The tire bead is the portion (or component) of the tire that sits on the rim of the wheel. Tire beads are steel wire bundles that are coated with a specific rubber compound and secure the tire to the metal wheel.

**Bead Filler**: A rubber compound placed above the bead that may be used between the body plies, which wrap around the bead to enhance ride and handling characteristics.

**Belts**: Belts provide stability to the tread area of the tire, which minimizes wear and contributes to vehicle handling and traction. Typically, two belts with steel cords laid at opposing angles form a hoop under a tire's tread. The steel belt is coated with a rubber compound that is called a belt coat or belt skim compound.

**Body Plies**: Body plies function as the base structure of the tire and provide the strength to contain the inflation pressure. Most car tires have one or two body plies, each typically comprised of textile cords within a rubber layer. Truck and bus tires typically use steel cords for body plies.

**Curing Systems**: Sulfur, chemical accelerators (often derivatives of benzothiazole), stearic acid, and zinc oxide are crucial ingredients for vulcanization, which transforms soft uncured rubber into a solid elastic article during tire curing. Curing systems not only enable vulcanization, but also shorten the vulcanization time and impact the length and number of crosslinks in the rubber matrix which, in turn, affects the rubber's properties.

**Dynamic Load Performance**: Antioxidants and antiozonants with dynamic operation modes protect the tire while it is in motion and being flexed.

**Endurance**: Tire endurance is a measurement of how long a tire can withstand severe conditions before reaching its limit. Endurance can be tested by varying the speed, load, inflation pressure, temperature, and/or number of cycles.

**Field Testing**: Tire manufacturers may conduct field testing to obtain performance data for tires operated under real-life conditions for an extended period of time. Field testing is typically performed by a contracted fleet with routine monitoring by the tire manufacturer.

**Fillers**: Multiple grades of carbon black and coupled/uncoupled precipitated amorphous silica are used as fillers to reinforce the rubber and modify its properties, resulting in improved wear performance and traction.

**Gravel Chip/Tear**: For passenger and light truck tires that are intended to be driven off road, an evaluation is conducted on a vehicle on a gravel route to assess chipping and tearing of tread elements.

Handling: Handling is a result of tire/vehicle interactions in response to various driver inputs.

Inner Liner: A rubber compound used to retain the inflation pressure inside the tire.

Irregular Wear: Uneven or abnormal wear features observed during a wear test.

**Natural Rubber**: Natural rubber provides specific performance characteristics to tires, such as tear and fatigue crack resistance. Some tires, especially truck and bus tires, use natural rubber in tread compounds to provide reduced rolling resistance (the resistance the tire encounters when rolling down the road – an important consideration for fuel efficiency). Natural rubber is a form of polyisoprene which is obtained by tapping rubber trees (*Hevea brasiliensis*).

Non-Pneumatic Tire: A type of tire that is airless.

**Original Equipment (OE)**: Equipment supplied on a vehicle at its time of purchase.

**Original Equipment Manufacturer (OEM)**: Manufacturer of original equipment (OE) supplied on a vehicle at its time of purchase.

**Original Equipment Tires (OE Tires)**: OE tires must meet specific, often numerous and complex performance requirements specific to the vehicle manufacturer. OE tires are designed to a specific vehicle model year/make/model/trim level combination, and any changes to the materials used to manufacture OE tires, or the tire design itself, would require approval from the vehicle manufacturer. While tires installed on new vehicles are not part of the Priority Product definition, OE tires are also considered replacement tires due to requirements in OE contracts for OE tires to be available as replacements, customer demand for OE tires in the replacement market, and to manage excess OE tire inventory. OE tires typically do not come with treadwear warranties. For purposes of this Stage 1 AA, OE tires are considered to be a subset of the replacement tire market and included in the analyses.

**Pneumatic Tire**: A type of tire that is filled with air.

**Processing Aids**: Bio-based oils, low aromatic petroleum oils, pine tar, and resins are the most common softening agents used in rubber compounding. Tackifying resins can be added to increase the rubber compound stickiness (tack), which helps the various tire components stick together assembly of tire components.

**Replacement Tires**: Tires designed for the replacement market to perform well on a wide range of vehicles – often as many as 30 different vehicle applications are appropriate for a single tire service description (tire size/speed rating/load index combination). Passenger and light truck replacement tires can be installed by a tire dealer or other tire service professional without original equipment manufacturer (OEM) approval.

**Rolling Resistance**: The force necessary to keep a tire rolling.

**Rolling Resistance Coefficient (RRC)**: Rolling resistance is measured according to ISO 28580:2018 and is expressed in terms of rolling resistance coefficient (RRC). To measure rolling resistance, a load is placed on the tire while it is being forced to turn by the drum and the resistance force, which the tire generates to prevent it from turning, is measured.

**Sidewall**: A rubber compound used to cover the body plies on the sides of the tire, which provides abrasion, scuff, and weathering resistance.

**Static Load Performance**: Antioxidants and antiozonants with static operation modes form a coating that protect the tire when it is in its resting and stationary state.

**Synthetic Polymers**: The two main synthetic rubber polymers, or elastomers, used in tire manufacturing are butadiene rubber (BR) and styrene butadiene rubber (SBR). These synthetic rubber polymers are used in combination with natural rubber. The physical and chemical properties of these rubber polymers determine the performance of each component in the tire as well as the overall tire performance. Another important synthetic rubber is halogenated polyisobutylene rubber, commonly known as halobutyl rubber, which is used in the inner liner. This material causes the inner liner to have reduced air permeability, which helps to keep the tire inflated.

Tack: The stickiness of a green, or uncured, rubber compound.

**Tire**: As used in this document, "tire" refers to a pneumatic radial tire used with motor vehicles (*e.g.*, passenger cars and light duty trucks; heavy duty trucks and buses).

**Tire Identification Number (TIN)**: A string of letters and numbers on the tire sidewall that begins with the letters DOT. New passenger and some new light truck tires are required by federal law to have a full TIN on the intended outboard side and a partial TIN on the intended inboard side. All new commercial truck and bus tires, motorcycle tires, and some light truck tires are required to have a full TIN on the intended outboard side.

**Tire and Road Wear Particles (TRWP)**: Particles produced as the tire grips and interacts with the road surface during driving.

**Tread**: Located on the road-contacting portion of the tire, the tread rubber compound and tread pattern provide grip and abrasion resistance contributing to traction and treadwear.

**Vulcanization**: The process in which heat is applied to the green, or uncured, rubber compound causing a chemical reaction among sulfur, other chemicals, and polymers (elastomers) in the rubber compound. These reactions result in chemical bonds (cross links) between the polymer (elastomer) chains to produce cured tires.

**Wear Rate**: Usually measured in miles of travel per thousandth of an inch of tread depth loss (*i.e.*, miles per mil) or as tread loss per mileage increment (*i.e.*, mils/1,000 miles).

# Appendix B

List of Products Covered by This AA

#### America Kenda Rubber Ind Co. America Kenda Rubber Ind Co. Apollo Tyres Limited Apollo Tyres Limited **Apollo Tyres Limited** Apollo Tyres Limited Bridgestone Americas, Inc. **CEAT Limited CEAT Limited CEAT** Limited **CEAT** Limited **CEAT** Limited China Manufacturers Alliance, LLC China Manufacturers Alliance, LLC Continental Tire the Americas, LLC Continental Tire the Americas. LLC Continental Tire the Americas, LLC Continental Tire the Americas. LLC Continental Tire the Americas, LLC Continental Tire the Americas. LLC Continental Tire the Americas, LLC

**Responsible Entity Name** 

Brand/Trade Name Kenda Kenda Apollo Apollo Apollo Vredestein Bandag Bridgestone Bridgestone Bridgestone Firestone Firestone Fuzion SureDrive CEAT CEAT CEAT Private label Private label Double Coin Warrior Airfix Airfix Airfix Ameri\*Steel America Bandvulc bandvulc Barum Barum Barum BestDrive Blackstone Capitol Continental Continental Continental Continental Contire Contitread Cosmos Dunlop Dunlop Dunlop ESA+ Tecar Eurostone Eurotec Eurotyre Euzkadi Euzkadi Fate Feu Vert Flamingo General General General General General Tire Gislaved Global Hoosier Kingstone Kormoran Mabor

#### Tire Type

Motorcycle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires **Retread Material for Tires** Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Motorcycle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Dehicle tires Medium and Heavy Duty Dehicle tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Motorcycle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires **Retread Material for Tires** Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Motorcycle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and heavy duty tires **Retread Material for Tires** Light Duty Vehicle Tires Motorcycle Tires Medium Duty and Heavy Duty Tires **Retread Material for Tires** Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Motorcycle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Motorcycle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires

**Responsible Entity Name** Continental Tire the Americas, LLC Continental Tire the Americas. LLC Continental Tire the Americas, LLC Giti Tire Guizhou Tyre Co., Ltd. Guizhou Tyre Co., Ltd. Guizhou Tyre Co., Ltd. Guizhou Tyre Co., Ltd. Guizhou Tvre Co., Ltd. Guizhou Tvre Co.. Ltd. Hankook Tire America Corporation Hankook Tire America Corporation Hankook Tire America Corporation Hankook Tire America Corporation

Marongoni Matador Matador Mazama Midas Midas Minerva MYCAR Nichols Norauto Paxaro Platin Point S Retrak Roadhandler ROADHOG Sebring Semperit Semperit Semperit Sidewinder Sime Sime Tyres SIME TYRES SIMEX SIMEX SIMEX Speedy Sportiva Sumitomo Sumo Taxat Team Star TEAMSTAR Tecnotread Tiger Wheel Toyo Truckstar Tyfoon Ultrex IV Uniroyal Uniroyal Uniroyal Viking Yokohama Yokohama Dextero Giti Giti GT Radial GT Radial Primewell **Rocky Mountain** Runway ADVANCE ADVANCE SAMSON SAMSON TORNADO TORNADO Hankook Hankook Hankook Hankook & New Englander

Brand/Trade Name

#### Tire Type

Medium Duty and Heavy Duty Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Motorcycle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Motorcycle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires **Retread Material for Tires** Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty Vehicle Tires Heavy Duty Vehicle Tires Light Duty Vehicle Tires

#### **Responsible Entity Name** Hankook Tire America Corporation Hebei Wanda Tyre Co., Ltd. Jiangsu General Science Technology Co., Ltd.

Hankook & Traction Control Laufenn Laufenn Laufenn PathFinder Statewide AIR-LOC ALL STAR AMERITRAIL AMS ANTEGO BLACK TRAIL CADDIE MASTER CARAWAY DEWOSTONE ECO TRAIL ST EFX ELEVATE STR EVOLUTION EXCEL FREE COUNTRY GOLBALTRAX GOLDENWEST GOLFPROPLUS HAKUBA HI-RUN INTERCO JET STAR JIMEXS JOURNEY К9 LSI ELITE ΜΑΧΑUTO **MILESTAR** MODZ NANCO OBOR OCELOT OTR POWER KING PRIMEX RAINIER ST RHOX RUBBERMASTER SLINGSHOTXT SPEEDUTV STEELENG SUPER GRIP SYNERGY TEX STAR TRAC GARD TRAILFINDER TRAILQUEST TRAXION TREAD-STAR TUSK VISION WANDA WD-SIGNATURE WDT WD-VELOCITY

wolfpack

ZEEMAX

Achilles

Brand/Trade Name

#### Tire Type

Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty Vehicle Tires Heavy Duty Vehicle Tires Light Duty Vehicle Tires

#### **Responsible Entity Name** Jiangsu General Science Technology Co., Ltd. Jiangsu General Science Technology Co., Ltd.

Brand/Trade Name Advanta Advanta American Roadstar American Roadstar Antini Arroyo Arroyo Atlander Atlander Atturo Aufine Barkley Barklev CELIMO CELIMO Converse Duro Evertour Finalist Finalist Fury Off Road Gladiator Gladiator Goodtrip Goodtrip Gremax Gremax Gripmax Gripower Hillrock Knight Kwik Lancaster Lancaster Landspider Landspider Lenso Lexmont Longmarch Magna Mastertrack Mastertrack Matrax Matrix Mazzini Megalodon Milestar Miletrip Mud Claw Nama Navitrac Nebula Neoterra Neoterra Randhawa Rockfleet Sentinel Sentinel Supercargo TBBtires TBBtires Tekpro Tesche Tomoro

#### Tire Type

Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires

#### **Responsible Entity Name**

Jiangsu General Science Technology Co., Ltd. JK Tyre and Industries Limited Kumho Tire Kumho Tire Linglong Americas Linglong Americas

Brand/Trade Name Towmax Travelstar Travelstar Unigrip Vantage Venom Power Vitour Xcellent Xcellent Zenna Zenna Celestis Coker Ironman JK Tvre JK Tyre New Pride Pearly Tornel Vikrant Kumho Tire Kumho Tire Altenzo AMP Annex Atlas Atlas Aufine Black Bear Constellation Contender Crosswind Crosswind Custom Dynatrac Dynatrac Epic Tour Evoluxx Evoluxx Finalist Freedom Hauler Freedom Hauler Geostar Geostar Geo-Trac Giovanna Gladiator Gladiator **GREEN-Max GREEN-Max** Gripmax Gripmax Gripower Grit King Grit Master Hemisphere Hercules Hercules Hubtrac Ironman Ironman Leao Leao Linglong

#### **Tire Type** Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires

**Responsible Entity Name Linglong Americas** Linglong Americas **Linglong Americas** Linglong Americas Maxxis Technology Center (Cheng Shin USA Tech Center) Michelin North America Inc. Michelin North America Inc.

Maxxis Technology Center (Cheng Shin USA Tech Center)

Michelin North America Inc.

Michelin North America Inc.

Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK

Milestar Nama Omni Percheron Percheron Performer Pinnacle Pinnacle Predator Prometer Prometer Provato Provato Provider Provider Remington RoadOne Roadone Sentry Solidana Super Cargo Superior Suretrac Suretrac Symmetry Synergy Taskmaster Trailermaster Trailermaster TransEagle TransEagle Travelstar Travelstar Tri-Ace Venezia Venom Power Venom Power Vercelli Versatyre Wild Spirit Zenna Zenna Maxxis Maxxis **BF** Goodrich **BF** Goodrich Megamile Michelin Michelin Michelin Michelin Oliver Recamic Riken Uniroyal Uniroyal ACHILLES ACHILLES ATR 122 ATR RADIAL ATR SPORT ATR-K BELLAGIO

Brand/Trade Name

Linglong

#### Tire Type

Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires **Retread Material for Tires** Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Motorcycle Tires Retread Material for Tires **Retread Material for Tires Retread Material for Tires** Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Tires Light Duty Vehicle Tires

#### **Responsible Entity Name**

Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK Michelin North America, Inc. on behalf of PT Multistrada Arah Sarana TBK

NEXEN TIRE CORPORATION Nokian Tyres US Operations LLC Ohtani Radial Co., Ltd Otani Tire Co., Ltd Pirelli Tire LLC Pirelli Tire LLC Pirelli Tire LLC Prinx Chengshan Holdings Limited Prometeon Tyre Group Commercial Solutions, LLC Prometeon Tyre Group Commercial Solutions, LLC Qingdao Sentury Tire Co., Ltd. Qingdao Sentury Tire Co., Ltd.

#### Brand/Trade Name

CORSA

CORSA GOLDENBRIDGE MILESTAR MONTANA NEUTON PINSO RADAR SINGA RADIAL STRADA SYRON Aspen Geotour Hercules NEXEN Sceptor Solar Nokian Tyres Nokian Tyres Nokian Tyres Nordman Private brand Otani Otani Metzeler Pirelli Pirelli AMP/TWG Crossmax/Otai Fortune Fortune Kelly/Goodyear Muscle Power/Delta Pathraider/Otai Prinx Prinx Radar/Omni United Rainier St/Otai Sotera/Otai Synergy/Sutong Tourador/Otai UniRoyal/TBC Wellplus/Otai Nextroad Pirelli AEROTYRE ALBOURGH ALBOURGH AMERICAN ROAD STAR AMERICAN ROAD STAR AMERICAN TOURER ARROYO ARROYO AVANTECH Avantech BARKLEY BARKLEY DAVANTI DCENTI Delinte Delinte DELTA ELDORAD GREENTRAC

#### **Tire Type**

Motorcycle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires **Retread Material for Tires** Light Duty Vehicle Tires Light Duty Vehicle Tires Tires Tires Motorcycle Tires Light Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires

**Responsible Entity Name** Qingdao Sentury Tire Co., Ltd. Sailun Group Co., Ltd.

Brand/Trade Name GREENTRAC Groundspeed Groundspeed HOPEWAY KADO Kinforest KINFOREST LANDFLEET LANDFLEET LANDGOLDEN LANDGOLDEN LANDSAIL LANDSAIL LAZZAZ I FXANI LIONHART LIONHART MASTERSTEEL Mavis MOHAVE MUDDER TRUCKER MULTI-MILE NATIONAL NEBULA NEBULA NEOTERRA PACE PANTERA PATRIOT PHYRON PHYRON RADAR RADAR Raiden ROADHOG **ROLLING BIG POWER ROLLING BIG POWER** SENTURY SpeedMax SURETRAC SURETRAC TRANSMAX TRANSMAX TYFOON VANDERBILT VELOZZA VENEZIA VERCELLI Vercelli II Vercelli IV Wild Spirit ZEETEX ZETA ZETA Blackhawk Blackhawk Dynamo Dynamo Ironhead Ironhead Maxam Private brand Private brand Private brand

#### Tire Type

Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires

**Responsible Entity Name** Sailun Group Co., Ltd. SHANDONG HAOHUA TIRE CO., LTD. Shandong Jinyu Tire Co., Ltd. Sumitomo Rubber Industries, Ltd. Sumitomo Rubber Industries. Ltd. Sumitomo Rubber Industries. Ltd. Sumitomo Rubber Industries, Ltd. Sumitomo Rubber Industries, Ltd. Sumitomo Rubber Industries. Ltd. Sumitomo Rubber Industries, Ltd.

Private brand Roadx Roadx Rovelo Rovelo Sailun Sailun APLUS COMPASAL GRANDSTONE HAOHUA LANVIGATOR MAGNA POWERTRAC **ROYAL BLACK** WIDEWAY WINDFORCE Amulet Eudemon Evergreen Geoquest Vitour Aspen GT-AS Delta Doral Dunlop Dunlop Dunlop Eldorado Falken Falken Geotour Grand Spirit Mazama Multi-Mile National

Brand/Trade Name

#### **Tire Type**

Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Passenger Car Tire and Truck & Bus Tire Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires

Light Duty Vehicle Tires

**Responsible Entity Name** Sumitomo Rubber Industries, Ltd. Tianjin Wanda Tyre Group Co., Ltd.

Brand/Trade Name Ohtsu Ohtsu Sumitomo Sumitomo **Trailcutter AT4S** XTRRT Air-Loc All Star Ameritrail AMS Antego Antego Black Trail Caddie Master Caraway Dewostone Eco Trail ST EFX Elevate STR Evolution Excel Free Country Golbaltrax Goldenwest Golfproplus Hakuba HI-RUN Interco Jet Star limexs Journey К9 LSI Elite Maxauto Milestar Modz Nanco Obor Ocelot OTR Power King Primex Quadboss Rainier ST RHOX Rubbermaster Slingshotxt Speedutv Steeleng Super Grip Synergy Tex Star Trac Gard Trailfinder Trailquest Traxion Tread-Star Tusk Vision Wanda WD-Signature WDT WD-Velocity Wolfpack

#### Tire Type

Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires

Responsible Entity Name
Tianjin Wanda Tyre Group Co., Ltd.
The Goodyear Tire & Rubber Company
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Brand/Trade Name Zeemax ARIZONIAN Avon Avon Beltak BIG O Blackstone **BLUE STREAK TBA 8 BLUE STREAK TBA 9 BLUE STREAK-BLSTRK BLUE STREAK-GDYR** Centara Cooper Cooper Cooper Dean Dean Dean PB Debica Debica Debica Diamondback Diplomat Douglas Dunlop Dunlop Dunlop Dunlop Dunlop Dunlop Durun Falke Fulda Fulda Fulda Futura Geotred Geotred Goodyear Goodyear Goodyear Hercules Ironman Kelly Kelly Kelly Lemans Lexington Loadrunner Mastercraft Mastercraft **Mickey Thompson Mickey Thompson Mickey Thompson** Motomaster Motrio Non-core brand O'Green Omega Provato Remington Remington Roadmaster Roadstone

Tire Type Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Motorcycle Tires Retread Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Motorcycle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Retread Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires

#### **Responsible Entity Name** The Goodyear Tire & Rubber Company Toyo Tire Holdings of Americas Inc. Triangle Tyre Co., Ltd Yokohama Tire Corporation Yokohama Tire Corporation Yokohama Tire Corporation Yokohama TWS North America, Inc. ZC Rubber America Inc. 7C Rubber America Inc. ZC Rubber America Inc. ZC Rubber America Inc. 7C Rubber America Inc. ZC Rubber America Inc.

Brand/Trade Name Sava Sava Sava Sava Starfire Starfire Steelmark Ultima ARCTIC CLAW BIG O FIRESTONE IRONMAN NITTO ROAD HUGGER TOYO TOYO DIAMONDBACK DIAMONDBACK DIAMONDBACK DIAMONDBACK TRIANGLE TRIANGLE TRIANGLE TRIANGLE Galaxy Yokohama Yokohama Mitas ADVANTA ADVANTA ΔΟΛΔΝΤΦ AMERICAN TOURER AMERICAN TOURER AMERICAN TOURER AMERICUS AMERICUS AMERICUS AMP AMP AMP ANGLER ANGLER ANGLER ARISUN ARISUN ARISUN ARISUN BISON BISON BISON BULL BUH BULL CASTLE ROCK CASTLE ROCK CAVALRY CAVALRY CAVALRY CHAOYANG CHAOYANG CHAOYANG CHAOYANG COSMO COSMO

#### **Tire Type** Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Retread Light Duty Vehicle Tires Medium and Heavy Duty Vehicle Tires Special Tyre for Trailer Passenger Car Radial Tyres Light Truck Tyres Truck-Bus Truck-Bus Special Tyre for Trailer Passenger Car Radial Tyres **Light Truck Tyres** Medium Duty and Heavy Duty Vehicle Tires Light Duty Vehicle Tires Medium Duty and Heavy Duty Vehicle Tires Motorcycle Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires** Medium Truck Tires Motorcycle Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Motorcycle Tires **Passenger Car Tires Light Truck Tires**

**Responsible Entity Name** ZC Rubber America Inc. 7C Rubber America Inc. ZC Rubber America Inc. 7C Rubber America Inc. ZC Rubber America Inc. ZC Rubber America Inc. 7C Rubber America Inc. ZC Rubber America Inc.

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Brand/Trade Name

#### Tire Type

**Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires Passenger Car Tires Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Motorcycle Tires Passenger Car Tires Light Truck Tires Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires Light Truck Tires **Medium Truck Tires** Passenger Car Tires **Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires Light Truck Tires **Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires Medium Truck Tires Passenger Car Tires Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires Passenger Car Tires** Light Truck Tires

**Responsible Entity Name** ZC Rubber America Inc. ZC Rubber America Inc.

Brand/Trade Name RED FLAME

RISEN

RISEN

RISEN

SENTINEL

SENTINEL

SENTINEL

STERLING

STERLING

SUPERCARGO

SUPERCARGO

SUPERCARGO

SUPERMAX

SUPERMAX

SUPERMAX

TAMARACK

TAMARACK

TRAILFINDER

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#### Tire Type

Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires Light Truck Tires Medium Truck Tires Light Truck Tires **Medium Truck Tires** Passenger Car Tires Light Truck Tires Medium Truck Tires Passenger Car Tires **Light Truck Tires** Medium Truck Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires **Light Truck Tires** Medium Truck Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** Passenger Car Tires Light Truck Tires Medium Truck Tires Passenger Car Tires Light Truck Tires Medium Truck Tires Motorcycle Tires Passenger Car Tires Light Truck Tires Passenger Car Tires Light Truck Tires Medium Truck Tires Motorcycle Tires Passenger Car Tires **Light Truck Tires Medium Truck Tires** 

# Appendix C

SDS for Santoflex<sup>™</sup> 6PPD Pastilles



## Santoflex(TM) 6PPD Pastilles

Version 2.9 PRD	Revision Date: 05/04/2022	15	DS Number: 50000093128 55US / Z8 / 0528	Date of last issue: 04/28/2022 Date of first issue: 09/06/2016				
SECTION	1. IDENTIFICATION							
Product name		:	Santoflex(TM) 6PPD Pastilles					
Product code		:	P3408704					
Manu	Manufacturer or supplier's details							
Comp	Company name of supplier : Flexsys Chemicals Belgium NV							
Address		:	Scheldelaan 460, Haven 627 Antwerpen 2040					
Emergency telephone		:	CHEMTREC: +1 703-741-5970					
Recommended use of the chemical and restrictions on use								
Recor	nmended use	:	: antioxidant (industrial) Stabilizer					
Restri	ctions on use	:	None known.					

## SECTION 2. HAZARDS IDENTIFICATION

GHS classification in accor 1910.1200)	dar	nce with the OSHA Hazard Communication Standard (29 CFR
Acute toxicity (Oral)	:	Category 4
Skin sensitization	:	Category 1
Reproductive toxicity	•	Category 1B
GHS label elements		
Hazard pictograms	:	
Signal Word	:	Danger
Hazard Statements	:	H302 Harmful if swallowed. H317 May cause an allergic skin reaction. H360 May damage fertility or the unborn child.
Precautionary Statements	:	<b>Prevention:</b> P201 Obtain special instructions before use. P202 Do not handle until all safety precautions have been read and understood.



## Santoflex(TM) 6PPD Pastilles

Versior 2.9 PRD	n Revision Date: 05/04/2022	SDS Numb 150000093 SDSUS / Z8 /	128 Date of first issue: 09/06/2016				
		P264 W P270 Do P272 Co the work	ear protective gloves/ protective clothing/ eye protection/				
		CENTE P302 + P308 + attentior P333 + attentior	P312 + P330 IF SWALLOWED: Call a POISON R/ doctor if you feel unwell. Rinse mouth. P352 IF ON SKIN: Wash with plenty of soap and water. P313 IF exposed or concerned: Get medical advice/ n. P313 If skin irritation or rash occurs: Get medical advice/				
		<b>Storage</b> P405 St	e: ore locked up.				
		P501 Di	<b>Disposal:</b> P501 Dispose of contents/ container to an approved waste disposal plant.				
•	t <b>her hazards</b> one known.						
SECTI	SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS						
Su	ubstance / Mixture	: Substar	се				
Su	ubstance name	: N-(1,3-E	Dimethylbutyl)-N'-phenyl-p-phenylenediamine				
C	AS-No.	: 793-24-	793-24-8				

### Components

Chemical name	CAS-No.	Concentration (% w/w)
N-1,3-Dimethylbutyl-N'-phenyl-p-	793-24-8	>= 95 - <= 100
phenylenediamine		

Actual concentration is withheld as a trade secret

Flexsys is committed to the safety, health and environment of our employees, our customers, and the communities we operate within. As part of this commitment, Flexsys' Safety Data Sheets (SDS) are prepared in accordance with all applicable national and local regulations. The compositions of our documents reflect these requirements which include, but are not limited to, requirements under the Globally Harmonized System of Classification and Labeling (GHS). These compositions commonly involve the use of ranges versus specific analytical values. If you require a composition that is more specific , please refer to the Certificate of Analysis, sales specification, or contact your Customer Service Representative.

# SAFETY DATA SHEET



# Santoflex(TM) 6PPD Pastilles

Version 2.9 PRD	Revision Date: 05/04/2022	15	DS Number: 50000093128 DSUS / Z8 / 0528	Date of last issue: 04/28/2022 Date of first issue: 09/06/2016		
SECTION	N 4. FIRST AID MEASUR	RES				
If inhaled		:	breathing. If breathing is diff	Remove person to fresh air and keep comfortable for breathing. If breathing is difficult, give oxygen. Consult a physician if necessary.		
In case of skin contact		:	Wash off immediately with soap and plenty of water while removing all contaminated clothes and shoes. If skin irritation or rash occurs: Get medical advice/ attention. Wash contaminated clothing before reuse.			
In case of eye contact		:	In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention if symptoms occur.			
If swallowed : Rinse mouth. Get medical attention if symptoms occur. Do not induce vomiting unless directed to do so personnel. Never give anything by mouth to an unconscious		miting unless directed to do so by medical				
	t important symptoms effects, both acute and yed	:		ved. ergic skin reaction. lity or the unborn child.		
Note	es to physician	:	Treat symptomati	cally.		

# **SECTION 5. FIRE-FIGHTING MEASURES**

Suitable extinguishing media	:	Water spray Foam Dry powder Carbon dioxide (CO2)
Unsuitable extinguishing media	:	Do not use a solid water stream as it may scatter and spread fire.
Specific hazards during fire fighting	:	Do not allow run-off from fire fighting to enter drains or water courses.
Hazardous combustion prod- ucts	:	Carbon oxides Nitrogen oxides (NOx)
Further information	:	In case of fire and/or explosion do not breathe fumes.
Special protective equipment for fire-fighters	:	Wear an approved positive pressure self-contained breathing apparatus in addition to standard fire fighting gear.

# SECTION 6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protec- : Wear appropriate personal protective equipment.



Vers 2.9 PRD		Revision Date: 05/04/2022	15	DS Number: 0000093128 SUS / Z8 / 0528	Date of last issue: 04/28/2022 Date of first issue: 09/06/2016
tive equipment and emer- gency procedures			Avoid prolonged or repeated contact with skin. Avoid breathing dust/ fume/ gas/ mist/ vapors/ spray. Ensure adequate ventilation. Material can create slippery conditions. Local authorities should be advised if significant spillages cannot be contained.		
	Enviror	nmental precautions	<ul> <li>Prevent further leakage or spillage if safe to do so.</li> <li>Avoid release to the environment.</li> <li>Collect spillage.</li> </ul>		
		ds and materials for Iment and cleaning up	:	Sweep up and sh	ovel into suitable containers for disposal.

# SECTION 7. HANDLING AND STORAGE

Advice on safe handling	:	Wear appropriate personal protective equipment. Avoid breathing dust/ fume/ gas/ mist/ vapors/ spray. Handle product only in closed system or provide appropriate exhaust ventilation at machinery. Drain or remove substance from equipment prior to break-in or maintenance. Wash thoroughly after handling. Do not eat, drink or smoke when using this product. Contaminated work clothing should not be allowed out of the workplace.
Conditions for safe storage	:	Keep containers tightly closed in a dry, cool and well- ventilated place.

# SECTION 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

### Ingredients with workplace control parameters

Contains no substances with occupational exposure limit values.

Engineering measures : Good general ventilation (typically 10 air changes per hour) should be used. Ventilation rates should be matched to conditions. If applicable, use process enclosures, local exhaust ventilation, or other engineering controls to maintain airborne levels below recommended exposure limits. If exposure limits have not been established, maintain airborne levels to an acceptable level.

## Personal protective equipment

Respiratory protection	<ul> <li>Use a properly fitted, particulate filter respirator complying with an approved standard if a risk assessment indicates this is necessary.</li> </ul>
	Respirator selection, use, and maintenance must be in accordance with regulatory requirements, if applicable. If engineering controls do not maintain airborne concentrations below recommended exposure limits (where



Version 2.9 PRD	Revision Date: 05/04/2022	SDS Number: 150000093128 SDSUS / Z8 / 0528	Date of last issue: 04/28/2022 Date of first issue: 09/06/2016
			or to an acceptable level (in countries where nits have not been established), an approved ust be worn.
Hand	protection		
Remarks		After contain immediately	ed or repeated contact use protective gloves. hination with product change the gloves and dispose of them according to relevant local regulations.
Eye protection		be used whe to avoid exp	ear complying with an approved standard should on a risk assessment indicates this is necessary osure to liquid splashes, mists, gases or dusts. glasses with side shields (or goggles).
Skin and body protection		selected bas involved and handling this	otective equipment for the body should be sed on the task being performed and the risks I should be approved by a specialist before s product. I wash contaminated clothing before re-use.
Protec	ctive measures		eye flushing systems and safety showers are e to the working place.

SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES				
Appearance	:	Pastilles		
Color	:	purple, brown		
Odor	:	aromatic		
Odor Threshold	:	not determined		
рН	:	not determined		
Melting point/range	:	120 °F / 49 °C (1,013 hPa)		
Boiling point/boiling range	:	325 - 329 °F / 163 - 165 °C (1.33 hPa)		
Flash point	:	396 °F / 202 °C (1,013 hPa) Method: Pensky-Martens closed cup		
Evaporation rate	:	not determined		
Flammability (solid, gas)	:	Not classified as hazardous.		
		5 / 15		



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Upper explosion limit / Upper flammability limit		:	not determined		
	Vapor	pressure	:	0.000036 hPa (7	7 °F / 25 °C)
	Relativ	e vapor density	:	not determined	
	Relativ	e density	:	0.995 (122 °F / 5	0 °C)
	Density	/	:	995 kg/m3 (122	°F / 50 °C)
	Solubil	ity(ies)			
	Wat	ter solubility	:	0.001 g/l(122 °ł	= / 50 °C)
	Solu	ubility in other solvents	:	soluble Solvent: Hydroca	arbons
	Partitio octano	n coefficient: n- I/water	:	log Pow: 4.68 (6	8 °F / 20 °C)
	Autoigr	nition temperature	:	1022 °F / 550 °C Method: VDI 226 Dust	
	Decom	position temperature	:	> 392 °F / > 200	°C
	Viscosi Visc	ity cosity, kinematic	:	not determined	
	Explos	ive properties	:	Not classified	
	Oxidizi	ng properties	:	Not classified	
	Molecu	ılar weight	:	268.44 g/mol	
850					
SEC	TION 1	0. STABILITY AND RI	EAC		
	Reactiv	<i>v</i> ity	:	None reasonably	r foreseeable.

Redolivity	•	None reasonably foreseeable.
Chemical stability	:	Stable under normal conditions.
Possibility of hazardous reac- tions	:	None known.
Conditions to avoid	:	Heating in air.
Incompatible materials	:	Strong oxidizing agents
Hazardous decomposition products	:	Emits acrid smoke and fumes when heated to decomposition.

# SAFETY DATA SHEET



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SECTION	I 11. TOXICOLOGICA	L INFORMATION	
	<b>e toxicity</b> nful if swallowed.		
Prod	luct:		
Acute	e oral toxicity	: LD50 Oral (Rat): 8	93 mg/kg
Acute	e dermal toxicity	: LD50 Dermal (Rat	obit): > 7,940 mg/kg
Com	ponents:		
N-1.3	3-Dimethvlbutvl-N'-p	nenyl-p-phenylenediamii	ne:
	e oral toxicity	: LD50 Oral (Rat): 8	
Acute	e dermal toxicity	: LD50 Dermal (Rat	bbit): > 7,940 mg/kg
-	corrosion/irritation	ailable information.	
Prod	luct:		
Spec	cies osure time	: Rabbit : 72 h : No skin irritation	
<u>Com</u>	ponents:		
N-1.3	3-Dimethylbutyl-N'-p	nenyl-p-phenylenediamii	ne:
Spec		: Rabbit	
	osure time	: 72 h	
Resu	ılt	: No skin irritation	
Serio	ous eye damage/eye	irritation	
	classified based on av		
Prod	luct:		
Spec		: Rabbit	
Resu	ılt	: slight	
Expo	osure time	: 72 h	
Com	ponents:		
N-1,3	3-Dimethylbutyl-N'-p	nenyl-p-phenylenediamii	ne:
Spec		: Rabbit	
Resu		: slight	
Expo	osure time	: 72 h	



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# Respiratory or skin sensitization

### Skin sensitization

May cause an allergic skin reaction.

# **Respiratory sensitization**

Not classified based on available information.

### Product:

\_

Species	:	Skin sensitization Guinea pig May cause sensitization by skin contact.
51		Human experience May cause sensitization by skin contact.

## **Components:**

# N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

Species	:	Skin sensitization Guinea pig May cause sensitization by skin contact.
	:	Human experience May cause sensitization by skin contact.

# Germ cell mutagenicity

Not classified based on available information.

### Product:

Genotoxicity in vitro :		Test Type: Mutagenicity - Bacterial Metabolic activation: +/- activation Method: Bacterial Reverse Mutation Assay Result: negative
		Metabolic activation: +/- activation Method: In vitro Mammalian Chromosome Aberration Test Result: positive
Genotoxicity in vivo	:	Test Type: various Species: Rat Result: negative

# **Components:**

# N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

Genotoxicity in vitro	: Test Type: Mutagenicity - Bacterial Metabolic activation: +/- activation Method: Bacterial Reverse Mutation Assay Result: negative

Metabolic activation: +/- activation Method: In vitro Mammalian Chromosome Aberration Test



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		Result: positive	9
Geno	toxicity in vivo	: Test Type: vari Species: Rat Result: negativ	
Carci	nogenicity		
Not cl	assified based on avai	lable information.	
<u>Produ</u>	uct:		
Speci Applic Metho Rema	cation Route	<ul> <li>Rat, male and</li> <li>Ingestion</li> <li>OECD Test Gu</li> <li>Based on avail</li> </ul>	
Comp	<u>oonents:</u>		
N-1.3	-Dimethylbutyl-N'-pho	envl-p-phenvlenedia	mine:
Speci	• • •	: Rat, male and	
	ation Route	: Ingestion	
Metho Rema		: OECD Test Gu : Based on avail	ideline 451 able data, the classification criteria are not me
IARC			ent at levels greater than or equal to 0.1% is r confirmed human carcinogen by IARC.
OSHA		ent of this product pre ist of regulated carcir	sent at levels greater than or equal to 0.1% is nogens.
NTP			ent at levels greater than or equal to 0.1% is ed carcinogen by NTP.
Repro	oductive toxicity		
•	lamage fertility or the ι	inborn child.	
Produ	uct:		
	s on fertility	General Toxici Fertility: NOAE Early Embryon	
		General Toxicit Fertility: NOAE Early Embryon	



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PRD		SDSUS / Z8 / 0528	

## Components:

# N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

App Gen Gen Fert Earl	cies: Rat, male ication Route: Oral eral Toxicity Parent: NOAEL: 60 mg/kg/day eral Toxicity F1: NOAEL: 60 mg/kg/day lity: NOAEL: 60 mg/kg/day / Embryonic Development: NOAEL: 20 mg/kg/day iod: OECD Test Guideline 443
-----------------------------------	---

Species: Rat, female Application Route: Oral General Toxicity Parent: NOAEL: 60 mg/kg/day General Toxicity F1: NOAEL: 60 mg/kg/day Fertility: NOAEL: 7 mg/kg/day Early Embryonic Development: NOAEL: 20 mg/kg/day Method: OECD Test Guideline 443

#### STOT-single exposure

Not classified based on available information.

### Components:

# N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

Assessment : Not classified

# STOT-repeated exposure

Not classified based on available information.

#### Components:

#### N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

Assessment : Not classified

### **Repeated dose toxicity**

#### **Components:**

#### N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

Species NOAEL Application Route Exposure time	:	Rat, male and female 20 mg/kg by gavage 28 days
Species NOAEL Application Route Exposure time	:	Rat, male and female 13.5 mg/kg in feed 2 year

# Aspiration toxicity

Not classified based on available information.



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	Produc Not cla	<u>et:</u> ssified			
E	Experi	ence with human exp	osı	ıre	
	Produc				
Ir	nhalati	on	:	Remarks: None k	nown.
S	Skin co	ontact	:	Remarks: May ca	use an allergic skin reaction.
E	Eye co	ntact	:	Remarks: None k	nown.
Ir	ngestio	on	:	Remarks: Harmfu	l if swallowed.

# SECTION 12. ECOLOGICAL INFORMATION

# Ecotoxicity

## Components:

# N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

Toxicity to fish	:	LC50 (Oryzias latipes (Japanese medaka)): 0.028 mg/l Exposure time: 96 h
Toxicity to daphnia and other aquatic invertebrates	:	EC50 (Daphnia magna (Water flea)): 0.13 mg/l Exposure time: 48 h Remarks: Read-across from a similar material
Toxicity to algae/aquatic plants	:	EC50 (Pseudokirchneriella subcapitata (green algae)): 0.335 mg/l Exposure time: 72 h Remarks: Read-across from a similar material
		NOEC: 0.23 mg/l Exposure time: 72 h Remarks: Read-across from a similar material
Toxicity to fish (Chronic tox- icity)	:	NOEC (Oryzias latipes (Japanese medaka)): 0.0037 mg/l Exposure time: 30 d
Toxicity to daphnia and other aquatic invertebrates (Chron-ic toxicity)	:	NOEC (Daphnia magna (Water flea)): 0.007 mg/l Exposure time: 21 d Remarks: Read-across from a similar material

# Persistence and degradability

# Components:

# N-1,3-Dimethylbutyl-N'-phenyl-p-phenylenediamine:

Biodegradability	:	Method: Ready Biodegradability: Modified MITI Test (I)
		Remarks: Not readily biodegradable.



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Stabil	ity in water	:	Degradation ha Hydrolysis: at 2			
Bioad	cumulative potential					
Comp	oonents:					
N-1,3	-Dimethylbutyl-N'-phe	nyl-	p-phenylenedia	mine:		
Bioac	cumulation	:		n factor (BCF): 569 ccumulation is unlikely.		
	Partition coefficient: n- octanol/water		log Pow: 4.68 (68 °F / 20 °C)			
Mobil	lity in soil					
Comp	oonents:					
N-1,3	-Dimethylbutyl-N'-phe	nyl-	p-phenylenedia	mine:		
	oution among environ- al compartments	:	log Koc: 3.45 Method: QSAR	model		
Other	adverse effects					
	available	DEF	ATIONS			
ECTION		DEF	ATIONS			
ECTION Dispo	13. DISPOSAL CONSI	DEF :		ccordance with local regulations.		
BECTION Dispo Waste	13. DISPOSAL CONSI	:	Dispose of in a	ccordance with local regulations.		
ECTION Dispo Waste	13. DISPOSAL CONSI osal methods e from residues	:	Dispose of in a	ccordance with local regulations.		
ECTION Dispo Waste ECTION Interr IATA	13. DISPOSAL CONSI osal methods e from residues 14. TRANSPORT INFO national Regulations	:	Dispose of in a	ccordance with local regulations.		
SECTION Dispo Waste SECTION Interr IATA- UN/ID	13. DISPOSAL CONSI osal methods e from residues 14. TRANSPORT INFO national Regulations	:	Dispose of in a	cordance with local regulations.		
ECTION Dispo Waste ECTION Interr IATA- UN/ID Prope	<ul> <li>13. DISPOSAL CONSI</li> <li>Desal methods</li> <li>e from residues</li> <li>14. TRANSPORT INFO</li> <li>mational Regulations</li> <li>DGR</li> <li>D No.</li> <li>er shipping name</li> </ul>	:	Dispose of in a ATION UN 3077 Environmentally (N-1,3-Dimethy			
ECTION Dispo Waste ECTION Interr IATA- UN/ID Prope Class	<ul> <li>13. DISPOSAL CONSI</li> <li>Dosal methods</li> <li>e from residues</li> <li>14. TRANSPORT INFO</li> <li>national Regulations</li> <li>DGR</li> <li>D No.</li> <li>er shipping name</li> </ul>	:	Dispose of in a ATION UN 3077 Environmentally (N-1,3-Dimethy 9	/ hazardous substance, solid, n.o.s.		
ECTION Dispo Waste ECTION Interr IATA- UN/ID Prope Class Packi Label	13. DISPOSAL CONSI osal methods e from residues 14. TRANSPORT INFO national Regulations DGR D No. er shipping name	:	Dispose of in a ATION UN 3077 Environmentally (N-1,3-Dimethy 9 III Miscellaneous	/ hazardous substance, solid, n.o.s.		
ECTION Dispo Waste ECTION Interr IATA- UN/ID Prope Class Packi Label: Packi	13. DISPOSAL CONSI osal methods e from residues 14. TRANSPORT INFO national Regulations -DGR D No. er shipping name ng group s ng instruction (cargo	:	Dispose of in a ATION UN 3077 Environmentally (N-1,3-Dimethy 9 III	/ hazardous substance, solid, n.o.s.		
ECTION Dispo Waste ECTION Interr IATA- UN/ID Prope Class Packi Label: Packi aircra	13. DISPOSAL CONSI psal methods e from residues 14. TRANSPORT INFO national Regulations DGR D No. er shipping name ng group s ng instruction (cargo ft) ng instruction (passen-	:	Dispose of in a ATION UN 3077 Environmentally (N-1,3-Dimethy 9 III Miscellaneous	/ hazardous substance, solid, n.o.s.		
ECTION Dispo Waste ECTION Interr IATA- UN/ID Prope Class Packi Label Packi aircra Packi ger ai	13. DISPOSAL CONSI psal methods e from residues 14. TRANSPORT INFO national Regulations DGR D No. er shipping name ng group s ng instruction (cargo ft) ng instruction (passen-	:	Dispose of in a ATION UN 3077 Environmentally (N-1,3-Dimethy 9 III Miscellaneous 956	/ hazardous substance, solid, n.o.s.		
SECTION Dispo Waste SECTION Interr IATA- UN/ID Prope Class Packi Label: Packi aircra Packi aircra Packi ger ai UN nu	13. DISPOSAL CONSI psal methods e from residues 14. TRANSPORT INFO national Regulations DGR D No. er shipping name ng group s ng instruction (cargo ft) ng instruction (passen- rcraft)	:	Dispose of in a ATION UN 3077 Environmentally (N-1,3-Dimethy 9 III Miscellaneous 956 956 UN 3077	/ hazardous substance, solid, n.o.s.		

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# Santoflex(TM) 6PPD Pastilles

Vers 2.9 PRD	ion Revision 05/04/20	022	150	9 <b>S Number:</b> 0000093128 SUS / Z8 / 0528	Date of last issue: 04/28/2022 Date of first issue: 09/06/2016
	Class Packing group Labels EmS Code Marine pollutant Transport in bu		:	9 III 9 F-A, S-F yes Annex II of MARP	OL 73/78 and the IBC Code
	Not applicable for	-			
	Domestic regulation				
	<b>49 CFR</b> Not regulated as Remarks	s a dangerous ç	goc :	Shipping in packa	ge sizes of less than 5 L (liquids) or 5 KG to a non-regulated classification.
	Special precaut	tions for user			
	Remarks		:		ge sizes of less than 5 L (liquids) or 5 KG to a non-regulated classification.
	based upon the	properties of th	ne u	unpackaged materi	r informational purposes only, and solely al as it is described within this Safety Data ode of transportation, package sizes, and

variations in regional or country regulations.

# **SECTION 15. REGULATORY INFORMATION**

### **CERCLA Reportable Quantity**

This material does not contain any components with a CERCLA RQ.

#### SARA 304 Extremely Hazardous Substances Reportable Quantity

This material does not contain any components with a section 304 EHS RQ.

## SARA 302 Extremely Hazardous Substances Threshold Planning Quantity

This material does not contain any components with a section 302 EHS TPQ.

SARA 311/312 Hazards	:	Respiratory or skin sensitization Reproductive toxicity Acute toxicity (any route of exposure)
SARA 313	:	This material does not contain any chemical components with known CAS numbers that exceed the threshold (De Minimis) reporting levels established by SARA Title III, Section 313.

# California Prop. 65

This product does not contain any chemicals known to the State of California to cause cancer, birth, or any other reproductive defects.

The ingredients of this product are reported in the following inventories:									
TCSI	:	On the inventory, or in compliance with the inventory							
TSCA	:	All substances listed as active on the TSCA inventory							



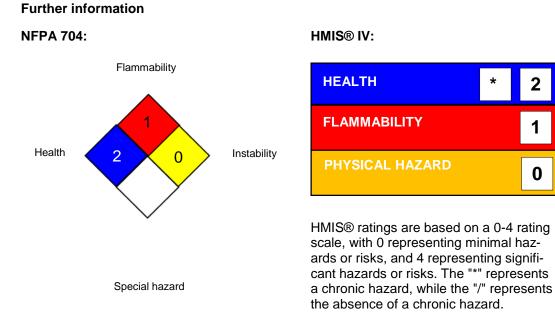
Version 2.9 PRD	Revision Date: 05/04/2022	1500	Number: 000093128 JS / Z8 / 0528	Date of last issue: 04/28/2022 Date of first issue: 09/06/2016
AIIC		: 0	On the inventory,	or in compliance with the inventory
DSL		: A	All components of	f this product are on the Canadian DSL
ENCS		: 0	On the inventory,	or in compliance with the inventory
ISHL		: 0	On the inventory,	or in compliance with the inventory
KECI		: 0	On the inventory,	or in compliance with the inventory
PICCS	3	: 0	On the inventory,	or in compliance with the inventory
IECSC	;	: 0	On the inventory,	or in compliance with the inventory
NZIoC	:	: 0	On the inventory,	or in compliance with the inventory
TECI		: 0	On the inventory,	or in compliance with the inventory

# **TSCA** list

No substances are subject to a Significant New Use Rule.

No substances are subject to TSCA 12(b) export notification requirements.

# **SECTION 16. OTHER INFORMATION**



# Full text of other abbreviations

AIIC - Australian Inventory of Industrial Chemicals; ASTM - American Society for the Testing of Materials; bw - Body weight; CERCLA - Comprehensive Environmental Response, Compensation,

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# Santoflex(TM) 6PPD Pastilles

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and Liability Act; CMR - Carcinogen, Mutagen or Reproductive Toxicant; DIN - Standard of the German Institute for Standardisation; DOT - Department of Transportation; DSL - Domestic Substances List (Canada); ECx - Concentration associated with x% response; EHS - Extremely Hazardous Substance; ELx - Loading rate associated with x% response; EmS - Emergency Schedule; ENCS - Existing and New Chemical Substances (Japan); ErCx - Concentration associated with x% growth rate response; ERG - Emergency Response Guide; GHS - Globally Harmonized System; GLP - Good Laboratory Practice; HMIS - Hazardous Materials Identification System; IARC -International Agency for Research on Cancer; IATA - International Air Transport Association; IBC - International Code for the Construction and Equipment of Ships carrying Dangerous Chemicals in Bulk; IC50 - Half maximal inhibitory concentration; ICAO - International Civil Aviation Organization; IECSC - Inventory of Existing Chemical Substances in China; IMDG - International Maritime Dangerous Goods; IMO - International Maritime Organization; ISHL - Industrial Safety and Health Law (Japan); ISO - International Organisation for Standardization; KECI - Korea Existing Chemicals Inventory; LC50 - Lethal Concentration to 50 % of a test population; LD50 - Lethal Dose to 50% of a test population (Median Lethal Dose); MARPOL - International Convention for the Prevention of Pollution from Ships; MSHA - Mine Safety and Health Administration; n.o.s. - Not Otherwise Specified; NFPA - National Fire Protection Association; NO(A)EC - No Observed (Adverse) Effect Concentration; NO(A)EL - No Observed (Adverse) Effect Level; NOELR - No Observable Effect Loading Rate; NTP - National Toxicology Program; NZIoC - New Zealand Inventory of Chemicals; OECD - Organization for Economic Co-operation and Development; OPPTS - Office of Chemical Safety and Pollution Prevention; PBT - Persistent, Bioaccumulative and Toxic substance: PICCS - Philippines Inventory of Chemicals and Chemical Substances: (Q)SAR - (Quantitative) Structure Activity Relationship; RCRA - Resource Conservation and Recovery Act; REACH - Regulation (EC) No 1907/2006 of the European Parliament and of the Council concerning the Registration, Evaluation, Authorisation and Restriction of Chemicals; RQ - Reportable Quantity; SADT - Self-Accelerating Decomposition Temperature; SARA - Superfund Amendments and Reauthorization Act; SDS - Safety Data Sheet; TCSI - Taiwan Chemical Substance Inventory; TECI - Thailand Existing Chemicals Inventory; TSCA - Toxic Substances Control Act (United States); UN - United Nations; UNRTDG - United Nations Recommendations on the Transport of Dangerous Goods; vPvB - Very Persistent and Very Bioaccumulative

Revision Date

: 05/04/2022

The information provided in this Safety Data Sheet is correct to the best of our knowledge, information and belief at the date of its publication. The information given is designed only as a guidance for safe handling, use, processing, storage, transportation, disposal and release and is not to be considered a warranty or quality specification. The information relates only to the specific material designated and may not be valid for such material used in combination with any other materials or in any process, unless specified in the text.

US / Z8

# Appendix D

Survey Concerning 6PPD Alternatives Sent to Consortium Members

# **Consortium Survey**

# Background

In order to assist the consortium in preparing an Alternative Analysis in California on behalf of the consortium members, the consortium members are required to respond to information surveys. Companies that joined the consortium when it was first formed have already completed this survey. We now ask that companies that have newly joined complete the survey as well.

The DTSC Safer Consumer Products Regulations define "alternative" to mean any of the following:

- "Removal of Chemical(s) of Concern from a Priority Product, with or without the use of one or more replacement chemicals;
- Reformulation or redesign of a Priority Product and/or manufacturing process to eliminate or reduce the concentration of Chemical(s) of Concern in the Priority Product;
- Redesign of a Priority Product and/or manufacturing process to reduce or restrict potential exposures to Chemical(s) of Concern in the Priority Product; or
- Any other change to a Priority Product or a manufacturing process that reduces the potential adverse impacts and/or potential exposures associated with the Chemical(s) of Concern in the Priority Product, and/or the potential adverse waste and end-of-life effects associated with the Priority Product."

As part of the alternatives analysis process, the consortium is required to assess all aspects of the definition of alternative listed above. Since information regarding removal of 6PPD, reformulation, or redesign may be considered confidential business information, outside counsel is assisting us in collecting and aggregating information to ensure consortium member proprietary data and information remains confidential.

Individual company responses to survey questions, submitted to outside counsel, will not be shared with USTMA. Outside counsel will summarize the information received from companies and will then work with individual companies to obtain approval for the anonymized summary responses to each question. Summary, genericized responses will be shared with USTMA and consortium members.

Please provide responses to the survey questions below in Microsoft Word. **Please send your responses** to [contact information] by [DUE DATE]. Please do not copy any USTMA staff on your response to outside counsel.

# **Categorizing Possible Alternatives**

The Consortium Alternatives Analysis Working Group has identified a preliminary list of possible alternatives. The Working Group has also been working to identify and collect information regarding the performance and toxicity of these potential alternatives. Based on the information the working group was able to identify in the public domain regarding performance and toxicity, the working group has grouped possible alternatives into four categories (based on information regarding chemical performance):

- 1. Data/information confirms these chemicals are not suitable alternatives
- 2. Chemical structure confirms these chemicals are not suitable alternatives
- 3. Chemical structure of these chemicals indicates they may be possible alternatives but there is no performance or toxicity data/information to confirm
- 4. Chemical structure and initial data/information for these chemicals demonstrates they could be possible alternative

# **Survey Questions**

# For survey questions 1-4, please consult list of potential alternatives in responding to the questions below.

- 1) Chemical Alternative Approach: An addition of a different chemical or chemicals in place of 6PPD.
  - a. Category 3 and Category 4 may be possible alternatives. Please review the Category 3 and Category 4 chemicals listed in the attached. Category rankings are found in column l:

# [Attachment containing list of chemicals under consideration from Gradient]

- b. Please provide any data/information on chemicals on the alternatives list or related compounds that would change the conclusion of Category 3 or Category 4.
- c. Please provide any data/information on chemicals that are on the alternatives list, or related compounds that would **support** the overall conclusion of Category 3 or Category 4.
- d. Please provide any data/information on chemical toxicity of the chemicals on the alternatives list or related compounds that would change or support the overall conclusion of Category 3 or Category 4.
- e. In general: If known, please provide any data/information on other chemical approaches that are a lower hazard, **available, functionally acceptable, and technically feasible** alternative that would eliminate the need for 6PPD.

# 2) The Safer Consumer Products Regulations require the consortium to consider whether 6PPD can be <u>removed</u> from tires, with or without the use of one or more replacement chemicals.

- Please review the list of potential alternatives and add any additional chemicals that your company recommends the consortium consider as part of the alternatives analysis. Note, alternatives should have a lower hazard profile and should be functionally acceptable, technically feasible, and economically viable.
- b. Does your company have any information/data that demonstrates that 6PPD can be removed from tires?
- c. Please review the list of potential alternatives and provide any information that would change the categorization of the potential alternative.
- d. For possible alternative chemicals that have the needed chemical structure and that have initial data/information on performance and/or toxicity which demonstrates these chemicals may be possible alternatives please provide any additional information your company might have regarding performance and/or toxicity of these chemicals.

# 3) The Safer Consumer Products Regulations require the consortium to consider whether tires can be reformulated or redesigned to eliminate or reduce the concentration of 6PPD in tires.

- a. Please provide any data and/or information that your company has regarding the ability to reformulate tires to reduce the amount of 6PPD used in tread or sidewall (not including saturated polymers).
  - i. What worked? What did not work?
  - ii. Is it currently used in production or planned for future production? If planned for future production, what is the timing for that launch?
- b. Please provide any data/information on reformulation approaches using SATURATED polymers that reduce the amount of 6PPD used in tread or sidewall.
  - i. What worked? What did not work?
  - ii. Is it currently used in production or planned for future production? If planned for future production, what is the timing for that launch?
- 4) The Safer Consumer Products Regulations require the consortium to consider whether tires can be redesigned to reduce or restrict potential exposures to 6PPD. This may include a change to the configuration or construction of a tire to reduce or eliminate the need for 6PPD.
  - a. Please provide any data/information on a construction feature that would eliminate or reduce the amount of 6PPD used in a tire.
    - i. What has worked? What has not worked?
    - ii. Is it currently used in production or planned for future production? If planned for future production, what is the timing for that launch?

# 5) Other Approaches

a. Please share any data/information regarding any other approaches beyond those listed above that can be used to reduce/eliminate/reformulate/reconfigure tires to remove/reduce 6PPD. For example - exterior coating or veneer applications, improved tread wear/lower skid, other sidewall solutions that may not work for tread.

# 6) Use of 6PPD

- a. USTMA has gathered preliminary information regarding the use of 6PPD. Please review the statement below and provide any recommended edits based on the use of 6PPD by your company.
  - i. "Some consortium members began using 6PPD in tire manufacturing in the mid 1960's and early 1970's. However, as tire wear life increased in the 1990s, 6PPD became more widely used by consortium members in the early 2000s."

# Appendix E

Derivation of Estimated Tire Shipments into the State of California

# Derivation of Estimated Tire Shipments into the State of California

The SCP regulations require companies preparing an AA to include data on the number of product units sold in the State of California. Consortium members do not have data on products sold in California because products are typically sold by third parties. USTMA does collect data on total US shipments of tires by year, but data are not available at the statewide level. This data nonetheless allows us to approximate tire shipments to the State of California, which can be used as a proxy for tire sales. The USTMA shipment data for 2022 are shown below.

Vehicle Category	US Tire Shipments in 2022 (Excluding Exports)
Passenger cars/Light Truck	298,847,000
Heavy-duty Truck/Bus	33,139,000
Total	331,986,000

Notes:

US = United States; USTMA = U.S. Tire Manufacturers Association. Source: U.S. Tire Manufacturers Association (USTMA) Factbook 2024

To determine what percentage of total US tire shipments are attributable to California, 2022 passenger car, bus, and motorcycle vehicle registrations for the US and California were obtained directly from the Federal Highway Administration (FHWA)

# Table E.2 FHWA Motor Vehicle Registration Data for 2022

Jurisdiction	Automobiles Buses Truc		Trucks <sup>1</sup>	Motorcycles	All Vehicles		
California	13,796,109	95,965	16,424,539	802,500	31,119,113		
US Total	99,946,870	954,119	172,932,334	9,567,664	283,400,986		

Notes:

FHWA = Federal Highway Administration; US = Unites States.

(1) Trucks as described by the FHWA includes light duty trucks, heavy duty trucks, vans, and sport vehicle utilities (SUVs). Source: US Dept. of Transportation (US DOT), Federal Highway Administration (FHWA). 2022. "Highway Statistics 2022." Accessed on March 6, 2024 at <a href="https://www.fhwa.dot.gov/policyinformation/statistics/2022/">https://www.fhwa.dot.gov/policyinformation/statistics/2022</a>.

The FHWA motor vehicle registration data reports total "trucks" registrations to include heavy duty trucks, light duty trucks, vans, and SUVs; whereas USTMA separates data for heavy duty and light duty trucks. The total number of "trucks" reported in the FHWA data use percentages from a US National Highway Traffic Safety Administration (NHTSA) report, which provided more detailed classifications (*i.e.*, passenger cars, light duty trucks, large trucks, motorcycles, and busses) but only at the national level. The latest publicly available vehicle registration data from NHTSA is from 2021, published in 2023 (NHTSA, 2023). Unfortunately, 2022 data is not yet publicly available.

Vehicle Type	2021 US Motor Vehicle Registrations	Percent of Total		
Passenger cars	107,934,093	35.65%		
Light duty trucks	170,108,546	56.19%		
Large trucks ( <i>i.e.</i> , heavy duty trucks)	13,859,181	4.58%		
Motorcycles	9,881,414	3.26%		
Buses	939,219	0.31%		
Calculations:				
Combined light and heavy duty trucks	183,967,727	60.77%		
Percent of light duty trucks out of		92.5%		
combined truck numbers				
Percent of heavy duty out of combined		7.5%		
truck numbers				

Table E.3 Latest NHTSA US Motor Vehicle Registration Data (2021)

Notes:

NHTSA = National Highway Traffic Safety Administration; US = United States.

Source: US Dept. of Transportation (US DOT), National Highway Traffic Safety Administration (NHTSA), National Center for Statistics and Analysis (NCSA). December 2023. "Traffic Safety Facts 2021: A Compilation of Motor Vehicle Crash Data." DOT HS 813 527. 225p. Accessed on March 20, 2024 at https://crashstats.nhtsa.dot.gov/Api/Public/ViewPublication/813527.

The abovementioned NHTSA data indicates that light duty trucks were 92.5% of the combined truck number and heavy duty trucks were 7.5% of the total truck number. These percentages were applied to the FHWA reported California "truck" registrations to allocate these trucks to the proper category. The passenger car/light duty truck (again, including vans and SUVs) and heavy duty truck/bus combined categories, were then constructed by adding the appropriate numbers. The California percentage of national vehicle registrations in each category was then multiplied against US tire shipments in those categories to arrive at California specific estimates of tires that were shipped and likely to be used in California.

Vehicle Category	Vehicle	e Registrations ir	n 2022	USTMA Tire Shipments in 2022				
	US	US CA (		US	CA (est.)			
Passenger/Light Duty Truck	259,909,278	28,988,808	11.2%	298,847,000	33,332,000			
Heavy Duty Truck/Bus	13,924,044	1,327,805	9.5%	33,139,000	3,160,000			
Total	273,833,322	30,316,613	11.1%	331,986,000	36,492,000			

Table E.4 Estimated Annual S	Shipments of the Priority	Product in California
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Notes:

CA = California; calc. = Calculated; est. = Estimated; US = United States; USTMA = U.S. Tire Manufacturers Association.

As indicated in the table above, an estimated 33,332,000 passenger car/light duty truck tires were shipped to California in 2022 and an estimated 3,160,000 heavy duty truck and bus tires were shipped to California in that year, for an estimated total number of tires shipped to California in 2022 at 36,492,000 units. Data on motorcycle tires are not included in this count because USTMA does not collect data on motorcycle tires. As shown above in the NHTSA data, motorcycle tires represent a very small portion of the overall vehicle fleet and this would not be expected to significantly different for California.

# **Appendix F**

List of All Candidate Alternatives Identified and Reviewed by the Consortium

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
1	Phenylene Diamine		6PPD	N1-(4-Methylpentan-2-yl)-N4- phenylbenzene-1,4-diamine N-(1,3-dimethylbutyl)-N'-phenyl- 1,4-benzenediamine N-(1,3-dimethylbutyl)-N'-phenyl-p- phenylenediamine DMBPPD	793-24-8	CC(C)CC(C)NC1=CC= C(C=C1)NC2=CC=CC= C2	Fatigue Oxygen Ozone	NA	NA	3	Yes	Most effective antiozonant and antioxidant discovered in the past 50 years	Not applicable	1,2,3, 124
2	Phenylene Diamine	NH NH	77PD	N,N'-Bis(1,4-dimethylpentyl)- phenylenediamine; Tenamene 4; Santoflex 77; Antioxidant 4030; 1- N,4-N-Bis(5-methylhexan-2- yl)benzene-1,4-diamine; UOP 788	3081-14-9	CC(C)CCC(C)NC1=CC =C(C=C1)NC(C)CCC(C )C	Fatigue Oxygen Ozone	1	5	2	Yes	Migrates faster than 6PPD, so not as effective in long term protection.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 1	7, 8, 146
3	Phenylene Diamine		Flexzone 8L UOP 88 DEMPD	N,N'-Bis(1-ethyl-3-methylpentyl)-p- phenylenediamine	139-60-6	CCC(C)CC(CC)NC1=C C=C(C=C1)NC(CC)CC( C)CC	Fatigue Oxygen Ozone	No data but expected to perform similar to 77PD based on very close structural similarities	N/A	3	Probable	Expected to migrate faster than 6PPD. Reported as a commercial antiozonant in the 1970s.	No, due to compound effectiveness score of 77PD (score of 1)	91, 92, 95
4	Phenylene Diamine	HaC CHa CHa	UOP 688; 8PPD	N-1-Methylheptyl-N'-phenyl-p- phenylenediamine	15233-47-3	CCCCCCCC(C)NC1=CC= C(C=C1)NC2=CC=CC= C2	Fatigue Oxygen Ozone	No data but expected to perform similar to 7 PPD based on very close structural similarities	N/A	2	Probable	Same as 7PPD but 1 more carbon. Reported as a commercial antiozonant in the 1970s.	Yes, due to compound effectiveness score of 7PPD (score of 4)	90, 92
5	Phenylene Diamine	H <sub>2</sub> C HN CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	UOP 288	Di-2-octyl-p-phenylenediamine Elastozone 30	103-96-8	CCCCCCC(C)NC1=CC= C(C=C1)NC(C)CCCCC C	Fatigue Oxygen Ozone	No data but expected to perform similar to 77PD based on very close structural similarities	N/A	2	Probable	Expected to migrate faster than 6PPD. Reported as a commercial antiozonant in the 1970s.	No, due to compound effectiveness score of 77PD (score of 1)	92,125
6	Phenylene Diamine		N' -Phenyl-N-Fluorenyl- Para-Phenylenediamine	N' -Phenyl.N-Fluorenyl-Para- Phenylenediamine	Not available	C1=C(NC(C=CC2)=CC =2)C=CC(=C1)NC(C(C 1=CC2)=CC=2)C(=CC= C2)C1=C2	Fatigue Oxygen Ozone	4	3	1	Probable	In static ozone tests the compound was equivalent to 77PD	Yes, due to compound effectiveness score of 4	93
7	Phenylene Diamine	H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C	N-(p- phenylthiomethylphenyl)- N'-(1,3 dimethyl-butyl)-p- phenylenediamine	N-(p-phenylthiomethylphenyl)-N'- (1,3 dimethyl-butyl)-p- phenylenediamine	Not available	C1=CC=C(C=C1)SCC1 C=CC(=CC=1)NC1=CC =C(C=C1)NC(C)CC(C) C	Fatigue Oxygen Ozone	4	2	1	Probable	Equivalent to 6PPD in dynamic ozone testing - similar in other tests.	Yes, due to compound effectiveness score of 4	94
8	Phenylene Diamine	CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	7PPD	N-(1,4-Dimethylpentyl)-N'-phenyl- p-phenylendiamine	3081-01-4	CC(C)CCC(C)NC1=CC =C(C=C1)NC2=CC=CC =C2	Fatigue Oxygen Ozone	4	Unclear, maybe 5	3	Probable	Similar antiozonant properties to 6PPD; virtually equivalent dynamic ozone in ESBR.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	9, 122

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
9	Phenylene Diamine		DAPD BENPAT Wingstay 100	Wingstay 100; 1,4- Benzenediamine,N,N'-diphenylj- methyl derivatives; 1,4- Benzenediamine, N,N'-mixed tolyl and xylyx derivatives; Hydroquinone,o-toludine,xylidine condensate	68953-84-4	N/A, Multi- Constituent	Fatigue Oxygen Ozone	4	5	3	Yes	Already used commercially in combination with 6PPD. Could not function on itself because it's slower to migrate than 6PPD. Not as effective an antiozonant as 6PPD.	Yes, due to compound effectiveness score of 4	1, 10, 138, 118, 119
10	Phenylene Diamine		Flexzone 6H Vulkacit 4010 CPPD	N-Cyclohexyl-N'-phenyl-p- phenylenediamine	101-87-1	C1CCC(CC1)NC2=CC= C(C=C2)NC3=CC=CC= C3	Fatigue Oxygen Ozone	3	Unclear, maybe 5	2	Probable	Antioxidant and antiozonant similar to 6PPD.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 3	55,56,57,123
11	Phenylene Diamine		UOP 26 CCPD	N,N'-Dicyclohexyl-p- phenylenediamine	4175-38-6	C1CCC(CC1)NC2=CC= C(C=C2)NC3CCCCC3	"Although no specific uses for N,N'-Dicyclohexyl-4- phenylenediamine (CCPD) were identified, based on similarity to other phenylenediamines, it is presumed to be used as an antioxidant / antiozonant, fuel additive, and in monomer distillation" ref. 77 p. i	3	Unclear	2	Probable	Not commonly used as a polymer stabilizer. Expected to have characteristics similar to 77PD.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 3	76,77
12	Phenylene Diamine		IPPD	N-Isopropyl-N'-phenyl-1,4- phenylenediamine; Stanguard IPPD	101-72-4	CC(C)NC1=CC=C(C=C 1)NC2=CC=CC=C2	Fatigue Oxygen Ozone	4	5	3	Yes	More water soluble than 6PPD; quinone formation probable.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	1,2
13	Phenylene Diamine	H <sub>3</sub> C	None	4-(2,5-Dimethyl-1H-pyrrol-1-yl)-N- phenylaniline; N-Phenyl-4-(2,5- dimethyl-1H-pyrrole-1-yl)aniline	Not available	C1=CC=C(C=C1)NC1C =CC(=CC=1)N(C1C)C( =CC=1)C		4	3	1	Probable, but predicted to be much less than 6PPD	Shown to be an effective antiozonant in rubber. May have less tendency to form quinone because of aromatic pyrrole substructure.	Yes, due to compound effectiveness score of 4	2 (Table 6.2), 14, 15
14	Phenylene Diamine		None	N,N - (Ethane-1,2-diyl ) bis (N- phenylbenzene-1 4-diamine; 1-N- [2-(4-anilinoanilino)ethyl]-4-N- phenylbenzene-1,4-diamine	Not available	C1C=CC=C(C=1)NC(C =CC1NCCNC(C=CC2N C(=CC=C3)C=C3)=CC= 2)=CC=1	Fatigue Oxygen Ozone	4	3	1	Probable, but size of quinone molecule may affect toxicity	This and similar materials were shown to be as effective as 6PPD for ozone protection of nitrile rubber. Quinones from this material may be less toxic due to size of molecule.	Yes, due to compound effectiveness score of 4	24,25, 98 (Note the patents contain references to similar materials)
15	Phenylene Diamine		None	4-N-(2,3-Dimethylphenyl)-1-N- phenylbenzene-1,4-diamine- R1 and R2 are methyl	Not available	C1=CC=C(C=C1)NC1= CC=C(C=C1)NC(C1)=C (C(C)=CC=1)C	Fatigue Oxygen Ozone	3	3	1	Probable	No comment	Yes, due to compound effectiveness score of 3	99
16	Phenylene Diamine	X Y X and Y are NH-Aryl and R' is alkyl or aryl	None	N,N'-Diphenyl-2-(butylthio)-p- phenylenediamine if R' is n-butyl	Not available	C1C=C(C=CC=1)NC(C =C1)=C(C=C1NC(C=C C1)=CC=1)SCCCC	Fatigue Oxygen Ozone	1	3	1	Possible	Data indicates poorer performance for dynamic ozone	No, due to compound effectiveness score of 1	100, 101

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
17	Phenylene Diamine (Kruger)	$HO_{C_6H_{13}} HO_{C_6H_{13}} HO_{$	RU997 Irgazone 997	Reaction product of N-phenyl-N'- (1,3 dimethylbutyl)-p- phenylenediamine with an alkyl glycidylthioether.	444992-04-5	C1=CC(=CC=C1)NC1C =CC(=CC=1)N(CC(O)C SC(C)(C)CCCCCCCCC) C(CC(C)C)C	Fatigue Oxygen Ozone	3	1	1	Probable	Non-staining and had been approved for some food use in Europe.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 3	46,47,48
18	Phenylene diamine	HO	4-[4-(4-Methylpentan-2- ylamino)anilino]phenol	Not available	Not available	C1(O)=CC=C(C=C1)N C1=CC=C(C=C1)NC(C) CC(C)C	Oxygen but also claimed as antiozonant	3	3	1	Probable	Prediction of performance based on bond strength; no ozone data.		117, 124
19	Phenylene diamine	$(OH)_m$ $H$ $(OH)_n$ $(OH)_n$ $(R')_p$ $R_1$ $R_2$	This is a class of compounds - Reference uses case where R1 and R2 are methyl; n,p and q are zero and m=1 and is in the para position	Reference example is 4-((4- (dimethylamino)phenyl)amino)ph enol	6358-22-1	CN(C)C1=CC=C(C=C1) NC2=CC=C(C=C2)O	Fatigue Oxygen Ozone	34	4	2	Probable	Recent patent – good static and dynamic ozone resistance in natural rubber black compound.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	78,79
20	Phenylene diamine		N,N'-Di-2-naphthyl-p- phenylenediamine	DNPD; AgeRite W	93-46-9	C1=CC=C2C=C(C=CC2 =C1)NC3=CC=C(C=C3 )NC4=CC5=CC=CC=C 5C=C4	Fatigue Oxygen Ozone	3	1	2	Probable	Listed in a review article as an early commercial antiozonant.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 3	86,87
21	Phenylene Diamine related		6QDI	N-4-(1,3-Dimethylbutyl)imino-2,5- cyclohexadien-1-ylidene benzenamine; 4-N-(4- methylpentan-2-yl)-1-N- phenylcyclohexa-2,5-diene-1,4- diimine	52870-46-9	CC(C)CC(C)N=C1C=C C(=NC2=CC=CC=C2)C =C1	Fatigue Oxygen Ozone	4	4	2	Yes	Forms 6PPD during use. Partially attaches to polymer on mixing and in service.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	2,12
22	Dihydroquinoline	CH <sub>3</sub> CH <sub>2</sub> O	Ethoxyquin	Ethoxyquin; 6-Ethoxy-2,2,4- trimethyl-1,2-dihydroquinoline; Santoquin; Antioxidant EC; Santoflex AW	91-53-2	CCOC1=CC2=C(C=C1) NC(C=C2C)(C)C	Fatigue Oxygen Ozone	1	3	3	Yes	Used as very early antiozonant; heavily staining; used in fish food as an antioxidant but the authorization for that application has been suspended in the EU. Not as effective as 6PPD.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 1	4,5,6
23	Dihydroquinoline	n (n=ca. 3)	TMQ	2,2,4-Trimethyl-1 <i>H</i> -quinoline	26780-96-1	N/A Polymer	Oxygen Ozone (in combination with 6PPD)	1	Unclear, maybe 5	3	Unlikely	Very low antiozonant activity by itself but acts synergistically with 6PPD.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 1	1,2
24	Dihydroquinoline	R <sup>1</sup> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>	N,N-diethyl-2,2,4-trimethyl- 1H-quinolin-6-amine (R= N(C2H5)2	N,N-Diethyl-2,2,4-trimethyl-1H- quinolin-6-amine (R= N(C2H5)2	Not available	C1(C)(C)C=C(C)C(C2N 1)=CC(=CC=2)N(CC)C C	Fatigue (?) Oxygen Ozone	4	2	1	Probable	Amino derivatives of ethoxyquin have been shown to be better antiozonants than ethoxyquin in lab testing.	Yes, due to compound effectiveness score of 4	29, 30, 31 (for morpholine derivative), 32 (comparison of amine derivatives in lab tests)

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
25	Diphenyl amine	N N N N N N N N N N N N N N N N N N N	Dioctyl diphenylamine DOPD	Vanox ODP; Standguard ODP	101-67-7	CCCCCCCC1=CC=C( C=C1)NC2=CC=C(C=C 2)CCCCCCCC	Oxygen	4	1	3	Unlikely	Used in chloroprene rubber for improved flex resistance. 15% better ozone resistance than ethoxyquin and 15% better than DTPD in a Tier 1 type test	Yes, due to compound effectiveness score of 4	31
26	Diphenyl amine		Wingstay 29	4-(1-Phenylethyl)-N -[4-(1- phenylethyl)phenyl]aniline	68442-68-2	CC(C1=CC=CC=C1)C2 =CC=C(C=C2)NC3=CC =C(C=C3)C(C)C4=CC= CC=C4	Owigon	2	1	3	Unlikely	Good antioxidant, but minimal antiozonant activity.	No, due to compound effectiveness score of 2	11, 126
27	Hindered amine	R <sup>2</sup> , R <sup>1</sup> R <sup>4</sup> , N, R <sup>3</sup>	Mixed xylene diamines	N,N'-Dibenzyl-p-xylene- alpha,alpha'-diamine and N,N'- Dibenzyl-m-xylene-alpha,alpha'- diamine perform the best	N,N'-Dibenzyl-p- xylene-alpha,alpha'- diamine- 25790-41-4	N/A Mixture	Fatigue (not known, probably low) Oxygen (not known) Ozone	4	3	2	No, unless R groups are phenyl	Patent claims better crack growth inhibitor than PPDs on exposure to ozone. Quinone formation much less likely than with 6PPD. Best performance was with N,N'-dibenzyl-m,p- xylylenediamine.	Yes, due to compound effectiveness score of 4	16, 17 , 18
28	Hindered amine (HALS)		Tinuvin 770	Bis(2,2,6,6-tetramethyl-4- piperidyl) sebacate	52829-07-9	CC1(CC(CC(N1)(C)C) OC(=0)CCCCCCCCC(= 0)OC2CC(NC(C2)(C)C )(C)C)C	Oxygen	2	1	3	No	These are based on 2,2,6,6- tetramethylpiperidine and are used as light stabilizers. They have no ability to protect against ozone.	No, due to compound effectiveness score of 2	48, 49, 81, 127
29	Sulfur compound	O (CH <sub>2</sub> ) <sub>11</sub> Me S O (CH <sub>2</sub> ) <sub>11</sub> Me	DLTDP	Dodecyl 3-(3-dodecoxy-3- oxopropyl)sulfanylpropanoate	123-28-4	o=c(occcccccccc c)ccsccc(=0)occcc ccccccc		2	1	3	No	Does not act as an antiozonant.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 2	13
30	Triazine	$\begin{array}{c} CH_{3}\\ CH_{3}\\ H_{N}\\ \mathsf$	Durazone 37 TAPDT	2,4,6-Tris-(N-1,4-dimethylpentyl- para-phenylenediamino)-1,3,5 triazine; TAPDT	121246-28-4	CC(C)CCC(C)NC1=CC =C(C=C1)NC2=NC(=N C(=N2)NC3=CC=C(C= C3)NC(C)CCC(C)C)NC 4=CC=C(C=C4)NC(C)C CC(C)C	Oxygen (not known, but expected to be good)	4	Unclear, maybe 5	3	Probable, but size of quinone molecule may affect toxicity	Good solubility in natural rubber but limited solubility in butadiene rubber and SBR. Works as antiozonant at low levels in sidewall with phenolic resin but no comparison to 6PPD. Most likely to migrate too slowly.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	19, 20
31	Triazinethione		Tetrahydro-1,3,5-tri-n- butyl-(S)-triazinethione	Tetrahydro-1,3,5-tri-n-butyl-(S)- triazinethione	Not available	N1(CCCC)NN(C[C@H ](CCCC)C1=S)CCCC	Oxygen Ozone	1	3	1	No	Used as a non-staining antiozonant. Extremely scorchy/cures too fast for rubber to be workable.	No, due to compound effectiveness score of 1. A triazine ( <i>e.g.</i> , Durazone 37) is evaluated in Preliminary (Stage 1) AA.	27, 28

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
32	Polymeric antioxidant	HO W N HO W N Acetone Diphrylamine Condensates	ADPA (not to be confused with p-aminodiphenyl- amine)	Acetone-Diphenylamine condensate; Accinox BL 75S	68412-48-6	CC(=O)C.C1=CC=C(C= C1)NC2=CC=CC=C2	Fatigue Oxygen Ozone (poor)	2	1	3	Unknown	Mainly used as an antioxidant in lower cost applications. Poor antiozonant.	No, due to compound effectiveness score of 2	26, 80, 137
33	Polymeric antioxidant	$ \begin{array}{c} \begin{array}{c} H_{3}C \\ H_{3}C \\ H_{3}C \end{array} \\ \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \end{array} \\ \end{array} \\ \end{array} $ \\ \begin{array}{c} O \\ H_{3}C \end{array} \\ \end{array} \\ \end{array} \\ \end{array}  \\ \end{array}  \\ \end{array}  \\ \end{array}  \\ \end{array}  \\ \end{array}  \\ \end{array}  \\ \end{array}  \\ \\ \end{array}  \\ \end{array} \\ \end{array}	Poly(4-hydroxy-2,2,6,6- tetramethyl-1- piperidineethanol-alt-1,4- butanedioic acid)	2-(4-Methoxy-2,2,6,6- tetramethylpiperidin-1-yl)ethyl 4- oxopentanoate; Photo-stabilizer BW-10LD	65447-77-0	CC(=0)CCC(=0)OCCN 1C(CC(CC1(C)C)OC)(C )C		2	1	2	No	Reference reports compound as an antiozonant for polybutadiene but only data shown was on polypropylene. Unlikely to be effective in diene-based polymers.	No, due to compound effectiveness score of 2	38, 128
34	Polymeric siloxane	$ \begin{array}{c} \begin{array}{c} CH_{i} & CH_{i} & CH_{i} \\ \hline SI = O & SI = O \\ (CH_{i})_{i} & (CH_{i})_{i} & CH_{i})_{i} \\ \hline O & O \\ \hline \\ H & H & H \\ \end{array} \\ \begin{array}{c} O \\ H \\ \end{array} \\ n \end{array} $	Uvasil 299	Poly-methylpropyl-3-oxy [4(2,2,6,6-tetramethyl)piperidinyl] siloxane	164648-93-5	N/A Polymer	Oxygen	2	1	1	No	This is a hindered amine light stabilizer and is not expected to have any antiozonant activity.	No, due to compound effectiveness score of 2	50,51
35	Metal dithiocarbamate		Nickel dibutyl dithiocarbaname	Perkacit NDBC; Naugard NBC	13927-77-0	CCCCN(CCCC)C(=S)[S ].CCCCN(CCCC)C(=S)[ S-].[Ni+2]		1	3	3	No	Reasonable antiozone activity but extremely scorchy, making it impractical in tire compounds. Contains high level of nickel.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 1	26, 52, 53 ,54
36	Hindered phenolic		None	2-Cyano-3-(3,5-ditert-butyl-4- hydroxyphenyl)propionic acid ethyl ester	132631-62-0	CCOC(=O)C(CC1=CC( =C(C(=C1)C(C)(C)C)O) C(C)(C)C)C#N		2	3	1	Probable	Paper claims antiozonant activity similar to IPPD and DPPD but only looks at carbonyl formation during ozonation.	No, due to compound effectiveness score of 2	41, 42
37	Bisphenol	OH OH	AO2246	2,2'-Methylenebis(6-tert-butyl-4- methylphenol)		CC1=CC(=C(C(=C1)C( C)(C)C)O)CC2=C(C(=C C(=C2)C)C(C)(C)C)O	Oxygen	2	1	3	Probable	Does not act as antiozonant but is a common antioxidant used as a raw polymer stabilizer; not typically used in final compound.	No, due to compound effectiveness score of 2	43,44, 82, 129
38	Hydroquinone	ИО ОН	Santovar A	2,5-Di-tert-amylhydroquinone	79-74-3	CCC(C)(C)C1=CC(=C( C=C10)C(C)(C)CC)O		2	1	3	Yes	Does not act as antiozonant but is a common antioxidant used as a raw polymer stabilizer; not typically used in final compound.	No, due to compound effectiveness score of 2	45,82,83,130
39	Phosphite	$(CH_{3})_{3}C$ $(CH_{3})_{3}$	Irgafos 168	Tris(2,4-ditert- butylphenyl)phosphite	31570-04-4	CC(C)(C)C1=CC(=C(C= C1)OP(OC2=C(C=C(C =C2)C(C)(C)C)C(C)(C) C)OC3=C(C=C(C=C3)C (C)(C)C)C(C)(C)(C)C(C) (C)C	Fatigue Oxygen	2	1	3	Unlikely	Used as raw polymer stabilizer. Highly hindered phosphites have some antifatigue activity, although most phosphites are decomposed during vulcanization.	No, due to compound effectiveness score of 2	61,62,84 131

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
40	Nitroxyl compounds	$H_3C$ $CH_3$ $H_3C$ $N$ $CH_3$ $O^{\bullet}$	ТЕМРО	2,2,6,6-Tetramethylpiperidine 1- oxyl,	2564-83-2	CC1(CCCC(N1[O])(C) C)C	UV protection	2	1	2	Not from TEMPO but potentially from other HALS compounds depending on the structure	TEMPO is not used as a polymer stabilizer – it is a precursor to a hindered amine light stabilizer. The nitroxyl species is not an antiozonant.	No, due to compound effectiveness score of 2	67,68, 132
41	Phenylnaphthyl amines	HN	N-Phenyl-1-naphthylamine	1-Anilinonaphthalene Phenyl-α-naphthylamine	90-30-2	C1=CC=C(C=C1)NC2= CC=CC3=CC=CC=C32	Fatigue Oxygen Ozone	3	1	2	Probable	Used in neoprene but not common with diene rubbers used in tires.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 3	64, 65
42	Phenylnaphthyl amines		N-Phenyl-2-naphthylamine	2-Anilinonaphthalene; Phenyl-β- naphthylamine	135-88-6	C1=CC=C(C=C1)NC2= CC3=CC=CC=C3C=C2	Fatigue Oxygen Ozone	4	Unclear, maybe 5	2	Possible	An early antiozonant	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	63, 66
43	Ether		Vulcazon AFS	3,9-Dicyclohex-3-enyl-2,4,8,10- tetraoxaspiro[5.5]undecane	6600-31-3	C1CC(CC=C1)C2OCC3 (CO2)COC(OC3)C4CC C=CC4	Ozone (for chloroprene rubber)	2	3	3	No	Shown to be effective in chloroprene and claimed for "natural latex" but unlikely to perform in other diene rubbers. Ineffective with tire elastomers – does not have antiflex properties.	No, due to compound effectiveness score of 2	21,22,23,88
44	Phenol	OH S-C8H17	Irganox 1520 blend	2-Methyl-4,6- bis((octylthio)methyl)phenol (Irganox 1520)	110553-27-0	CCCCCCCCSCC1=CC( =C(C(=C1)C)O)CSCCC CCCCC	Ozone	4	3	3	Probable Irganox 1520	Shown to be effective in chloroprene and claimed for "natural latex". Shown to be effective in sidewall formulation in combination with Irganox 1520 phenolic antioxidant.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	21,22,23,85, 88
45	Thiosemicarbazide	H <sub>3</sub> C NH NH NH NH CH <sub>3</sub>	1,1,4 Tributyl thiosemicarbazide	1,1,4 Tributyl thiosemicarbazide	Not available	CCCCN(NC(=S)NCCCC )CCCC	Fatigue (unknown) Oxygen (unknown) Ozone	1	2	1	No	Trialkyl thiosemicarbazides have antiozonant properties but are extremely scorchy. Better reaction with ozone than substituted thioureas.	No, due to compound effectiveness score of 1	33
46	Inorganic		Graphene	Prophene™	1034343-98-0	N/A	Fatigue (unknown) Oxygen Ozone	2	3	2	No	Reported to reduce 6PPD when used as a filler in rubber compounds.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 2	34
47	Inorganic	AI2O3	Alumina (in combination with 6PPD)	Alumina (in combination with 6PPD)	1344-28-1	[O-2].[O-2].[O- 2].[Al+3].[Al+3]	Ozone	2	1	3	No	All examples had 6PPD in the formulation. Unlikely to be effective without 6PPD.	No, due to compound effectiveness score of 2	39, 133

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
48	Inorganic	MnO2	Manganese dioxide	Dioxy magnesium	1313-13-9	O=[Mn]=O	None in rubber. Acts to promote oxidation. Ligated manganese salts have been shown to decompose ozone, but no work has been done in rubber	2	1	3	No	Literature exists on manganese complexes to decompose ozone, but it is know that manganese salts promote oxidation of rubber compounds.	No, due to compound effectiveness score of 2	75,76, 134
49	Hydrazine	H <sub>3</sub> C NH CH <sub>3</sub> CH <sub>3</sub>	1,1' - Pentamethylenebis(2,2-Di- n- Butylhydrazine)	1,1' -Pentamethylenebis(2,2-Di-n- Butylhydrazine)	Not available	CCCCN(CCCC)NCCCC CNN(CCCC)CCCC	Fatigue (unknown) Oxygen (unknown) Ozone	3	3	1	No	Shown to be an antiozonant in dynamic testing of rubber but not compared to conventional antiozonants.	Yes, due to compound effectiveness score of 3	35
50	Imine	CH=H	N-phenyl-meta- phenoxyphenylmethaneim ine; 1-(3-phenoxyphenyl)- N-phenylmethanimine	N-Phenyl-meta- phenoxyphenylmethaneimine; 1-(3-phenoxyphenyl)-N- phenylmethanimine	Not available	C1(/C=N/C(C=CC2)=C C=2)=CC(=CC=C1)OC( C=CC1)=CC=1	Fatigue (unknown) Oxygen (unknown) Ozone	1	1	1	Unlikely	Reported as an antiozonant in rubber.	No, due to compound effectiveness score of 1	36
51	Diamine	HN	1,3-Bis(4- piperidyl)propane; 4-(3-piperidin-4- ylpropyl)piperidine	1,3-Bis(4-piperidyl)propane; 4-(3-piperidin-4- ylpropyl)piperidine	16898-52-5	C1CNCCC1CCCC2CC NCC2	Fatigue (unknown) Oxygen (unknown) Ozone	2	1	2	Depends on structure	Shown to be improve crack resistance in static ozone tests on SBR.	No, due to compound effectiveness score of 2	37, 135
52	Nitrone	HO	α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	α- C-4- Hydroxy- 3,5- dimethylphenyl-N-tert. butyl nitrone	Not available	C1(/C=[N+](\[O- ])C(C)(C)C)=CC(C)=C( C(C)=C1)O	Fatigue Oxygen (unknown) Ozone	4	4	1	Depends on structure, but probable from a phenolic nitrone	Compared to IPPD – reasonable ozone performance, some antifatigue activity, synergistic with phenolic antioxidants. May be able to replace some 6PPD with this material.	Yes, due to compound effectiveness score of 4	40
53	Nitrone + Phenolic AO	Hand Land Contraction	α- C-4- Hydroxy- 3,5- dimethylphenyl - N-isopropyl nitrone and Lowinox WSP	I dimethylphenyl I	Lowinox WSP - 77-6; 3; No CAS number for nitrone	2. N/A Mixture	Fatigue Oxygen Ozone	4	4	Nitrone - 1 Lowinox WSP-3	Probable for both the nitrone and the Lowinox WSP	Blend of two materials gave ozone protection equal to IPPD in static ozone testing. A film was noted on ozone treated rubber.	Yes, due to compound effectiveness score of 4	40, 89
54	Enol ether		Cyclohexen-3- ylidenemethyl benzyl ether Vulkazon® AFD	Benzyl-3-cyclohexen-1- ylidenemethyl ether	22428-48-4	C1CC(=COCC2=CC=C C=C2)CC=C1	Ozone	2	1	1	Unlikely	Claimed to be a good antiozonant for light colored rubber. Not as good as PPD for fatigue. Primarily used with chloroprene.	No, due to compound effectiveness score of 2	69,70,136
55	Phenothiazine		N-(4-methylpentan-2-yl)- 10H-phenothiazin-3-amine	3-(1,3- Dimethylbutylamino)phenothiazin e	Not available	C1=CC(=CC(=C1NC1= CC2)SC1=CC=2)NC(C) CC(C)C	Ozone	3	1	1	Unlikely	Reported as an antiozonant in rubber but no data in patent.	Yes, due to compound effectiveness score of 3	96,97
56	Unsaturated alcohol	HC <sub>3</sub> CH <sub>3</sub> HC <sub>3</sub> HC <sub>3</sub> CH <sub>3</sub> HC <sub>3</sub> HC <sub>3</sub>	Vitamin A	Retinol	11103-57-4	CC1=C(C(CCC1)(C)C) C=CC(=CC=CC(=CCO) C)C	Weak food antioxidant; important to biochemical processes	2	1	3	Unlikely	The material would be expected to be decomposed by ozone reacting with the double bonds in the molecule.	No, due to compound effectiveness score of 2	102

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57	Phenolic	$HO + CH_3 + CH$	Vitamin E	α-Tocopherols (all isomers)	59-02-9	CC1=C(C2=C(CCC(O2 (C)CCCC(C)CCCC(C)C CCC(C)C)C(=C1O)C)C	important in biochemical	2	1	3	Probable	As a phenolic, this material would not be expected to act as an effective antiozonant, and there is no data indicating it works in rubber in that capacity. It has been used as an antioxidant in plastics.	No, due to compound effectiveness score of 2	103, 104
58	Conventional lignin	$ \begin{array}{c} HO \\ HO $	Lignin	Not available	9005-53-2 (other "lignins" may occur as different CAS numbers)	N/A polymer	Mild antioxidant. Has been used in rubber as a filler and a stabilizer. No known antiozonant properties.	2	1	3	Probable, but the quinone would have very high molecular weight	Lignin is complex polymeric phenolic material which also occurs in a sulfonated form with different metal ions which can play a large role in its antioxidant and filler properties. It was compared to octylated diphenyl amine in nitrile rubber. No known antiozonant properties.	No, due to compound effectiveness score of 2	105, 106, 107
59	Alcohol/Acid		Vitamin C	Ascorbic acid	50-81-7	C(C(C1C(=C(C(=O)O1 O)O)O)O	Mild antioxidant but important in biochemical processes. Has been used as an antiozonant for plants.	2	1	3	No	Used in plants but not expected to have any antiozonant activity in compounded rubber.	No, due to compound effectiveness score of 2	108, 109
60	Amine		7-(4-Methylpentan-2- ylamino)-2,3,4,10- tetrahydro-1H-acridin-9- one	Not available	Not available	C1C=C(C=C(C=1N1)C =O)C(=C1CC1)CC1)N C(C)CC(C)C		4	2	1	Unknown	Compound has better oxidation onset temperature than 6PPD, but no ozone data.	Yes, due to compound effectiveness score of 4	110, 115,116
61	Amine		2-Cyclohexyl-N-(4- methylpentan-2-yl)-1H- indol-5-amine	Not available	Not available	C1CCC(CC1)C1=CC2= CC(=CC=C2N1)NC(CC (C)C)C	Patent claims antiozonant activity but compared with 6PPD in simple oxidation test and compounding after air aging.	4	2	1	Unknown	Compound has better oxidation onset temperature than 6PPD, but no ozone data.	Yes, due to compound effectiveness score of 4	111, 112
62	Amine		4-(1H-Indol-2-yl)-N-(4- methylpentan-2-yl)aniline	Not available	Not available	C1C=C2C=C(NC2=CC =1)C1C=CC(=CC=1)N C(CC(C)C)C		4	2	1	Unknown	Compound has better oxidation onset temperature than 6PPD, but no ozone data.	Yes, due to compound effectiveness score of 4	113, 114
63	Phenylene Diamine		DPPD	1-N,4-N-Diphenylbenzene-1,4- diamine	74-31-7	C1=CC=C(C=C1)NC2= CC=C(C=C2)NC3=CC= CC=C3		4	3	3 (as mixture)	Probable	Part of CAS 68478-45-5. Low solubility in rubber; not as good as an antiozonant as 6PPD.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	75, 76, 118, 1

Count	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
64	Phenylene Diamine	CH <sub>3</sub> H H CH <sub>3</sub> CH <sub>3</sub>	DTPD	1-N,4-N-Bis(2- methylphenyl)benzene-1,4- diamine 1,4-Benzenediamine, N,N'- bis(2-methylphenyl)-	60052 04 4	CC1=CC=C(C=C1)NC2 =CC(=CC=C2)NC3=CC =C(C=C3)C		4	3	3 (as mixture)	Probable	Part of CAS 68478-45-5. Low solubility in rubber; not as good as an antiozonant as 6PPD.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	119,1
65	Polymeric amine functionalized lignin	H <sub>2</sub> N NH OH HO OH HO OCH <sub>3</sub> H <sub>3</sub> CO OH HH NH <sub>2</sub> OH	Amine functionalized lignin	Not available	Not available	N/A polymer	Antioxidant Antiozonant	4	4	1	Probable, but would be polymer bound	Ozone testing was static, but comparable to GPPD. Fatigue was similar to GPPD. Since there is no blooming or reservoir, it is unlikely to provide long term protection.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	120
66	Phenolic	+ Calcium salt	Calcium salt from lignin rice straw black liquor	Not available	Not available	Not available	Antioxidant	2	1	1	Probable, but would be polymer bound	Reference only claims material as antioxidant; no work with ozone was done.	No, due to compound effectiveness score of 2	121
67	Gallate related	Not available	Rambutan peel extract	Not available	Not available	N/A complex mixture	Antioxidant Antiozonant	4	4	2	Unknown	Static ozone testing showed comparable crack resistance to 6PPD. No dynamic ozone data are available.	Yes, due to compound effectiveness score of 4	139
68	Gallate related		Octyl Gallate	Octyl 3,4,5-trihydroxybenzoate	1034-01-1	CCCCCCCCCC(=0)C1 =CC(=C(C(=C1)0)0)0	Antioxidant	3	1	3	Unknown	Compound used in food applications. Similar compound, propyl gallate has shown to be active against ozone in protecting biological systems, but no data in tires or tire compounding was found. Octyl gallate was included for evaluation over propyl gallate since octyl gallate has a better melting point for tire compounding.	Yes, similar compound identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 3	140

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69	Gallate related	Not available	Gallate ester of castor oil	Not available	Not available	Not available	Antioxidant	2	1	2	Unknown	Synthesized via esterification between castor oil and gallic acid. Only data as an antioxidant compared to IPPD was available. No ozone data are available.	No, due to compound effectiveness score of 2	143
70	Gallate related	CaCO <sub>3</sub> Ca <sup>2+</sup> Ca <sup>2+</sup> Ca <sup>2+</sup> Ca <sup>2+</sup> absorbed CaCO <sub>3</sub> Ca <sup>2+</sup> Ca <sup>2+</sup> absorbed CaCO <sub>3</sub> Ca <sup>2+</sup> Ca <sup>2+</sup> absorbed CaCO <sub>3</sub> Ca <sup>2+</sup> Ca <sup>2+</sup>	Nano calcium carbonate surface modified by gallic acid	Not available	Not available	Not available	Antioxidant Antiozonant	4	3	2	Unknown	Synthesized by surface coating nano calcium carbonate with gallate ions. Using Irganox 1010 as the control, compound showed improved static ozone resistance. No dynamic ozone data are available.	Yes, due to compound effectiveness score of 4	144
71	Coatings	Mixture	Coating	Mixture	Mixture	Mixture	Ozone (static)	1	3	2	Would depend on coating material, but probably not	In general, coatings wear off during use and would expose the underlying rubber to ozone attack. For examples, tire sidewall coatings will be scuffed when tires brush against curbs. Tire tread coatings will wear off almost immediately. Additionally, performance is also an issue. An example formulated nitrile coated natural rubber compound (Nipol 1312 with several other ingredients) showed significant cracking after ozone exposure.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 1	141
72	Hydrocarbon wax	H <sub>a</sub> C-(CH <sub>a</sub> ) <sub>a</sub> -CH <sub>a mot</sub> n=24	microcrystalline wax paraffinic wax	Nochek (blend of waxes for tires)	Not available	CCCCCCCCCCCCCCC CCCCCCCCCCCCCC CCCC(CC)CCCCCCCC	Ozone (static)	1	5	3	No	In general, waxes bloom to the surface and will wear off during active use. Currently, blends of microcrystalline and paraffin waxes are already used in combination with 6PPD in tires. Waxes alone only provide static ozone protection.	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC, but compound effectiveness score is 1	58,59,60

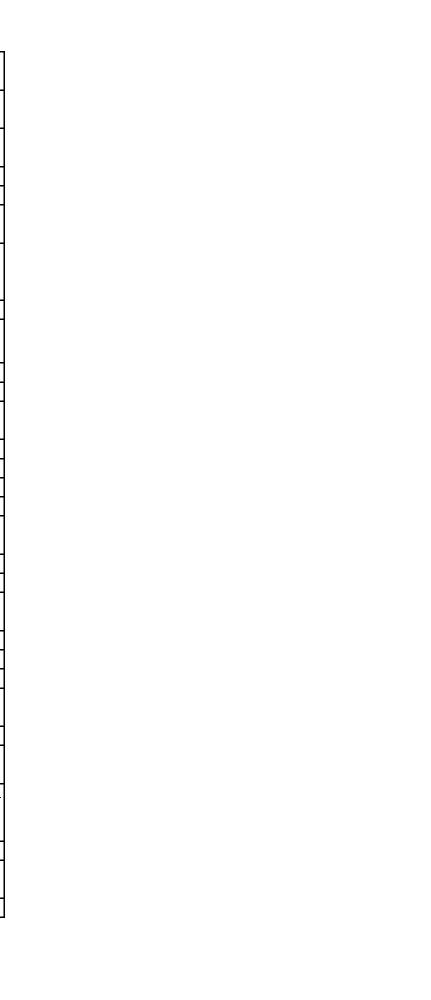
c	ount	Class of Compounds	Compound	Common Name	Other Names	CAS No.	SMILES	Primary Usage Property - Antidegradant	Compound Effectiveness	Quality of Data	Commercial Availability	Quinone Formation on Oxidation/Ozonati on?	Comments	Evaluate Further in Preliminary (Stage 1) AA?	References
	73	Inorganic	Mixture	MOLECULAR REBAR® carbon nanotubes	Not available	Not available	N/A	Anti-static Reinforcing agent	4	3	2-3 (more than lal quantities, but insufficient to ever partially replace a significant amount of GPPD)	No	Preliminary static ozone data is positive. It is not known if carbon nanotube would work by themselves as antiozonants in dynamic ozone tests over prolonged period. May need to be used in conjunction with 6PPD in tires. Additionally, material is highly reinforcing and difficult to mix. Major adjustments required if to be used in tire conventional compounds ( <i>e.g.,</i> sidewall)	Yes, identified as a possible alternative by DTSC or sources referenced by DTSC. Also compound effectiveness score is 4	142, 145

Notes:

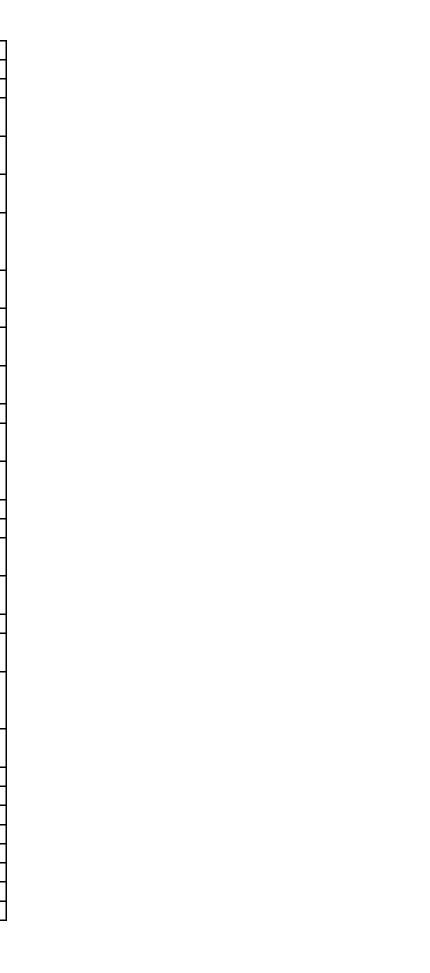
BR = Butadiene Rubber; CAS No. = Chemical Abstracts Service Number; DTSC = Department of Toxic Substances Control; ESBR = Emulsion Styrene Butadiene Rubber; HALS = Hindered Amine Light Stabilizer; N/A = Not Applicable; SBR = Styrene Butadiene Rubber.

	1. Have data and doesn't work					
	1. Have data and doesn't work					
Compound Effectiveness	<ol><li>Have no data but chemical molecule will not work</li></ol>					
Against Ozone	3. Have no data to say either way					
	4. Have some positive data but not enough to say yes or no					
	1. Not available					
Commercial Availability	2. Laboratory availability only					
	3. Multi ton lots					
	1. Listed as an antioxidant or antiozonant; little or no data on					
	ozone					
Quality of Data	2. Chemical data only (reaction with ozone)					
Quality of Data	3. Compounding data (ozone resistance)					
	4. Compounding data (ozone/fatigue)					
	5. Tire data					

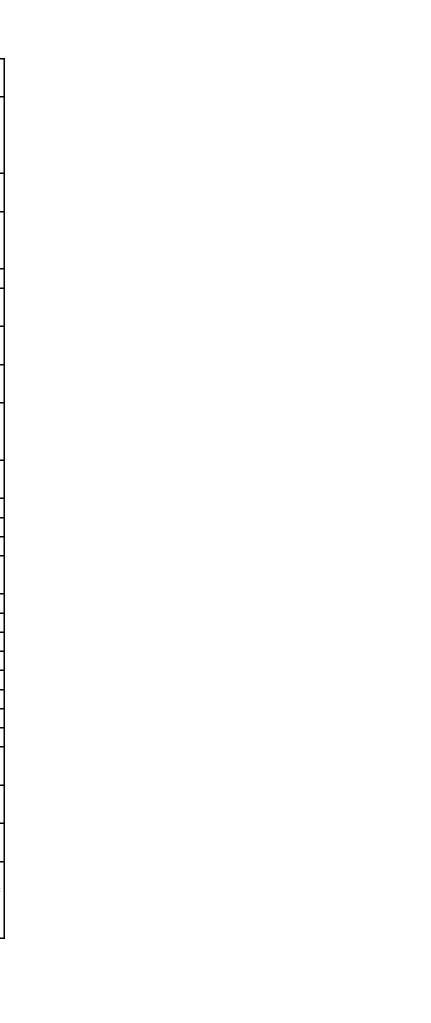
Count	Author Names and Publication	Reference List
	Date	
1	Hong <i>et al.</i> , 2016	S. Hong, "Antioxidants and Other Protectant Systems" in Brendan Rodgers' Rubber Compounding Chemistry and Applications, 2nd Ed., CRC
		Press, 2016 p. 419-459
2	Huntink, 2003	N. Huntink Durability of Rubber Products, PhD Thesis, University of Twente, 2003
		(https://ris.utwente.nl/ws/portalfiles/portal/6072556/thesis_Huntink.pdf)
3	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/compound/13101#section=Safety-and-Hazards
4	Stahly, 1968	Comparison of Ethoxyquine with various 6PPD type antiozonants - E.E. Stahly, US Patent 3.391,107, 1968
5	Kristin Hals and Sofia Helena	https://aquafeed.co.uk/entrada/ethoxyquin-ban-in-the-eu-are-there-viable-alternatives-21749
	Lindahl, 2020	
6	Byrne, 2017	Reason for suspending Ethoxyquin authorization in feed -
		https://www.feednavigator.com/Article/2017/06/12/Ethoxyquin-authorization-suspended-in-the-EU-are-there-viable-
		alternatives#:~:text=Last%20week%20saw%20the%20European,Official%20Journal%20of%20the%20EU.
7	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/compound/18320
8	ToxServices LLC, 2021	Greenscreen assessment - https://www.ezview.wa.gov/Portals/_1962/Documents/6ppd/GreenScreenExecutiveSummaryFor77PPD.pdf
9	Pubchem, 2024	For a number of patents cited on this material see https://pubchem.ncbi.nlm.nih.gov/compound/92190
10	Lippincott & Peto Inc, 1993	https://www.thefreelibrary.com/Review+of+antiozonants-a014459625
11	Engels <i>et al.</i> , 2011	Engels, Hans-Wilhelm, Herrmann-Josef Weidenhaupt, Manfred Pieroth, Werner Hofmann, Karl-Hans Menting, Thomas Mergenhagen, Ralf
		Schmoll, and Stefan Uhrlandt. "Rubber, 9. Chemicals and additives." Ullmann's Encyclopedia of industrial chemistry (2000). P.24
12	Huntink, 2004	"Addressing Durability of Rubber Compounds" N. Huntink et al, RC&T (2004) 77 (3): 476-511
13	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/compound/31250
14	KR 20090100673A	KR 20090100673A "Tire Sidewall Composition"
15	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/substance/164110485
16	Masatomo <i>et al.</i> , 1972	E. Masatomo, et al. US 3,634,316 "Sulfur-Vulcanizable Natural, and Synthetic Rubbery Polymers Containing Xylylene Diamines as Antiozonants"
	,	
17	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/compound/15320161
18	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/compound/284842
19	Huntink, 2003	N. Huntink Durability of Rubber Products, PhD Thesis, University of Twente, 2003 p. 45 and listed references within
		(https://ris.utwente.nl/ws/portalfiles/portal/6072556/thesis_Huntink.pdf)
20	Pender, 2008	M. Pender, US 8,329,788 B2, Tire having enhanced ozone resistance
21	Lanxess, 2019	https://lanxess.com/en/Media/Press-Releases/2019/07/Safe-protection-against-oxidation
22	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/compound/98143
23	Bruck <i>et al.</i> , 1985	D. Bruck; H. Konigshofen and L. Ruetz, "The Action of Antiozonants in Rubber " Rubber Chemistry and Technology (1985) 58 (4): 728–739.
24	Pubchem, 2024	https://pubchem.ncbi.nlm.nih.gov/compound/129273577
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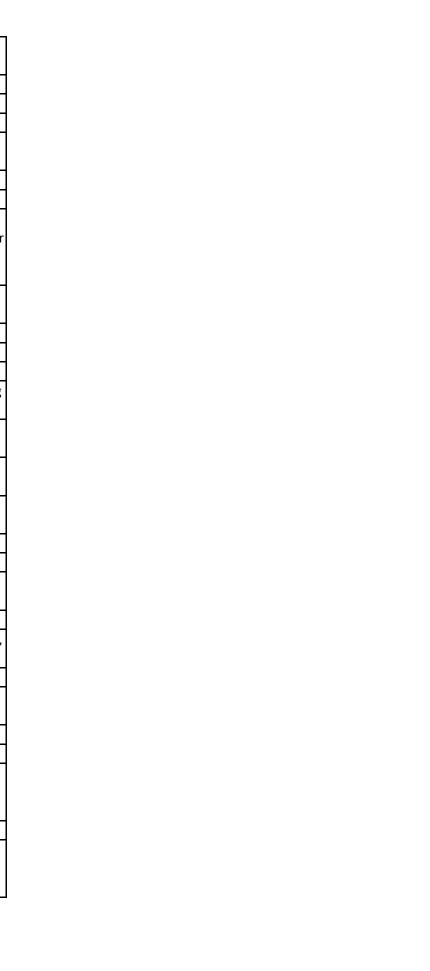
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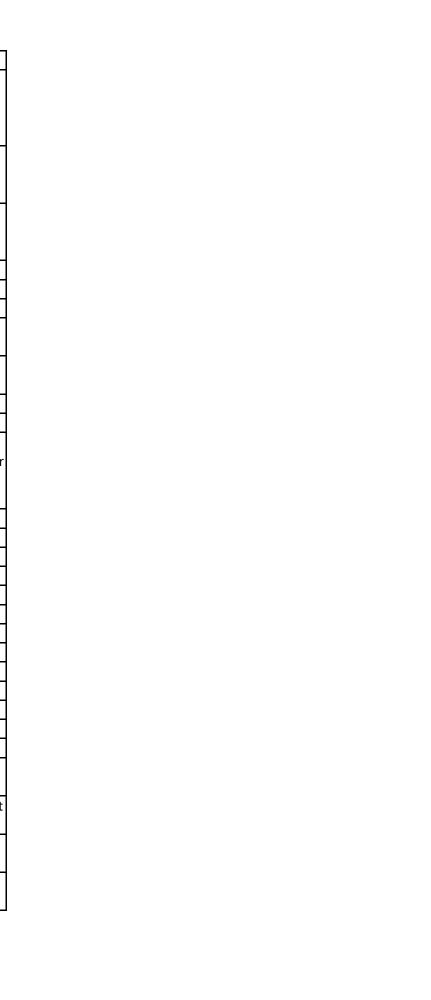
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